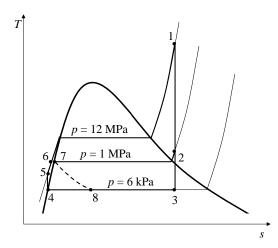
8.52 As indicated in Fig. P8.52, a power plant similar to that in Fig. 8.11 operates on a regenerative vapor power cycle with one closed feedwater heater. Steam enters the first turbine stage at state 1 where pressure is 12 MPa and temperature is 560°C. Steam expands to state 2 where pressure is 1 MPa and some of the steam is extracted and diverted to the closed feedwater heater. Condensate exits the feedwater heater at state 7 as saturated liquid at a pressure of 1 MPa, undergoes a throttling process through a trap to a pressure of 6 kPa at state 8, and then enters the condenser. The remaining steam expands through the second turbine stage to a pressure of 6 kPa at state 3 and then enters the condenser. Saturated liquid feedwater exiting the condenser at state 4 at a pressure of 6 kPa enters a pump and exits the pump at a pressure of 12 MPa. The feedwater then flows through the closed feedwater heater, exiting at state 6 with a pressure of 12 MPa. The net power output for the cycle is 330 MW. For isentropic processes in each turbine stage and the pump, determine

- (a) the cycle thermal efficiency.
- (b) the mass flow rate into the first turbine stage, in kg/s.
- (c) the rate of entropy production in the closed feedwater heater, in kW/K.
- (d) the rate of entropy production in the steam trap, in kW/K.

KNOWN: A regenerative vapor power cycle with one closed feedwater heater operates with steam as the working fluid. Operational data are provided.

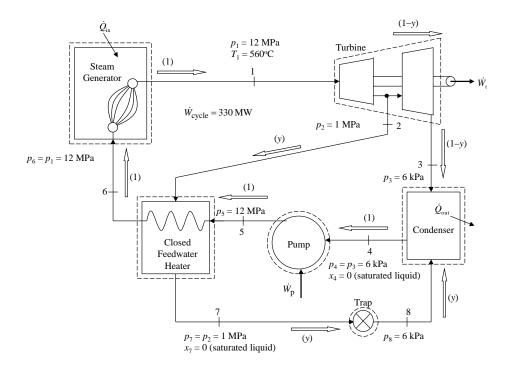
FIND: Determine (a) the cycle thermal efficiency, (b) the mass flow rate into the first turbine stage, in kg/s, (c) the rate of entropy production in the closed feedwater heater, in kW/K, and (d) the rate of entropy production in the steam trap, in kW/K.

SCHEMATIC AND GIVEN DATA:



State	p (kPa)	T (°C)	h (kJ/kg)	s (kJ/kg·K)	x
1	12,000	560	3506.2	6.6840	
2	1,000		2823.3	6.6840	
3	6		2058.2	6.6840	0.7892
4	6		151.53	0.5210	0
5	12,000		163.60	0.5210	
6	12,000		606.61	1.7808	
7	1,000		762.81	2.1387	0
8	6		762.81	2.4968	0.2530

P8.52



ENGINEERING MODEL:

- 1. Each component of the cycle is analyzed as a control volume at steady state. The control volumes are shown on the accompanying sketch by dashed lines.
- 2. All processes of the working fluid are internally reversible except for heat transfer through a finite temperature difference in the closed feedwater heater and throttling through the trap.
- 3. The turbines, pump, closed feedwater heater, and steam trap operate adiabatically.
- 4. Kinetic and potential energy effects are negligible.
- 5. Saturated liquid exits the closed feedwater heater, and saturated liquid exits the condenser.

ANALYSIS:

(a) Applying energy and mass balances to the control volume enclosing the closed feedwater heater, the fraction of flow, y, extracted at location 2 is

$$y = \frac{h_6 - h_5}{h_2 - h_7} = \frac{(606.61 - 163.60) \text{ kJ/kg}}{(2823.3 - 762.81) \text{ kJ/kg}} = 0.2150$$

For the control volume surrounding the turbine stages

$$\frac{\dot{W}_{t}}{\dot{m}_{1}} = (h_{1} - h_{2}) + (1 - y)(h_{2} - h_{3})$$

$$\frac{\dot{W}_{t}}{\dot{m}_{1}} = (3506.2 - 2823.3) \frac{kJ}{kg} + (1 - 0.2150)(2823.3 - 2058.2) \frac{kJ}{kg} = 1283.5 \text{ kJ/kg}$$

For the pump

$$\frac{\dot{W}_{\rm p}}{\dot{m}_1} = (h_5 - h_4)$$

$$\frac{\dot{W}_{\rm p}}{\dot{m}_{\rm l}} = (163.60 - 151.53) \frac{\rm kJ}{\rm kg} = 12.07 \text{ kJ/kg}$$

For the working fluid passing through the steam generator

$$\frac{\dot{Q}_{\text{in}}}{\dot{m}_1} = h_1 - h_6 = (3506.2 - 606.61) \frac{\text{kJ}}{\text{kg}} = 2899.6 \text{ kJ/kg}$$

Thus, the thermal efficiency is

$$\eta = \frac{\dot{W}_{t} / \dot{m}_{l} - \dot{W}_{p} / \dot{m}_{l}}{\dot{Q}_{in} / \dot{m}_{l}} = \frac{(1283.5 - 12.07) \text{ kJ/kg}}{2899.6 \text{ kJ/kg}} = \frac{\textbf{0.438 (43.8\%)}}{\textbf{0.438 (43.8\%)}}$$

(b) The *net* power developed is

$$\dot{W}_{\text{cycle}} = \dot{m}_1 (\dot{W}_{\text{t}} / \dot{m}_1 - \dot{W}_{\text{p}} / \dot{m}_1)$$

Thus,

$$\dot{m}_{1} = \frac{\dot{W}_{\text{cycle}}}{(\dot{W}_{\text{t}} / \dot{m}_{1} - \dot{W}_{\text{p}} / \dot{m}_{1})}$$

$$\dot{m}_1 = \frac{330 \,\mathrm{MW}}{(1283.5 - 12.07) \frac{\mathrm{kJ}}{\mathrm{kg}}} \left| \frac{1000 \frac{\mathrm{kJ}}{\mathrm{s}}}{1 \,\mathrm{MW}} \right| = \frac{259.6 \,\mathrm{kg/s}}{1000 \,\mathrm{kg/s}}$$

(c) The rate of entropy production in the closed feedwater heater is determined using the steady-state form of the entropy rate balance:

$$0 = \sum_{j} \frac{\dot{Q}_{j}}{T_{j}} + \sum_{i} \dot{m}_{i} s_{i} - \sum_{e} \dot{m}_{e} s_{e} + \dot{\sigma}_{cv}$$

Since the feedwater heater is adiabatic, the heat transfer term drops. Thus,

$$\dot{\sigma}_{cv} = \sum_{e} \dot{m}_{e} s_{e} - \sum_{i} \dot{m}_{i} s_{i} = \dot{m}_{6} s_{6} + \dot{m}_{7} s_{7} - \dot{m}_{5} s_{5} - \dot{m}_{2} s_{2}$$

$$\dot{\sigma}_{cv} = \dot{m}_1[s_6 - s_5 + y(s_7 - s_2)]$$

$$\dot{\sigma}_{\text{cv}} = 259.6 \frac{\text{kg}}{\text{s}} [1.7808 - 0.5210 + (0.2150)(2.1387 - 6.6840)] \frac{\text{kJ}}{\text{kg} \cdot \text{K}} \left| \frac{1 \text{ kW}}{1 \text{ kJ/s}} \right| = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{73.35 kW/K}}{\textbf{K}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kW}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | \frac{1 \text{ kJ/s}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | = \frac{\textbf{K}}{1 \text{ kJ/s}} | = \frac$$

Heat transfer between a finite temperature difference within the closed feedwater heater is a source of irreversibility that produces entropy.

(d) The rate of entropy production in the steam trap is determined using the one-inlet, one-exit, steady-state form of the entropy rate balance:

$$0 = \sum_{i} \frac{\dot{Q}_{j}}{T_{j}} + \dot{m}(s_{i} - s_{e}) + \dot{\sigma}_{cv}$$

where \dot{m} is the mass flow rate through the steam trap.

Since the steam trap is adiabatic, the heat transfer term drops. Thus,

$$\dot{\sigma}_{cv} = \dot{m}(s_e - s_i) = \dot{m}_7(s_8 - s_7) = y\dot{m}_1(s_8 - s_7)$$

$$\dot{\sigma}_{cv} = (0.2150) \left(259.6 \frac{kg}{s} \right) (2.4968 - 2.1387) \left| \frac{kJ}{kg \cdot K} \right| \frac{1 \, kW}{1 \, kJ/s} = \frac{19.99 \, kW/K}{1 \, kJ/s}$$

The throttling process in the steam trap is a source of irreversibility that produces entropy.