

DOUBLE ELIMINATION 1

TOSS-UP

1) Longtermism - *Short Answer* Which 6th century pandemic killed approximately 6 million people and contributed to a loss of territory by empires?

ANSWER: The Plague of Justinian [jus-TIH-nee-in]

BONUS

1) Longtermism - *Multiple Choice* The plague of Justinian was caused by *Yersinia pestis* [yur-SIH-nee-uh PEH-stis], which, despite its rare status in the modern day, still has not been eradicated. Why is it not considered a candidate for eradication?

- W) It is spread by animal vectors
- X) There is no vaccine against it
- Y) It can remain on surfaces for long periods of time
- Z) Outbreaks are too infrequent to target it for eradication

ANSWER: W) It is spread by animal vectors

TOSS-UP

2) Math - *Short Answer* $\frac{3^{2022}+3^{2020}}{3^{2022}-3^{2020}}$ [the fraction with numerator 3 to the 2022 plus three to the 2020 and denominator 3 to the 2022 minus 3 to the 2020] is equivalent to what real number?

ANSWER: $\frac{5}{4}$ (ACCEPT: 1.25)

BONUS

2) Math - *Multiple Choice* What is the eccentricity of the ellipse with equation $16x^2 + 25y^2 = 100$?

W) $\frac{3}{4}$

X) $\frac{3}{5}$

Y) $\frac{4}{5}$

Z) $\frac{5}{4}$

ANSWER: X) $\frac{3}{5}$

TOSS-UP

3) Chemistry - *Multiple Choice* Since molecular bromine readily adds across double bonds, which of the following bromine containing reagents is typically used to add bromine to the allyl [A-lil] position of an alkene?

- W) NBS
- X) AgBr
- Y) HOBr
- Z) CH₃MgBr

ANSWER: W) NBS

BONUS

3) Chemistry - *Multiple Choice* Which of the following best describes the character of the boron-boron bond in diboron according to molecular orbital theory?

- W) 1 sigma bond with bond order 1
- X) 2 sigma bonds with bond order $\frac{1}{2}$
- Y) 1 pi bond with bond order 1
- Z) 2 pi bonds with bond order $\frac{1}{2}$

ANSWER: Z) 2 pi bonds with bond order $\frac{1}{2}$

TOSS-UP

4) Earth and Space - *Multiple Choice* Which of the following ores would be least likely to produce acid mine drainage?

- W) Bornite
- X) Pyrite
- Y) Magnetite
- Z) Sphalerite [SFA-ler-ite]

ANSWER: Y) Magnetite

BONUS

4) Earth and Space - *Multiple Choice* Which of the following conditions would most likely create a mesocyclone?

- W) Strong wind shear
- X) Strong downdraft
- Y) Conditional instability
- Z) High-altitude convergence

ANSWER: W) Strong wind shear

TOSS-UP

5) Biology - *Short Answer* What is the name for the reduced mitochondria found in the Parabasalid [PAIR-uh-BAY-suh-lid] group?

ANSWER: Hydrogenosomes [hi-DRAW-jen-uh-zomes]

BONUS

5) Biology - *Short Answer* Identify all of the following three parts of the kidney where the osmolarity of the extracellular fluid increases: 1) Ascending limb of the loop of Henle [HEN-lee]; 2) Collecting duct; 3) Descending limb of the loop of Henle.

ANSWER: 2 and 3

TOSS-UP

6) Physics - *Short Answer* What type of magnetism that is least affected by changes in temperature is most common in ionic crystals and has a negative magnetic susceptibility?

ANSWER: Diamagnetism

BONUS

6) Physics - *Multiple Choice* Cookie dough is spread on a conveyor belt moving at velocity v to the right. A circular cookie cutter stamps out cake as it passes by underneath. Taking relativistic lengths into account, what shape does the stamped out cake make?

W) Circular

X) Squashed in the direction of of the belt's motion

Y) Stretched in the direction of the belt's motion

Z) It depends on the speed of the conveyor belt

ANSWER: Y) Stretched in the direction of the belt's motion

TOSS-UP

7) Longtermism - *Short Answer* Researcher and essayist Brian Tomasik [tuh-MAH-shik] has been influential in writing about a new type of existential risk, in which the concern is catastrophic amounts of future suffering rather than the end of civilization. What is this risk called?

ANSWER: S-risk

BONUS

7) Longtermism - *Multiple Choice* Which of the following existential risks is not considered an s-risk?

- W) Far-future harms to animals
- X) Risk from bioengineering of viruses
- Y) Misaligned artificial intelligence
- Z) Far-future value drift

ANSWER: X) Risk from bioengineering of viruses

TOSS-UP

8) Math - *Short Answer* Let x be a three digit number with three equal digits and let y be a three digit number with three distinct digits. What is the minimum possible difference between the two numbers?

ANSWER: 2

BONUS

8) Math - *Multiple Choice* According to Vieta's [vee-EH-duhs] formulas, the product of the roots r_1 , r_2 , and r_3 of the polynomial $P(x) = ax^3 + bx^2 + cx + d$ equals which of the following?

W) $-\frac{b}{a}$

X) $\frac{c}{a}$

Y) $-\frac{c}{a}$

Z) $-\frac{d}{a}$

ANSWER: Z) $-\frac{d}{a}$

TOSS-UP

9) Chemistry - *Multiple Choice* Which of the following molecules is IR active?

- W) Methane
- X) Diatomic oxygen
- Y) Diatomic nitrogen
- Z) Diatomic hydrogen

ANSWER: W) Methane

BONUS

9) Chemistry - *Multiple Choice* Which of the following gases has the highest b van der Waals [VAN der VALLS] parameter?

- W) Argon
- X) Chlorine
- Y) Carbon Dioxide
- Z) Ammonia

ANSWER: X) Chlorine

TOSS-UP

10) Earth and Space - *Short Answer* What cosmological theory of galaxy formation states that small variations in the primordial universe created so-called seeds that built up the large-scale structures we see today?

ANSWER: Bottom-up theory

BONUS

10) Earth and Space - *Short Answer* Sirius is a binary system consisting of two stars, one with twice the mass of the other. Measuring from the barycenter [BEAR-ih-center], what is the ratio of the distance from the larger star to the distance from the smaller star?

ANSWER: 1 to 2 (ACCEPT: $\frac{1}{2}$)

TOSS-UP

11) Biology - *Multiple Choice* Immunophenotyping [ih-MEW-no-FEE-nuh-typing] involves the characterization of the cell types present in a patient's immune system. Modern lab techniques allow exquisite specificity, sorting millions of cells into distinct populations. What technique is most commonly used in immunophenotyping studies?

W) MALDI-TOF [MAHL-dee toff] mass spectrometry

X) Flow cytometry [sigh-TAW-muh-tree]

Y) Immunoprecipitation

Z) Affinity chromatography

ANSWER: X) Flow cytometry

BONUS

11) Biology - *Short Answer* Many muscles are set up in antagonistic pairs, where one muscle contracts while the other relaxes. A motor neuron causes contraction of one muscle while an inhibitory interneuron prevents the motor neuron innervating the other muscle from causing contraction. Identify all of the following four neurotransmitters that the inhibitory interneuron could release: 1) GABA; 2) Glutamate [GLOO-duh-mate]; 3) Acetylcholine [uh-SEE-dil-KO-leen]; 4) Glycine [GLY-seen].

ANSWER: 1 and 4

TOSS-UP

12) Physics - *Short Answer* A cylinder has an outer radius B and an inner radius A. By what factor is the rate of heat transfer multiplied by if both the inner and outer radius are squared?

ANSWER: $\frac{1}{2}$

BONUS

12) Physics - *Multiple Choice* Water's index of refraction is 1.3, and the critical angle for the interface between medium A and water is 30 degrees. If light has a critical angle of 60 degrees, in an interface between medium A and B, which of the following is closest to the index of refraction of medium B?

W) 1.12

X) 1.75

Y) 2.25

Z) 2.6

ANSWER: Y) 2.25

TOSS-UP

13) Longtermism - *Multiple Choice* Which of the following diseases, contrary to its name, is not caused by a poxvirus?

- W) Monkeypox
- X) Smallpox
- Y) Cowpox
- Z) Chickenpox

ANSWER: Z) Chickenpox

BONUS

13) Longtermism - *Short Answer* Chickenpox is in the same family as which virus, which is considered a major cause of myalgic encephalomyelitis [my-AL-jik en-SEF-a-luh-my-EL-uh-dus]?

ANSWER: Epstein-Barr virus

TOSS-UP

14) Math - *Short Answer* How many arithmetic sequences of three terms exist such that all three terms are part of the set 1, 2, 3, 4, 5?

ANSWER: 13 (DO NOT ACCEPT: 8)

BONUS

14) Math - *Multiple Choice* What is the approximation for $f(1)$ using Euler's [OY-lers] method with two equal steps given the initial condition $f(0) = 1$ and the differential equation $\frac{dy}{dx} = 2y + 1$?

W) 2.5

X) 3.5

Y) 4.5

Z) 5.5

ANSWER: Z) 5.5

TOSS-UP

15) Chemistry - *Short Answer* What is the degree of unsaturation of maleic [muh-LAY-ic] anhydride?

ANSWER: 4

BONUS

15) Chemistry - *Multiple Choice* The bond between oxygen and hydrogen in water is said to be polar covalent since there is higher electron density on oxygen than hydrogen. Which of the following differences best explains this difference?

W) Valence electrons in hydrogen are in an $n = 1$ orbital, giving them a lower energy than those in oxygen

X) Valence electrons in hydrogen are in an $l = 0$ orbital, giving them lower energy than those in oxygen

Y) Valence electrons in oxygen are in an $n = 1$ orbital, giving them lower energy than those in hydrogen

Z) Valence electrons in oxygen are in an $l = 1$ orbital, so the lower effective nuclear charge gives them a lower energy than those in hydrogen

ANSWER: Z) Valence electrons in oxygen in an $l = 1$ orbital, so the lower effective nuclear charge gives them a lower energy than those in hydrogen

TOSS-UP

16) Earth and Space - *Short Answer* In which of Earth's four compositional layers do earthquake waves typically travel the fastest?

ANSWER: Mantle

BONUS

16) Earth and Space - *Multiple Choice* Pyritization [PIE-rih-dih-ZAY-shun] of fossils would most likely occur in which of the following depositional environments?

W) Oxygen-rich lacustrine [luh-KOOS-treen]

X) Anoxic lacustrine

Y) Oxygen-rich marine

Z) Anoxic marine

ANSWER: Z) Anoxic marine

TOSS-UP

17) Biology - *Short Answer* Histone modifications play an extremely important role in the regulation of gene expression. What amino acid, which can be both methylated and acetylated [uh-SEE-duh-lay-did], is most commonly modified in histones?

ANSWER: Lysine [LIE-seen]

BONUS

17) Biology - *Short Answer* Based on their listed functions, identify all of the following three compounds that would inhibit protein synthesis in bacteria but not in eukaryotes: 1) Tetracycline, inhibits binding of charged tRNAs to the 30S ribosomal subunit; 2) Chloramphenicol [KLOR-am-FEH-ni-kall], inhibits peptide bond formation by binding to the 70S ribosome; 3) Cycloheximide [SIGH-klo-HEX-ih-mide], blocks exit side of the 60S ribosomal subunit.

ANSWER: 1 and 2

TOSS-UP

18) Physics - *Multiple Choice* What type of spectroscopy uses radio waves to excite nuclei, which gives important information on a molecule's structure?

- W) X-ray
- X) NMR
- Y) Raman
- Z) Spark emission

ANSWER: X) NMR

BONUS

18) Physics - *Short Answer* An object with a mass of 2 grams and a charge of 1 coulomb is moving perpendicular to a magnetic field with a strength of 4×10^{-3} teslas. Due to centripetal acceleration, the object moves in a circular path with a radius of 25 centimeters. What is the object's velocity, in meters per second?

ANSWER: 0.5

TOSS-UP

19) Longtermism - *Short Answer* Which thought experiment suggests that in utilitarianism, the welfare of a being that experiences sufficiently massive amounts of happiness would be put above all other considerations?

ANSWER: Utility monster

BONUS

19) Longtermism - *Short Answer* Which analogous thought experiment suggests that a world containing some number of people living in utopian conditions would be worse than a world containing a sufficiently large number of people experiencing only 'Muzak and potatoes'?

ANSWER: Repugnant conclusion

TOSS-UP

20) Math - *Multiple Choice* Which of the following double-angle identities does not equal $\cos 2x$?

W) $\cos^2 x - \sin^2 x$

X) $2 \cos x \sin x$

Y) $2 \cos^2 x - 1$

Z) $1 - 2 \sin^2 x$

ANSWER: X) $2 \cos x \sin x$

BONUS

20) Math - *Short Answer* Compute $\int_{\frac{\pi}{4}}^{\frac{9\pi}{4}} \sin^2 x + \cos^2 x dx$ [the definite integral from pi over 4 to 9pi over 4 of sine squared of x plus cosine squared of x dx]

ANSWER: 2π

TOSS-UP

21) Chemistry - *Multiple Choice* Which of the following elements exhibits the weakest inert pair effect?

- W) Tin
- X) Lead
- Y) Gallium
- Z) Polonium

ANSWER: Y) Gallium

BONUS

21) Chemistry - *Short Answer* Identify all of the following four functional groups that are ortho-para directors in electrophilic aromatic substitutions: 1) Hydroxyl; 2) Trifluoromethyl [try-FLOOR-uh-METH-ull]; 3) Chloro; 4) Aldehyde.

ANSWER: 1 and 3

TOSS-UP

22) Earth and Space - *Multiple Choice* The solar tachocline [TACK-uh-kline] separates which of the following regions of the Sun?

- W) Chromosphere and photosphere
- X) Photosphere and convective zone
- Y) Convective zone and radiative zone
- Z) Radiative zone and core

ANSWER: Y) Convective zone and radiative zone

BONUS

22) Earth and Space - *Short Answer* Identify all of the following three star types that are typically Population I stars: 1) RR Lyrae [LIE-ray] variables; 2) B-type main sequence stars; 3) Classical Cepheid [SEH-fee-id] variables.

ANSWER: 2 and 3

TOSS-UP

23) Biology - *Multiple Choice* Which of the following statements is true about transcription and post-transcriptional processing?

- W) Addition of the 5 prime cap occurs after introns are removed
- X) Alternative splicing occurs after pre-mRNA is exported to the cytoplasm
- Y) The TATA [TAH-TAH] box is not found in bacteria
- Z) Silencers are typically found immediately upstream of the gene they regulate

ANSWER: Y) The TATA box is not found in bacteria

BONUS

23) Biology - *Short Answer* Identify all of the following four statements that are true regarding human chorionic gonadotropin [go-NAD-oh-TRO-fin], abbreviated as hCG: 1) hCG is produced by the trophoblast [TRO-fo-blast], which eventually forms the placenta; 2) hCG stimulates the production of progesterone [pro-JES-chur-own] by supporting the corpus luteum [LOO-dee-um]; 3) A primary method of testing for pregnancy is by measuring hCG in the blood or urine.

ANSWER: All

TOSS-UP

24) Physics - *Short Answer* An observer throws a clock in a circle at a constant speed of $0.96c$. To the observer, how long does it appear to take, in seconds, for one second to elapse on the clock?

ANSWER: $\frac{25}{7}$ sec

BONUS

24) Physics - *Short Answer* A system of gas molecules is in equilibrium with its surroundings such that the root mean square speed of the molecules was found to be $2\sqrt{6}$ meters per second. In meters per second, what speed are the majority of the gas molecules within the system moving with?

ANSWER: 4