

{Question 6.145}

$T_1 = 190 [\text{K}]$
 $P_1 = 100 [\text{kPa}]$
 $P_2 = 330 [\text{kPa}]$
 $\text{efficiency} = 90.3$

$\text{efficiency} = W_s / W$
 $W_s = h_{2s} - h_1$
 $W = h_2 - h_1$

$h_1 = \text{enthalpy}(\text{Air}, T = \text{converttemp}(\text{K}, \text{C}, T_1))$
 $s_1 = \text{entropy}(\text{Air}, T = \text{converttemp}(\text{K}, \text{C}, T_1), P = P_1)$
 $s_1 = s_{2s}$
 $h_{2s} = \text{enthalpy}(\text{Air}, P = P_2, s = s_{2s})$

{Using the isentropic efficiency equation it is possible to find h_2 and then W }
 {Work Input: $W = 0.8597 \text{ kJ/kg}$ }

SOLUTION

Unit Settings: SI C kPa kJ mass deg

$\text{efficiency} = 90.3$

$h_{2s} = 267.7 [\text{kJ/kg}]$

$s_1 = 5.248 [\text{kJ/kg-K}]$

$W = 0.8597 [\text{kJ/kg}]$

$h_1 = 190.1 [\text{kJ/kg}]$

$P_1 = 100 [\text{kPa}]$

$s_{2s} = 5.248 [\text{kJ/kg-K}]$

$W_s = 77.63 [\text{kJ/kg}]$

$h_2 = 191 [\text{kJ/kg}]$

$P_2 = 330 [\text{kPa}]$

$T_1 = 190 [\text{K}]$

No unit problems were detected.