# MA1506 Mathematics II

Chapter 3
Basic Mathematical Modelling

## What is Modelling?

- Art of using mathematics to analyze simple situations
- To approximate complicated realistic situation

#### Mathematical Modelling

#### **Practical Problems**

- Improve efficiency of chemical reactor
- Maximize audio signal output
- Analyze impact of raising taxi fares

Develop a model

**Equations** 

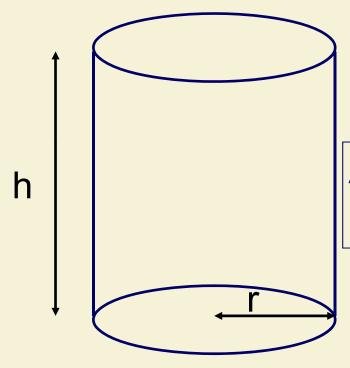
Validate / Implement

### How should cans be made?

- Aim: Minimize Costs
- Costs depends on
  - Raw material
  - Labour
  - Production costs
- Minimize amount of tin (aluminium)



#### Model 1: Minimize amount of tin



$$A = 2\pi r^2 + 2\pi rh$$

A is min when r = 0

Add Condition: Fixed Volume

$$V = \pi r^2 h$$

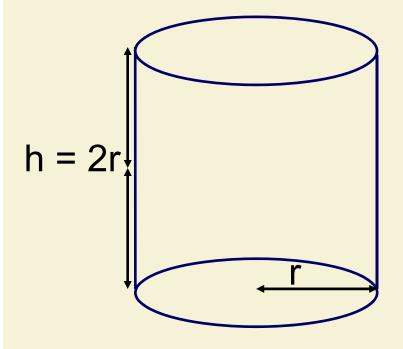
$$A = 2\pi r^2 + \frac{2V}{r}$$
$$A' = 4\pi r - \frac{2V}{r^2} = 0$$

$$V = 2\pi r^3$$



$$h=2r$$

## Model 1: h = 2r

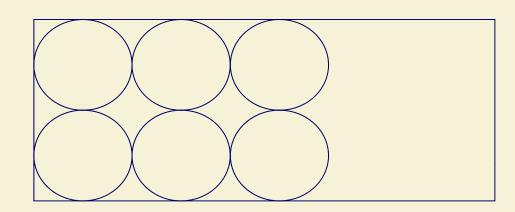




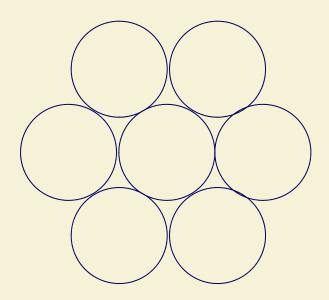
Something is not quite right!

## Model 2 : Model 1 + min wastage

Tops and bottoms



Better packing



## Model 2: Model 1 + min wastage

Wastage

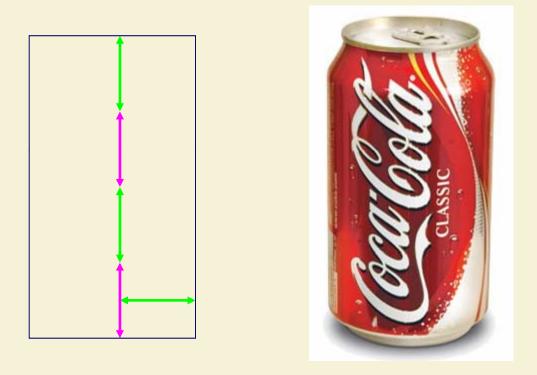
Area of hexagon =  $2\sqrt{3}r^2$ 

$$A = 4\sqrt{3}r^2 + \frac{2V}{r}$$

$$A' = 8\sqrt{3}r - \frac{2V}{r^2} = 0$$

$$\pi r^2 h = V = 4\sqrt{3}r^3$$

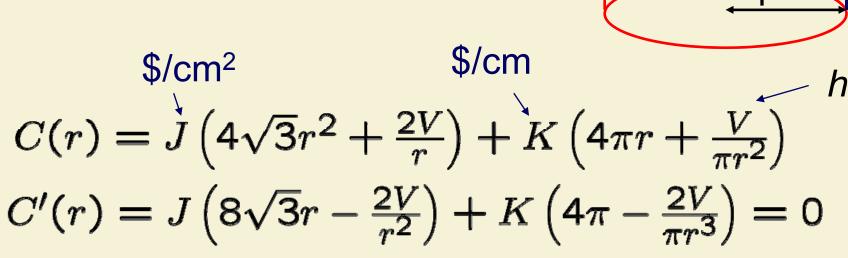
## Model 2 : h = 2.2r



Still not quite right!

## Model 3 : Model 2 + Manufacturing Process

Welding the top and bottom and side



$$\Rightarrow \frac{h}{r} = \frac{4\sqrt{3} + \frac{2\pi K}{rJ}}{\pi + \frac{K}{rJ}}$$

units : K/J = cm

#### Model 3: Model 2 + Manufacturing Process

K/J sets the scale

$$\frac{h}{r} = \frac{4\sqrt{3} + \frac{2\pi K}{rJ}}{\pi + \frac{K}{rJ}}$$

$$\frac{h}{r} \approx \frac{4\sqrt{3}}{\pi}$$

Model 2



$$\frac{h}{r} pprox rac{\frac{2\pi K}{rJ}}{\frac{K}{rJ}} = 2\pi$$



## Summary

 We have been constructing models, that is, very simple versions of a real problem. The real problem is very complicated, the model is just an <u>approximation</u>, but it is easier to understand.

 Basic Principle: begin with simple models, understand their weaknesses, and only then make them more complicated!

#### 3.2 Malthus Model of Population

Total Population: N(t)

Per Capita Birth-Rate, B

# babies born in  $\delta t = BN \delta t$ 

Per Capita Death-Rate, D

# deaths in  $\delta t = DN \delta t$ 

 $\delta N = \#$  births - # deaths =  $(B-D)N\delta t$ 



Thomas Malthus 1766 -1834

#### 3.2 Malthus Model of Population

$$\frac{dN}{dt} = (B - D)N = kN$$

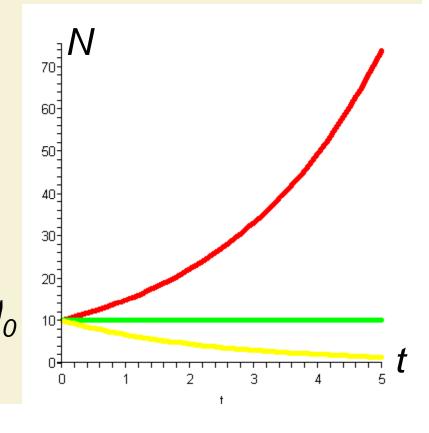
$$\int \frac{dN}{N} = \int kdt = k \int dt = kt + c$$

$$N(t) = Ae^{kt} = N_0 e^{kt}$$

k > 0 : population explosion

k = 0: stable

k < 0 : extinction



 $e^{kt}$  Grows too quickly

Competition

Death rate is a function of population

D = sN (logistic) Assumption

Simple models before complicated ones!

$$D = sN$$
 (logistic) Assumption

Unit of 
$$D = \frac{\text{\# death per sec}}{\text{\# population}} = \sec^{-1}$$

Unit of  $s = sec^{-1}$ 

$$D = sN$$
 (logistic) Assumption

$$\frac{dN}{dt} = BN - DN = BN - sN^2$$
logistic equation

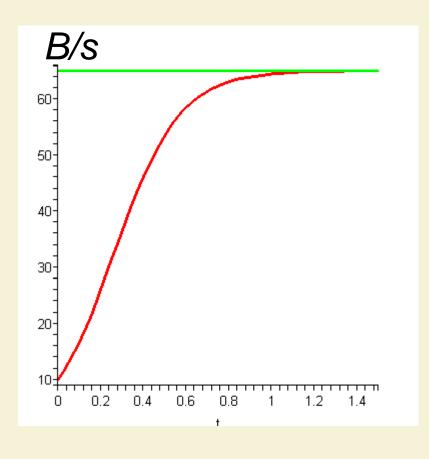
Small 
$$N \longrightarrow \frac{dN}{dt} \approx BN$$
 (s $N^2$  small)

$$N(t) \approx \hat{N}e^{Bt}$$

Initially grows exponentially

$$\frac{dN}{dt} = BN - sN^2$$

Small 
$$N \rightarrow \frac{dN}{dt} \approx BN \rightarrow N(t) \approx \hat{N}e^{Bt}$$



as *N* increases, *N*<sup>2</sup> grows faster

Grow rate decreases!

until  $BN = s N^2$ 

$$\frac{dN}{dt} = BN - sN^2$$

$$t = \int \frac{dN}{N(B - \underline{s}\underline{N})} + c$$
 non zero

Use partial fraction

$$\frac{1}{N(B-sN)} = \frac{\alpha}{N} + \frac{\beta}{B-sN}$$

$$1 = \alpha(B - sN) + \beta N = \alpha B + (\beta - \alpha s)N$$

$$1 = \alpha B, \beta = \alpha s$$
Unknown

$$\frac{dN}{dt} = BN - sN^2$$

$$t = \int \frac{dN}{N(B - sN)} + c$$

$$\int \frac{dN}{N(B-sN)} = \frac{1}{B} \int \frac{dN}{N} + \frac{s}{B} \int \frac{dN}{B-sN}$$
$$= \frac{1}{B} \ln |N - \frac{1}{B} \ln |B - sN| + c.$$

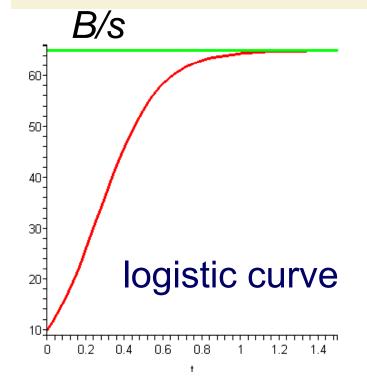
assume B-sN>0

$$\frac{N}{B-sN} = Ke^{Bt}$$

$$\frac{dN}{dt} = BN - sN^2$$

$$\frac{N}{B - sN} = Ke^{Bt}$$

$$N(0) = \hat{N}$$



$$\frac{N}{B - sN} = \frac{\hat{N}}{B - s\hat{N}}e^{Bt}$$

$$N(t) = \frac{B}{s + \left(\frac{B}{\widehat{N}} - s\right)e^{-Bt}}$$

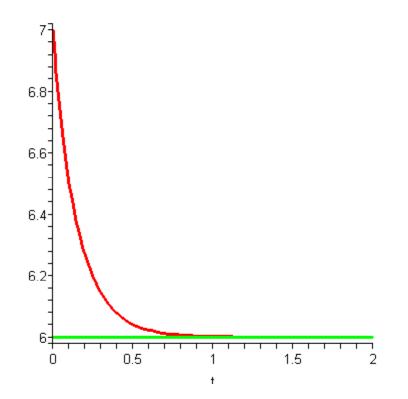
assumed B-sN>0

$$\frac{dN}{dt} = BN - sN^2$$

assume B-sN<0

$$t = \frac{1}{B} \ln N - \frac{1}{B} \ln |B - sN| + c$$

$$= \frac{1}{B} \ln \frac{N}{sN-B} + c$$



$$N(t) = \frac{B}{s - \left(s - \frac{B}{\hat{N}}\right)e^{-Bt}}$$

$$\frac{dN}{dt} = BN - sN^2$$

## B/s is the carrying capacity or sustainable population

set 
$$N_{\infty} = B/s$$

$$N(t) = \frac{N_{\infty}}{1 + \left(\frac{N_{\infty}}{\hat{N}} - 1\right)e^{-Bt}}$$

$$N(t) = \frac{N_{\infty}}{1 - \left(1 - \frac{N_{\infty}}{\hat{N}}\right)e^{-Bt}}$$

$$N(t) = N_{\infty}$$

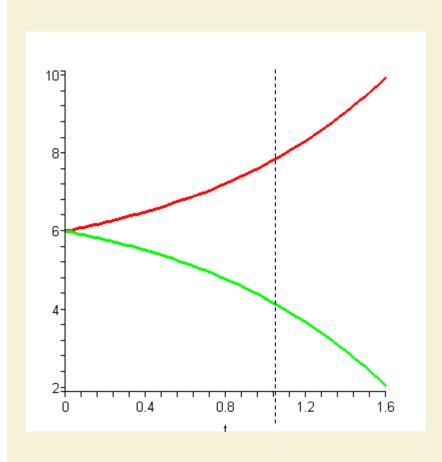
$$\hat{N} < N_{\infty}$$

$$\hat{N} > N_{\infty}$$

$$\hat{N} = N_{\infty}$$

#### Remark

## Will such a situation happen?



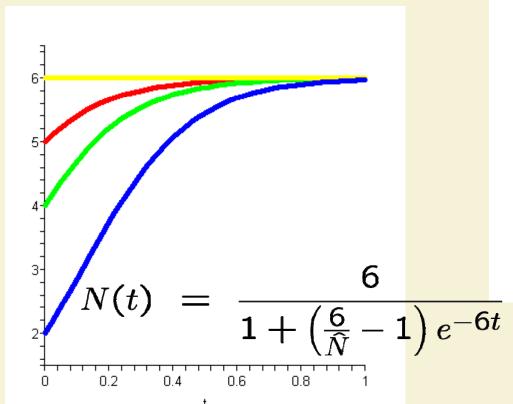
Not for real world situation

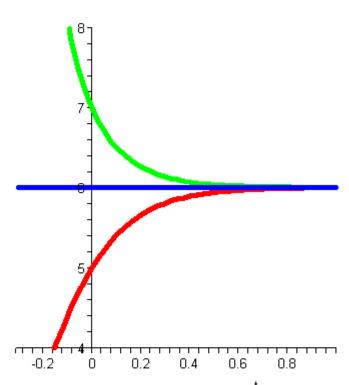
- choose a solution
- use a new model

#### 3.4 No Crossing Principle

## Solution curves of a first order ode should never cross each other.

If N > B/s initially, it will never cross B/s





#### 3.5 Harvesting

## **Assumptions**

- Fish population : N Logistic model
- Constant rate of fishing, E

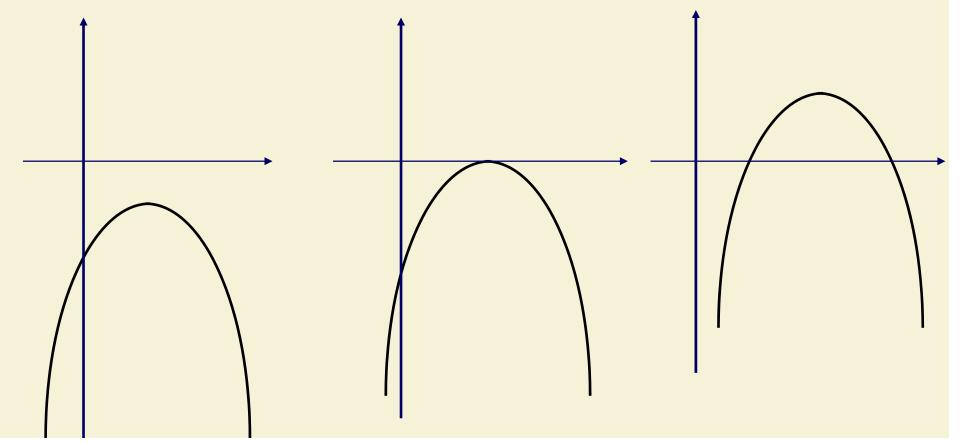
$$\frac{dN}{dt} = (B - sN)N - E$$

**Basic Harvesting Model** 

#### 3.5 Harvesting

$$\frac{dN}{dt} - F(N) = -sN^2 + BN - E$$

$$\operatorname{disc} = B^2 - 4(-s)(-E)$$



$$E>rac{B^2}{4s}$$

$$F(N) = -sN^2 + BN - E$$

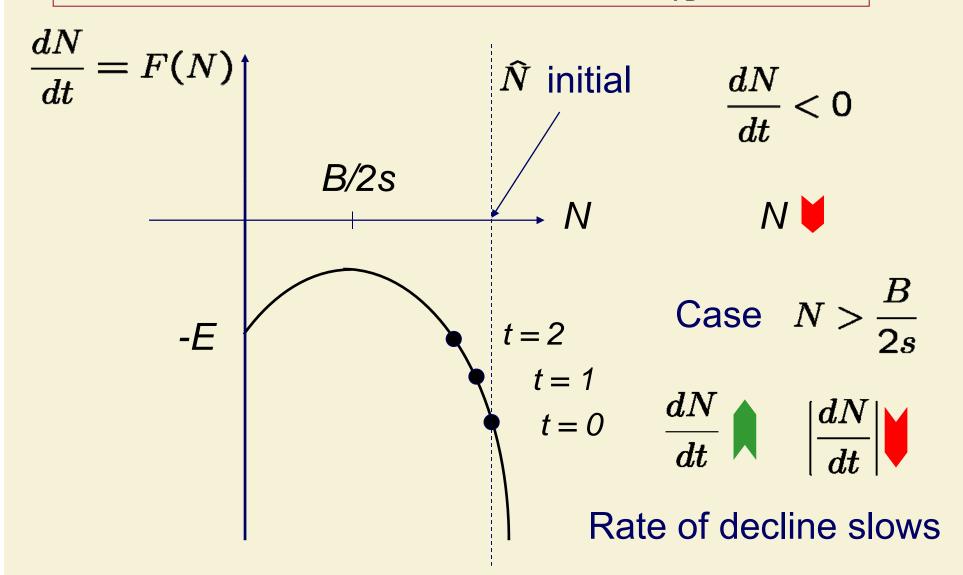
$$\frac{dN}{dt} = F(N)$$

$$B/2s$$

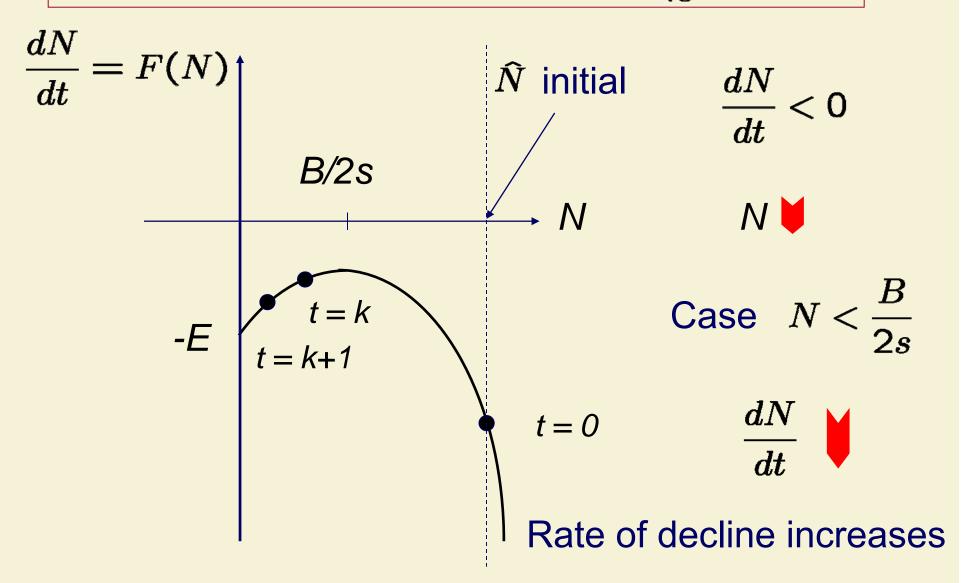
$$R > \frac{B^2}{4s}$$

$$E > \frac{B^2}{4s}$$
Where is  $t$ ?

$$E > \frac{B^2}{4s}$$

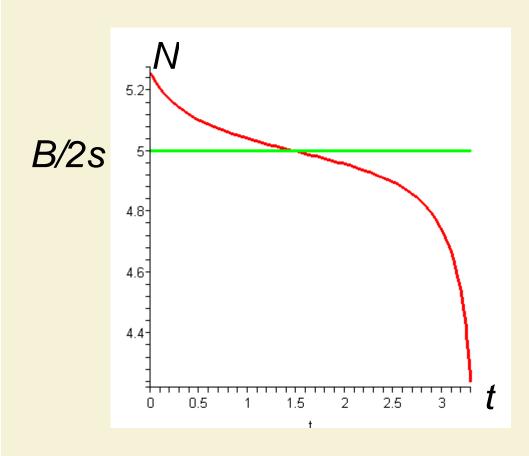


$$E > \frac{B^2}{4s}$$

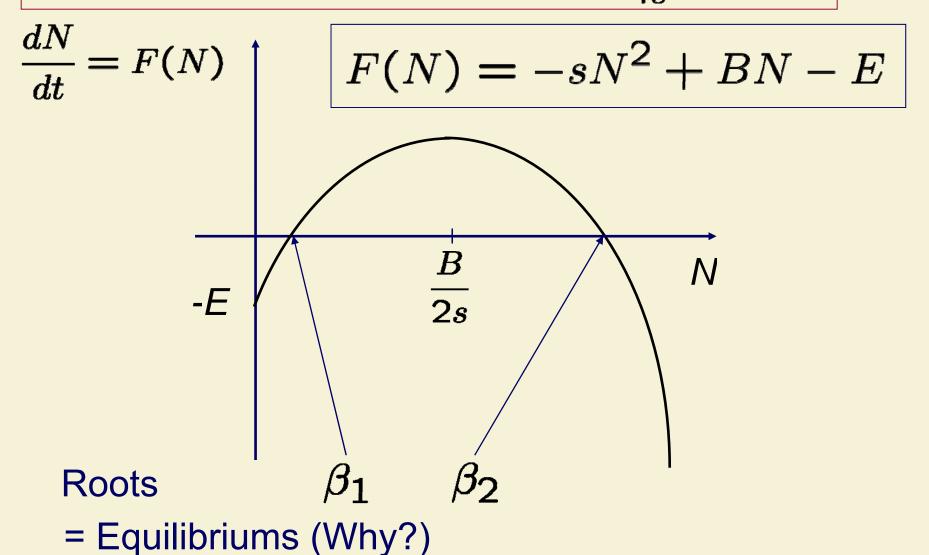


$$E > \frac{B^2}{4s}$$

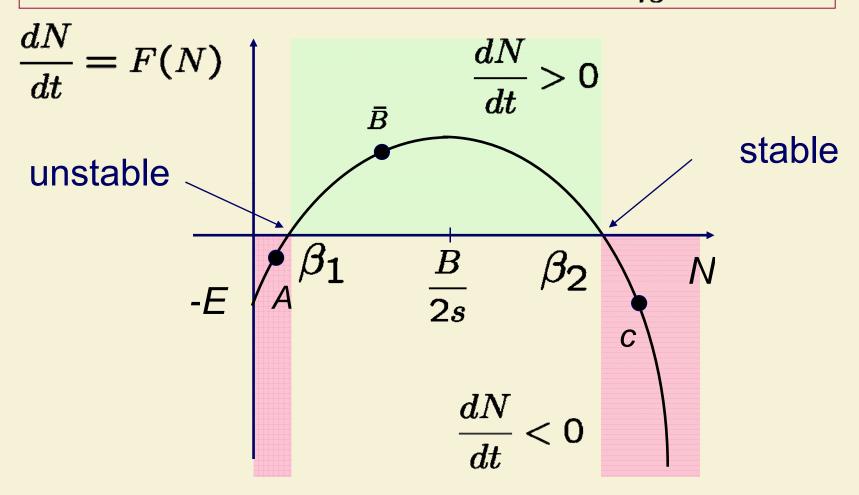
## What is the equation of this graph?



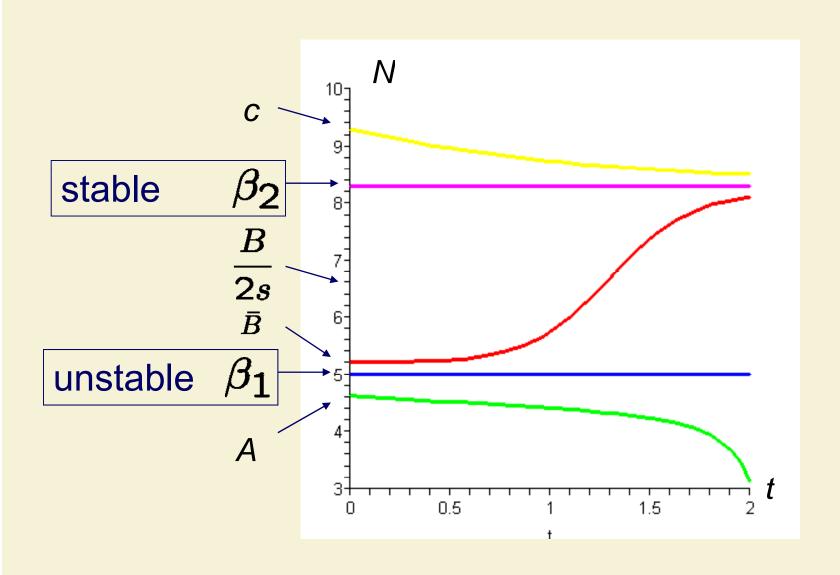
$$E<rac{B^2}{4s}$$



$$E<rac{B^2}{4s}$$



$$E<rac{B^2}{4s}$$



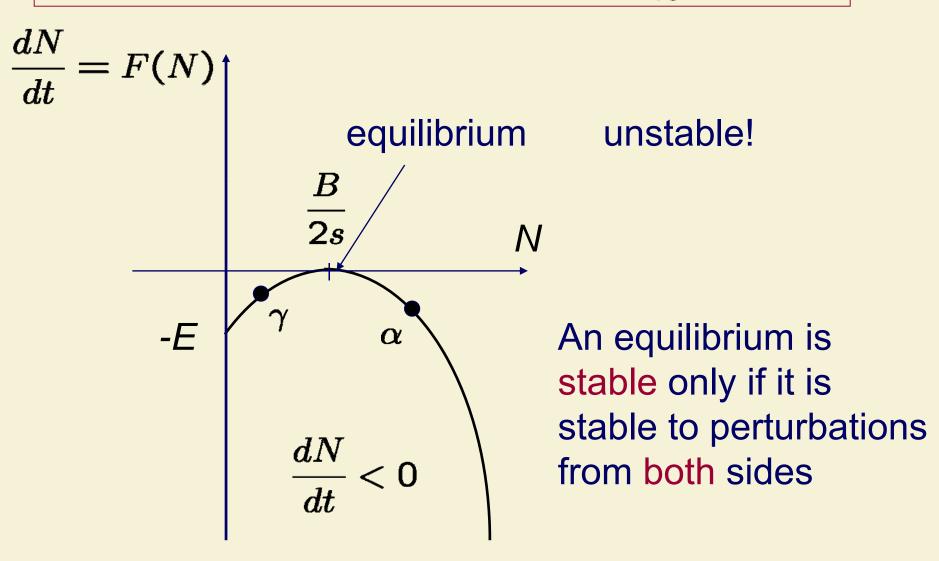
#### 3.5 Extinction Time

$$\frac{dN}{dt} = -sN^2 + BN - E$$

$$\int_0^T dt = T = \int_{\widehat{N}}^0 \frac{dN}{N(B - sN) - E}$$

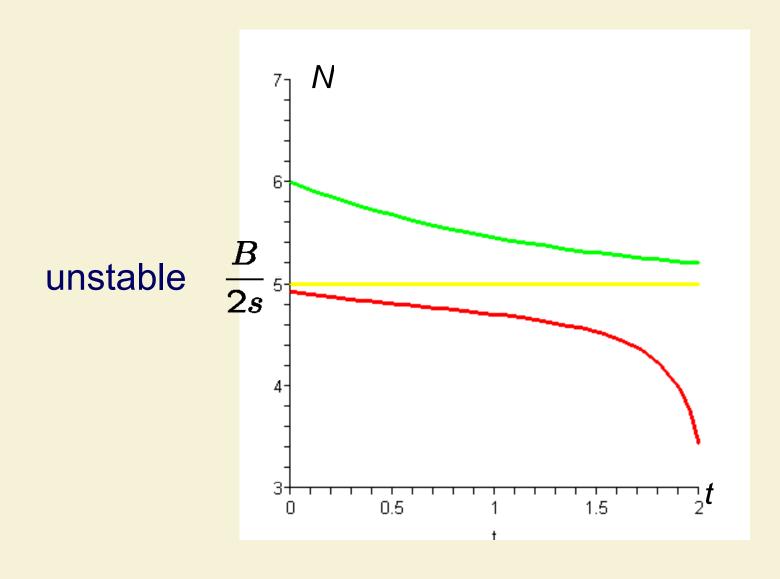
How to integrate?

$$E = \frac{B^2}{4s}$$



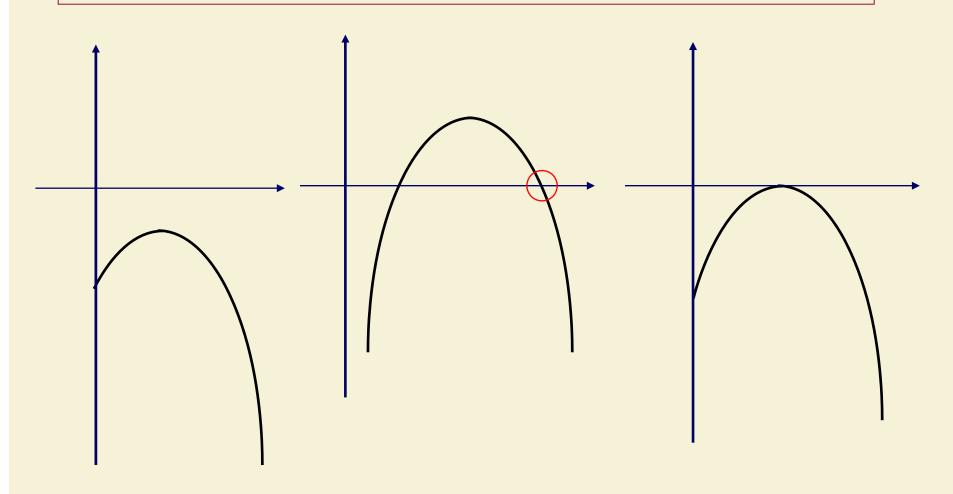
# 3.5 Harvesting Rate

$$E = \frac{B^2}{4s}$$



# 3.5 Stable Equilibrium

$$E < \frac{B^2}{4s}$$

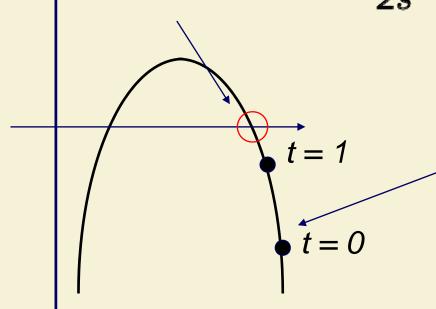


#### 3.5 Remark

Assume you started from no fishing.

$$\hat{N} = \frac{B}{s}$$
 carrying capacity

$$\beta_2 = \frac{B + \sqrt{B^2 - 4Es}}{2s} < \frac{B + \sqrt{B^2 - 0}}{2s} = \frac{B}{s}$$



In order to not overfish

Rate 
$$E < \frac{B^2}{4s}$$

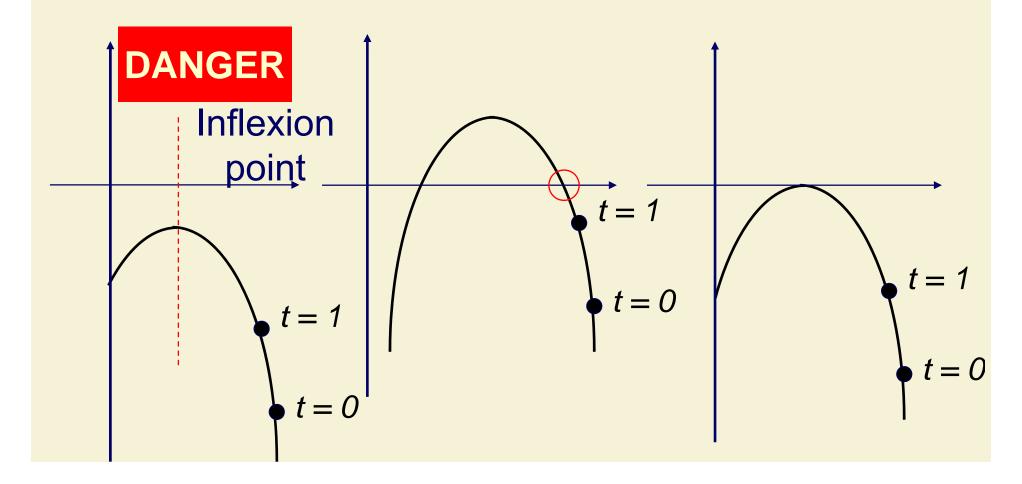
### 3.5 Remark

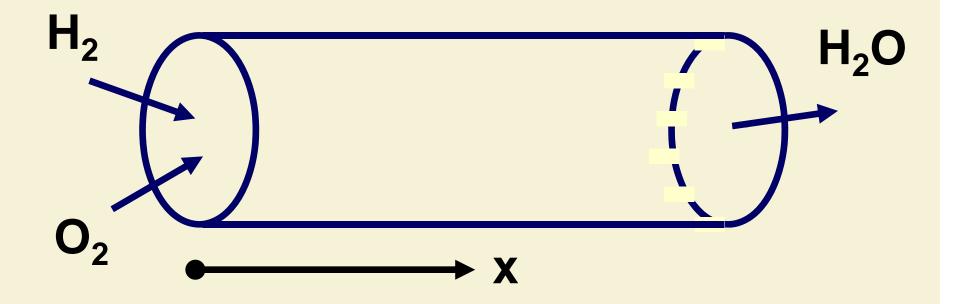
How to tell which situation are we in?

In all cases

$$\frac{dN}{dt}$$

even when you overfish!





Assumption: Oxygen is cheap

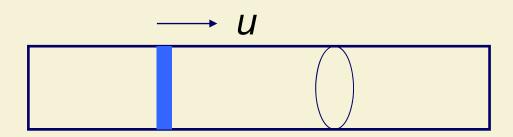
What is the concentration of Hydrogen?

# What is the concentration of Hydrogen?

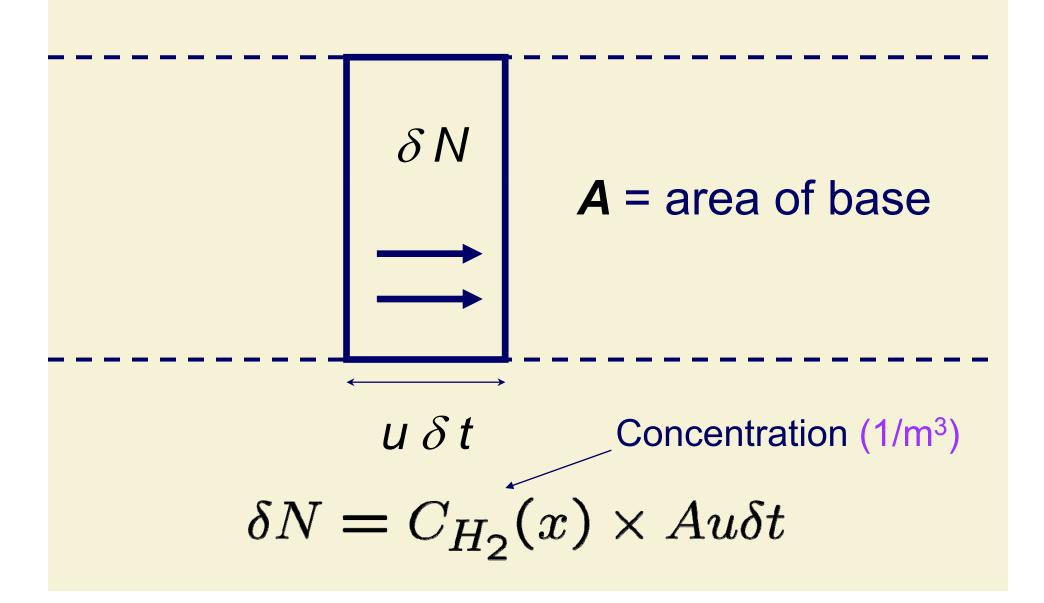
as a function of position in PFR

# **Assumptions:**

- 1. Reagents flow at constant speed u
- 2. Uniform cross section area, A
- 3. No mixing upstream or downstream
- 4. Temp constant



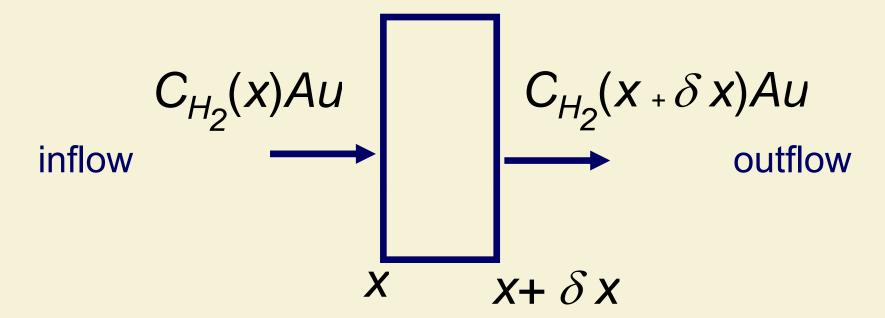
# 3.6 PFR: Counting H<sub>2</sub>



$$\delta N = C_{H_2}(x) \times Au\delta t$$

$$\frac{dN}{dt} = \lim_{\delta t \to 0} \frac{\delta N}{\delta t} = C_{H_2}(x)Au$$

$$\frac{dN}{dt} = C_{H_2}(x)Au$$

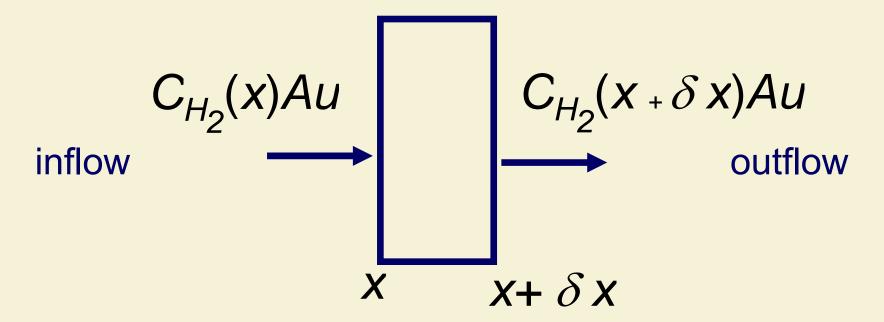


Hydrogen molecules destroyed at rate:

$$-2rA\delta x$$

 $r = \text{rate of chemical reaction per unit volume } (1/\text{sm}^3)$ 

$$\frac{dN}{dt} = C_{H_2}(x)Au$$



$$C_{H_2}(x)Au - 2\eta A\delta x = C_{H_2}(x + \delta x)Au$$

$$2H_2 + O_2 = 2H_2 O$$

$$C_{H_2}(x)Au - C_{H_2}(x + \delta x)Au - 2rA\delta x = 0$$

concentration

change in 
$$\delta C_{H_2} A u = -2rA\delta x$$

$$u\frac{dC_{H_2}}{dx} = u\lim_{\delta x \to 0} \frac{\delta C_{H_2}}{\delta x} = -2r$$

Depends on ... temp, conc of  $H_2$  ...

Assume 
$$r = kC_{H_2}(x)$$

unit of k (1/s)

$$u\frac{dC_{H_2}}{dx} = -2r = -2kC_{H_2}$$
 
$$\frac{dC_{H_2}}{C_{H_2}} = -\frac{2k}{u}dx$$
 dimensionless

$$C_{H_2} = C_{H_2}(0)e^{\underbrace{-2kx}{u}}$$

x/u = Time

Let X = length of PFR, T = total time, u = X/T

$$C_{H_2}(exit) = C_{H_2}(entrance)e^{-2kT}$$

- PFRs are efficient
- Our model did not consider temperature

### **Digression: Chemical Reactions**

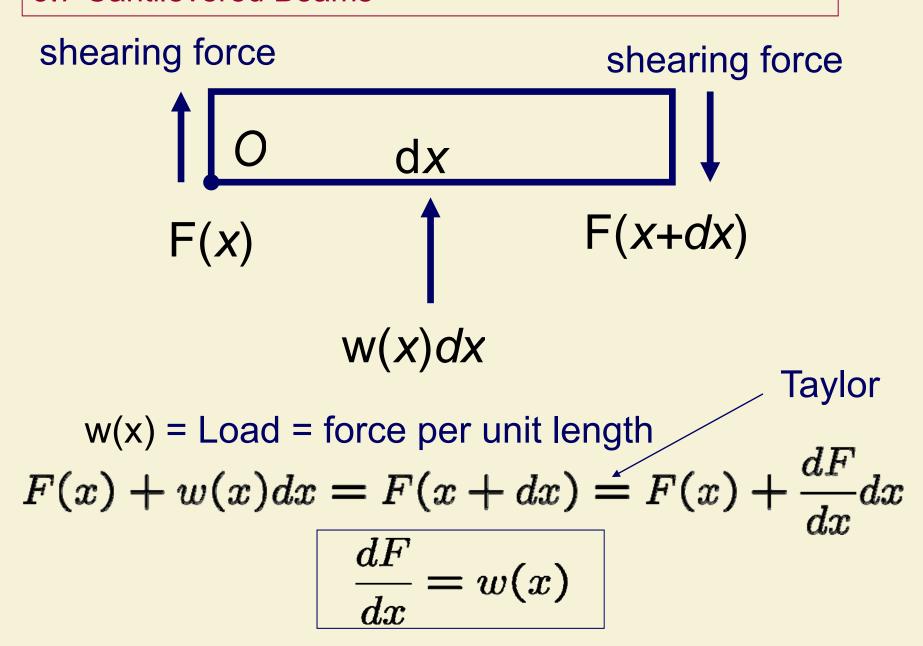
Assume 
$$r = kC_{H_2}(x)$$

• First order reaction  $\frac{dX}{dt} = kX$ 

t-butyl chloride into t-butyl alcohol

• Second order reaction  $\frac{dX}{dt} = k(\alpha - X)(\beta - X)$ 

methyl chloride into methyl alcohol



Torque (moment) = rotational force

$$M(x) = \text{force x dist}$$

$$M(x+dx)$$

$$O \quad dx$$

$$F(x) \qquad F(x+dx)$$

$$W(x)dx$$

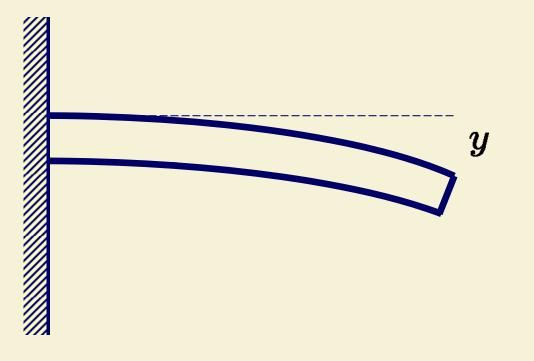
$$M(x) + F(x+dx)dx = M(x+dx) + (w(x)dx)\frac{dx}{2}$$

$$M(x) + F(x+dx)dx = M(x+dx) + (w(x)dx)\frac{dx}{2}$$

$$M(x) + F(x)dx + \frac{dF}{dx}(dx)^{2} \leftarrow \text{small}$$

$$= M(x) + \frac{dM}{dx}dx + \frac{1}{2}w(x)(dx)^{2}$$

$$\frac{dM}{dx} = F \qquad \qquad \frac{d^2M}{dx^2} = \frac{dF}{dx} = w(x)$$



Deflection

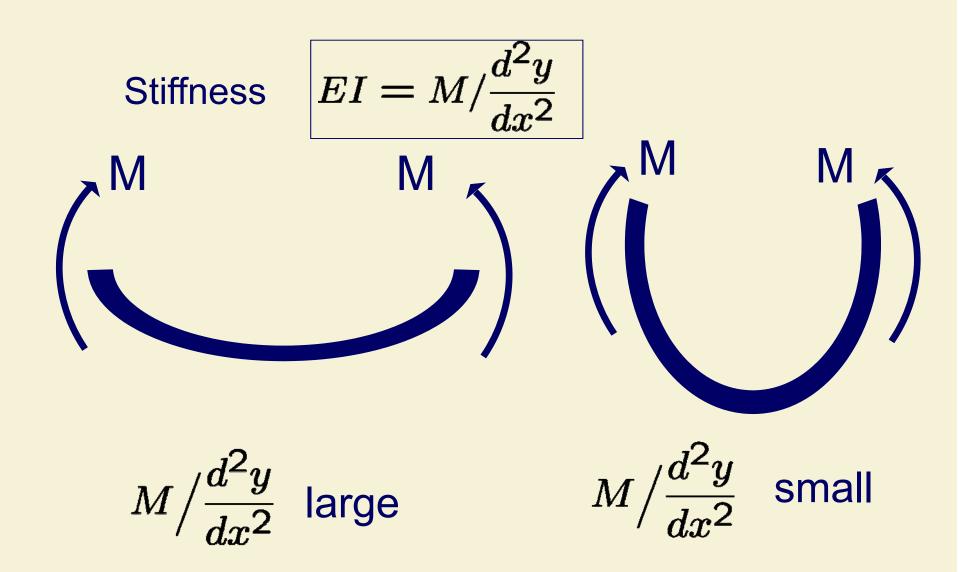
Depends on

Stiffness

# Stiffness depends on

- Material: Young's modulus, E
- Shape of cross section , I





$$EI = M / \frac{d^2y}{dx^2}$$

$$\frac{d^2y}{dx^2} = \frac{M}{EI}$$

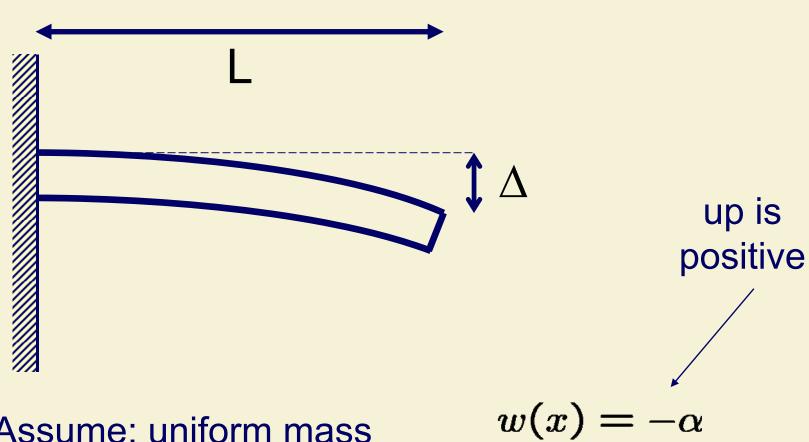
$$\frac{d^2y}{dx^2} = \frac{M}{EI}$$

$$\frac{d^4y}{dx^4} = \frac{1}{EI} \frac{d^2M}{dx^2} = \frac{w(x)}{EI}$$

$$\frac{d^4y}{dx^4} = \frac{w(x)}{EI}$$

http://en.wikipedia.org/wiki/Euler-Bernoulli beam equation

$$\frac{d^4y}{dx^4} = \frac{w(x)}{EI}$$



Assume: uniform mass

$$\frac{d^4y}{dx^4} = \frac{w(x)}{EI}$$

$$\frac{d^4y}{dx^4} = \frac{-\alpha}{EI} \quad \Longrightarrow \quad \frac{d^3y}{dx^3} = -\frac{\alpha x}{EI} + A$$

Recall 
$$\frac{d^3y}{dx^3} = \frac{1}{EI}\frac{dM}{dx} = \frac{F(x)}{EI}$$

$$F(L) = 0 = EI\frac{d^3y}{dx^3}(L) = EI\left(-\frac{\alpha L}{EI} + A\right)$$

No shearing force at L.

$$A = \frac{\alpha L}{EI}$$

$$\frac{d^3y}{dx^3} = -\frac{\alpha x}{EI} + \frac{\alpha L}{EI} \implies \frac{d^2y}{dx^2} = -\frac{\alpha x^2}{2EI} + \frac{\alpha Lx}{EI} + B$$

Recall 
$$\frac{d^2y}{dx^2} = \frac{M}{EI}$$

$$M(L) = 0 = EI\frac{d^2y}{dx^2}(L)$$

No bending moment at L

$$= EI\left(-\frac{\alpha L^2}{2EI} + \frac{\alpha L^2}{EI} + B\right)$$

$$\Rightarrow B = \frac{\alpha L^2}{2EI} - \frac{\alpha L^2}{EI} = -\frac{\alpha L^2}{2EI}$$

$$\Rightarrow B = \frac{\alpha L^2}{2EI} - \frac{\alpha L^2}{EI} = -\frac{\alpha L^2}{2EI}$$

$$\frac{d^4y}{dx^4} = \frac{w(x)}{EI}$$

$$\frac{d^2y}{dx^2} = -\frac{\alpha x^2}{2EI} + \frac{\alpha Lx}{EI} - \frac{\alpha L^2}{2EI}$$

$$\frac{dy}{dx}(0) = 0 \quad \longrightarrow \quad C=0$$

No curvature at x=0

$$y = -\frac{\alpha x^4}{24EI} + \frac{\alpha L x^3}{6EI} - \frac{\alpha L^2 x^2}{4EI} + D$$

# Cantilever deflection formula

$$y = -\frac{\alpha x^4}{24EI} + \frac{\alpha L x^3}{6EI} - \frac{\alpha L^2 x^2}{4EI} + D$$

No deflection at x=0

$$\rightarrow$$
 D=0

$$y = \frac{\alpha L^4}{2EI} \left( -\frac{1}{12} \left( \frac{x}{L} \right)^4 + \frac{1}{3} \left( \frac{x}{L} \right)^3 - \frac{1}{2} \left( \frac{x}{L} \right)^2 \right)$$

$$\Delta = y(L) = \frac{\alpha L^4}{2EI} \left( -\frac{1}{12} + \frac{1}{3} - \frac{1}{2} \right)$$
$$= -\frac{\alpha L^4}{8EI}$$