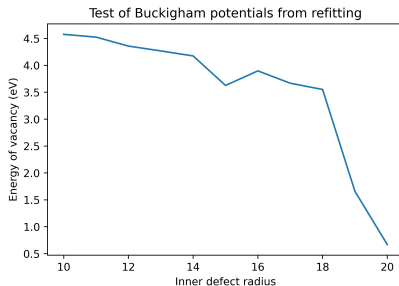
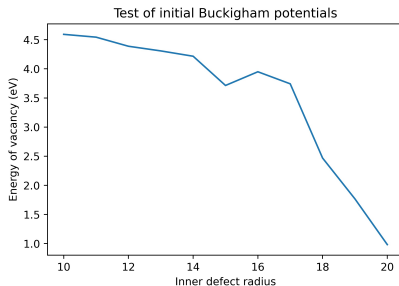


Post-Christmas Update

Ben Goldmann

January 3, 2021

Initial vs Refitted



Buckingham potentials

Na-O	1226.84	0.307	0	1225.11	0.307	0
Na-Cl	2314.70	0.290	0	2292.53	0.290	0

The structural stability, lattice dynamics, electronic, thermophysical, and mechanical properties of the inverse perovskites A_3OX : A comparative first-principles study

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Funding Information
United Arab Emirates University

Summary

We present a comparative study on the structural, electronic, elastic, and thermodynamic properties of the cubic inverse-perovskites A_3OX (where $A = Li, Na, K$ and $X = Cl, Br, I$) by density functional theory (DFT). The cohesive, vibrational, and elastic properties analysis indicates that all studied materials are chemically, thermodynamically, and mechanically stable. Electronic properties reveal that all the inverse A_3OX perovskites are direct bandgap semiconductors except Li_3OCl and Li_3OBr with indirect nature which is confirmed by electron localization function (ELF) analysis. We have also calculated Debye temperature (Θ_D) and Grüneisen parameter (γ) to determine the lattice thermal conductivity for all the A_3OX materials. Furthermore, thermodynamic (TE) properties are explored by calculating the Seebeck coefficient (S), electronic thermal conductivity, power factor (PF), electrical conductivity (σ), lattice thermal conductivity, and zT value. Our investigated A_3OX inverse-perovskites provide a fertile base that can improve the overall TE performance for TE applications and green energy production.

KEYWORDS

inverse-perovskite, formation energy, mechanical properties, phonon, p -type semiconductors, structural properties, thermodynamic properties

Sattar et al. (2020)



Ab initio study of electronic structure, elastic and optical properties of anti-perovskite type alkali metal oxyhalides

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ABSTRACT

We report structural, elastic, electronic, and optical properties of antiperovskite alkali metal oxyhalides Na_3XCl , Na_3XBr , and K_3XBr using two different density functional methods: either generalized gradient approximation (GGA) or PBE with spin-orbit coupling (SOC) method. We have used to calculate the ground state structural and elastic properties while the electronic structure and optical properties are calculated by employing the full potential linearized augmented plane wave method. We have used the calculated ground state properties of the investigated compounds agree well with the available experimental data. The predicted elastic constants using both PBE and PBE+SOC methods are in good accord with each other and show that the materials are mechanically stable. The low values of the elastic modulus indicate that these materials are soft in nature. The bulk properties such as shear modulus, Young's modulus, and Poisson's ratio are derived from the calculated elastic constants. The Na_3XBr exhibits better electronic band structure compared to these materials are direct band gap semiconductors. The complete dielectric function of the total oxyhalide compounds have been calculated and the observed prominent peaks are analyzed through the 10-ml dielectric function. By using the knowledge of complex dielectric function other important optical properties including absorption, reflectivity, refractive index and loss function have been obtained as a function of energy.

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Elastic Properties of Alkali Superionic Conductor Electrolytes from First Principles Calculations

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In this work, we present a comprehensive investigation of the elastic properties of the full structure (unit cell, shape and Young's modulus) and Poisson's ratios of 23 well-known ceramic alkali superionic conductor electrolytes (SECEs) using first principles calculations. We find that the computed elastic moduli are in good agreement with experimental data and observed banding patterns. The atom species and structural framework has a significant influence on the elastic properties, and the relative elastic moduli of the various classes of SECEs follow the order: fluorides > oxides > sulfates > phosphates > nitrides > perovskites. Within the same framework structure, we observe that Na SECEs are softer than Li analogs. We discuss the implications of these findings on the context of fabrication, thermal expansion, and making a useful model. The data compiled in this work will also serve as a useful reference for future experiments as well as theoretical modeling of SECEs for rechargeable solid ion batteries. © 2015 The Electrochemical Society. [DOI: 10.1149/1.5022476] All rights reserved.

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Deng et al. (2016)

Electronic structure, thermomechanical and phonon properties of inverse perovskite oxide (Na_3OCl): An ab initio study

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Summary

Within first principles calculations, the electronic structure, thermodynamic, mechanical stability, magnetism, and phonon properties of the inverse perovskite (Na_3OCl) have been examined up to the Birch-Murnaghan derived bulk constant and bond-lengths are identical, when compared to the experimental data. A direct energy gap of 2.18 eV observed from the band structure reveals the semiconducting nature of the present oxide. Also, the application of strain on electronic properties predicts the decrease in bandgap with respect to compressive strain and vice versa. The constituent nonmagnetic atoms in its crystal possess the total magnetic moment to be zero and the same is supported by susceptibility data. In addition to the negative Cauchy's pressure, the small bulk modulus compared to Young's modulus determined from elastic constants, possibly claims it as a brittle material. Also, the temperature dependent Grüneisen parameter (1.59) and Debye temperature (362.2 K) are determined to reveal the lattice thermal conductivity ($\kappa \approx 0.46$ W/mK) at room temperature.

KEYWORDS

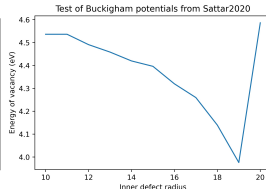
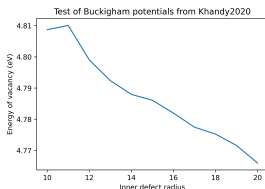
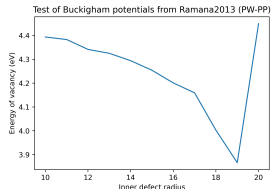
electronic structure, mechanical strength, phonon properties, structural properties, thermal conductivity

Khandy et al. (2020)

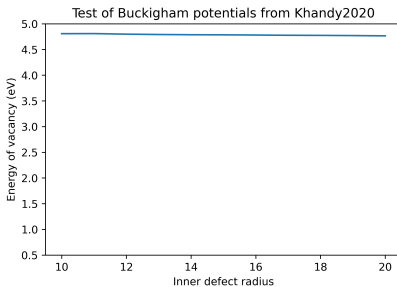
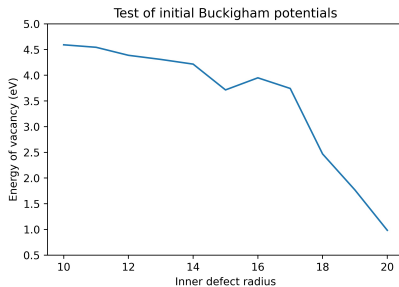
Ramana et al. (2013)

Data used and potentials calculated

Paper	Model	Bulk, GPa	Shear, GPa	Na-O Buckingham	Na-Cl Buckingham	Variation
Original	N/A	N/A	N/A	1226.84 0.307 0	2314.70 0.290 0	N/A
Ramana2013	FP-LAPW GGA	32.5	21.9	322.01 0.388 0	1727.87 0.297 0	3.55
Ramana2013	PW-PP GGA	34.2	22.9	369.22 0.376 0	1775.12 0.300 0	0.58
Deng2016	PAW GGA	36.4	24.6	1042.96 0.310 0	1591.38 0.288 0	20.35
Khandy2020	FP-LAPW GGA	33.45	26.87	588.38 0.338 0	1170.41 0.315 0	0.04
Sattar2020	FP-LAPW GGA	32.53	25.42	477.56 0.354 0	1270.12 0.309 0	0.61

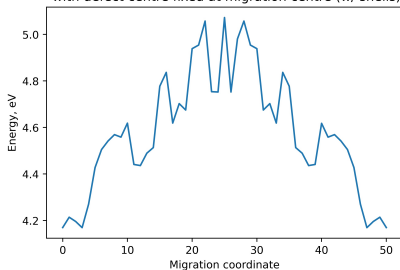


Comparison of Initial and Khandy potentials

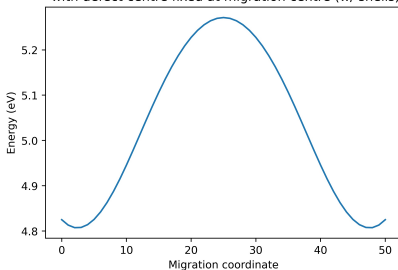


Comparison of Initial and Khandy potentials

Na vacancy migration using initial potentials
with defect centre fixed at migration centre (w/ shells)

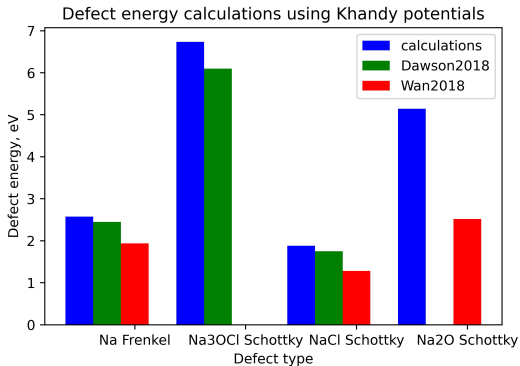


Na vacancy migration using Khandy potentials
with defect centre fixed at migration centre (w/ shells)



Calculations using potentials derived from Khandy2020

Parameter	Calc.	Comp. GGA	Comp. LDA	Comp. GULP	Experimental
lattice parameter, Å	4.41	4.54 ¹ , 4.538 ² , 4.53 ³ , 4.543 ⁴ , 4.514 ⁴ , 4.541 ⁵	4.382 ² , 4.381 ⁴ , 4.31 ⁶	4.501 ⁷	4.504 ⁸ , 4.496 ⁹ , 4.500 ¹⁰ , 4.4908 ¹¹
Na Frenkel, eV	2.58	1.94 ¹ , 2.45 ⁷			
NaCl Schottky, eV	1.88	1.28 ¹ , 1.75 ⁷			
Na ₂ O Schottky, eV	5.14	2.52 ¹			
Na ₃ OCl Schottky, eV	6.74	6.10 ⁷			
Na vacancy migration, eV	0.46	0.61 ¹ , 0.426 ¹¹ , 0.29 ⁸			0.63 ¹¹ , 1.04 ⁸



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