# Atomic-scale investigation of doping effects in the anti-perovskite Na3OCI sodium-ion battery material







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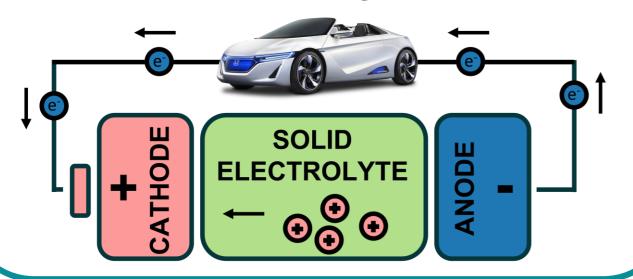
#### **Solid-State Batteries**

#### Advantages:

- stability
- safety
- energy density

#### **Issues:**

- ↓ ionic conductivity
- dopant effects
- interfaces
- grain boundaries



#### Sodium vs. Lithium

#### **Sodium batteries:**

- abundance
- cost
- sustainability
- for renewables

## **Lithium batteries:**

- energy density
- for portable electronics and electric vehicles





### **Project Aims**

**Atomistic modelling** of Na<sub>3</sub>OCl solid electrolyte to gain insight into:

- cation doping to increase relevant defect concentration
- Na-ion conduction mechanism and performance

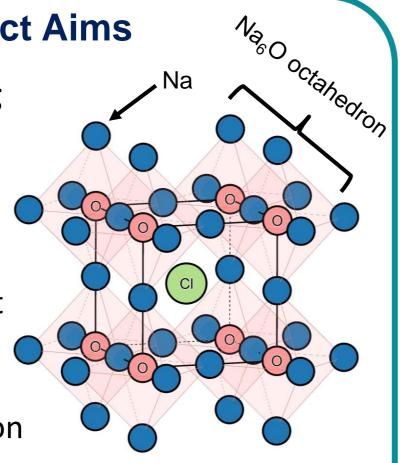


Fig. 1 – The anti-perovskite Na<sub>3</sub>OCI structure.

### Doping and ion conduction

- Na-ion conduction via vacancy migration along curved pathway at octahedral edges
- Aliovalent doping to induce 1.2% Na-vacancy conc. → increased ionic conductivity
- 500 -1.0 log<sub>10</sub>[α] -1.5 -2.0-2.5 $(1000/T)(K^{-1})$

Fig. 2 – Temperaturedependent Na+ conductivities doped Na<sub>3</sub>OCI with 1.2% vacancy concentration.

## **Clustering effects**

- Increased defect clustering in doped materials compared to undoped
- Leads to higher E<sub>a</sub> of Na-ion migration

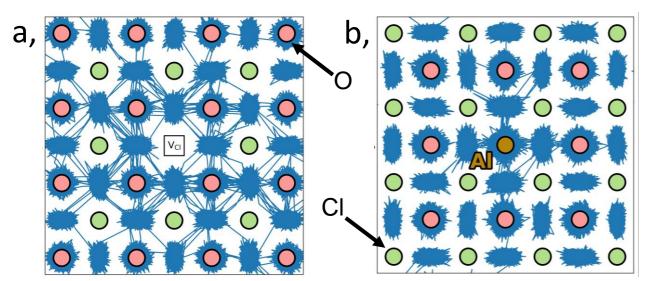


Fig. 3 – Na ion trajectories (blue) in (a) undoped and (b) Al-doped Na<sub>3</sub>OCl.

### **Conclusions and publication**

- Atomistic model reproduces Na<sub>3</sub>OCl structure accurately
- Favourable dopants include Mg<sup>2+</sup>, Ca<sup>2+</sup>, Al<sup>3+</sup> and Ga<sup>3+</sup>
- Smallest clustering effect and highest conductivity with Mg<sup>2+</sup> dopant at 1.2% vacancy conc.



Atomic-scale investigation of cation doping and defect clustering in the anti-perovskite Na<sub>3</sub>OCl sodium-ion conductor†

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