

- RNMC: kinetic Monte Carlo implementations for
- 2 complex reaction networks
- Laura Zichi 10 1,2*, Daniel Barter 10 3*, Eric Sivonxay 10 3*, Evan Walter Clark
- Spotte-Smith 6, Rohith Srinivaas Mohanakrishnan^{1,4}, Kristin Aslaug
- **5** Persson^{4,5¶}, and Samuel M. Blau^{3¶}
- 1 Materials Science Division, Lawrence Berkeley National Laboratory, Berkeley, CA, USA 94720 2
- 7 Department of Physics, University of Michigan Ann Arbor, Ann Arbor, MI, USA 48109 3 Energy
- 8 Storage and Distributed Resources, Lawrence Berkeley National Laboratory, Berkeley, CA USA 94720 4
- 9 Department of Materials Science and Engineering, University of California Berkeley, CA, USA 94720 5
- 10 Molecular Foundry, Lawrence Berkeley National Laboratory, Berkeley, CA, USA 94720 ¶ Corresponding
- author * These authors contributed equally.

DOI: 10.xxxxx/draft

Software

- Review 🗗
- Repository 🖸
- Archive 🗗

Editor: Open Journals ♂ Reviewers:

@openjournals

Submitted: 01 January 1970 **Published:** unpublished

License

Authors of papers retain copyright and release the work under a 24 Creative Commons Attribution 4.0 International License (CC BY 4.0)

Summary

17

Macroscopic chemical and physical phenomena are driven by microscopic interactions at the atomic and molecular scales. In order to capture complex processes with high fidelity, simulation methods that bridge disparate time and length scales are needed. While techniques like molecular dynamics and *ab initio* simulations capture dynamics and reactivity at high resolution, they cannot be used beyond relatively small length (hundreds to thousands of atoms) and time scales (picoseconds to microseconds). Kinetic Monte Carlo (kMC) approaches overcome these limitations to bridge length and time scales across several orders of magnitude while retaining relevant microscopic resolution, making it a powerful and flexible tool.

Here, we present RNMC, an easy-to-use, modular, high-performance kMC simulation framework that enables modeling of complex systems. RNMC consists of a core module defining the common features of kMC algorithms, including an implementation of the Gillespie algorithm (Gillespie, 1977), input/output operations leveraging SQLite databases, threading logic for parallel execution, and dependency graphs for efficient event propensity updates. In addition, there are currently three modules defining kMC implementations for different types of applications. The GMC (Gillespie Monte Carlo) module enables simulations of reaction networks in a homogeneous (well-mixed) environment. GMC is a basic tool that is appropriate for general simulations of solution-phase chemistry. The NPMC (NanoParticle Monte Carlo) module enables simulation of dynamics in nanoparticles with 3D statistical field theory and supports one- and two-site interactions. Finally, the LGMC (Lattice Gillespie Monte Carlo) module is designed for simulations of multi-phase systems (especially at solid-fluid interfaces) where chemical and electrochemical reactions can occur between a lattice region and a homogeneous region. We have designed RNMC to be easily extensible, enabling users to add additional kMC modules for other diverse chemical and physical systems.

Statement of need

Three are many existing kMC implementations, including several open source examples (e.g. the Stochastic Parallel PARticle Kinetic Simulator or SPPARKS(Garcia Cardona et al., 2009) and kmos).(Hoffmann et al., 2014) RNMC began as a fork of SPPARKS but differs in several important ways. First, because RNMC uses the widely supported SQLite database engine for simulation inputs and outputs, it is facile to automate simulations using RNMC. Second, RNMC has a focus



- on modularity; it is designed such that users can quickly develop new types of kMC simulations using a common core library.
- 44 The simulation modules already implemented in RNMC provide unique capabilities that are not
- 45 widely available in other open source codes. NPMC is specially designed for 3D simulations of
- 46 interactions in nanocrystals. For instance, it can be used to simulate the upconversion of infrared
- to ultraviolet light via interactions between spatially distributed dopants in nanoparticles. (Gnach
- $_{48}$ & Bednarkiewicz, $_{2012}$) LGMC is also somewhat unique in that it can simulate multi-phase
- 49 systems and electrochemical processes. Simulations using LGMC can include a lattice region and
- a homogeneous solution region which can interact via interfacial reactions. Electrochemcial
- reactions can be treated using Marcus theory (Marcus, 1965) or Butler-Volmer kinetics. (Newman
- 52 & Balsara, 2021)
- We have already used the GMC module in a number of prior works in applications related to
- Li-ion and Mg-ion batteries. (Barter et al., 2023; Spotte-Smith et al., 2022, 2023) We note
- that these simulations included tens of millions of reactions, demonstrating that RNMC is able
- to scale to large and complex reaction networks.

Acknowledgements

- 58 This project was intellectually led by the Laboratory Directed Research and Development
- Program of Lawrence Berkeley National Laboratory under U.S. Department of Energy Contract
- No. DE-AC02-05CH11231. L.Z. was supported in part by the U.S. Department of Energy,
- office of Science, Office of Workforce Development for Teachers and Scientists (WDTS)
- under the Science Undergraduate Laboratory Internships Program (SULI). E.W.C.S.-S. was supported by the Kavli Energy NanoScience Institute Philomathia Graduate Student Fellowship.
- Additional support came from the Joint Center for Energy Storage Research (JCESR), an
- Energy Innovation Hub funded by the U.S. Department of Energy, Office of Science, Basic
- Energy Sciences. This code was developed and tested using computational resources provided
- by the National Energy Research Scientific Computing Center (NERSC), a U.S. Department of
- Energy Office of Science User Facility under Contract No. DE-AC02-05CH11231, the Eagle and
- 59 Swift HPC systems at the National Renewable Energy Laboratory (NREL), and the Lawrencium
- HPC cluster at Lawrence Berkeley National Laboratory.

References

- Barter, D., Spotte-Smith, E. W. C., Redkar, N. S., Khanwale, A., Dwaraknath, S., Persson,
 K. A., & Blau, S. M. (2023). Predictive stochastic analysis of massive filter-based electrochemical reaction networks. *Digital Discovery*, 2(1), 123–137.
- Garcia Cardona, C., Wagner, G. J., Tikare, V., Holm, E. A., Plimpton, S. J., Thompson, A. P., Slepoy, A., Zhou, X. W., Battaile, C. C., & Chandross, M. E. (2009). *Crossing the mesoscale no-mans land via parallel kinetic monte carlo*. Sandia National Laboratories (SNL), Albuquerque, NM,; Livermore, CA.
- Gillespie, D. T. (1977). Exact stochastic simulation of coupled chemical reactions. *The Journal of Physical Chemistry*, *81*(25), 2340–2361.
- Gnach, A., & Bednarkiewicz, A. (2012). Lanthanide-doped up-converting nanoparticles: Merits and challenges. *Nano Today*, 7(6), 532–563.
- Hoffmann, M. J., Matera, S., & Reuter, K. (2014). Kmos: A lattice kinetic monte carlo framework. *Computer Physics Communications*, *185*(7), 2138–2150.
- Marcus, R. A. (1965). On the theory of electron-transfer reactions. VI. Unified treatment for homogeneous and electrode reactions. *The Journal of Chemical Physics*, *43*(2), 679–701.



- Newman, J., & Balsara, N. P. (2021). Electrochemical systems. John Wiley & Sons.
- Spotte-Smith, E. W. C., Blau, S. M., Barter, D., Leon, N. J., Hahn, N. T., Redkar, N. S.,
 - Zavadil, K. R., Liao, C., & Persson, K. A. (2023). Chemical reaction networks explain
- gas evolution mechanisms in mg-ion batteries. Journal of the American Chemical Society,
- 91 *145*(22), 12181–12192.
- Spotte-Smith, E. W. C., Kam, R. L., Barter, D., Xie, X., Hou, T., Dwaraknath, S., Blau, S.
- M., & Persson, K. A. (2022). Toward a mechanistic model of solid–electrolyte interphase
- formation and evolution in lithium-ion batteries. ACS Energy Letters, 7(4), 1446–1453.

