

## {Question 4.007}

 $P_1 = 10 [\text{bar}]$  $T_1 = 20 [\text{C}]$  $P_2 = 3.0 [\text{bar}]$  $h_1 = \text{enthalpy}(\text{Ammonia}, P=P_1 * \text{convert}(\text{bar}, \text{kPa}), T=T_1)$  $h_2 = h_1$  $T_2 = \text{temperature}(\text{Ammonia}, P=P_2 * \text{convert}(\text{bar}, \text{kPa}), h=h_2) \{\text{Temperature: } T_2 = -9.231 \text{C}\}$  $x_2 = \text{quality}(\text{Ammonia}, P=P_2 * \text{convert}(\text{bar}, \text{kPa}), h=h_2) \{\text{Quality: } x_2 = 10.54 \%\}$ 

## SOLUTION

## Unit Settings: SI C kPa kJ mass deg

 $h_1 = 293.9 [\text{kJ/kg}]$  $h_2 = 293.9 [\text{kJ/kg}]$  $P_1 = 10 [\text{bar}]$  $P_2 = 3 [\text{bar}]$  $T_1 = 20 [\text{C}]$  $T_2 = -9.231 [\text{C}]$  $x_2 = 0.1054$ 

No unit problems were detected.

EES suggested units (shown in purple) for  $h_2$ .