

Aerospace Propulsion

Lecture 7

Compressible Flows: Part I

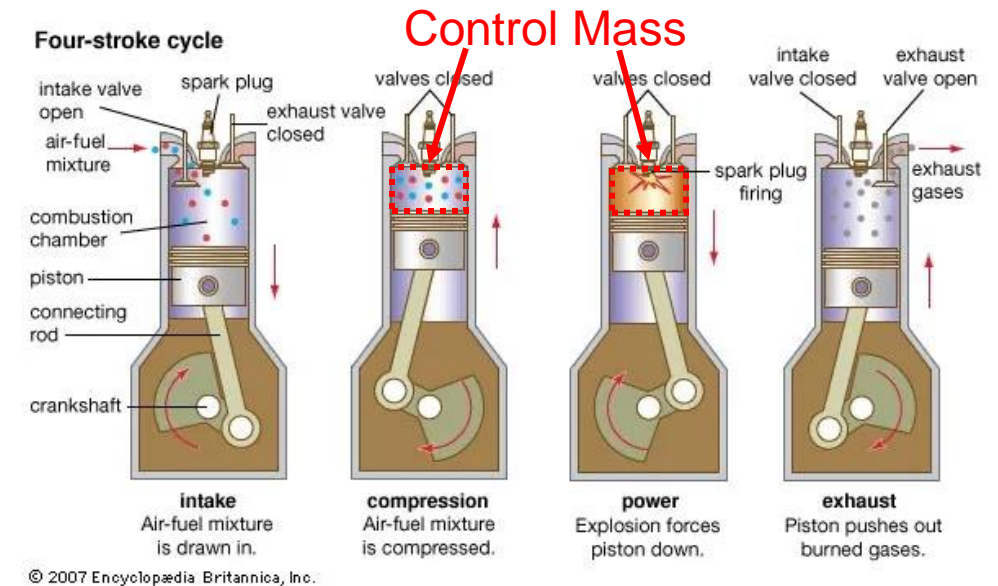
Compressible Flows: Part I

- Fluid Conservation Principles
- Speed of Sound
- Isentropic Flow
- Stagnation Properties

Fluid Conservation Principles

- Control Mass (Closed system)

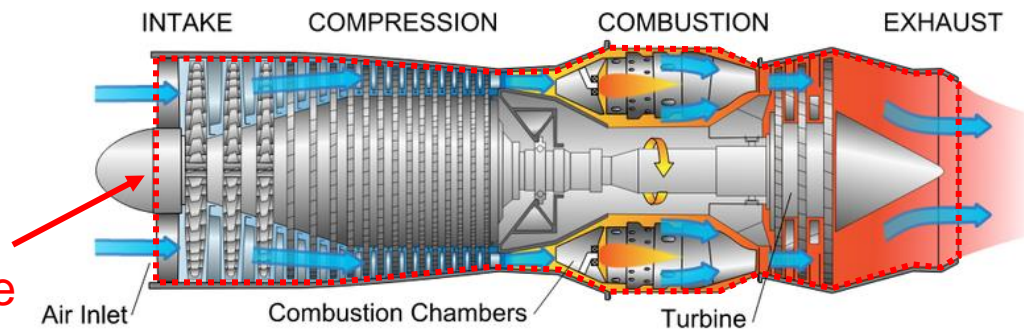
- Mass within system is fixed
 - Mass does not cross boundaries
- Example: Gasoline Engine



- Control Volume (Open system)

- Volume within system is fixed
 - Mass freely crosses boundaries
- Example: Gas turbine

Control Volume



Fluid Conservation Principles

- Reynold's transport theorem

$$\underbrace{\frac{DN}{Dt}}_{\text{Total rate of change of } N} = \underbrace{\frac{\partial}{\partial t} \iiint_{CV} \eta \rho dV}_{\text{Change of } N \text{ in control volume}} + \underbrace{\iint_{CS} \eta \rho (\vec{V} \cdot \hat{n}) dA}_{\text{Flux of } N \text{ across boundaries}}$$

Total rate of
change of N

Change of N in
control volume

Flux of N across
boundaries

V is volume
 \vec{V} is velocity
for now...

* \hat{n} points out

- N = extensive property
- $\eta = N/m$

Fluid Conservation Principles

- Mass

- Generally: $\frac{Dm}{Dt} = 0$

- $N = m$

- $\eta = \frac{m}{m} = 1$

- $0 = \frac{\partial}{\partial t} \iiint_{CV} \rho dV + \iint_{CS} \rho (\vec{V} \cdot \hat{n}) dA$

- $0 = \iint_{CS} \rho (\vec{V} \cdot \hat{n}) dA \quad (\text{steady state})$

Fluid Conservation Principles

- Momentum

- Generally: $\frac{Dm\vec{V}}{Dt} = \vec{F}_{net}$

Newton's
First
Law

- $N = m\vec{V}$

- $\eta = \vec{V}$

- $\vec{F}_{net} = \frac{\partial}{\partial t} \iiint_{CV} \rho \vec{V} dV + \iint_{CS} \rho \vec{V} (\vec{V} \cdot \hat{n}) dA$

- $\vec{F}_{net} = \iint_{CS} \rho \vec{V} (\vec{V} \cdot \hat{n}) dA$ (steady state)

Fluid Conservation Principles

• Energy

First Law of Thermo

$$\bullet \text{ Generally: } \frac{DE}{Dt} = \dot{Q} - \dot{W} = \dot{Q} - \underbrace{\oint_{CS} p (\vec{V} \cdot \hat{n}) dA}_{\text{Pressure work on surface}} - \underbrace{\dot{W}_{\text{other}}}_{\text{Shaft, viscous, etc.}}$$

$$\bullet N = E$$

$$\bullet \eta = e = u + \frac{1}{2} \vec{V} \cdot \vec{V}$$

$$\bullet \dot{Q} - \dot{W}_{\text{other}} = \frac{\partial}{\partial t} \iiint_{CV} \rho \underbrace{\left(u + \frac{\vec{V} \cdot \vec{V}}{2} \right)}_e dV + \oint_{CS} \rho \underbrace{\left(u + \frac{\vec{V} \cdot \vec{V}}{2} + \boxed{\frac{p}{\rho}} \right)}_e (\vec{V} \cdot \hat{n}) dA$$

$$\bullet \dot{Q} - \dot{W}_{\text{other}} = \oint_{CS} \rho \left(u + \frac{\vec{V} \cdot \vec{V}}{2} + \frac{p}{\rho} \right) (\vec{V} \cdot \hat{n}) dA \quad (\text{steady state})$$

Fluid Conservation Principles

- Example: Steady pipe flow

- Mass

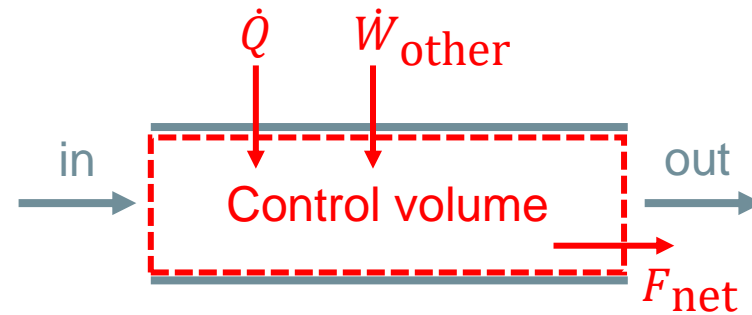
- $\dot{m}_{out} = \dot{m}_{in}$

- Momentum

- $F_{net} = (\dot{m}V)_{out} - (\dot{m}V)_{in}$

- Energy

- $\dot{Q} + \dot{W}_{other} = \left[\dot{m} \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) \right]_{out} - \left[\dot{m} \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) \right]_{in}$



Mass (Steady)

$$0 = \oint \rho (\vec{V} \cdot \hat{n}) dA$$

$$-(\rho V A)_{in} + (\rho V A)_{out} = 0$$

$$-\dot{m}_{in} + \dot{m}_{out} = 0$$

Momentum (Steady)

$$\oint \rho \vec{V} (\vec{V} \cdot \hat{n}) dA = F_{net}$$

$$-(\rho V^2 A)_{in} + (\rho V^2 A)_{out} = F_{net}$$

Energy (Steady)

$$\dot{Q} + \dot{W}_{other} = \oint \rho \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) (\vec{V} \cdot \hat{n}) dA$$

$$\dot{Q} + \dot{W}_{other} = - \left[\rho \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) V A \right]_{in} + \left[\rho \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) V A \right]_{out}$$

$$\dot{Q} + \dot{W}_{other} = - \left[\dot{m} \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) \right]_{in} + \left[\dot{m} \left(u + \frac{V^2}{2} + \frac{p}{\rho} \right) \right]_{out}$$

Speed of Sound

$$\gamma = \frac{C_p}{C_v}$$

- Sound waves are infinitesimal pressures waves
* Reversible and adiabatic (therefore also isentropic)
- Isentropic speed of sound a
 - $a \equiv \sqrt{\frac{\partial p}{\partial \rho}_s}$ } definition
 - For an Ideal Gas (constant specific heats)
 - $a = \sqrt{\gamma \frac{p}{\rho}} = \sqrt{\gamma R T}$
 - Mach Number: $M = V/a$

Isentropic
ideal gas
constant C_p

$$\left(\frac{p}{p_0}\right) = \left(\frac{\rho}{\rho_0}\right)^\gamma$$

$$\Rightarrow p = C \rho^\gamma$$

$$\left(\frac{\partial p}{\partial \rho}\right)_s = C \gamma \rho^{\gamma-1}$$

$$= C \gamma \frac{p}{\rho}$$

$$= \frac{\gamma p}{\rho}$$

Ideal gas Law: $p = \rho R T$

$$a = \sqrt{\left(\frac{\partial p}{\partial \rho}\right)_s} = \sqrt{\frac{\gamma p}{\rho}} = \sqrt{\gamma R T}$$

Isentropic Flow

- “One-dimensional” flow, no friction, no heating

- Mass

Mass (Steady)

$$\oint \rho(\vec{V} \cdot \hat{n}) dA = 0$$

$$(p+dp)(v+dv)(A+dA) - \rho VA = 0$$

$$\boxed{\rho VA} + \rho V dA + \rho dVA + \underbrace{\rho dv dA}_{\text{H.O.T.}} + d\rho VA + \underline{d\rho V dA} + \underline{d\rho dv A} + \underline{d\rho dv dA} - \boxed{\rho VA} = 0$$

$$\rho V dA + \rho dVA + d\rho VA = 0$$

divide ρVA

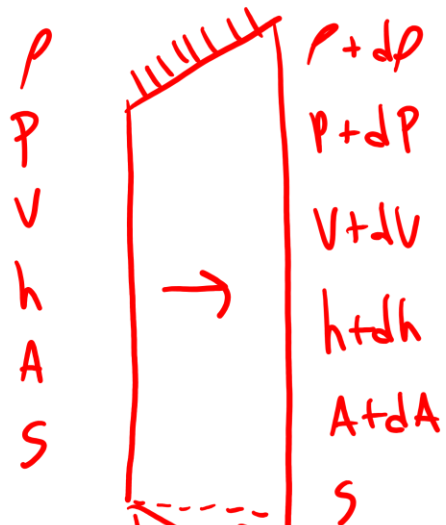
$$\frac{dA}{A} + \frac{dv}{v} + \frac{d\rho}{\rho} = 0$$

chain rule

$$d(\rho VA) = 0$$

$$d\dot{m} = 0$$

$$\bullet \frac{d\rho}{\rho} + \frac{dv}{v} + \frac{dA}{A} = 0$$



Fluid element

* Recall, $\frac{dA}{A} + \frac{dV}{V} + \frac{dp}{\rho} = 0$

Isentropic Flow

- “One-dimensional” flow, no friction, no heating
 - Momentum (Steady, X-direction)

$$\oint \rho \vec{v} (\vec{v} \cdot \hat{n}) dA = \sum F_x$$

$$(P+dp)(V+dV)(V+dV)(A+dA) - \rho V^2 A = \sum F_x$$

Expanded, neglect HOT

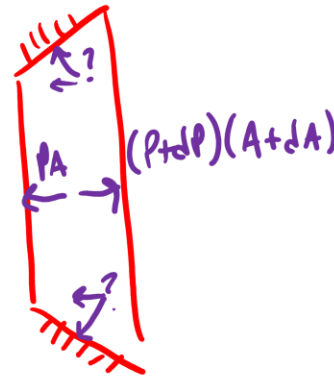
$$\boxed{\rho V^2 A} + \rho V^2 dA + 2\rho V A dV + V^2 A dp - \boxed{\rho V^2 A} = \sum F_x$$

$$\rho V^2 A \left(\frac{dA}{A} + \frac{2dV}{V} + \frac{dp}{\rho} \right) = \sum F_x$$

$$\rho V^2 A \left(\frac{dV}{V} \right) = \sum F_x$$

$$\bullet dp + \rho V dV = 0$$

$$\rho V^2 A \frac{dV}{V} = -A dp$$



Assume:

- θ is very small
- Pressure varies linearly across the wall
- Average pressure on each wall $P + \frac{dp}{2}$

Trig:

$$\sin \theta = \frac{dA/2}{A_{wall}} \Rightarrow A_{wall} = \frac{dA}{2 \sin \theta}$$

$$\sin \theta = \frac{F_x}{F} \Rightarrow F_x = F \sin \theta$$

$$\sum F_x = PA - (P+dp)(A+dA) + \left(P + \frac{dp}{2} \right) \left(\frac{dA}{2 \sin \theta} \right) (2) \sin \theta$$

$$\sum F_x = \boxed{PA - PA} - \boxed{Adp} - \boxed{PdA} - \boxed{dAdA} + \boxed{PdA} + \boxed{dPdA/2}$$

HOT HOT

$$\sum F_x = -Adp$$

Pressure A_{wall} 2 walls

F

F_x

Isentropic Flow

- “One-dimensional” flow, no friction, no heating
 - Energy (Steady)

$$\oint \rho(e + \frac{1}{2}V^2)(\vec{V} \cdot \hat{n})dA = 0$$

$$\oint \rho e(\vec{V} \cdot \hat{n})dA + \oint P(\vec{V} \cdot \hat{n})dA = 0$$

$$(\rho + d\rho)(e + de)(V + dV)(A + dA) - \rho e V A + (P + dP)(V + dV)(A + dA) = 0$$

Expand, neglect H.O.T., use continuity

$$de = -\frac{P}{\rho} \left[\frac{d\rho}{\rho} - \frac{d\rho}{\rho} \right] \quad (1)$$

$$\bullet dh + VdV = 0 \Rightarrow d \left(\underbrace{\frac{\gamma}{\gamma-1} RT}_{C_p} + \frac{1}{2} V^2 \right) = 0$$

for open system, prefer enthalpy

$$e = u + \frac{1}{2}V^2 \quad \left. \begin{array}{l} e = h - \frac{P}{\rho} + \frac{1}{2}V^2 \\ h = u + \frac{P}{\rho} \end{array} \right\}$$

$$de = dh - \frac{dP}{\rho} + \frac{P d\rho}{\rho^2} + VdV \quad (2)$$

$$(1) = (2)$$

$$-\frac{P}{\rho} \left[\frac{d\rho}{\rho} - \frac{d\rho}{\rho} \right] = dh - \frac{dP}{\rho} + \frac{P d\rho}{\rho^2} + VdV$$

$$dh + VdV = 0$$

Isentropic Flow

- “One-dimensional” flow, isentropic, **no work or heat**
 - $d\left(h + \frac{1}{2}V^2\right) = d\left(\frac{\gamma}{\gamma-1}RT + \frac{1}{2}V^2\right) = 0$
- First Law (open system)
 - $d\left(h + \frac{1}{2}V^2\right) = \delta q - \delta w$
 - Complicated accounting
 - If we add heat/work, does it change the kinetic energy or enthalpy?
- We can avoid the distinction by creating a new property that accounts for enthalpy and kinetic energy

Stagnation Properties

$$C_p = \frac{R}{\gamma - 1}$$

$\gamma = \frac{C_p}{C_v}$ $R = C_p - C_v$

- Thermodynamic properties of a gas brought to rest isentropically and adiabatically

constant C_p

- Stagnation Enthalpy

- $h_t = h + \frac{1}{2} V^2$

$$h_t = h + \frac{1}{2} V^2$$

$$h_t - h = \frac{1}{2} V^2$$

$$C_p (T_t - T) = \frac{1}{2} V^2$$

ideal gas
 $a^2 = \gamma R T$

$$T_t = T + \frac{V^2}{2 C_p}$$

$$T_t = T + \frac{V^2}{2} \frac{\gamma - 1}{\gamma R}$$

$$T_t = T \left(1 + \frac{V^2}{2} \frac{\gamma - 1}{\gamma} \frac{1}{R T} \right)$$

$$T_t = T \left(1 + \frac{V^2}{2} \frac{\gamma - 1}{a^2} \right)$$

$$T_t = T \left(1 + M^2 \frac{\gamma - 1}{2} \right)$$

- $T_t = T \left(1 + \frac{\gamma - 1}{2} M^2 \right)$

Stagnation Properties

- Thermodynamic properties of a gas brought to rest isentropically and adiabatically

- Stagnation Pressure

$$p_t = p \left(1 + \frac{\gamma-1}{2} M^2 \right)^{\frac{\gamma}{\gamma-1}}$$

Isentropic
flow
equation

- Stagnation Density

$$\rho_t = \rho \left(1 + \frac{\gamma-1}{2} M^2 \right)^{\frac{1}{\gamma-1}}$$

- What happens to these quantities in an isentropic process?

No work, No heat \rightarrow all remain constant

Stagnation Properties

- Recall we said for the First Law (open system)
 - $d\left(h + \frac{1}{2}V^2\right) = \delta q - \delta w$
- Using the stagnation concept, we instead write
 - $dh_t = \delta q - \delta w$
- We no longer need to worry which term the heat and work modify (worry about it later)