

Particle in a Box
Schrodinger Equation and Solution

Recapitulate

Under separation of variables, wave function can be written as a product

$$\Psi(x, t) = \psi(x)\phi(t) \qquad \phi(t) = e^{-iEt/\hbar}$$

$$\Psi(x, t) = \psi(x)e^{-iEt/\hbar}$$

Now find E from time independent Schrödinger Equation

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x)\psi(x) = E\psi(x)$$

Time independent Schrodinger equation (TISE)

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x)\psi(x) = E\psi(x)$$

Important points to note:

Given $V(x)$, solve TISE to obtain E and $\psi(x)$

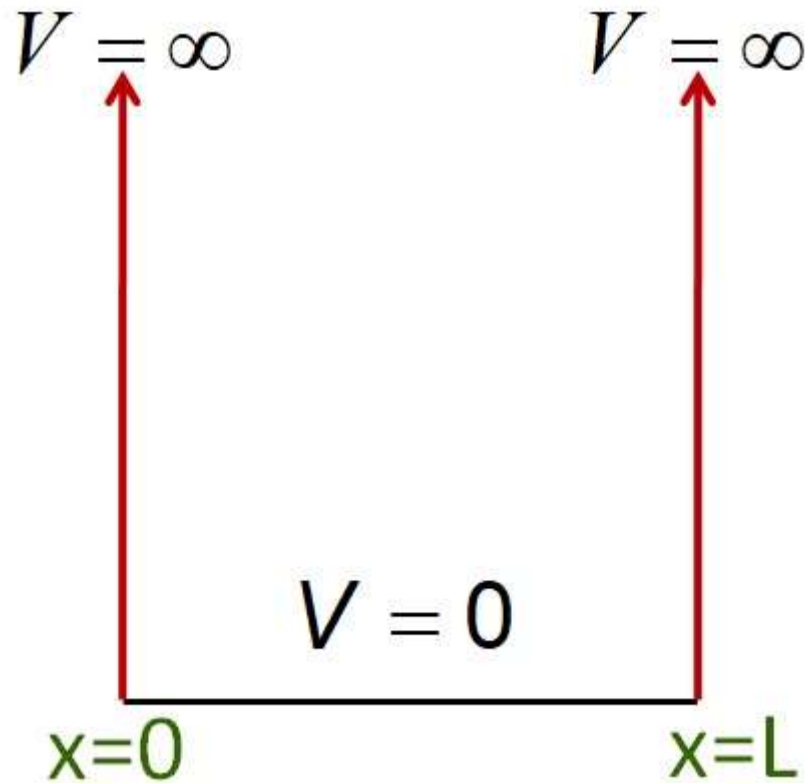
Solution is subjected to the 'boundary' conditions.

Acceptable solution $[\psi(x)]$ must be continuous, single valued, and its derivative must be continuous.

*Ideas on '**How to proceed to find the acceptable solution**' were discussed while working with 'Free Particle'.*

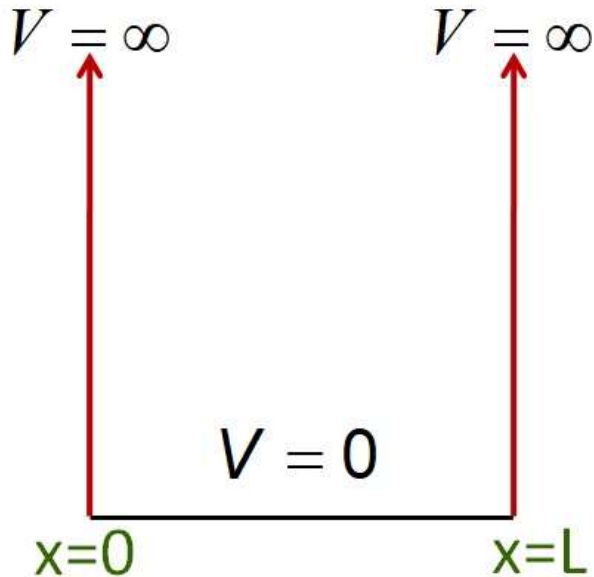
Particle in a Box

$$V(x) = 0 \quad \text{for} \quad 0 \leq x \leq L$$
$$= \infty \quad \text{for} \quad x < 0 \quad \text{or} \quad x > L$$



*Particle of mass **m** is placed in the potential*

Particle is free to move inside the box; however at the boundary, it experiences a strong force.



Time independent Schrodinger equation

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} + V(x)\psi(x) = E\psi(x)$$

$$\frac{\partial^2 \psi(x)}{\partial x^2} = -\frac{2m(E - V(x))}{\hbar^2} \psi(x)$$

for $x < 0$ and $x > L$, $V(x) = \infty \Rightarrow \psi(x) = 0$

Particle can not exists for $x < 0$ and $x > L$

for $0 \leq x \leq L$, $V(x) = 0 \Rightarrow \frac{\partial^2 \psi(x)}{\partial x^2} = -\frac{2mE}{\hbar^2} \psi(x)$

$$\frac{\partial^2 \psi(x)}{\partial x^2} = -\frac{2mE}{\hbar^2} \psi(x) \quad \text{for } 0 \leq x \leq L,$$

The general solution is

$$\psi(x) = A \sin(kx) + B \cos(kx)$$

$$k^2 = \frac{2mE}{\hbar^2} \quad \text{Since } E \geq 0, \quad k \text{ is real}$$

Note: We could choose a general solution of the form

$$\psi(x) = Ae^{ikx} + Be^{-ikx}$$

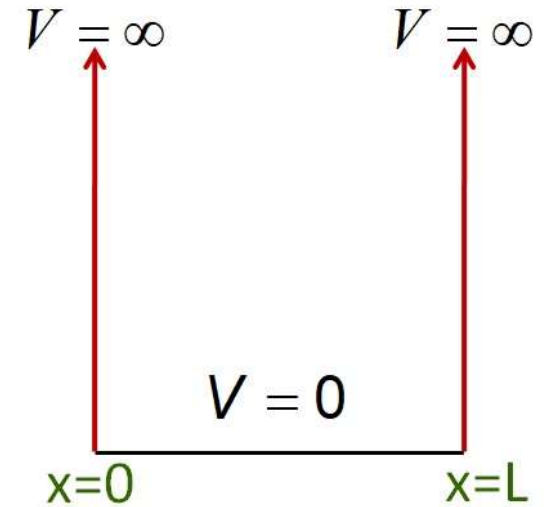
This choice will give same results since $\sin(kx)$ and $\cos(kx)$ are each superpositions of $\exp(\pm kx)$. Algebra will be a little more involved.

$$\psi(x) = A \sin(kx) + B \cos(kx)$$

Boundary Conditions

The wave function must be continuous

$$\psi(x) = 0 \quad \text{for} \quad x < 0 \quad \text{and} \quad x > L,$$



$$\Rightarrow \psi(x=0) = \psi(x=L) = 0$$

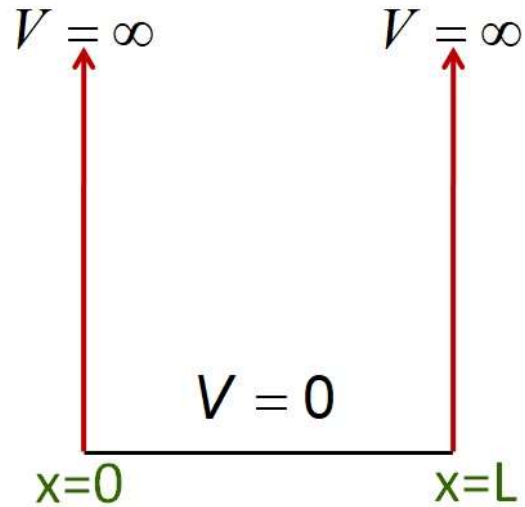
$$\psi(0) = 0 \Rightarrow A \sin(0) + B \cos(0) = 0 \Rightarrow B = 0$$

$$\psi(L) = 0 \Rightarrow A \sin(kL) = 0 \Rightarrow kL = n\pi$$

$$\therefore \frac{\sqrt{2mE}}{\hbar} L = n\pi \Rightarrow E = \frac{n^2 \pi^2 \hbar^2}{2mL^2} \quad n \geq 1$$

Energy is quantized!

What we obtain is



$$\psi(x) = A \sin(kx) \quad k = \frac{n\pi}{L}$$

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2} \quad n \geq 1$$

Normalization of Wave function

$$\int_{-\infty}^{\infty} \psi^*(x) \psi(x) dx = |A|^2 \int_0^L \sin^2(kx) dx = 1$$

$$\therefore |A|^2 \int_0^L \sin^2\left(\frac{n\pi}{L}x\right) dx = 1$$

$$|A|^2 \int_0^L \sin^2\left(\frac{n\pi}{L}x\right) dx = 1$$

$$\text{Use } \sin^2\theta = (1 - \cos^2\theta)/2$$

$$\therefore \frac{|A|^2}{2} \int_0^L \left[1 - \cos\left(\frac{2n\pi}{L}x\right)\right] dx = 1$$

$$\therefore \frac{|A|^2}{2} \left[x - \frac{L}{2n\pi} \sin\left(\frac{2n\pi x}{L}\right) \right]_0^L = 1 \quad \therefore \frac{|A|^2}{2} L = 1$$

$$A = e^{i\theta} \sqrt{\frac{2}{L}} \quad \text{Let us take real } A, \text{ i.e., } A = \sqrt{\frac{2}{L}}$$

$$\psi(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right) \quad \text{for } 0 \leq x \leq L$$

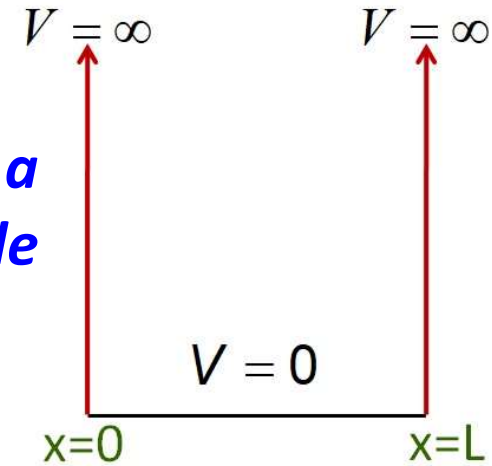
$$= 0 \quad \text{elsewhere}$$

Now let us start with the wave function

$$\psi(x) = Ae^{ikx} + Be^{-ikx}$$

The particle is a free particle inside the box!

To solve



$$\frac{\partial^2 \psi(x)}{\partial x^2} = -k^2 \psi(x) \quad \text{for } 0 \leq x \leq L, \quad k^2 = \frac{2mE}{\hbar^2}$$

$$\psi(x) = 0 \quad \text{for } x < 0 \text{ and } x > L,$$

$$\psi(0) = 0 \Rightarrow A + B = 0 \Rightarrow A = -B \Rightarrow \psi(x) = 2iA \sin(kx)$$

$$\psi(L) = 0 \Rightarrow 2iA \sin(kL) = 0 \Rightarrow kL = n\pi$$

$$\therefore \frac{\sqrt{2mE}}{\hbar} L = n\pi$$

$$\therefore E = \frac{n^2 \pi^2 \hbar^2}{2mL}$$

Normalization of wave function $\psi(x) = 2iA \sin(kx)$

$$\int_{-\infty}^{\infty} \psi^*(x) \psi(x) dx = 4|A|^2 \int_0^L \sin^2(kx) dx = 1$$

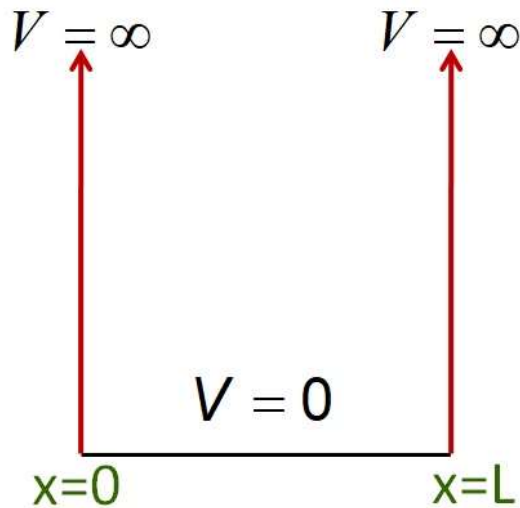
$$\therefore 2|A|^2 L = 1 \qquad A = \frac{e^{i\theta}}{\sqrt{2L}}$$

$$\psi(x) = \frac{2ie^{i\theta} \sin(kx)}{\sqrt{2L}} = \sqrt{\frac{2}{L}} e^{i(\theta+\pi/2)} \sin(kx)$$

Now choose phases to make wave function real

$$\psi(x) = \sqrt{\frac{2}{L}} \sin(kx) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L} x\right) \qquad \textbf{Answer is same.}$$

You can choose the phase since what matters is $\psi^(x) \psi(x)$*



Summary of Results

Since the energy levels are quantized ($n=1,2,3..$), we denote energy (E) and wave function [$\psi(x)$] by subscript 'n'

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right) \quad \text{for } 0 \leq x \leq L \quad E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$
$$= 0 \quad \text{elsewhere}$$

We may also write

$$H\psi_n(x) = E_n\psi_n(x)$$

E_n is the eigenvalue corresponding to wave function $\psi_n(x)$

How do these wave functions look like?

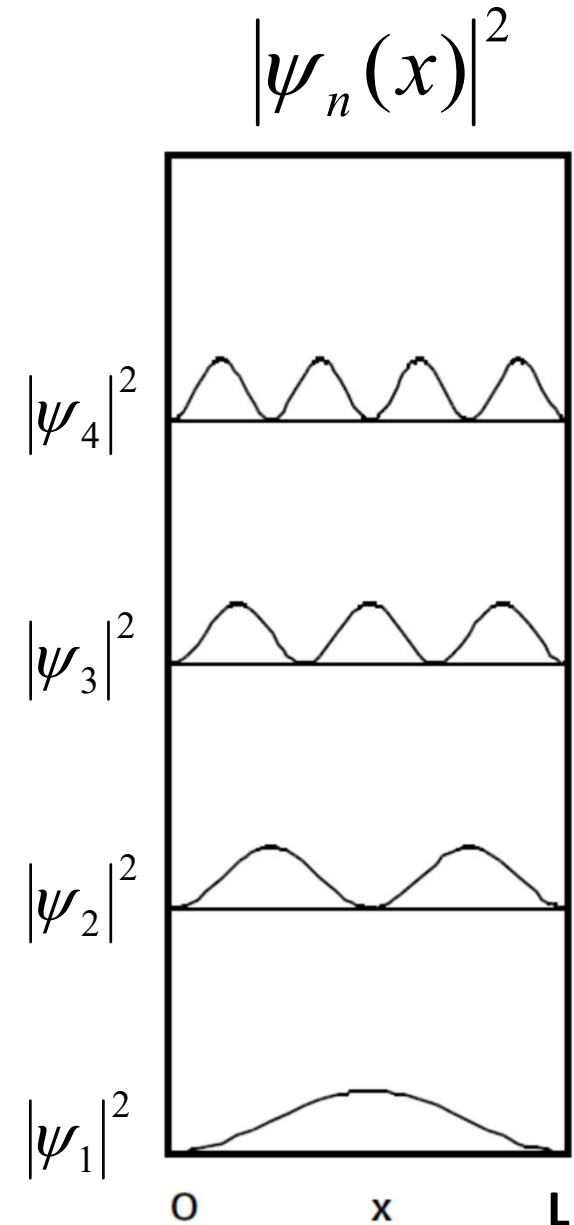
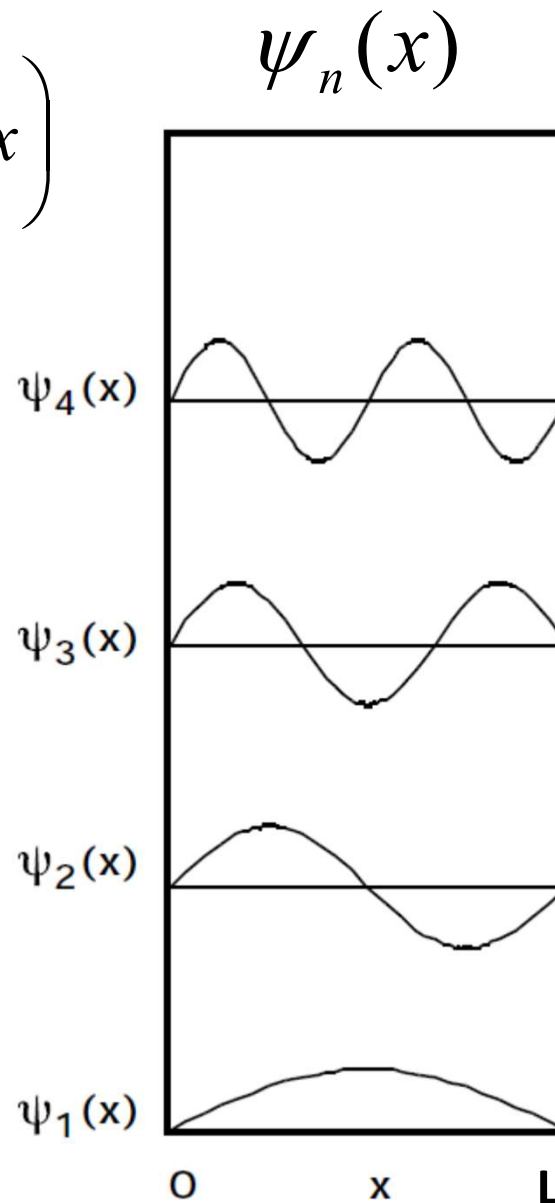
$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right)$$

$$\psi_4(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{4\pi}{L}x\right)$$

$$\psi_3(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{3\pi}{L}x\right)$$

$$\psi_2(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{2\pi}{L}x\right)$$

$$\psi_1(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{\pi}{L}x\right)$$



How are the energy levels organized?

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2} \quad n \geq 1$$

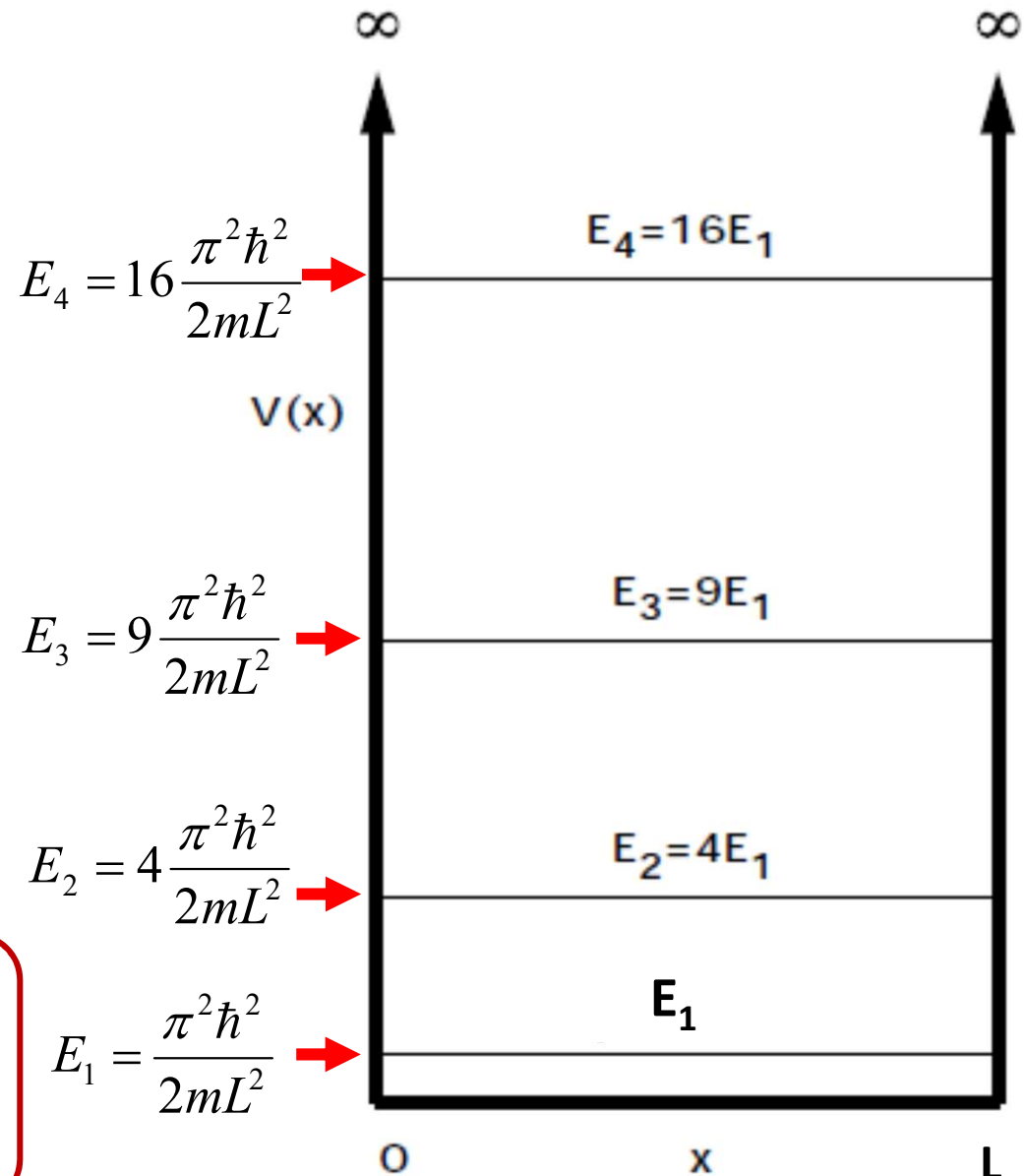
- Occurrence of quantized energy levels.

- Lowest energy

$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2} > 0$$

Classically, $E_1 = 0$

Quantum particle possesses “Zero point energy”



Energy levels of a particle in a box can be obtained using de Broglie equation

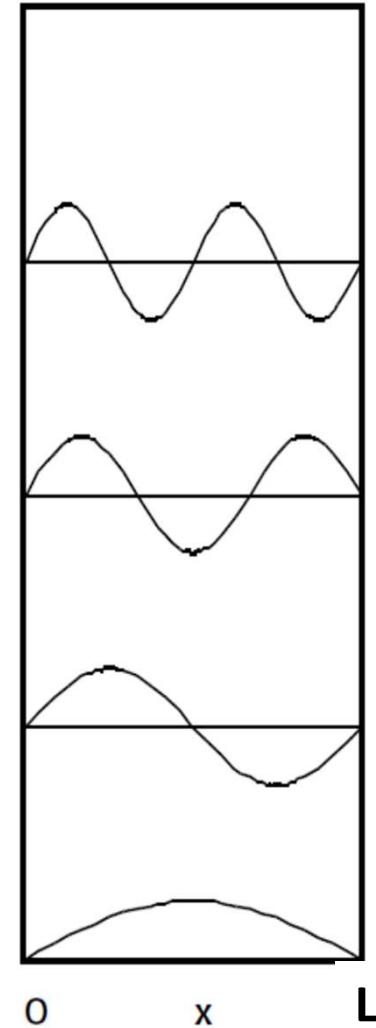
$$\lambda_{dB} = \frac{h}{p}$$

Amplitude of oscillation at both ends must be zero. Therefore, integer number of half waves must fit into the box

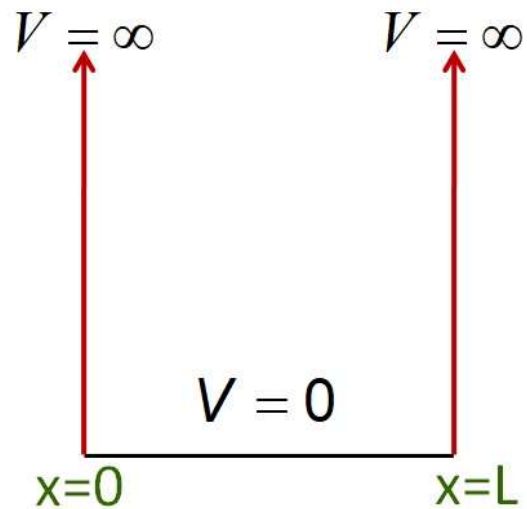
$$n \frac{\lambda_{dB}}{2} = L \quad \Rightarrow \quad n \frac{h}{2p} = L \quad \Rightarrow \quad p = \frac{nh}{2L}$$

Since $V=0$, all energy is kinetic energy

$$E = \frac{p^2}{2m} \quad \Rightarrow \quad E = \frac{n^2 h^2}{8mL^2} = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$



‘Zero point energy’ is a consequence of ‘Uncertainty Relation’



Particle is restricted within the box of dimension L

$$\Delta x = L$$

Momentum uncertainty

$$\Delta p = \frac{\hbar}{2\Delta x} = \frac{\hbar}{2L}$$

For the minimum energy state,

$$p = \Delta p = \frac{\hbar}{2L}$$

$$E_{\text{minimum}} = \frac{p^2}{2m} = \frac{\hbar^2}{8mL^2} > 0$$

Numerical Example:

An electron confined in a box of dimension 0.5 nm. Find lowest energy level and the energy difference between the second and first level.

$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2} = \left(\frac{\hbar^2}{2m_e} \right) \left(\frac{\pi}{5 \times 10^{10}} \right)^2 = 2.4 \times 10^{-19} J \sim 1.5 eV$$

Separation between second and first energy levels

$$E_2 = 4 \frac{\pi^2 \hbar^2}{2mL^2}$$

$$E_2 - E_1 = 3 \frac{\pi^2 \hbar^2}{2mL^2} = 3E_1 \sim 4.5 eV$$

Orthogonality

Two functions $\psi_1(x)$ and $\psi_2(x)$ are said to be **orthogonal** if

$$\int_{-\infty}^{\infty} \psi_1^*(x) \psi_2(x) dx = 0$$

For a particle in box

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right) \quad \text{and energy is} \quad E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

Let us test the orthogonality of $\psi_l(x)$ and $\psi_n(x)$

We need to find out

$$\int_{-\infty}^{\infty} \psi_l^*(x) \psi_n(x) dx = \int_0^L \psi_l^*(x) \psi_n(x) dx$$

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right)$$

$$\int_0^L \psi_l^*(x) \psi_n(x) dx = \frac{2}{L} \int_0^L \sin\left(\frac{l\pi}{L}x\right) \sin\left(\frac{n\pi}{L}x\right) dx$$

$$\text{Use } \sin A \sin B = [\cos(A - B) - \cos(A + B)] / 2$$

$$= \frac{1}{L} \int_0^L \left[\cos \frac{(l-n)\pi}{L}x - \cos \frac{(l+n)\pi}{L}x \right] dx = \delta_{ln}$$

Kronecker-Delta function

$$\delta_{ln} = \begin{cases} 1 & \text{for } l = n \\ 0 & \text{for } l \neq n \end{cases}$$

***Eigen functions belonging to different eigenvalues
are orthogonal***

Time dependent wave functions

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right)$$

$$\Psi(x,t) = \psi(x)e^{-iE/\hbar}$$

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

$$\Psi_n(x,t) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right) e^{-iE_n/\hbar}$$

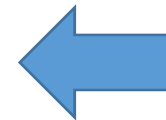
$$\Psi_n^*(x,t) \Psi_n(x,t) = \frac{2}{L} \sin^2\left(\frac{n\pi}{L}x\right)$$

*Probability of
observing the
particle at x*



$\Psi_n(x,t)$ *is a Stationary state*

***Independent
of time***



Particle in a box: Quantum Particle vs Classical Particle

- For a quantum particle the energy levels are quantized.

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

A classical particle can assume any value of E

- For a quantum particle the lowest energy is

$$E_1 = \frac{\pi^2 \hbar^2}{2mL^2} > 0$$

Quantum particle possesses “Zero point energy”

For a classical particle the lowest energy = 0

- **Probability of finding a particle in an interval $[L/4, 3L/4]$**

$$P_n[L/4, 3L/4] = \int_{L/4}^{3L/4} \psi_n^*(x) \psi(x) dx = \frac{2}{L} \int_{L/4}^{3L/4} \sin^2\left(\frac{n\pi x}{L}\right) dx$$

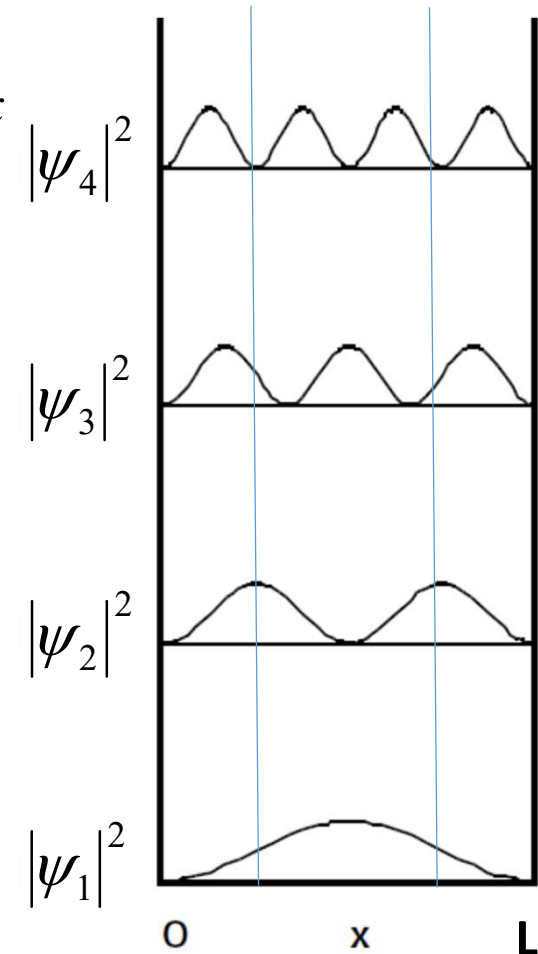
$$= \frac{1}{2} - \frac{1}{2n\pi} \left[\sin\left(\frac{3}{2}n\pi\right) - \sin\left(\frac{1}{2}n\pi\right) \right]$$

$$P_n[L/4, 3L/4] = \frac{1}{2} + \frac{1}{\pi} \quad \text{for } n = 1$$

$$P_n[L/4, 3L/4] = \frac{1}{2} \quad \text{for } n = \text{even}$$

$$P_n[L/4, 3L/4] = \text{Oscillating around } 1/2 \quad \text{for } n = \text{odd}$$

$$P_n[L/4, 3L/4] = \frac{1}{2} \quad \text{as } n \rightarrow \infty$$



For a classical particle, the probability of finding it at any x is $1/L$

$$\therefore P_{\text{class}}[L/4, 3L/4] = 1/2$$

For large n , quantum probability tends to classical probability

- **Momentum uncertainty**

$$\begin{aligned}\langle p_x \rangle &= \int_0^L \psi_n^*(x) \left(-i\hbar \frac{\partial}{\partial x} \right) \psi(x) dx = -\frac{2i\hbar}{L} \int_0^L \frac{n\pi}{L} \sin\left(\frac{n\pi x}{L}\right) \cos\left(\frac{n\pi x}{L}\right) dx \\ &= 0\end{aligned}$$

$$\begin{aligned}\langle p_x^2 \rangle &= \int_0^L \psi_n^*(x) \left(-\hbar^2 \frac{\partial^2}{\partial x^2} \right) \psi(x) dx = \frac{2\hbar^2}{L} \left(\frac{n\pi}{L} \right)^2 \int_0^L \sin^2\left(\frac{n\pi x}{L}\right) dx \\ &= \left(\frac{n\hbar\pi}{L} \right)^2\end{aligned}$$

$$\Delta p_x = \sqrt{\langle p^2 \rangle - \langle p \rangle^2} = \frac{n\hbar\pi}{L}$$

- **Position uncertainty**

$$\langle x \rangle = \int_0^L \psi_n^*(x) x \psi(x) dx = \frac{2}{L} \int_0^L x \sin^2\left(\frac{n\pi x}{L}\right) dx = \frac{L}{2}$$

$$\langle x^2 \rangle = \int_0^L \psi_n^*(x) x^2 \psi(x) dx = \frac{2}{L} \int_0^L x^2 \sin^2\left(\frac{n\pi x}{L}\right) dx = \frac{L^3}{3} \left[1 - \frac{3}{2n^2\pi^2} \right]$$

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2} = \sqrt{\frac{L^2}{12n^2\pi^2} (n^2\pi^2 - 6)}$$

- **Uncertainty relation**

$$\Delta x = \sqrt{\frac{L^2}{12n^2\pi^2}(n^2\pi^2 - 6)}$$

$$\Delta p_x = \frac{n\hbar\pi}{L}$$

$$\Delta x \Delta p_x = \hbar \sqrt{\frac{(n^2\pi^2 - 6)}{12}}$$

$$= 0.57\hbar \quad \text{for } n = 1$$

$$= 1.67\hbar \quad \text{for } n = 2$$

- **Symmetry of wave functions**

$$\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi}{L}x\right)$$

Odd n $\psi_1, \psi_3, \psi_5, \dots$ $\psi(x) = \psi(-x)$

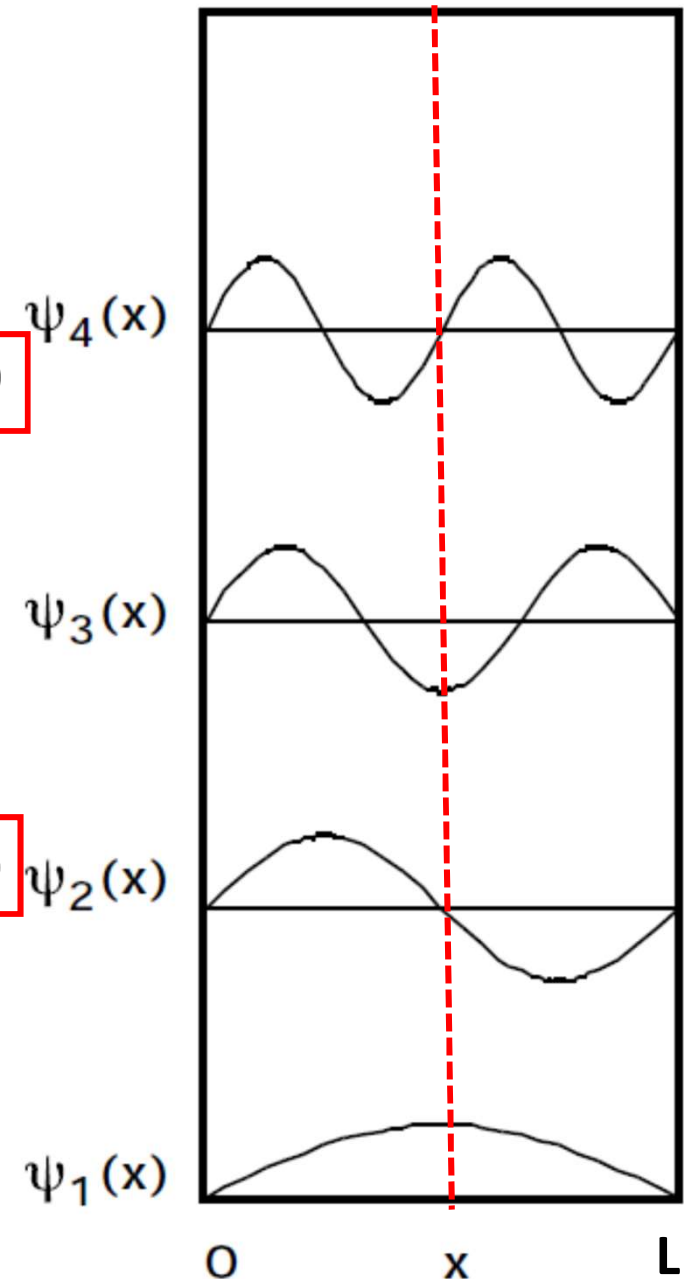
$\psi(x)$ on the right half is exactly mirror image of the $\psi(x)$ in the left half.

Such wave functions are called '**Even Parity**' wave functions

Even n $\psi_2, \psi_4, \psi_6, \dots$ $\psi(x) = -\psi(-x)$

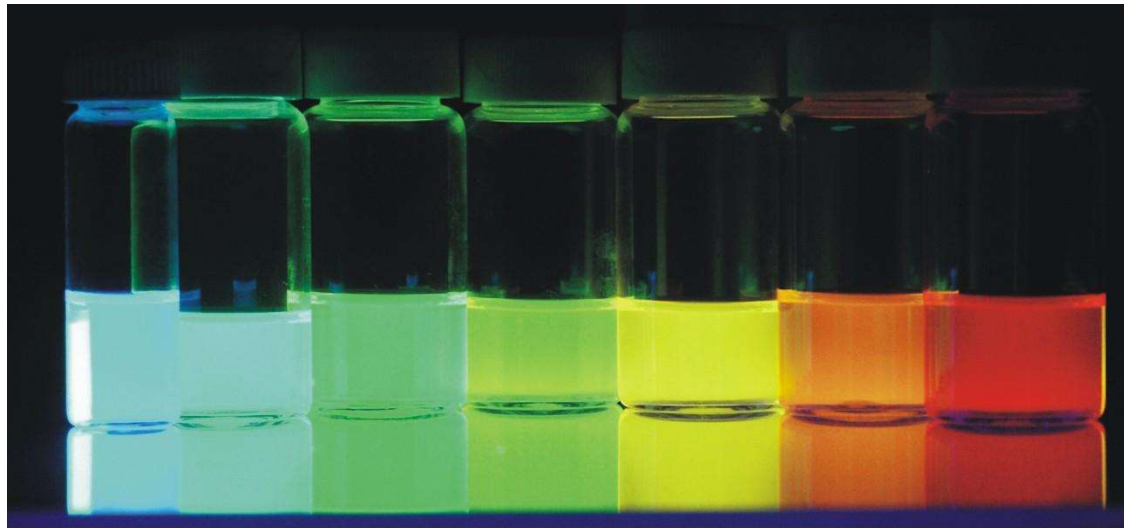
$\psi(x)$ on the right half is negative of the mirror image of the $\psi(x)$ in the left half.

Such wave functions are called '**odd Parity**' wave functions



Colours from Quantum Dots

A nanoscale semiconductor arrangement is called a quantum dot. They exhibit **size dependent colours**.



2.3 \longrightarrow 5.5
Size (nanometers)

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2}$$

L = size of the
quantum dot

$$E_2 - E_1 = \frac{3\pi^2 \hbar^2}{2m} \frac{1}{L^2}$$

*Size increases, emission shifts
to red side.*