

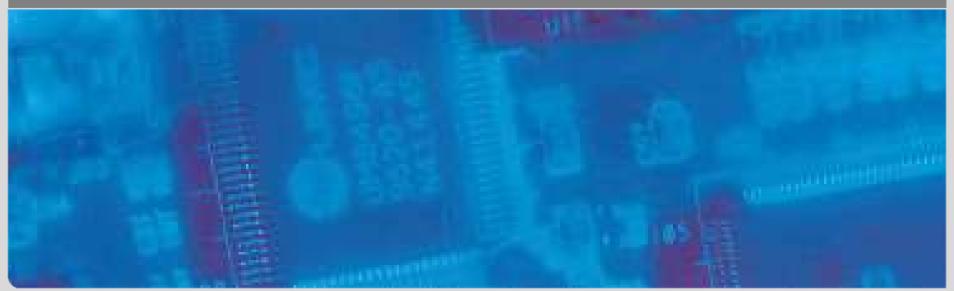


Lower Power Design

Lecture 1: Energy Sources

Anuj Pathania on behalf of Prof. Dr. Jörg Henkel Summer Semester 2017

CES – Chair for Embedded Systems



ces.itec.kit.edu





- Slides available for download -
 - http://cesweb.itec.kit.edu/teaching/LPD/s17/slides/
 - Username: student
 - Password: CES-Student
- Homework
 - Read a relevant scientific paper.
 - Discussion next class.
- Oral Exam
 - Make appointment with KIT CES secretary 6-8 weeks in advance.
 - Exam will be in English (or German if told in advance).
 - More information: http://ces.itec.kit.edu/972.php

Lectures



- 27.04.2017 Lecture 0: Introduction
- 04.05.2017 Lecture 1: Energy Sources
- 11.05.2017 Lecture 2: Battery Modelling Part 1
- 18.05.2017 Lecture 3: Battery Modelling Part 2
- 25.05.2017 Ascension Day (Holiday)
- 01.06.2017 TBA
- 08.06.2017 TBA
- 15.06.2017 Corpus Christi (Holiday)
- 22.06.2017 TBA
- 29.06.2017 TBA
- 06.07.2017 TBA
- 13.07.2017 TBA
- 20.07.2017 TBA
- 27.07.2017 TBA

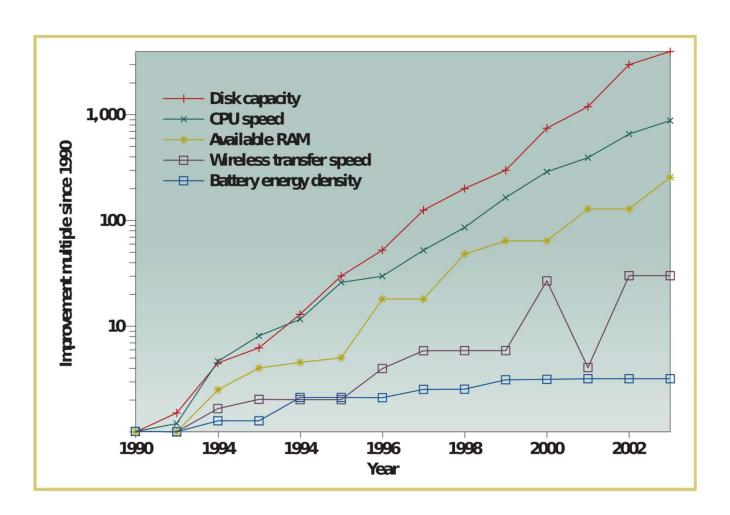
Overview for Today



- Fuel Cells
- Human-Generated Power for Portable Devices
- Solar Energy Harvesting
- Super Capacitors
- Hybrid Electric Storage System

Battery Gap





Source: Paradiso [2005]

Fuel Cells 1



- Direct conversion of fuel to electricity (direct current).
- High efficiency (~40-60%).
- Different types of fuel cells exist beside Hydrogen-Oxygen fuel cell.
- Hydrogen-Oxygen: Environmentally (mostly) clean; byproduct is water.
 - · Not yet mass produced.
- Solid Oxide Fuel Cells (SOFC):
 - Needs 800-850°C
- Proton Exchange Membrane (PEM)
 - Reaction positive electrode:

•
$$\frac{1}{2}$$
 O₂ + 2 H₃O⁺ + 2e⁻ \rightarrow 3 H₂O

Reaction negative electrode:

•
$$H_2 + 2 H_2O \rightarrow 2H_3O^+ + 2 e^-$$

Overall reaction:

•
$$H_2 + \frac{1}{2} O_2 \rightarrow H_2 O E_0 = 1.229 V$$

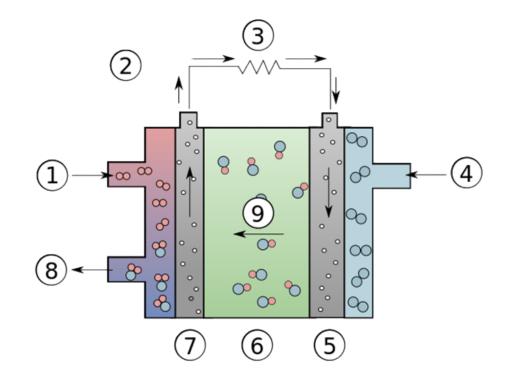


Fuel Cells 2: Alkaline Fuel Cell



- 1. Hydrogen
- 2. Electron Flow
- 3. Load
- 4. Oxygen
- 5. Cathode
- 6. Electrolyte
- 7. Anode
- 8. Water
- 9. Hydroxyl Ion

70% Efficiency



$$2H_2 + 4OH^- \rightarrow 4H_2O + 4e^-$$

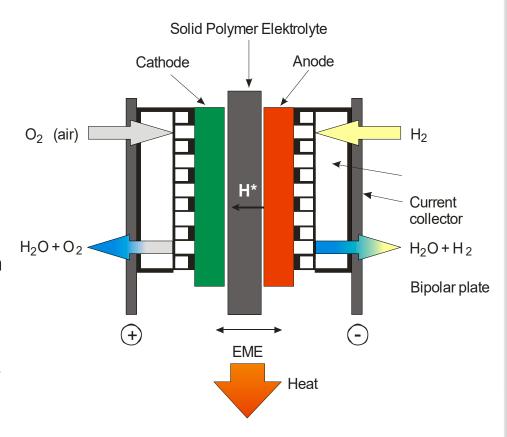
 $O_2 + 2H_2O + 4e^- \rightarrow 4OH^-$

Fuel Cells 3: Principle



Core parts

- Two electrodes separated by an ion-conducting polymeric membrane (electrolyte).
- Fuel (i.e. H₂) is transformed on catalytic sites at the negative electrode and form protons (H⁺) which cross the membrane and electrons on the other hand which produce a current outside the cell.
- Electrical energy is obtained when electrons recombine at the positive electrode with protons (H+) coming from the negative electrode and oxygen from the air.
 - Chemical reaction results in: electricity, water and heat.
 - A whole system is shown on the next page.

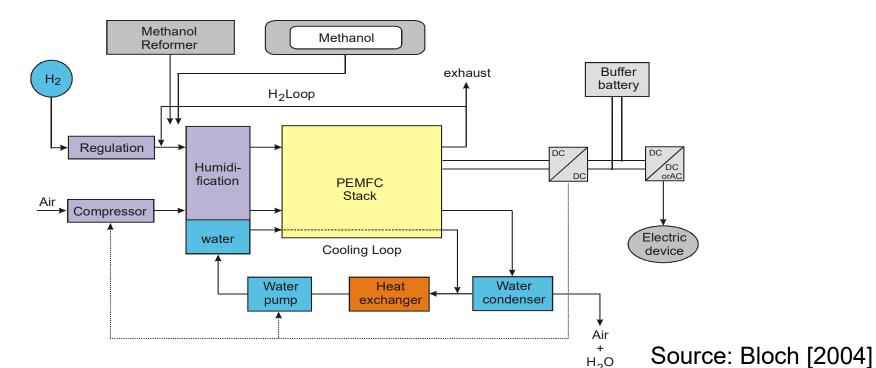


Source: Bloch [2004]





- Whole system contains besides the core (stack):
 - electrical management systems.
 - thermal management systems.
 - fluidic management systems.



Miniature Fuel Cells



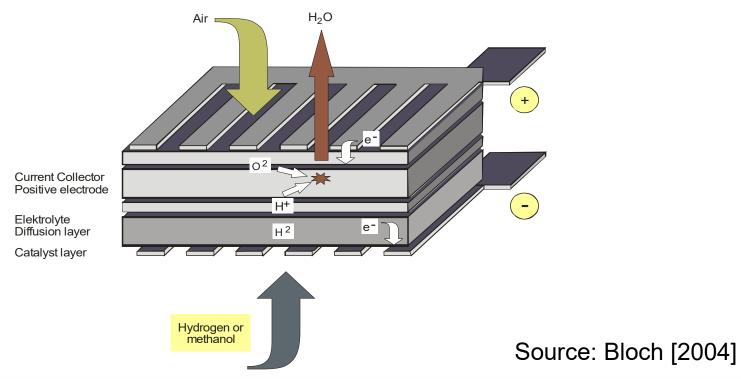
- Application domain: portable electronic devices (smartphone, etc.)
- Two approaches.
 - "Bipolar" Technology
 - Built with bipolar plates forming the fuel cell stack.
 - Typically 20-500 W.
 - Smaller stacks seem not to be competitive with Lithium-Ion batteries.
 - Example: SFC Energy fuel cells (sfc.com).
 - Various approaches with new concepts e.g. micro-fabrication techniques.
 - Typically 0.1 25 W.
 - Substrate (thin-film)-based.

Silicon Fuel Cells



Silicon Fuel Cells

- Silicon wafer; grown and treated with lithographic techniques.
- Often less than a centimeter wide.
- Available from various companies: Neah Power, Integrated Fuel Cell Technologies.



Human-Generated Power for Portable Devices



Can energy for portable electronic devices be harvested from humans?



1 gram of Fat = 37,700J of Energy

68 kg Human (with 15% fat) = 384 MJ !!!



Source: WarnerBros and ExtremeTech

Human Power Consumption for Various Activities

13

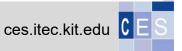


- A span of ~20x!
- Difficult to harvest.
- Need to be converted to DC/AC.
- Must be non-intrusive.

Human Energy Expenditures for Selected Activities

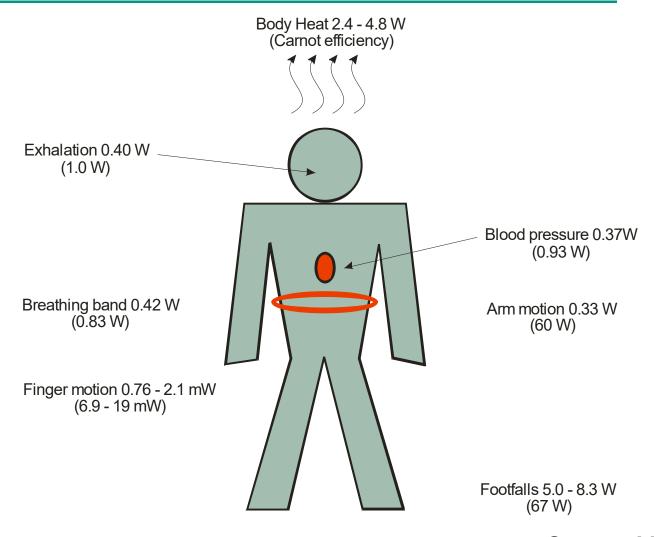
Activity	Kilocal/hr	Watts
Sleeping	70	81
Lying quietly	80	93
Sitting	100	116
Standing at case	110	128
Conversation	110	128
Eating a meal	110	128
Strolling	140	163
Driving a car	140	163
Playing the violin or piano	140	163
Housekeeping	150	175
Carpentry	230	268
Hiking, 4 mph	350	407
Swimming	500	582
Mountain climbing	600	698
Long-distance run	900	1048
Sprinting	1400	1630

Source: Morton [1952]



Human Power Consumption for Various Activities





Source: Morton [1952]

Human Power Sources 1

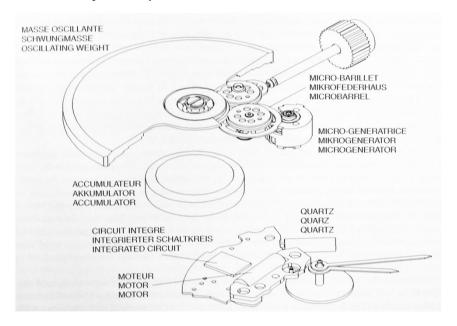


- Body Heat.
 - (T_Body T_Ambient)/T_body = (310K 293K)/310K = 5.5% [Carnot Efficiency]
 - Not very efficient (even theoretically); real-world only 0.8% efficient at best.
- Breathing.
 - Exploit difference breathing pressure and atmospheric pressure.
 - Only 2% difference.
- Blood Pressure.
- Vibrations from motion.

Human Power Sources 2



- Power from typing
 - Ex: 50 g key pressure, depress by 0.5 cm.
 - .05 kg/stroke * 9.8 m/s² * .005 m * 7.5 strokes/sec = 19 mW (too less).
 - User is not continuously typing.
 - Can at least power the keyboard (if not whole system).
- Inertial micro systems
 - Used for hundred of years in watches
- Electrical version
 - The mass winds a spring.
 - Drive a generator at 15000 rpm.
 - Yield 6 mA and 16 V for 50 ms.

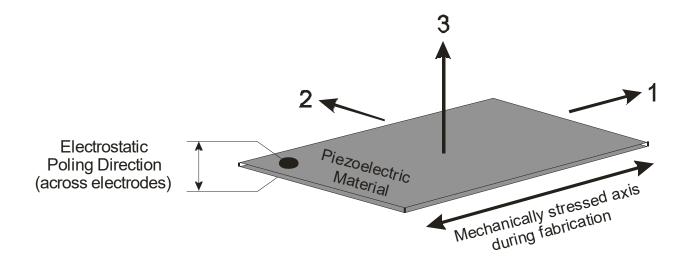


Source: Paradiso [2004]





- Walking (68 kg Human, 5.6 km/h) cost 324 Watts of power.
 - Most of this power is used to move legs.
- Power through the fall of the heel:
 - 68 kg * 9.8 m/s² * 0.05 m * 2 steps/sec = 67 Watts.
 - This power cannot be converted into electrical power w/o significant intrusion.
 - Use piezoelectric device like Quartz.



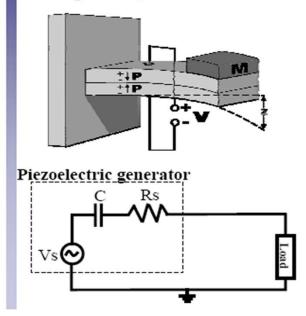
Source: Paradiso [2004]

Vibrations into Electricity



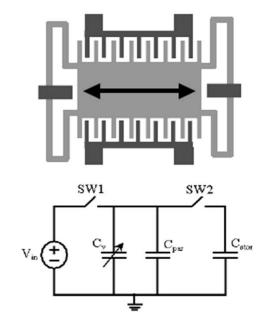
Piezoelectric

Strain in piezoelectric material causes a charge separation (voltage across capacitor)



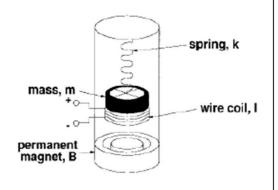
Capacitive

Change in capacitance causes either voltage or charge increase.



Inductive

Coil moves through magnetic field causing current in wire.



Amirtharajah et. al., 1998

Source: Hande [2007]



Karlsruhe Institute of Technology

Other power/energy sources

Energy Source	Power/Energy Density
Batteries (Zinc-Air, primary)	1050-1560 mWh/cm ³
Batteries (Li, rechargeable)	300 mWh/cm ³
Solar (outdoors)	15 mW/cm ² (direct sun)
	1 mW/cm ² (24 hour avg)
Solar (indoors)	0.006 mW/cm ² (office desk)
	0.57mW/cm ² (<60W desk lamp)
Vibrations	$0.01-0.1 \text{ mW/cm}^3$
Acoustic (noise)	3 e-6 mW/cm ² @ 75dB
	9.6 e-4 mW/cm ² @ 100dB
Miniature Fuel cells	0.1-500W

Source: Hande [2007]

Solar Energy



Energy from almighty Sun.



Floating Solar Power Plant in Japan (2.9 MW)

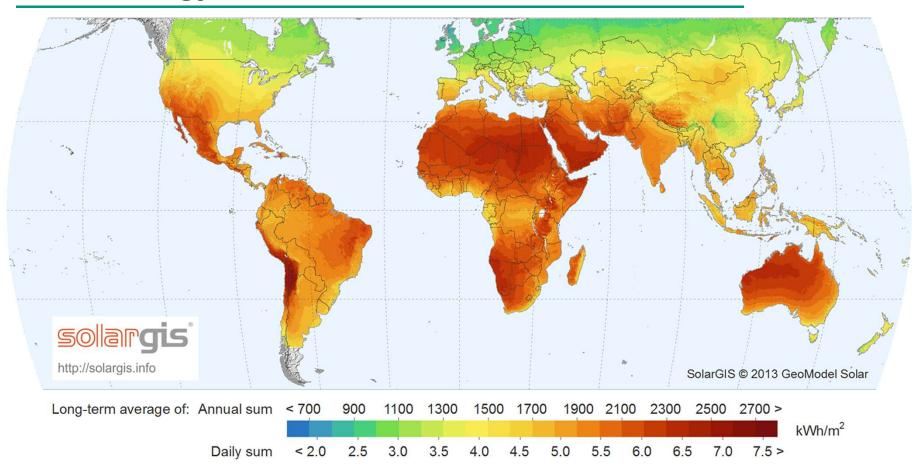
Canal Top Power Plant in India (10 MW)



Source: Kyocera and SSND Ltd.

Solar Energy Distribution





Outside Earth: Constant 1353 W/m²

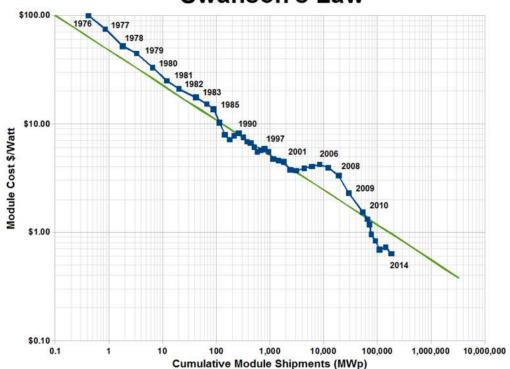


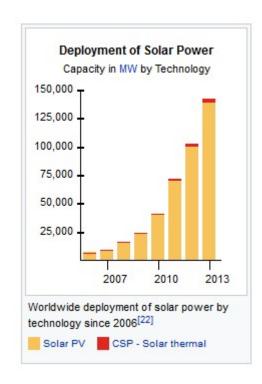
Swanson's Law



- Solar cell price drop 20% for doubling of cumulative cells shipped.
 - Half every ten years at current speed.

Swanson's Law





Solar Energy



- Energy harvesting through photo-voltaic conversion provides high power density.
 - Good for embedded systems that need some mW power.
 - Characteristics of solar cells need to be taken into consideration for system design.

Harvesting technology	Power density		
Solar cells (outdoors at noon)	$15mW/cm^2$		
Piezoelectric (shoe inserts)	$330\mu W/cm^3$		
Vibration	$116\mu W/cm^3$		
Thermoelectric (10°C gradient)	$40\mu W/cm^3$		
Acoustic noise (100dB)	$960nW/cm^3$		

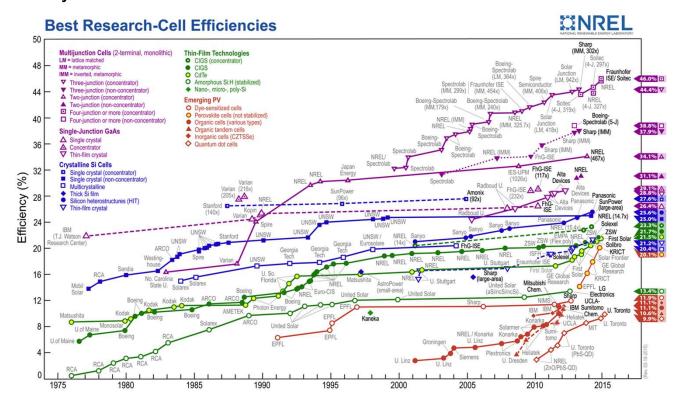
Shockley-Queisser Limit



Solar cell efficiency has a theoretical upper limit.

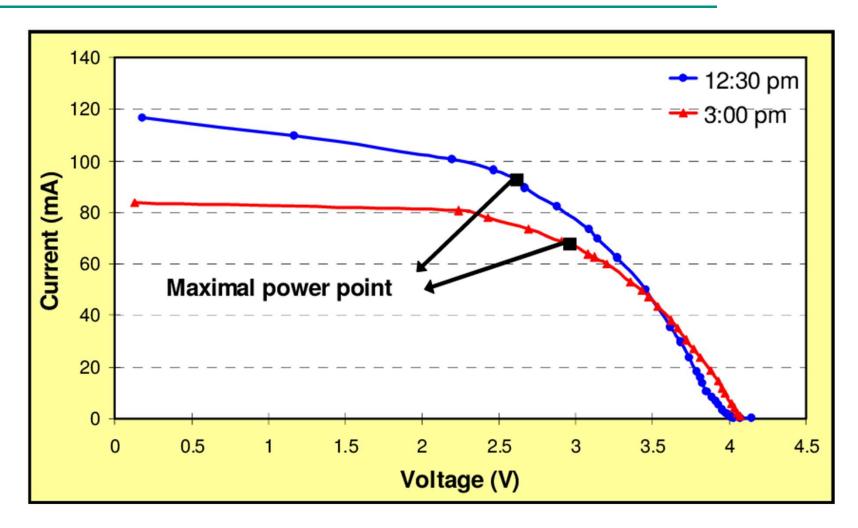
Single-Layer: 33.7%

• Multi-Layer: 86.8%



Solar Panels 2







Solar Panels 3



- Characteristics:
 - Solar panel behaves as a voltage limited current source.
 - Current tend be to constant, voltage vary over a wide range.
 - Remember: Battery is a voltage source.
 - There is an optimum operation point for maximum power extraction.
- Since it behaves like a current source (supply voltage depends on varying load), an energy storage element like a battery is necessary.

Super Capacitors (Ultra Capacitors)



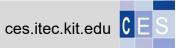
- Capacitors with very high capacitance (10x 100x normal capacitor).
- Bridge gap between normal capacitors and rechargeable batteries.
- Advantages
 - High power density.
 - Very long life (10 12 Years); charge/discharge cycles > 500,000 cycles.
 - No danger of overcharge.
- Disadvantage
 - Low energy density.
 - Relatively expensive.
 - High self-discharge (20% per Day).



Super Capacitors 2



Parameter	Aluminum Electrolytic Capacitors	Double-layer Capacitors for Memory Backup	Super- Capacitors for Power Applications	Pseudo and Hybrid Capacitors (Li-Ion capacitors)	Lithium-Ion Batteries
temperature range (°C)	-40 to 125	-20 to +70	-20 to +70	-20 to +70	-20 to +60
cell voltage (V)	4 to 550	1.2 to 3.3	2.2 to 3.3	2.2 to 3.8	2.5 to 4.2
charge/discharge cycles	unlimited	10 ⁵ to 10 ⁶	10 ⁵ to 10 ⁶	2*10 ⁴ to 10 ⁵	500 to 104
capacitance range (F)	≤1	0.1 to 470	100 to 12000	300 to 3300	
energy density (Wh/kg)	0.01 to 0.3	1.5 to 3.9	4 to 9	10 to 15	100 to 265
power density (kW/kg)	> 100	2 to 10	3 to 10	3 to 14	0.3 to 1.5
self discharge time at room temperature	short (days)	middle (weeks)	middle (weeks)	long (month)	long (month)
efficiency (%)	99	95	95	90	90
life time at room temperature (years)	> 20	5 to 10	5 to 10	5 to 10	3 to 5



Super Capacitors 3



Type: Storage Capacitor

Execution: Gold-Cap

Material: Cadmium Free

Capacity: 22 Faraday

Tension DC: 2.3 Volts

Dimensions: 18.00 mm

• Price: 5.60 Euros

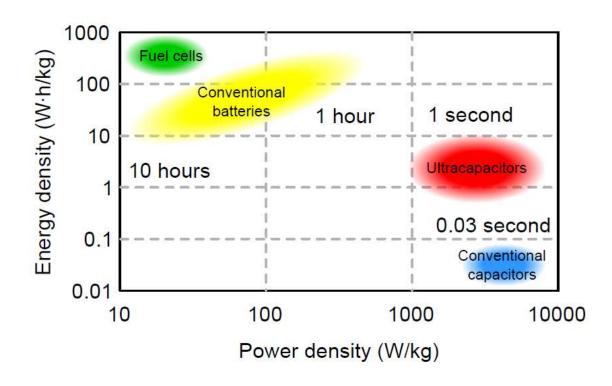


Source: reichelt.de

Ragone Chart



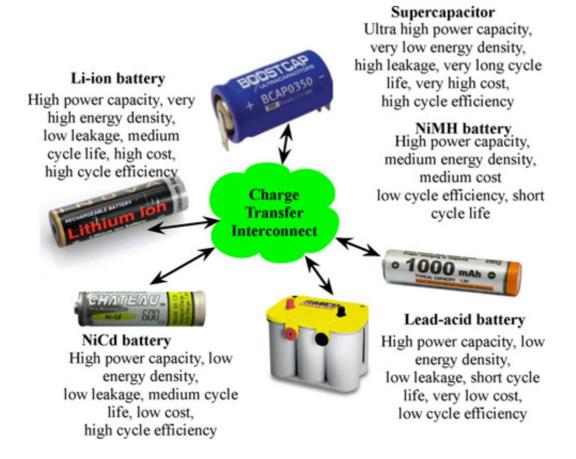
Performance comparison of various energy-storage devices.



Energy density =
$$\frac{V \times I \times t}{m}$$
,
Power density = $\frac{V \times I}{m}$,

Hybrid Electric Storage System

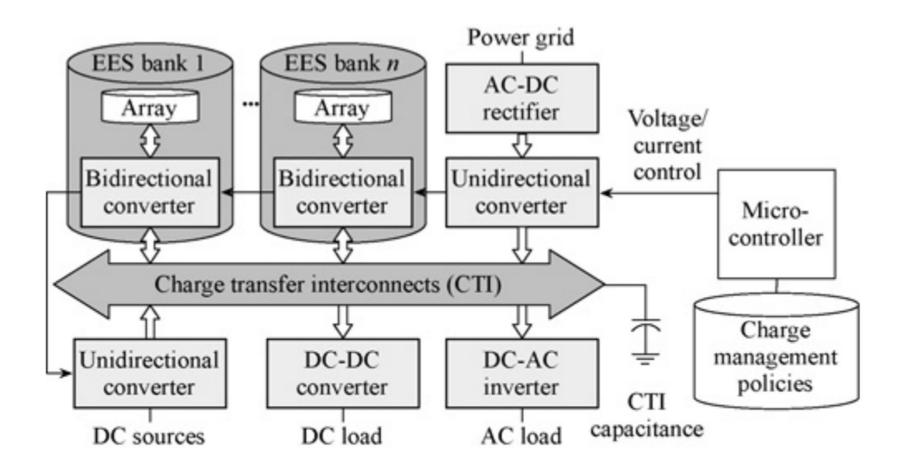




Source: Xie [2013]

Hybrid Electric Storage System 2: Design

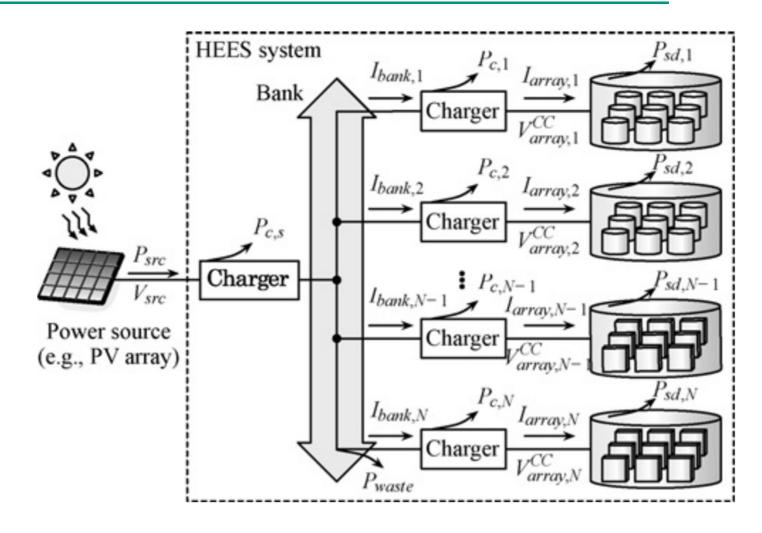




Source: Xie [2013]



Hybrid Electric Storage System 3: Process



Source: Xie [2013]



Source



- Homework >> Paradiso, Joseph A., and Thad Starner. "Energy scavenging for mobile and wireless electronics." *IEEE Pervasive computing* 4.1 (2005): 18-27.
- Bloch, Didier. "Miniature fuel cells for portable applications." *Low-Power Electronics Design*. CRC Press, 2004. 44-1.
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- Hande, Abhiman, et al. "Indoor solar energy harvesting for sensor network router nodes." *Microprocessors and Microsystems* 31.6 (2007): 420-432.
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