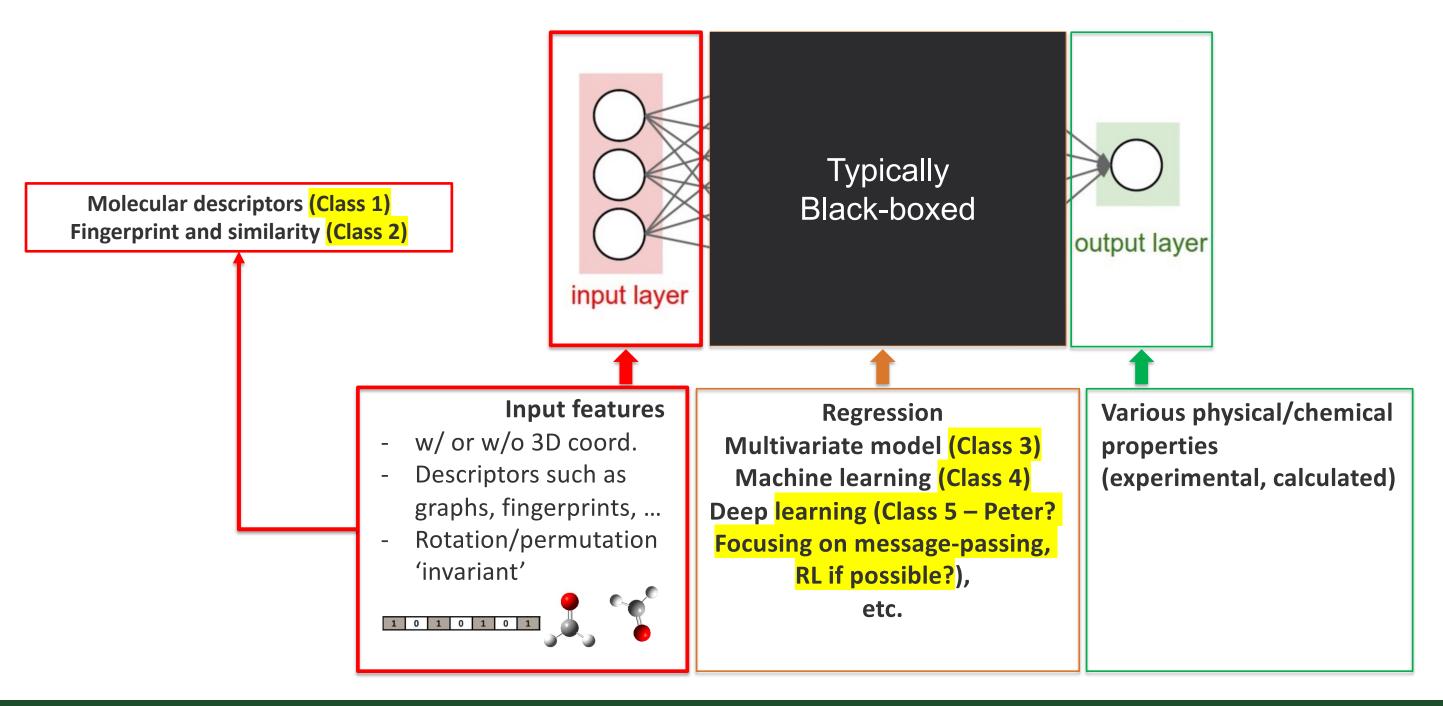
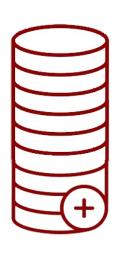
Class 0

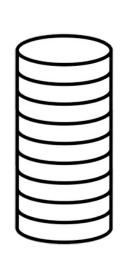
Conda, rdkit, numpy, pandas, matplotlib, seaborn ... Jupyter notebook installation instructions

Overview – Machine Learning in Chemistry



Overview – Machine Learning in Chemistry







- Data points > 10^4 , 10^5
 - → Minimal information is given in the input layer (SMILES, atom, bond feature, connectivity, etc.)
 - → Minimal human intervention
 - → Deep learning, so that the computer automatically recognizes and learns some 'patterns' that are consistent (sometimes not consistent) with our chemical intuition
- Data points < 10³
 - → Molecular descriptors can be generated (from QM calculations, Python packages, etc.) and chosen based on our intuition
 - → Target property = f(chosen molecular descriptors/features)

^{**} May contain oversimplification, the perspective in this slide is not always right.

Class 1 – Molecular Descriptors

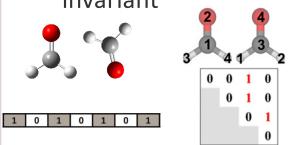


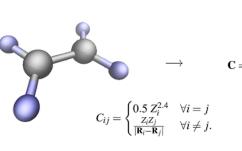
Features/Descriptors As Input

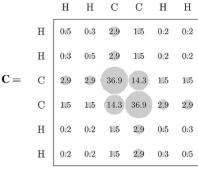
Input features using 3D coordinates (Continuous)

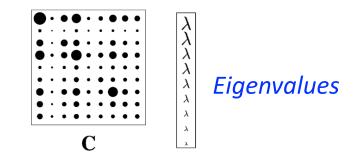


- w/ or w/o 3D coord.
- Descriptors such as graphs, fingerprints, ...
- Rotation/permutation 'invariant'





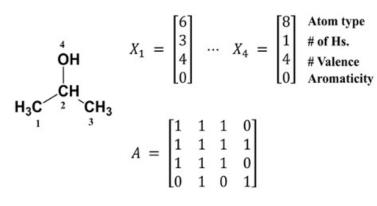


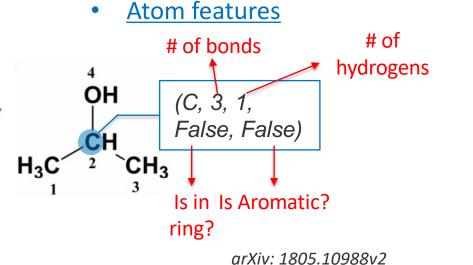


$$C_{ij} = \begin{cases} 0.5 Z_i^{2.4} & \forall i = j \\ \frac{Z_i Z_j}{|\mathbf{R}_i - \mathbf{R}_j|} & \forall i \neq j. \end{cases}$$

J. Chem. Theory Comput., 2013, 9, 3404.

Input features w/o 3D coordinates (Discrete)





Topological Descriptors

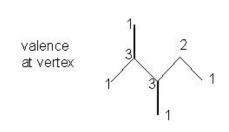
Molecular shape → Real number

Randić indices

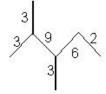
$$\chi_0(G) = (\text{Sum of Deg.'s for vertices } i)^{-1}$$

$$\chi_1(G) = \sum_{edges \ i-j} (\text{Deg}(i)\text{Deg}(j))^{-1/2}$$

$${}^h \chi(G) = \sum_{\text{paths of length } h} (\text{Deg}(i)\text{Deg}(j)...\text{Deg}(h+1))^{-1/h}$$



bond values as product of above



edge term as reciprocal of square rooot of above bond values

Sum of edge terms

3.179

Topological polar surface area

• Bertz complexity index

Balaban's J index

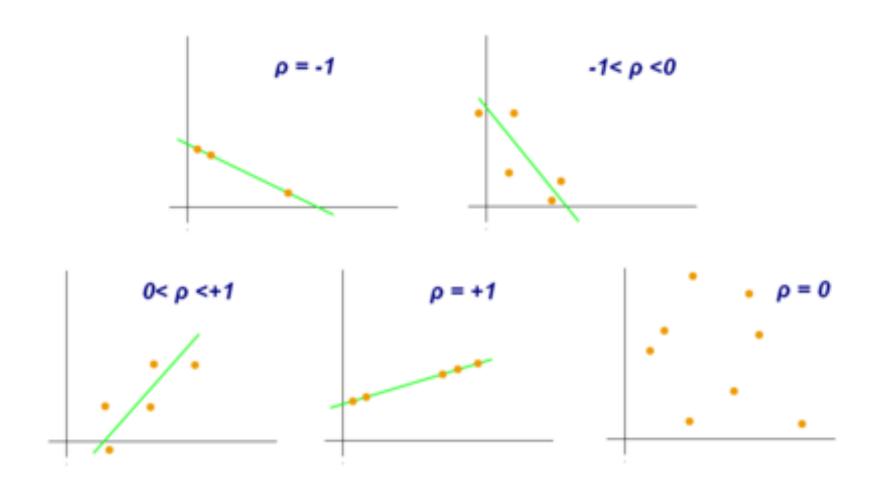
$$J = \frac{m}{\gamma + 1} \sum_{(i,j) \in E(G)} (D_i D_j)^{-1/2},$$
$$(\gamma = m - n + 1,$$
$$m: \# \text{ of bonds, } n: \# \text{ of atoms})$$

An index defined to quantify complexity, extent of branching of a molecule

https://www.rdkit.org/docs/source/rdkit.Chem.GraphDescriptors.html https://www.rdkit.org/docs/source/rdkit.Chem.rdMolDescriptors.html And refs therein

Pearson Correlation Coefficient

np.corrcoef



Jupyter Notebook Demo – Find the correlation between Topological descriptors vs. Boiling point

Class 2 – Fingerprint and Similarity



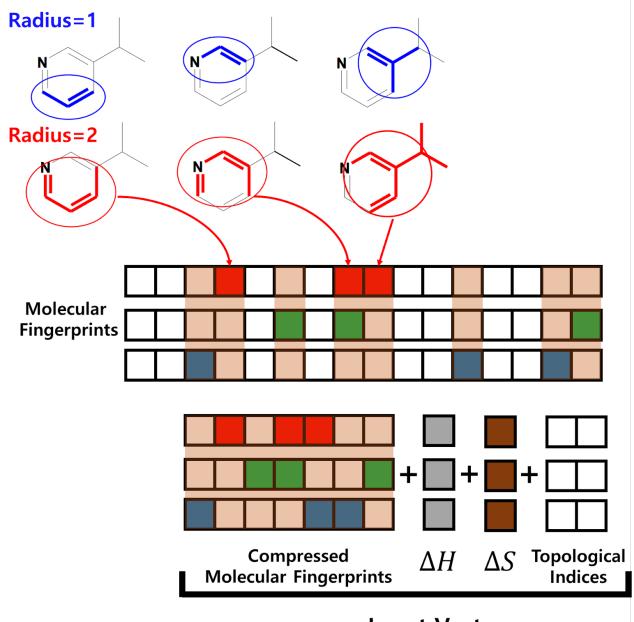
Why Molecular Fingerprints?

• How can we make a (complete) vector representing structural features of a molecule, without using Cartesian coordinates?

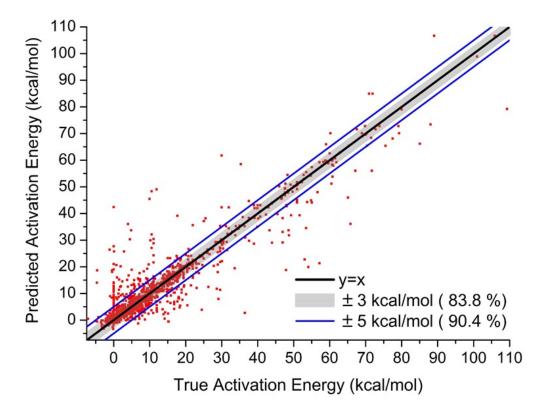
How can we find the molecules sharing the common substructure(s)?

How can we evaluate similarity between two molecules?

Morgan Fingerprint



Input Vector



MAE: 1.95 kcal/mol, RMSE: 4.49kcal/mol R²: 0.89

Reaction data are obtained from Reaction Mechanism Generator (RMG) Database 12,704 gas-phase reactions related to combustion (80% for training, 20% for test set)

Chem. Eur. J. 2018, 24, 12354-12358.

Class 3 – Multivariate Analysis



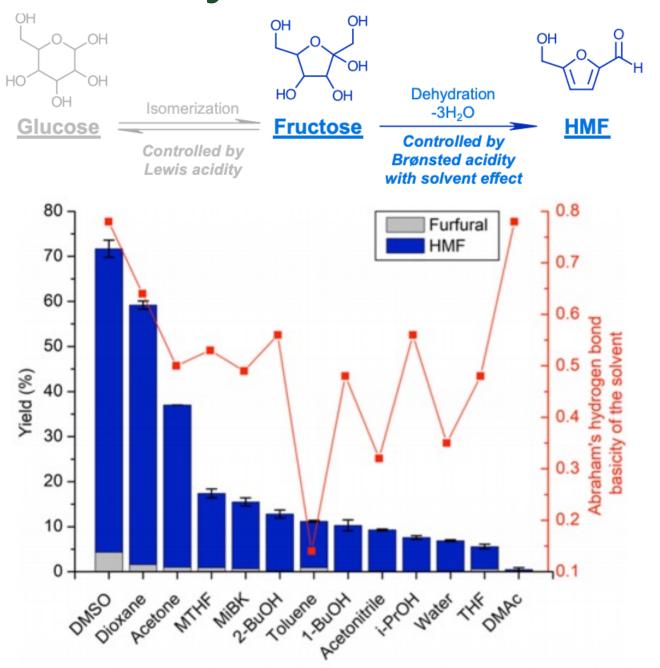
Multivariate Analysis - Motivation

• How can we choose appropriate descriptors with minimizing human bias (intervention)?

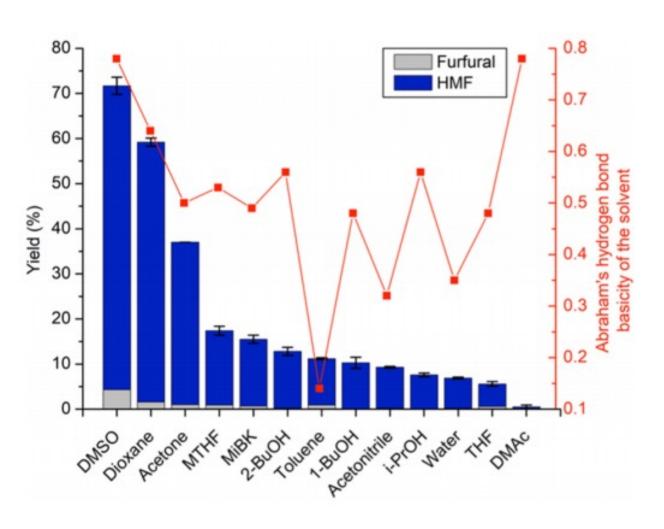
How can we overcome a limited number of data points (e.g. < 100?)

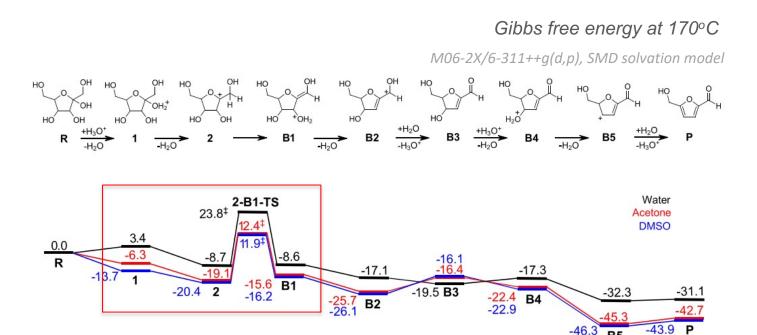
How can we gain chemical insights from the model?
 (Although it could be ad hoc and subject to change with more data points)

Multivariate Analysis - Motivation

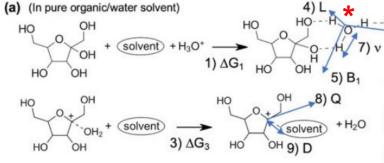


Multivariate Analysis - Motivation

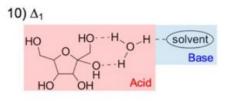


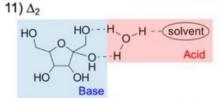


Multivariate Analysis



- - Complexation free energy of fructose, hydronium ion and one solvent molecule (at 170°C)
 - 2) Free energy of protonation and dehydration (at 170°C)
 - Free energy of dehydration and coordination of one solvent molecule (at 170°C)
 - 4~6) Sterimol parameters
 - 7) O-H vibration frequency
 - 8) NPA charge of the carbocation center
 - 9) Bond distance between the carbocation center and the solvent





- 10~11) Difference of chemical hardness between the acid and the base parts $(\Delta = |\eta_{acid} \eta_{base}|)$
- 12) $\Delta G_{\text{stab,HMF}}$: Stabilization energy of HMF by one explicit solvent from the most stable structure (at 170°C)
- 13) Δ_{LUMO} : Change in LUMO energy when an explicit solvent is added

(b) (In 2:1 organic:water co-solvent, at 170°C)

(c)

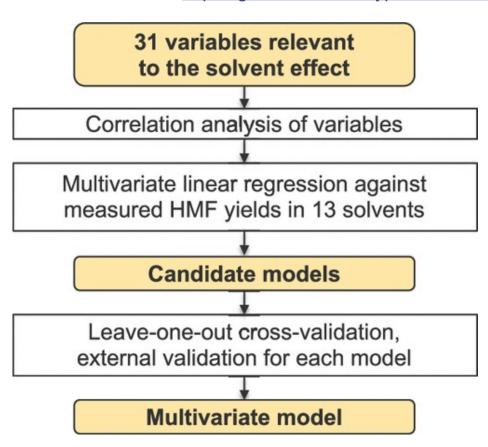
- · Calculated values
- 18~19) Calculated dipole moment (μ), polarizability (α_{pol})
- Calculated atomic charge of the electronegative center (Q⁻_{sol})
- · Experimental values
- 21~22) Abraham's H-bond acidity (α) and basicity (β) at 25°C
- 23) Dielectric constant (E) at 25°C
- 24) Refractive index (ε_{inf}) at 20°C
- 25) Dimroth-Reichardt parameter (E_T) at 25°C
- 26~27) Gutmann's acceptor (AN) and donor number (DN) at 25°C
- 28) Z-value at 25°C
- 29~31) Kamlet-Taft parameters at 25°C
- (H-bond donor (a), acceptor ability (b), polarizability (π^*)





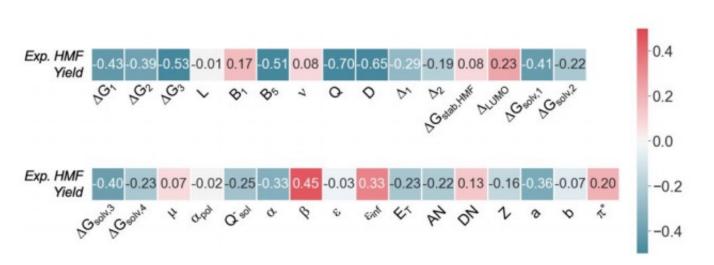


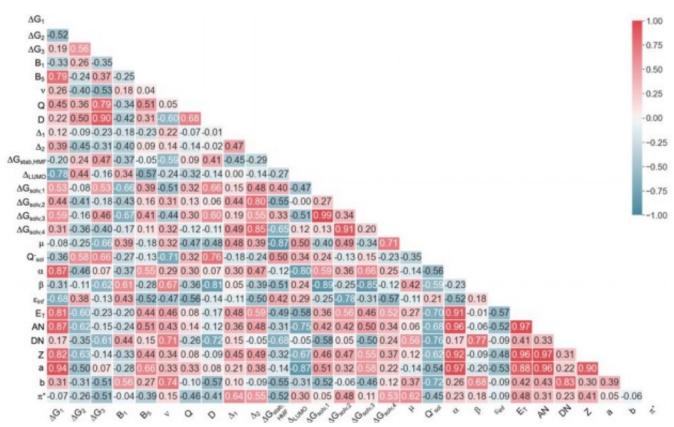
https://github.com/bobbypaton/Sterimol https://github.com/bobbypaton/wSterimol https://github.com/bobbypaton/DBSTEP



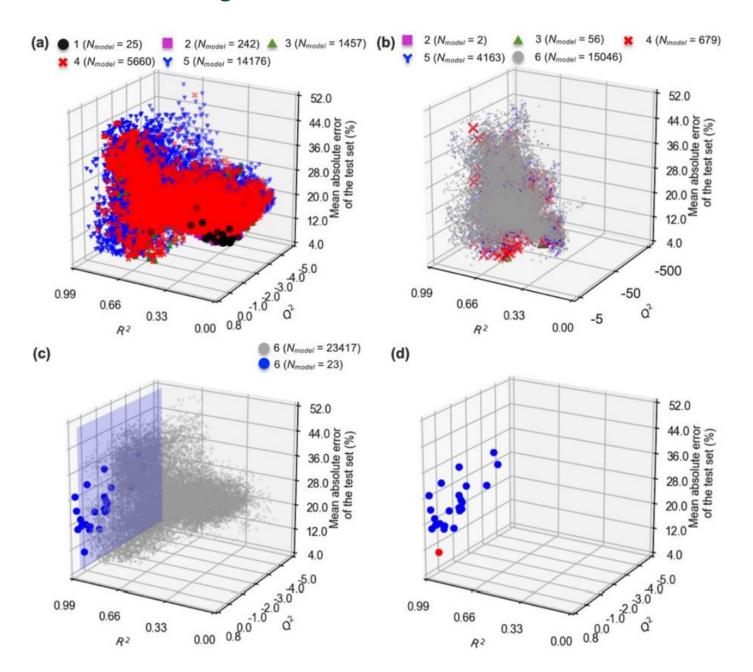
Training set: 13
Test set: 4 solvents

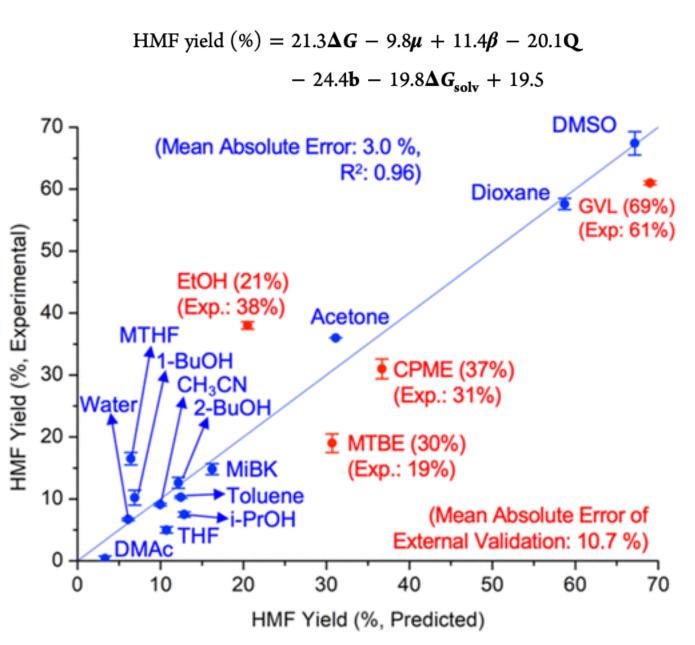
Correlation Analysis





Analysis of Candidate Models

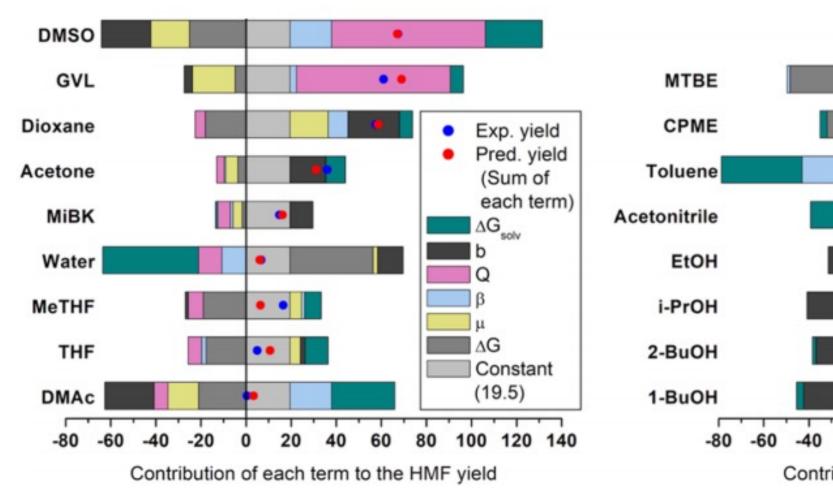


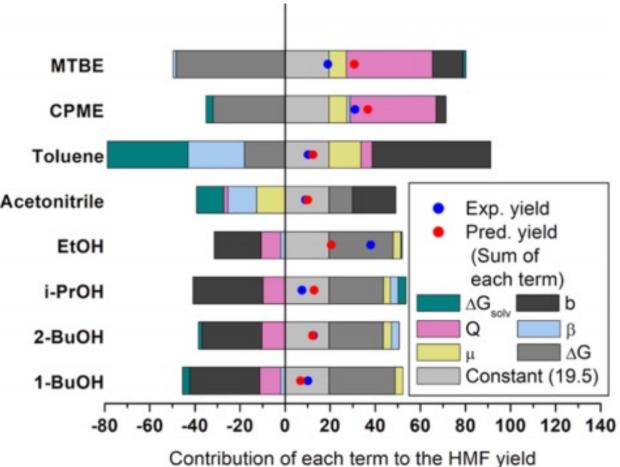


Chemical Explanation of the Model

HMF yield (%) =
$$21.3\Delta G - 9.8\mu + 11.4\beta - 20.1\mathbf{Q}$$

- $24.4\mathbf{b} - 19.8\Delta G_{\text{solv}} + 19.5$





Assignment

Apply multivariate analysis to your ongoing research

(OR)

• Develop more efficient workflow than the code I showed you today (e.g., minimizing iterations, using more scikit-learn functions, etc.)

(OR)

- Try another linear regression example:
 - Pick G16 output files of one reactant and one transition state
 - Obtain k at different temperatures using GoodVibes
 - Obtain A, Ea through linear regression

$$k_{rxn} = \frac{k_B T}{h} \exp\left(-\frac{\Delta G^{\ddagger}}{RT}\right) = A \exp\left(-\frac{E_a}{RT}\right) \qquad \text{In k = In A - Ea/RT}$$
(From G16)

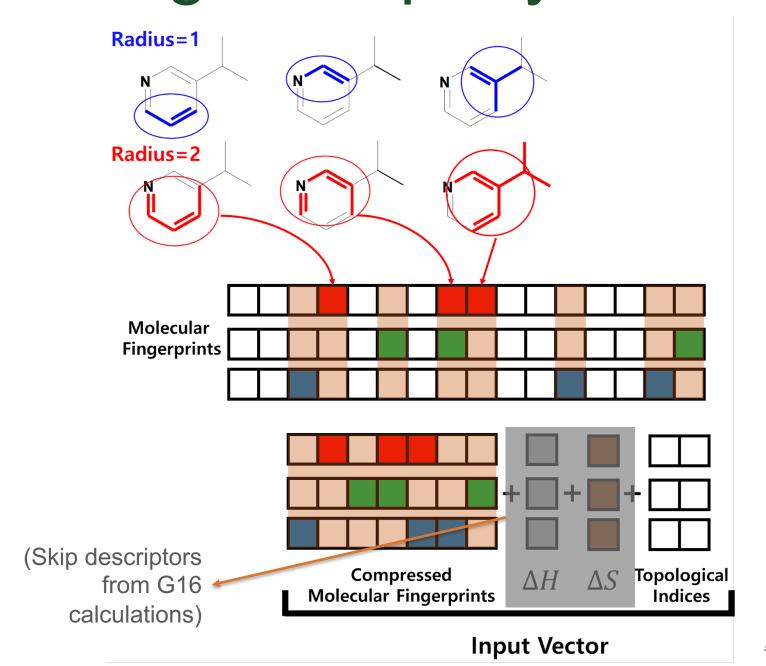
If it is too easy, try fitting to modified Arrhenius equation with non-linear regression

$$k = AT^n e^{-E_a/(RT)}$$
. https://docs.scipy.org/doc/scipy/reference/generated/scipy.optimize.curve_fit.html

Class 4 – Machine Learning



Target Property and ML Models



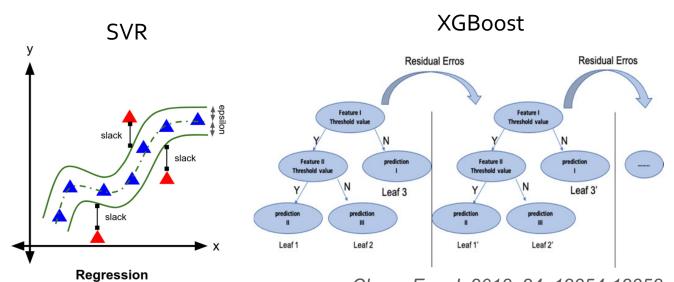
Predicting activation energy of gas phase reactions

Reaction data are obtained from Reaction Mechanism Generator (RMG) Database

2,386 gas-phase reactions related to combustion (80% for

training, 20% for test set)

ANN (multilayer perceptron)

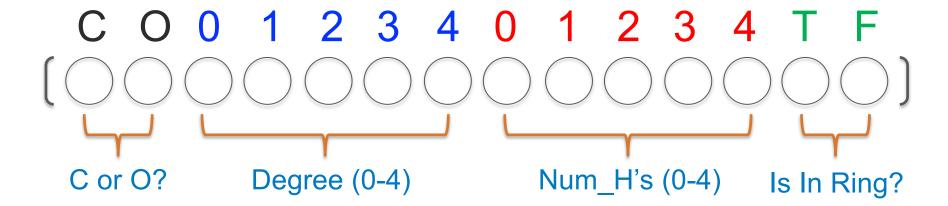


Chem. Eur. J. 2018, 24, 12354-12358. https://medium.com/it-paragon/support-vector-machine-regression-cf65348b6345 https://doi.org/10.1016/j.asej.2020.11.011

Miscellaneous



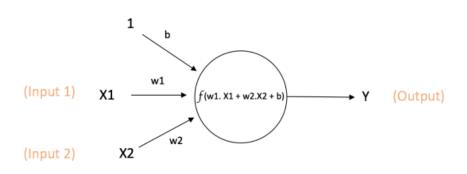
One-hot Vector - Example



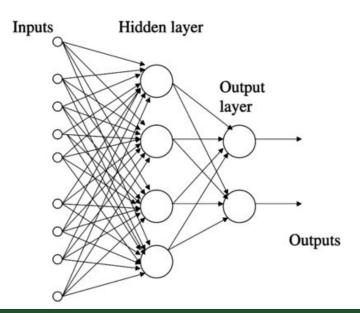
e.g.) Carbon, Degree 4, 3 hydrogens, not in the ring (1,0, 0,0,0,1, 0,0,0,1,0, 0,1)

A Brief Glimpse of Neural Nets

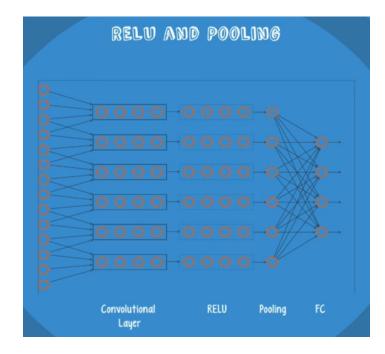
Artificial neural network

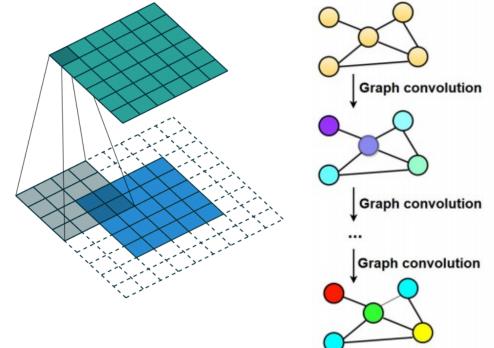


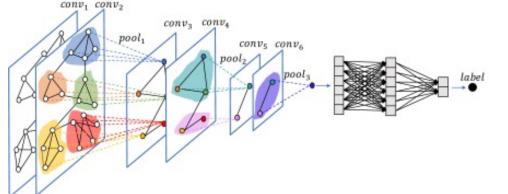
Output of neuron = Y = f(w1. X1 + w2. X2 + b)



Convolutional neural network







- Node atoms, edges bonds
- Each node and edge has its feature vector iteratively trained to predict the desired molecular properties

https://medium.com/@jayeshbahire/perceptron-and-backpropagation-970d752f4e44 https://www.groundai.com/project/graph-convolutional-networks-with-eigenpooling/1 arXiv: 1805.10988

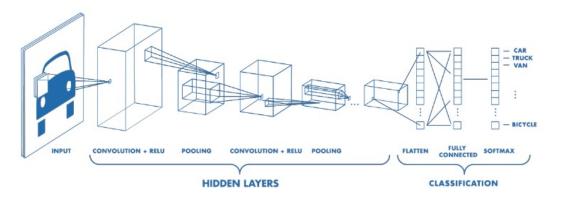
CNN and **GNN**

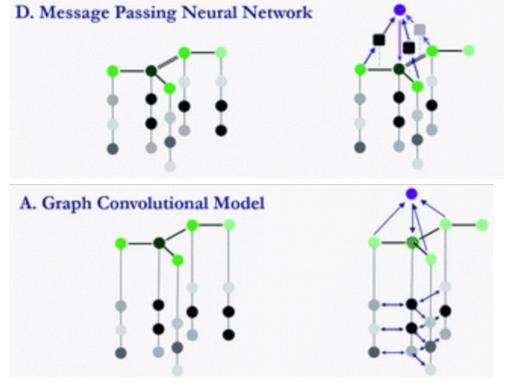
- Various types of neural networks
 - Convolutional neural network (CNN)

 Message-passing neural network (MPNN)

- Graph convolutional network (GCN)
- Graph attention network (GAT)

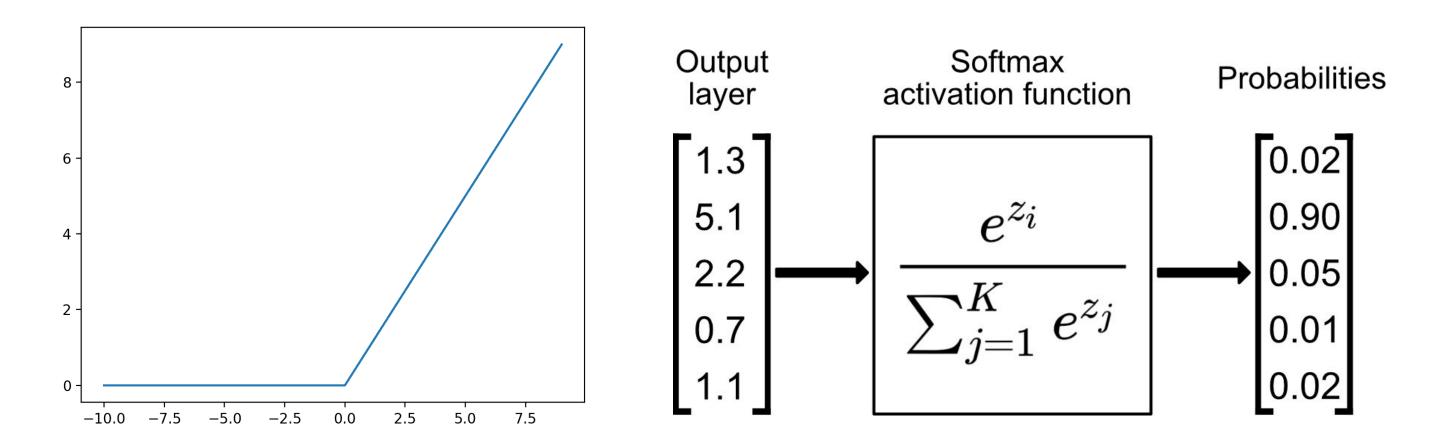
And other techniques...





Chem. Sci., 2018, 9, 513-530.

Activation Functions



- (1) To introduce non-linearity
- (2) To resolve vanishing gradient problem
- (3) For a specific purpose (Regressor->classifier, target values are all positive, etc.)