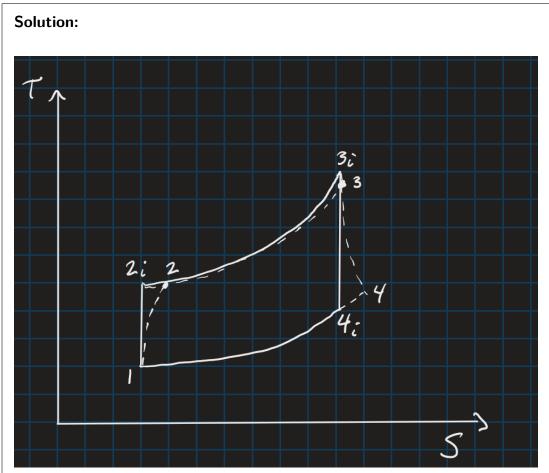
1. A jet engine is traveling through the air with the forward velocity of 300 m/s. The exhaust gases leave the nozzle with an exit velocity of 800 m/s with respect to the nozzle. If the mass flow rate through the engine is  $10~\rm kg/s$ , determine the jet engine thrust. The exit plane static pressure is  $80~\rm kPa$ , inlet plane static pressure is  $20~\rm kPa$ , ambient pressure surrounding the engine is  $20~\rm kPa$ , and the exit plane area is  $4.0~\rm m^2$ .

## **Solution:**

$$\begin{cases} V_{\infty} = 300 \frac{m}{s} \\ V_{e} = 800 \frac{m}{s} \\ P_{e} = 80 \text{kPa} \\ P_{\text{atm}} = 20 \text{kPa} \\ A_{e} = 4 \text{ m}^{2} \\ \dot{m} = 10 \frac{\text{kg}}{\text{s}} \end{cases}$$

$$\begin{split} \mathsf{T} &= \dot{\mathsf{m}} \mathsf{V}_{\varepsilon} - \dot{\mathsf{m}} \mathsf{V}_{\infty} + \mathsf{A}_{\varepsilon} (\mathsf{P}_{\varepsilon} - \mathsf{P}_{\mathsf{atm}}) \\ & \therefore \mathsf{T} = 245 \mathsf{kN} \end{split}$$

2. Describe the differences between Brayton Cycle and a Real Gas Turbine Cycle. Make diagrams to explain the losses associated with a real engine.



The first losses occur in the compressor from stage 1->2 due to the flow not being reversible. The combustion process is not completely isobaric. The process in the turbine is also not reversible.

3. Draw the T-s diagram and determine the turbine shaft power, and the air-fuel ratio

**Solution:** With all the given infomation the only equations used to find the shaft power were:

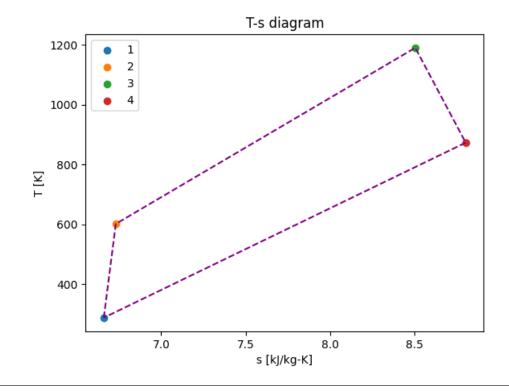
$$\begin{split} P_{02} &= r P_{01} \\ T_{02} &= T_{01} \left[ 1 + \frac{1}{\eta_c} \left( r^{\frac{\gamma - 1}{\gamma}} - 1 \right) \right] \\ W_* &= \frac{1}{\eta_*} c p_* (T_i - T_j) \\ P &= \dot{m}_{gas} (W_T - W_C) \end{split}$$

$$P_{\text{shaft}} = 5740.3282 \text{ [kW]}$$

The entropy starting values were pulled from Dr.Cizmas' textbook from the air tables for  $s_1$  and stoichiometric tables for  $s_3$ . The following equation was used to find the other two values:

$$s_i - s_j = c p_* \ln \frac{T_i}{T_j} - R \ln \frac{P_i}{P_j}$$

which then produces the following T-s graph.



The ideal fuel to air ratio was pulled from the 5th set of lecture slides:

$$\Delta t_c = 588.4864 \rightarrow f \approx 0.014$$

$$\therefore f^{-1} \approx 71.429$$