

ESBOT

2023

RR5

Tossup

- 1) *Energy – Short Answer:* Scientists at Savannah River National Lab are doing experiments on fireflies, since they are very common in the area. Specifically, they want to understand how they synchronize their lights. What enzyme is known to be responsible for the production of light in their abdomens?

ANSWER: Luciferase

Bonus

- 1) *Energy – Short Answer:* Researchers at Pacific Northwest National Lab are studying ways to improve the efficiency of industrial chemical production. When modeling multi-step reactions, the assumption is made that the concentration of intermediates stays constant as the reaction proceeds. What is the name of this assumption?

ANSWER: Steady-state assumption

Tossup

- 2) *Biology – Short Answer:* In a G protein coupled receptor, what molecule activates the G protein to induce signal transduction?

ANSWER: GTP

Bonus

- 2) *Biology – Short Answer:* Identify all of the following 3 statements that are incorrect about the process of DNA replication:

- I) DNA pol III synthesizes the leading strand
- II) DNA pol I synthesizes Okazaki fragments on the lagging strand
- III) Primase makes DNA primers on the leading and lagging strands

ANSWER: II and III

Tossup

3) *Math – Short Answer:* Identify all of the following 3 sets of side lengths that form an obtuse triangle:

- I) 2, 3, and 4
- II) 5, 7, and 8
- III) 6, 6, and 8

ANSWER: I only

Bonus

3) *Math – Short Answer:* All even numbers from 200 to 300, except for those ending in 0, are multiplied together. What is the units digit of the product?

ANSWER: 6

Tossup

4) *Chemistry – Multiple Choice:* What conclusion can be drawn about the signs of the ΔH and ΔS of a reaction with an equilibrium constant which increases with increasing temperature?

- W) ΔH is positive, ΔS is positive
- X) ΔH is negative, ΔS is positive
- Y) ΔH is positive, no conclusion can be drawn about ΔS
- Z) ΔH is negative, no conclusion can be drawn about ΔS

ANSWER: Y) ΔH is positive, no conclusion can be drawn about ΔS

Bonus

4) *Chemistry – Short Answer:* Serena, trying to charge her phone, creates a voltaic cell that reduces AuCl_4^- to solid Au and oxidizes solid Cd to Cd^{2+} . When balancing the net ionic equation for the whole cell, what is the ratio of the coefficients of Au and Cd?

ANSWER: 2 : 3

Tossup

5) *Physics – Multiple Choice:* A system consists of two metal blocks, one made of copper and one made of iron. The copper block is at 100 Kelvin and the iron one is at 200 Kelvin. The heat capacity of the copper block is 6 Joules per Kelvin and the heat capacity of iron is 2 Joules per Kelvin. The two blocks are allowed to touch and come to equilibrium. In Kelvin, what is the final temperature of the two blocks?

- W) 100
- X) 125
- Y) 150
- Z) 175

ANSWER: X) 125

Bonus

5) *Physics – Short Answer:* Consider a uniform rod with a mass of 2 kilograms and a moment of inertia of 24 meters about the axis perpendicular passing through the middle of the rod. What is the rotational inertia about an axis perpendicular passing through a point 3 meters away from the center of the rod, in kilogram meters squared?

ANSWER: 42

Tossup

6) *Biology – Short Answer:* Order the following 3 stages of the Ovarian cycle from first to last:

- 1) Luteal phase
- 2) Follicular phase
- 3) Ovulation

ANSWER: 2, 3, 1

Bonus

6) *Biology – Short Answer:* What plant hormone activates plant defenses against insect herbivores and microbial pathogens?

ANSWER: Jasmonic Acid (ACCEPT: Jasmonate, salicylic acid)

Tossup

7) *Earth and Space – Multiple Choice:* Which of the following is the most likely atmospheric condition in an area where the environmental lapse rate is 14 Kelvin per kilometer?

- W) Absolute Stability
- X) Conditional Stability
- Y) Absolute Instability
- Z) Conditional Instability

ANSWER: Y) Absolute Instability

Bonus

7) *Earth and Space – Short Answer:* Identify all of the following 3 rocks which are not composed primarily of minerals:

- I) Anthracite
- II) Pumice
- III) Anorthosite

ANSWER: I and II

Tossup

8) *Chemistry – Multiple Choice:* Which of the following acids or bases is correctly paired with its anhydride?

- W) Sulfurous acid; SO₃
- X) Magnesium hydroxide; MgO
- Y) Perchloric acid; HCl
- Z) Carbonate; CO₂

ANSWER: X) Magnesium hydroxide; MgO

Bonus

8) *Chemistry – Multiple Choice:* The compound Hg₂Cl₂ decomposes into HgCl₂ and Hg endothermically. Which of the following best describes the spontaneity of the decomposition reaction?

- W) Spontaneous at low temperature
- X) Spontaneous at high temperatures
- Y) Always spontaneous
- Z) Never spontaneous

ANSWER: X) Spontaneous at high temperatures

Tossup

9) *Math – Short Answer:* Set A contains 5 elements. How many proper subsets of A are there?

ANSWER: 31

Bonus

9) *Math – Short Answer:* How many factors of 1800 are divisible by 9?

ANSWER: 12

Tossup

10) *Energy – Multiple Choice:* Scientists at Lawrence Berkeley National Laboratory are searching for exotic phenomena predicted by supersymmetric theories. Which of the following is the supersymmetric partner of the electron?

- W) Superelectron
- X) Electrino
- Y) Selectron
- Z) Electron boson

ANSWER: Y) Selectron

Bonus

10) *Energy – Short Answer:* Scientists at Lawrence Livermore National Lab have recently achieved ignition at the National Ignition Facility. This means that they have released more energy from fusion than they put into their deuterium sample to make it fuse into Helium-4. This implies that the external energy put into the fusion process is less than the difference in what quantity between the two nuclides?

ANSWER: Mass Defect (ACCEPT: Nuclear Binding Energy)

Tossup

11) *Physics – Short Answer:* A 3 meter long conducting rod is placed perpendicular to a 5 millitesla magnetic field. The rod is given a velocity of 30 meters per second mutually perpendicular to the rod and the field. What is the magnitude of the voltage difference between the two ends of the rod in Volts?

ANSWER: 0.45

Bonus

11) *Physics – Multiple Choice:* Which of the following statements best describes how an increase in temperature affects the viscosity of gases and liquids?

- W) Viscosity increases for a gas and decreases for a liquid
- X) Viscosity decreases for a gas and increases for a liquid
- Y) Viscosity increases for both gasses and liquids
- Z) Viscosity decreases for both gasses and liquids

ANSWER: W) Viscosity increases for a gas and decreases for a liquid

Tossup

12) *Biology – Short Answer:* Identify all of the following three diseases that are caused by a gram-positive bacteria:

- I) Anthrax
- II) Botulism
- III) Tuberculosis

ANSWER: All

Bonus

12) *Biology – Short Answer:* Select all of the following three conditions in which the lac operon would be active:

- I) Low lactose, high glucose
- II) High lactose, low glucose
- III) Low lactose, low glucose

ANSWER: II only

Tossup

13) *Earth and Space – Short Answer:* What apparent wobbling of the moon allows earth observers to view 59% of the moon's surface rather than the expected 50% from its 1-to-1 spin-orbit resonance?

ANSWER: Lunar Libration (ACCEPT: Libration)

Bonus

13) *Earth and Space – Multiple Choice:* Rohit, attempting to avoid the sunlight, decides to go outside for the first time at night on Halloween. He notices that the moon is in its first quarter. Which of the following is a possible date of the next lunar eclipse?

- W) November 8th
- X) November 16th
- Y) November 23rd
- Z) November 30th

ANSWER: W) November 8th

Tossup

14) *Physics – Multiple Choice:* Which of the following is not a unit of power?

- W) Newton meters per second
- X) Joules per second
- Y) Electron volts per second
- Z) Kilogram meters per second

ANSWER: Z) Kilogram meters per second

Bonus

14) *Physics – Multiple Choice:* 1 meter by 1 meter square plates are made from each of the following materials. Light rays of fixed intensity are shone directly onto each of the following plates. Which of the following materials will cause the plate to experience the largest force from the light?

- W) Very dark paint that acts like a black-body
- X) Polished metal with perfect specular reflection
- Y) Matte finish with perfect diffuse reflection
- Z) Uniform glass with an index of refraction equal to that of air

ANSWER: X) Polished metal with perfect specular reflection

Tossup

15) *Chemistry – Short Answer:* Order the following 4 elements from lowest to highest melting point:

- 1) Sodium
- 2) Carbon
- 3) Potassium
- 4) Iron

ANSWER: 3, 1, 4, 2

Bonus

15) *Chemistry – Multiple Choice:* A reagent which is both a strong nucleophile and a strong base is reacted with a secondary alkyl halide at a high temperature. Which of the following reaction mechanisms dominates?

- W) Sn1
- X) Sn2
- Y) E1
- Z) E2

ANSWER: Z) E2

Tossup

16) *Earth and Space – Short Answer:* Order the following 3 bodies of water in order of increasing depth in a water column at 40 degrees south in the Atlantic Ocean:

- 1) North Atlantic Deep Water
- 2) Antarctic Bottom Water
- 3) Antarctic Intermediate Water

ANSWER: 3,1,2

Bonus

16) *Earth and Space – Short Answer:* Order the following three sulfate minerals from most to least resistant to weathering:

- 1) Gypsum
- 2) Celestite
- 3) Barite

ANSWER: 3, 2, 1

Tossup

17) *Math – Multiple Choice:* Some matrix A is right multiplied by a 2 by 3 matrix B to produce a 4 by 3 matrix C . Which of the following options could be the dimensions of matrix A ?

- W) 3 by 2
- X) 2 by 3
- Y) 4 by 2
- Z) 3 by 3

ANSWER: Y) 4 by 2

Bonus

17) *Math – Short Answer:* Nikhil is playing the hit mobile game Genshin Impact. In Genshin, you can sacrifice 3 artifacts to create 1 new artifact, which has a 0.01% chance of being good. Given that Nikhil has an infinite supply of artifacts, then what distribution does the number of artifacts that he must sacrifice to get the first good artifact follow?

ANSWER: Geometric

Tossup

18) *Chemistry – Short Answer:* What is the average bond order of the S-O bonds in the compound HSO_4^- ?

ANSWER: 1.5

Bonus

18) *Chemistry – Multiple Choice:* Which of the following is closest to the bond angle in degrees of a PH_3 molecule?

- W) 93
- X) 98
- Y) 103
- Z) 108

ANSWER: W) 93

Tossup

19) *Biology – Multiple Choice:* Which of the following traits is not characteristic of a K selected species?

- W) Found in high density environments
- X) High parental care
- Y) Relatively unstable populations
- Z) Include humans, bison, and elephants

ANSWER: Y) Relatively unstable populations

Bonus

19) *Biology – Short Answer:* What tissue in the human body, possessing an abundance of mitochondria, is responsible for non-shivering thermogenesis?

ANSWER: Brown fat (ACCEPT: brown adipose tissue, DO NOT ACCEPT: adipose tissue)

Tossup

20) *Earth and Space – Short Answer:* The merger of the Milky Way and Andromeda galaxies is expected to form what type of galaxy?

ANSWER: Elliptical

Bonus

20) *Earth and Space – Short Answer:* Comets are differentiated from asteroids through what feature which results from the sublimation of volatile ices at their surface?

ANSWER: Coma (ACCEPT: Gas Tail)

Tossup

21) *Physics – Multiple Choice:* A sound is produced by a point source and a pressure wave detector is placed some distance away from it. The amplitude of the detector signal is proportional to what power of the distance?

- W) 2
- X) 1
- Y) -1
- Z) -2

ANSWER: Y) -1

Bonus

21) *Physics – Short Answer:* Order the following 3 fundamental constants at STP from least to greatest:

- 1) Fine Structure constant
- 2) Weak Coupling constant
- 3) Strong Coupling constant

ANSWER: 2, 1, 3

Tossup

22) *Math – Multiple Choice:* Let $f(x) = x^3 - 6x^2 + 9x - 3$. What is the value of the product of the values of the three roots of f divided by the sum of the values of the three roots?

- W) 1/6
- X) 1/3
- Y) 1/2
- Z) 1

ANSWER: Y) 1/2

Bonus

22) *Math – Short Answer:* An ant stands on a 2D plane at the point $(0, 0)$ tied to a rope of length 2 that cannot be stretched further. If there is a very long barrier on the line $y = 1$, to the nearest integer, how much area is available to the ant?

ANSWER: 10

Tossup

23) *Energy – Short Answer:* Researchers at Pacific Northwest National Lab are studying ways to improve industrial chemical production. The production of benzene from non-oil based precursors involves the disproportionation of what other organic ring compound?

ANSWER: Toluene

Bonus

23) *Energy – Short Answer:* Scientists at the National Ignition Facility at Lawrence Livermore National Lab are studying nuclear fusion, which produces energy by forming increasingly stable nuclides. An isotope of which element on the periodic table has the highest binding energy per nucleon of any nuclide?

ANSWER: Nickel