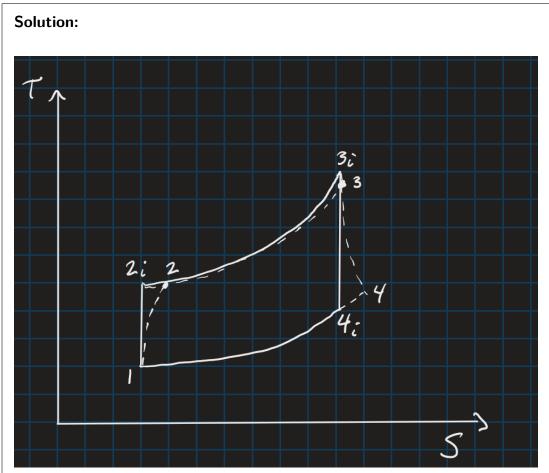
1. A jet engine is traveling through the air with the forward velocity of 300 m/s. The exhaust gases leave the nozzle with an exit velocity of 800 m/s with respect to the nozzle. If the mass flow rate through the engine is 10 kg/s, determine the jet engine thrust. The exit plane static pressure is 80 kPa, inlet plane static pressure is 20 kPa, ambient pressure surrounding the engine is 20 kPa, and the exit plane area is 4.0 m<sup>2</sup>.

## **Solution:**

$$\begin{cases} V_{\infty} = 300 \frac{m}{s} \\ V_{e} = 800 \frac{m}{s} \\ P_{e} = 80 \text{kPa} \\ P_{\text{atm}} = 20 \text{kPa} \\ A_{e} = 4 \text{ m}^{2} \\ \dot{m} = 10 \frac{\text{kg}}{\text{s}} \end{cases}$$

$$\begin{split} \mathsf{T} &= \dot{\mathsf{m}} \mathsf{V}_{\varepsilon} - \dot{\mathsf{m}} \mathsf{V}_{\infty} + \mathsf{A}_{\varepsilon} (\mathsf{P}_{\varepsilon} - \mathsf{P}_{\mathsf{atm}}) \\ & \therefore \mathsf{T} = 245 \mathsf{kN} \end{split}$$

2. Describe the differences between Brayton Cycle and a Real Gas Turbine Cycle. Make diagrams to explain the losses associated with a real engine.



The first losses occur in the compressor from stage 1->2 due to the flow not being reversible. The combustion process is not completely isobaric. The process in the turbine is also not reversible.

3. Draw the T-s diagram and determine the turbine shaft power, and the air-fuel ratio

**Solution:** With all the given infomation the only equations used to find the shaft power were:

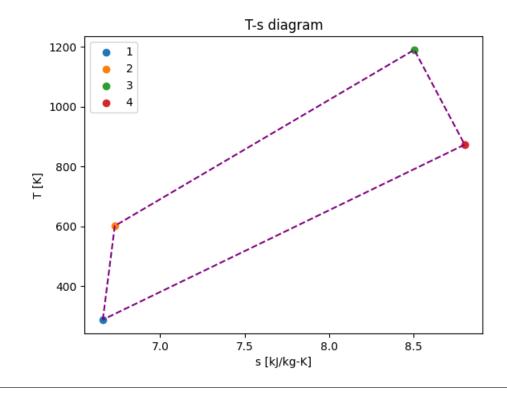
$$\begin{split} P_{02} &= r P_{01} \\ T_{02} &= T_{01} \left[ 1 + \frac{1}{\eta_c} \left( r^{\frac{\gamma - 1}{\gamma}} - 1 \right) \right] \\ W_* &= \frac{1}{\eta_*} c p_* (T_i - T_j) \\ P &= \dot{m}_{gas} (W_T - W_C) \end{split}$$

 $P_{\text{shaft}} = 5740.3282 \text{ [kW]}$ 

The entropy starting values were pulled from Dr.Cizmas' textbook from the air tables for  $s_1$  and stoichiometric tables for  $s_3$ . The following equation was used to find the other two values:

$$s_i - s_j = c p_* \ln \frac{T_i}{T_j} - R \ln \frac{P_i}{P_j}$$

which then produces the following T-s graph.



The ideal fuel to air ratio was pulled from the 5th set of lecture slides:

$$\Delta t_c = 588.4864 \rightarrow f \approx 0.014$$

$$\therefore f^{-1} \approx 71.429$$

```
In [1]:
         ### Benjamin Tollison ###
         import matplotlib.pyplot as plt
         import numpy as np
         import pandas as pd
         import scipy
         import sympy as sp
         from IPython.display import Latex, Math, display
         from sympy import (
             Eq,
             Function,
             Matrix,
             cos,
             cosh,
             exp,
             integrate,
             lambdify,
             рi,
             sin,
             sinh,
             symbols,
         from decimal import Decimal
         from sympy.solvers.pde import pdsolve
         from sympy.solvers.solveset import linsolve
         def displayEquations(LHS,RHS):
             left = sp.latex(LHS)
             right = sp.latex(RHS)
             display(Math(left + '=' + right))
             np.set_printoptions(suppress=True)
         def displayVariable(variable:str,RHS):
             left = sp.latex(symbols(variable))
             right = sp.latex(RHS)
             display(Math(left + '=' + right))
         def displayVariableWithUnits(variable:str,RHS,units):
             left = sp.latex(symbols(variable))
             right = sp.latex(RHS)
             latexUnit = sp.latex(symbols(units))
             display(Math(left + '=' + right + '\\;' +'\\left['+ latexUnit + '\\right]'))
         def format scientific(number:float):
             a = '%E' % number
             return a.split('E')[0].rstrip('0').rstrip('.') + 'E' + a.split('E')[1]
         deg2rad = np.pi/180
         rad2deg = 180/np.pi
         thrust = 10*(800-300) + 4*(80-20)*1000
         displayVariableWithUnits('T',thrust,'N')
```

```
In [2]:
```

T = 245000 [N]

3)

```
In [3]:
         T01 = 288 \# K
         P01 = 101325 \# Pa
         compressor_ratio = 10.3
         compressor_isentropic_efficiency = 0.87
         mechanical_efficiency = 0.99
         combustion_pressure_loss = 0.05
         LHV_kerosene = 41 \# J/kg
         combustion afficiency - 0 99
```

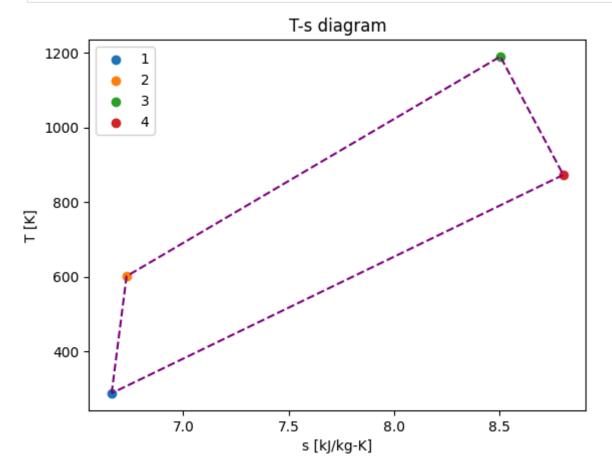
```
T03 = 1190 \# K
         turbine_efficiency = 0.88
         T4 = 873 \# K
         massflow_rate = 108 # kg/s
         cp_air = 1005 \# J/kg-K
         cp_gas = 1150 \# J/kg-K
         cycle_dict = {
           '1':{'T0': T01,'P0':P01}
         gamma_air = 1.4
         gamma_gas = 1.33
         display(pd.DataFrame(cycle_dict))
                1
       T<sub>0</sub>
              288
       P0
          101325
In [4]:
         P02 = compressor_ratio*P01
         T02 = T01 * (1 + (1/compressor_isentropic_efficiency)*(compressor_ratio**((gamma_air-1)
         displayVariableWithUnits('P_{02}',P02,'Pa')
         displayVariableWithUnits('T_{02}',T02,'K')
         compressor_work = cp_air*(T02-T01)
         displayVariableWithUnits('W_c',compressor_work,'J')
       P_{02} = 1043647.5 [Pa]
       T_{02} = 601.513567990099 [K]
       W_c = 315081.135830049 \ [J]
In [5]:
         turbine_work = cp_gas*(T03-T4)/mechanical_efficiency
         displayVariableWithUnits('W_T',turbine_work,'J')
       W_T = 368232.323232323 [J]
In [6]:
         shaft_power = massflow_rate*(turbine_work-compressor_work)
         displayVariableWithUnits('P_{shaft}',shaft_power,'W')
       P_{shaft} = 5740328.23944558 \ [W]
In [7]:
         501 = 6.6608
         S02 = cp_air*np.log(T02/T01)/1000 - .287*np.log(compressor_ratio) + S01
         displayVariableWithUnits('s_{02}',round(S02,4),'\\frac{kJ}{kgK}')
         503 = ((8.5067 - 8.4956)/(1193.16 - 1183.16))*(1190 - 1183.16) + 8.4956
         displayVariableWithUnits('s_{03}',round(S03,4),'\\frac{kJ}{kgK}')
         S04 = cp_gas*np.log(T4/T03)/1000 - .287*np.log(P01/(P02*0.95)) + S03
         s_{02} = 6.7316
      s_{03} = 8.5032 \quad \left| \frac{kJ}{kgK} \right|
```

combascion\_crricicy

```
s_{04} = 8.8016 \, \left[ rac{kJ}{kgK} 
ight]
```



```
In [8]:
    temperature_values = [T01,T02,T03,T4,T01]
    entropy_values = [S01,S02,S03,S04,S01]
    for i in range(len(temperature_values)-1):
        plt.scatter(entropy_values[i],temperature_values[i],label=f'{i+1}')
    plt.plot(entropy_values,temperature_values,label=None,color='purple',linestyle='--')
    plt.xlabel('s [kJ/kg-K]')
    plt.ylabel('T [K]')
    plt.legend()
    plt.title("T-s diagram")
    plt.show()
```



```
In [9]:
    displayVariableWithUnits('\\Delta{t_{c}}',round(T03-T02,4),'K')
    fuel_ratio_ideal = 0.014
    air_to_fuel = fuel_ratio_ideal**-1
    displayVariable('f^{-1}',air_to_fuel)
```

```
\Delta t_c = 588.4864 \; [K] f^{-1} = 71.4285714285714
```