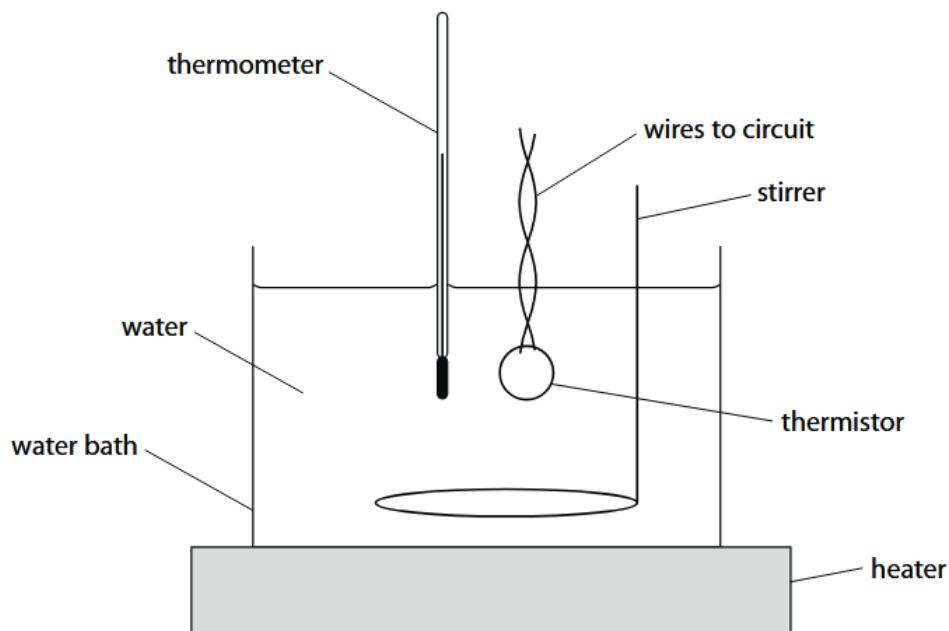


**8** A student investigates how the resistance of a thermistor varies with temperature.

- (a) The student uses the equipment shown in Figure 18 to measure the temperature of the thermistor.



**Figure 18**

- (i) Give **one** reason for using a water bath.

(1)

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- (ii) The equipment shown in Figure 18 is for investigations in the temperature range from 20 °C to 100 °C.

State **one** way the student could develop this experimental procedure to investigate temperatures outside this range.

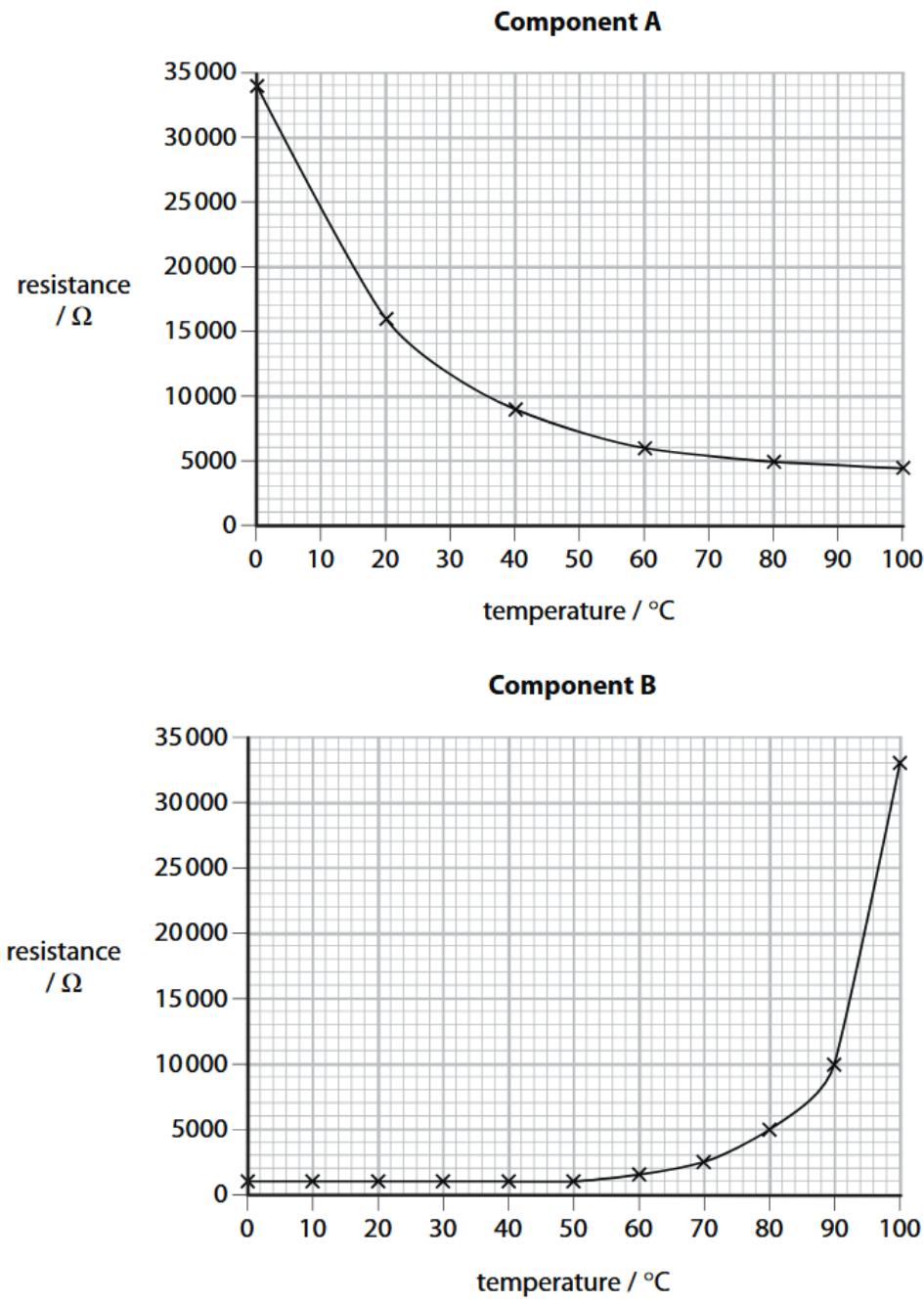
(1)

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(b) The student takes measurements for two other components, A and B.

The results for both these components are shown in Figure 19.



**Figure 19**

Compare and contrast how the resistances of component A and component B vary with temperature.

(3)

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\*(c) Describe how the student should carry out an experiment to determine the specific heat capacity of water.

(6)

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**(Total for Question 8 = 11 marks)**

<b>Question number</b>	<b>Answer</b>	<b>Additional guidance</b>	<b>Mark</b>
<b>8(a)(i)</b>	<p>Any one reason from:</p> <ul style="list-style-type: none"> <li>• the thermistor and the water are at the same temperature (1)</li> <li>• large volume of water gives a steady temperature rise (1)</li> </ul>	<p>accept idea that only small part of thermometer would be in contact with a thermistor in air</p> <p>accept difficult to control change in temperature of thermistor when heated in air</p>	<b>(1)</b>

<b>Question number</b>	<b>Answer</b>	<b>Additional guidance</b>	<b>Mark</b>
<b>8(a)(ii)</b>	<p>Any one of the following developments to the procedure:</p> <ul style="list-style-type: none"> <li>• add ice to increase lower limit of temperature range (1)</li> <li>• use liquid with higher boiling point to increase upper limit of temperature range (1)</li> </ul>	<p>accept named liquid with higher boiling point, e.g. oil</p>	<b>(1)</b>

Question number	Answer	Additional guidance	Mark
<b>8(b)</b>	<p>A comparison and contrast that must include at least <b>one</b> similarity and <b>one</b> difference from the following points to a maximum of three marks:</p> <p><b>Similarities</b></p> <ul style="list-style-type: none"> <li>• resistance of both changes with temperature (1)</li> <li>• both graphs show a non-linear relationship (1)</li> <li>• data comparison, e.g. both have the same resistance at 80 °C (1)</li> </ul> <p><b>Differences</b></p> <ul style="list-style-type: none"> <li>• resistance of A decreases with temperature but resistance of B increases with temperature (1)</li> <li>• for A, (largest slope/rate of change) is at lower temperature but for B, (largest slope/rate of change) is at higher temperature(s) (1)</li> <li>• for B, resistance is constant below 50 °C but for A resistance is roughly constant above 60 °C (1)</li> </ul>	accept (smallest slope/rate of change) for A is at higher temperature but (smallest slope/rate of change) for B is at lower temperature	(3)

Question number	Indicative content	Mark
<b>*8(c)</b>	<p>Answers will be credited according to candidate's deployment of knowledge and understanding of the material in relation to the qualities and skills outlined in the generic mark scheme.</p> <p>The indicative content below is not prescriptive and candidates are not required to include all the material which is indicated as relevant. Additional content included in the response must be scientific and relevant.</p> <p style="text-align: center;"><b>AO1 (6 marks)</b></p> <ul style="list-style-type: none"> <li>• Use of top pan balance to measure mass</li> <li>• Insulate beaker to reduce heat loss</li> <li>• Ammeter connected in series with heater</li> <li>• Voltmeter connected in parallel with heater</li> <li>• Use of <math>E = I \times V \times t</math> to determine energy supplied to the water</li> <li>• Accept use of joule-meter to measure energy supplied</li> <li>• Use of <math>\Delta E = m \times c \times \Delta\theta</math> to determine the specific heat capacity of the water</li> <li>• Measure p.d. across heater</li> <li>• Use stopwatch to measure time liquid is heating</li> <li>• Measure current in heater</li> <li>• Determine mass of water as mass of (beaker and water) – mass of beaker</li> <li>• Measure temperature before and after heating</li> </ul>	(6)

<b>Level</b>	<b>Mark</b>	<b>Descriptor</b>
	0	No awardable content.
Level 1	1–2	<p>Demonstrates elements of physics understanding, some of which is inaccurate. Understanding of scientific, enquiry, techniques and procedures lacks detail. (AO1)</p> <p>Presents a description which is not logically ordered and with significant gaps. (AO1)</p>
Level 2	3–4	<p>Demonstrates physics understanding, which is mostly relevant but may include some inaccuracies. Understanding of scientific ideas, enquiry, techniques and procedures is not fully detailed and/or developed. (AO1)</p> <p>Presents a description of the procedure that has a structure which is mostly clear, coherent and logical with minor steps missing. (AO1)</p>
Level 3	5–6	<p>Demonstrates accurate and relevant physics understanding throughout. Understanding of the scientific ideas, enquiry, techniques and procedures is detailed and fully developed. (AO1)</p> <p>Presents a description that has a well-developed structure which is clear, coherent and logical. (AO1)</p>