



# **Anchoring Synthetic DNA to a Membrane: A Novel Application of Free Energy Perturbations**

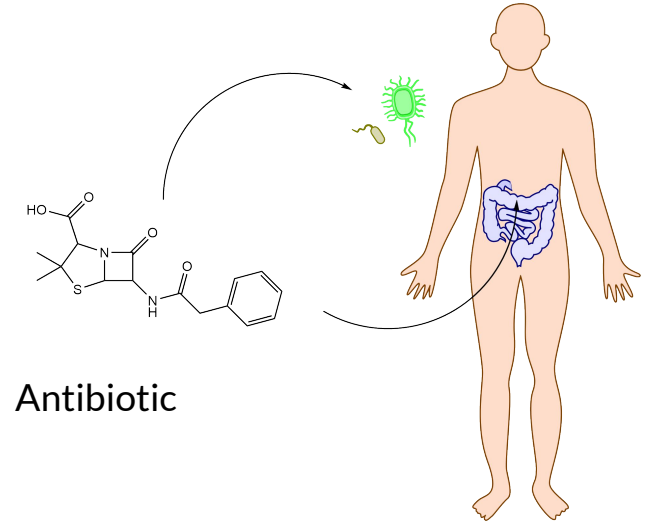
Ezry Santiago  
Seminar Presentation  
November 9, 2021

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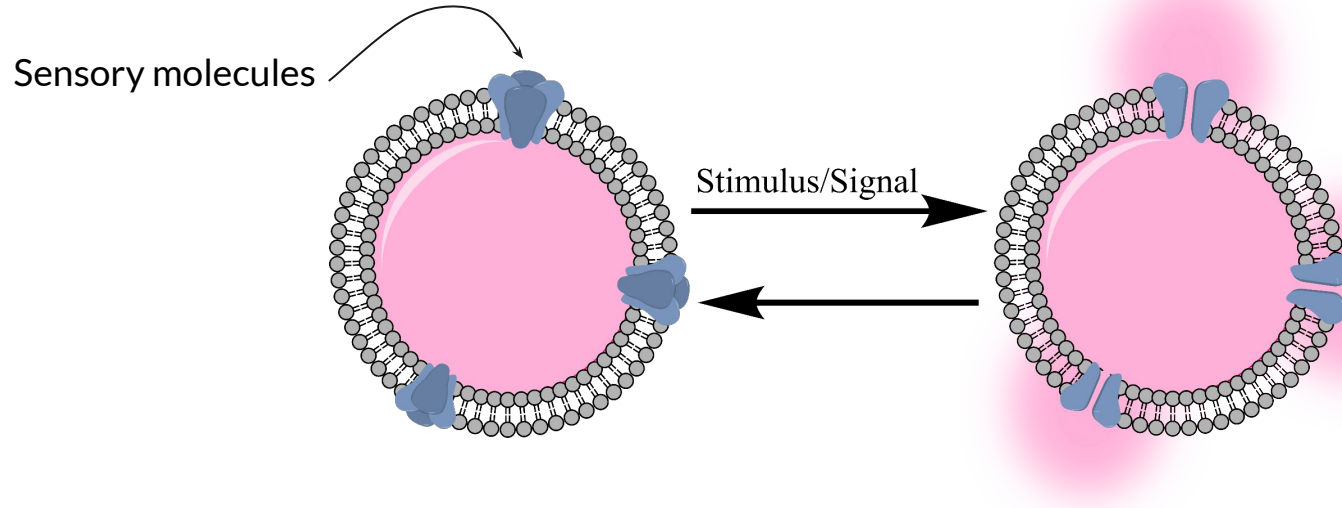
# Motivation: Drug Delivery

# The Problem of Drug Delivery

- The distribution of a drug in the body is hard to control.
- Non-specific delivery leads to side-effects.



# Simple Drug-Delivering Vesicle

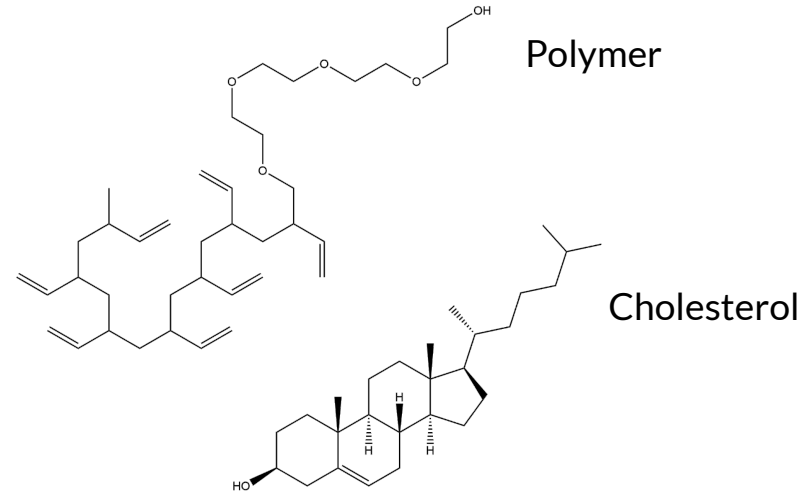
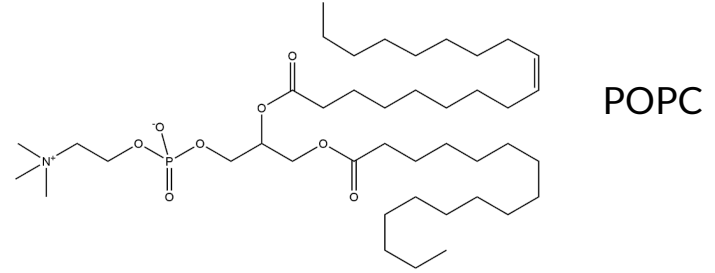


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# Bottom-Up Design

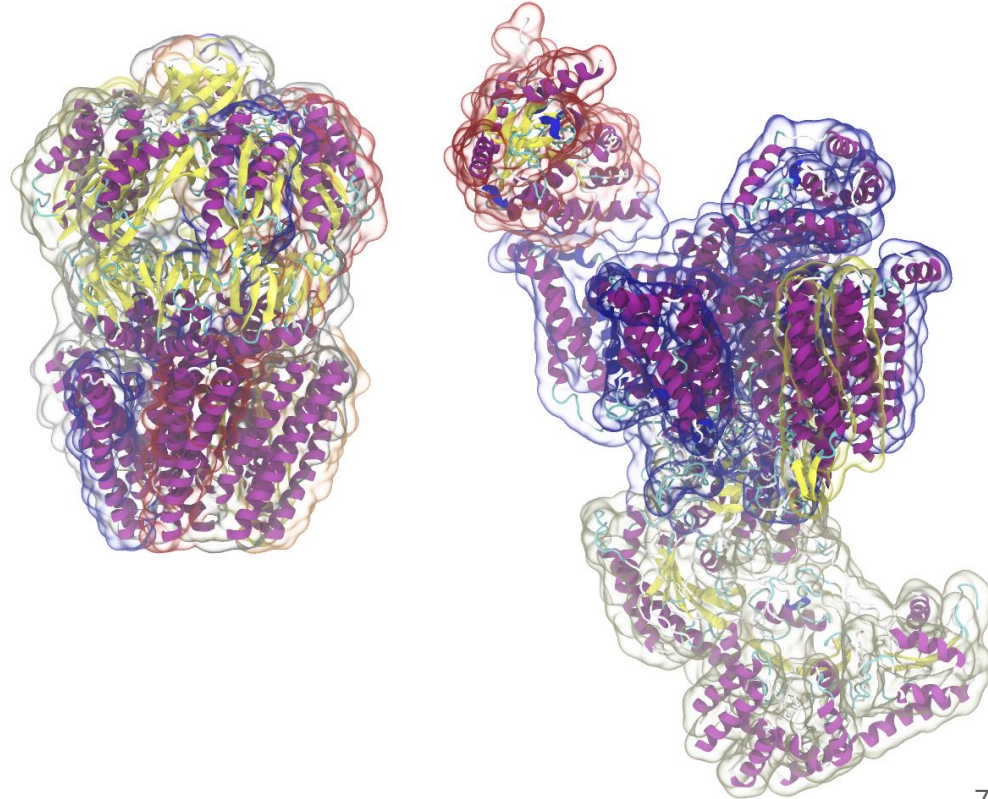
# Polar Lipids (and other amphipaths)

- Obvious choice for membranes
- Cheap
- Easy to manipulate
- Stable and biocompatible



# Proteins?

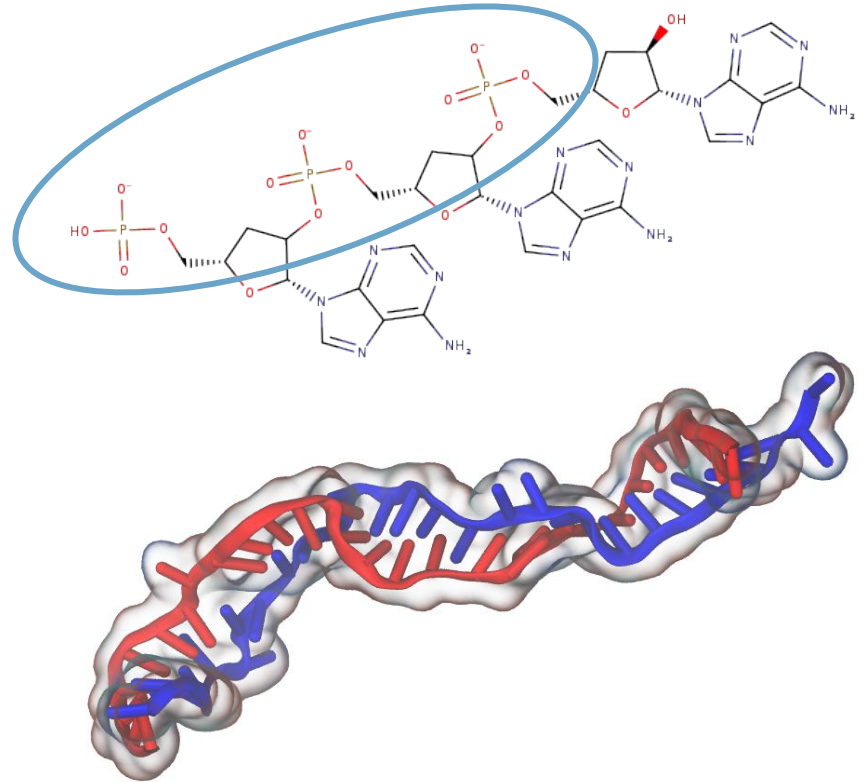
- Pros:
  - Structural and functional
  - Diverse conjugation chemistry
- Cons:
  - Difficult to design
  - Expensive to prototype



Structures: 2OAU (left) 3JBR (right)

# DNA nanotechnology

- Pros:
  - Tractable design due to base-pairing
  - Both structural and functional
  - Diverse conjugation chemistry
- Cons:
  - Less scalable
  - Polyanion





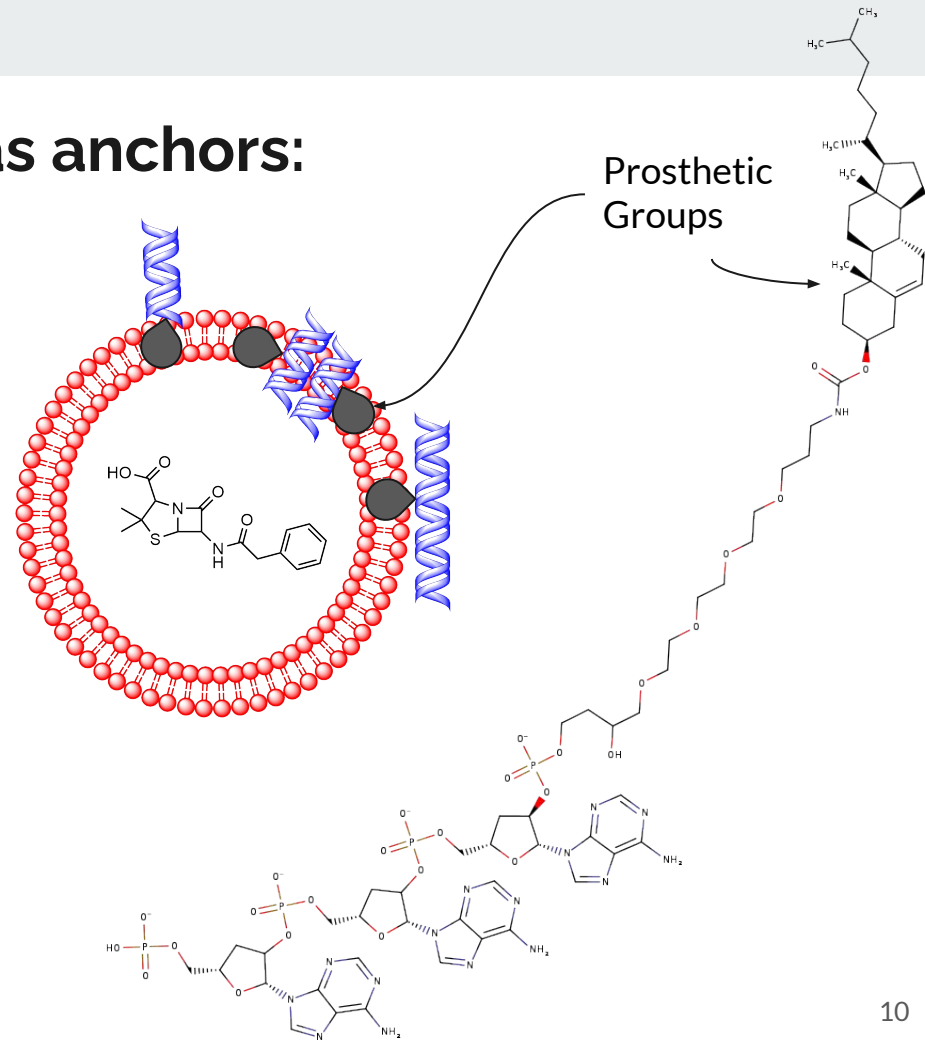
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## A Problem:

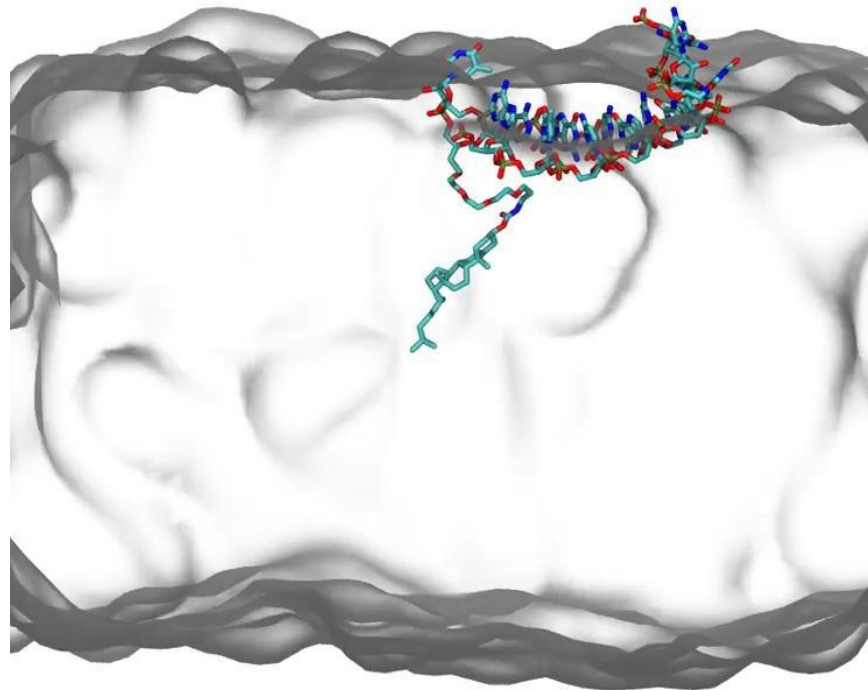
How can we get DNA to interact stably with the hydrophobic region of a membrane?

# Prosthetic groups can act as anchors:

- Chemical interfaces between DNA and other molecules



# DNA Patched to a Membrane (MD Simulation)



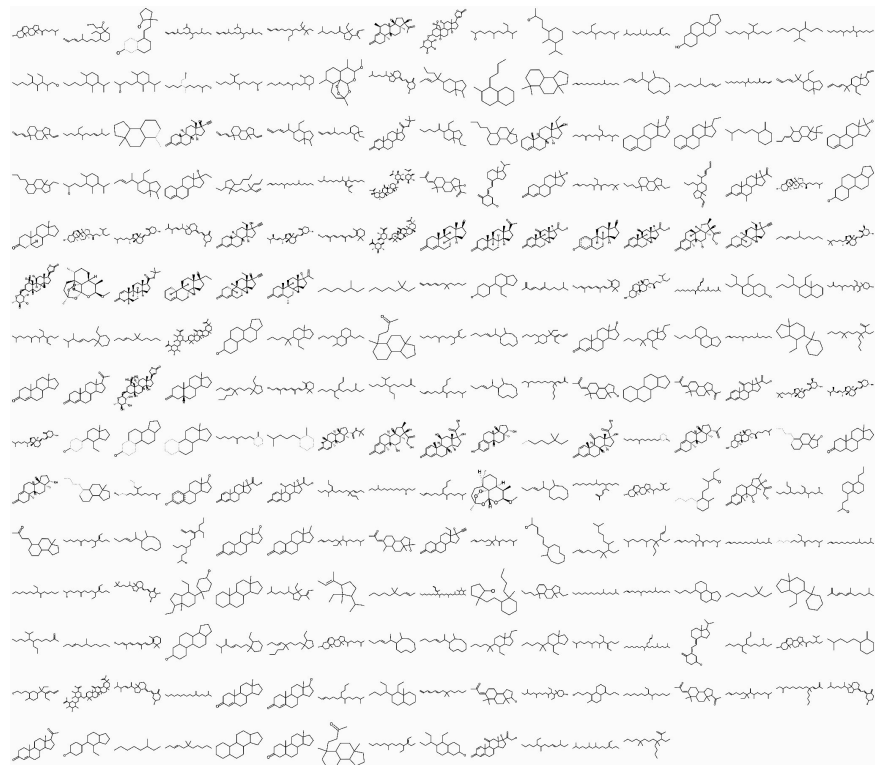
- Nitrogen
  - Oxygen
  - Carbon
  - Phosphorous
  - Lipids
- Water not shown

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**The question:**

**Which prosthetic group stabilizes the DNA-Membrane complex most?**

# Current Problem: Which prosthetic group is optimal?



a small selection of possible prosthetic groups  
(sterols and sphingosine derivatives)

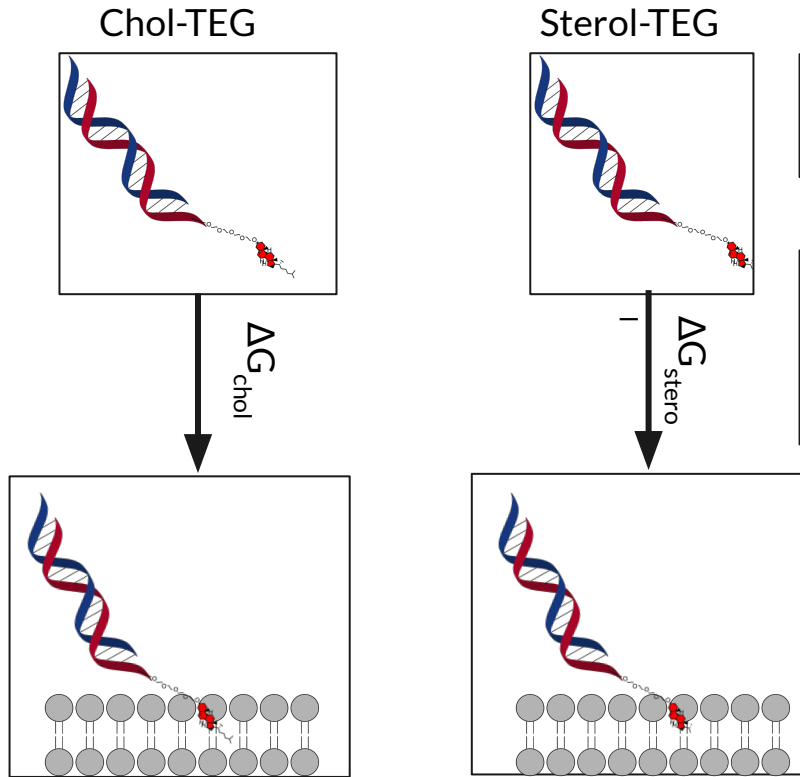


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**Our Approach:**

**Comparing Free Energies of Insertion**

# Free Energy Comparisons



$$\Delta\Delta G_{\text{chol} \rightarrow \text{sterol}} = \Delta G_{\text{sterol}} - \Delta G_{\text{chol}}$$

$\Delta\Delta G < 0$  Chol is more stable  
 $\Delta\Delta G > 0$  Sterol is more stable

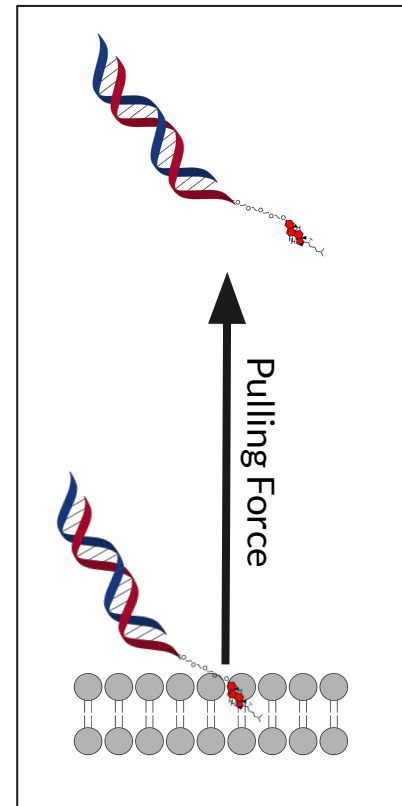


# Free Energy Calculations

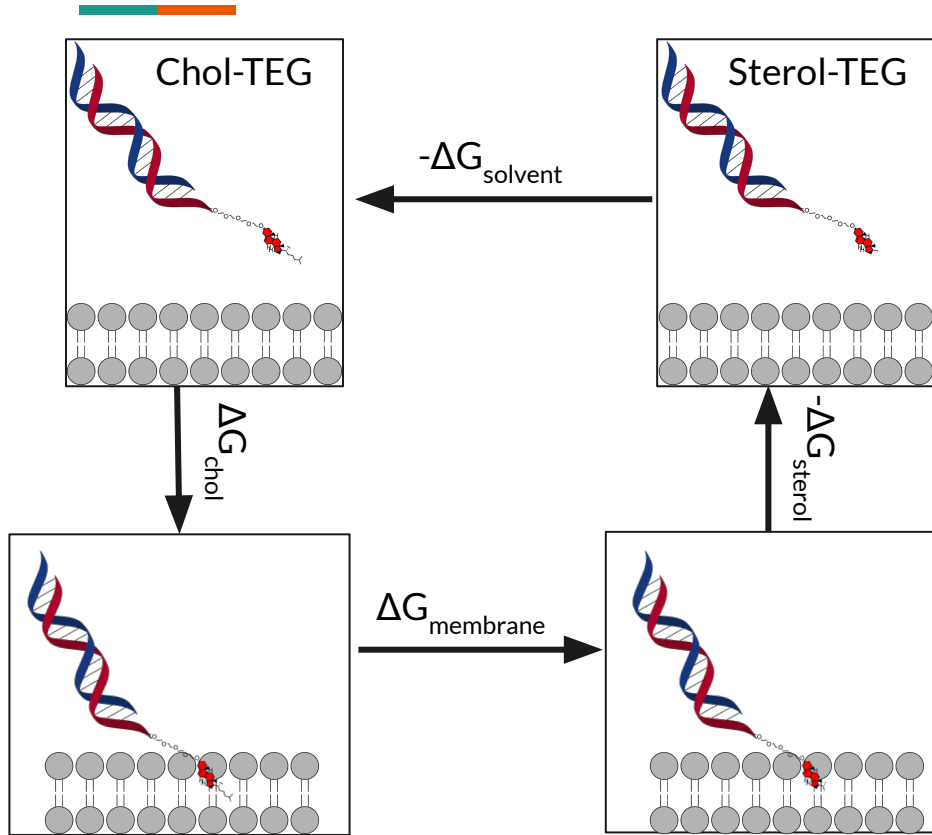
## Option 1: Spatial Method(s)

Example:

1. Push/pull the prosthetic group in/out of the membrane
2. Measure the resulting reactive forces to estimate the change in  $F$  over time
3. Sum the changes over  $dt$  to get the total
4. Repeat for each prosthetic group



# Free Energy Comparisons: Thermodynamic Cycle



$$\Delta G_{\text{chol}} + \Delta G_{\text{membrane}} - \Delta G_{\text{sterol}} - \Delta G_{\text{solvent}} = 0$$

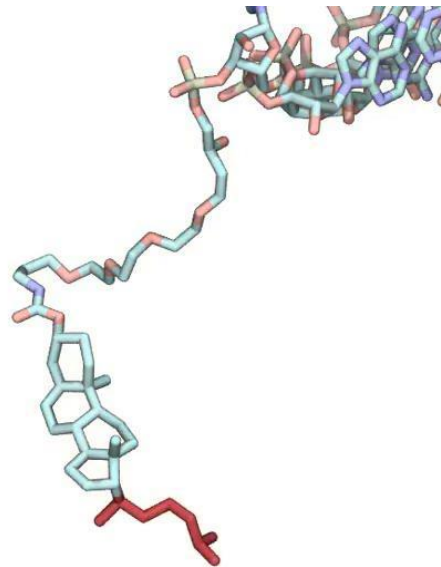
$$\Delta \Delta G_{\text{chol} \rightarrow \text{sterol}} = \Delta G_{\text{sterol}} - \Delta G_{\text{chol}}$$

$$\Delta \Delta G_{\text{chol} \rightarrow \text{sterol}} = \Delta G_{\text{membrane}} - \Delta G_{\text{solvent}}$$

# Getting $\Delta G_{\text{membrane}}$ and $\Delta G_{\text{solvent}}$ : Alchemical Free Energy Calculations (FEP)

$$\Delta G_{a \rightarrow b} = \sum_{\lambda_i=0}^1 \ln \langle \exp [-\beta [E_{\lambda_{i+1}} - E_{\lambda_i}]] \rangle_{\lambda_i}$$

1. Decompose the two states into N intermediates:  $\{\lambda_0, \lambda_1, \lambda_2 \dots \lambda_N\}$
2. Simulate in state  $\lambda_i$
3. Periodically calculate the internal energy (E) as if we were in state  $\lambda_{i+1}$
4. Use the Boltzmann distribution to estimate  $\Delta G(\lambda_i, \lambda_{i+1})$
5. Sum over all substates



# Uses and Challenges of Alchemical Free Energy Perturbations



## Uses:

- Traditionally used in drug discovery
- Application to membrane dynamics is very new (<5 years)
- When it works, it's much more efficient than spatial methods

## Challenges:

- Windows can't be too wide
- Each window must be simulated for a sufficient time

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**Question:**  
**Will FEP work in this system?**

## Two Main Criteria:



1. Convergence:

Does our estimate stabilize over simulation time?

2. Error:

How large is the remaining error?

# Hysteresis Plots and Convergence

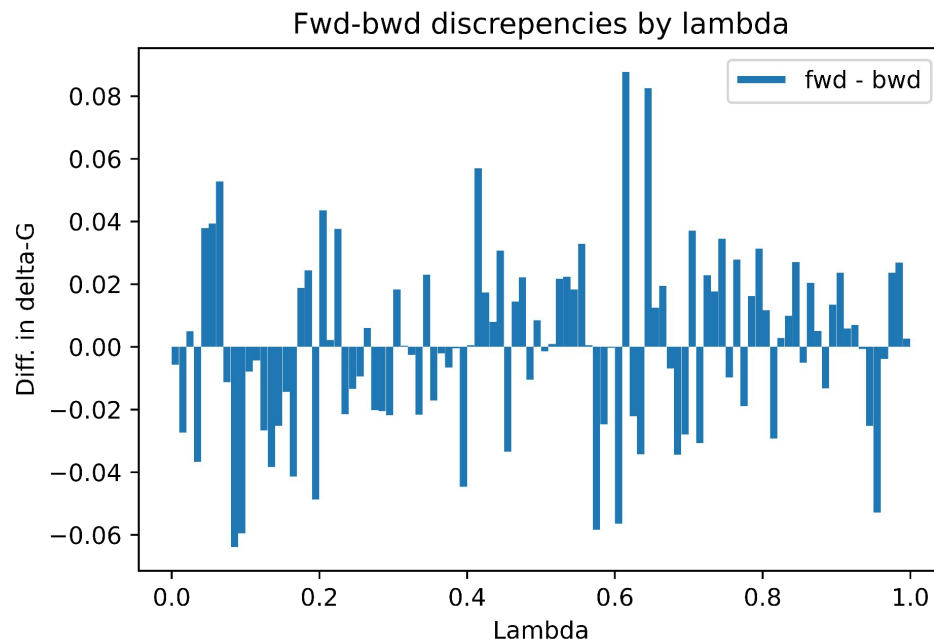
$$\Delta G_{a \rightarrow b} = \sum_{\lambda_i=0}^1 \ln \langle \exp [-\beta [E_{\lambda_{i+1}} - E_{\lambda_i}]] \rangle_{\lambda}$$

For every pair  $\lambda_i$  and  $\lambda_{i+1}$

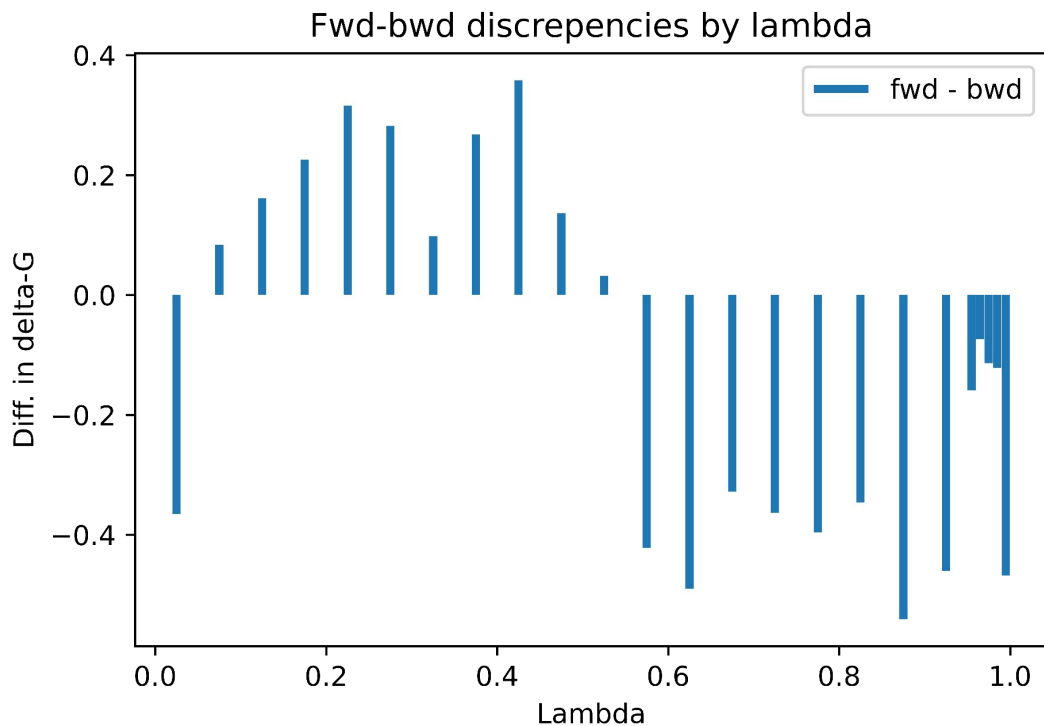
With sufficient simulation time:

$$\Delta G(\lambda_i, \lambda_{i+1}) = \Delta G(\lambda_{i+1}, \lambda_i)$$

I.e. The calculation **converges**



# First Attempt (July)

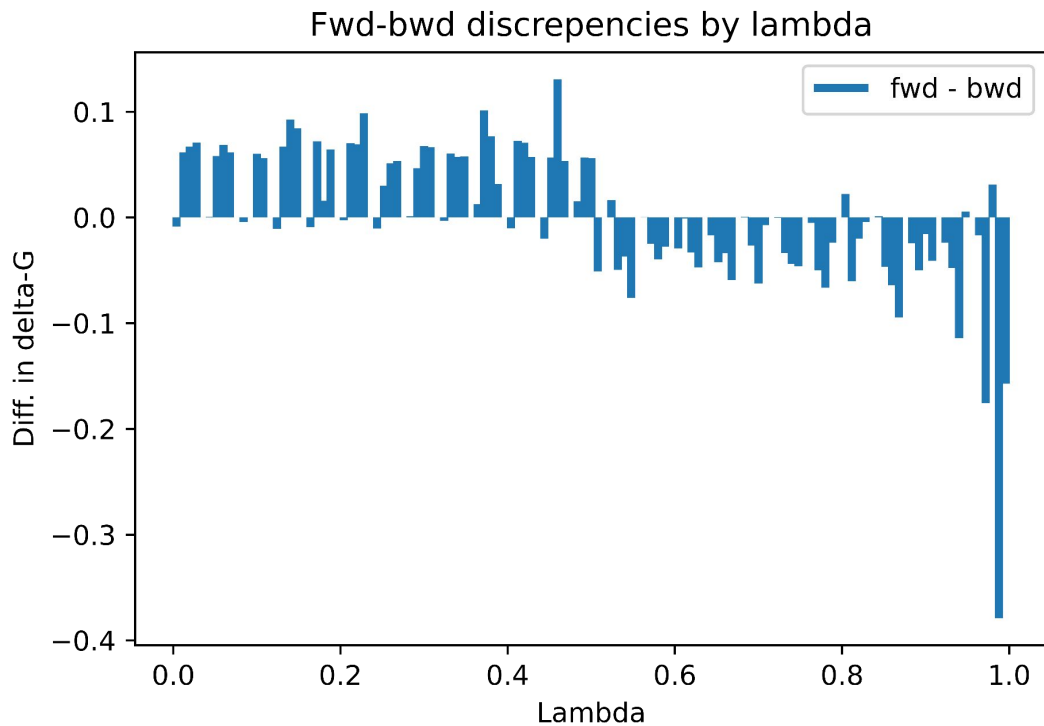


Issues:

1. Differences are large
2. Sign switch near 0.5



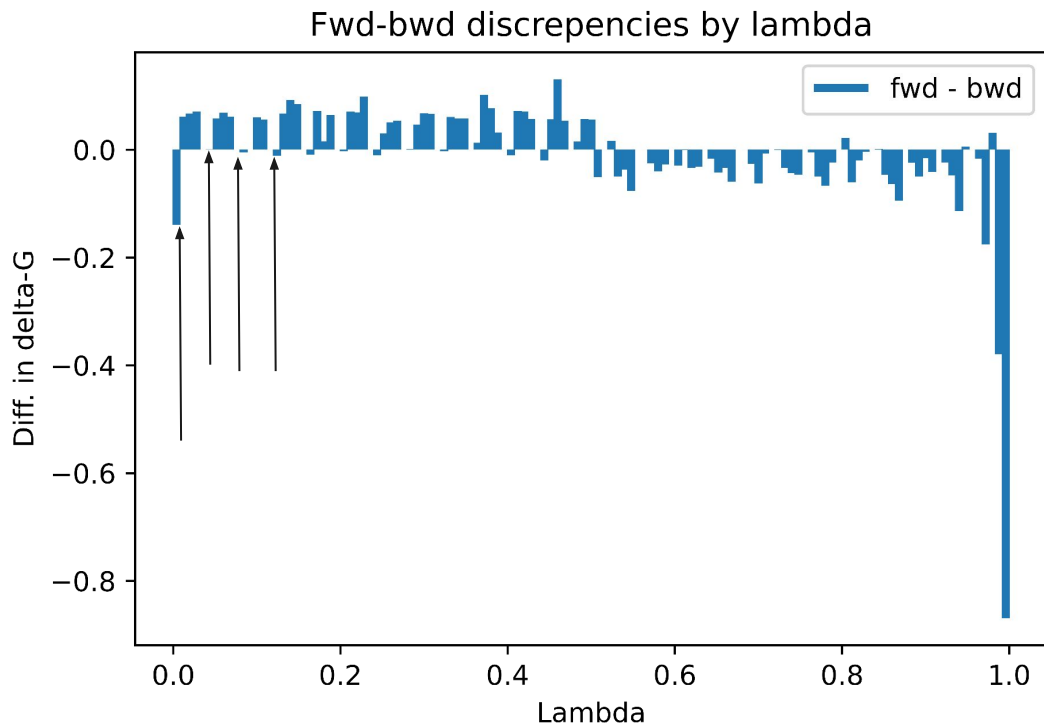
# Second Attempt



Issues:

1. Differences are large
2. Sign switch near 0.5
3. Stubborn difference at 1

# Penultimate Attempt




We tried everything

Issues:

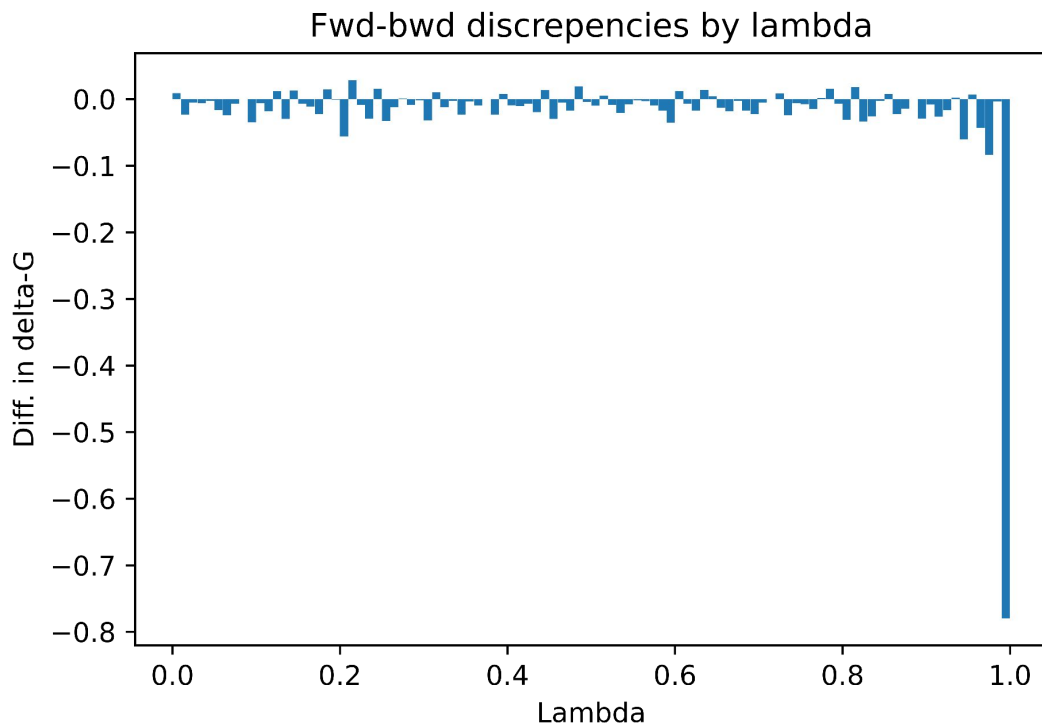
1. Differences are large
2. Sign switch near 0.5
3. Stubborn difference at 1

# There was a bug in NAMD

## With implications for many other labs

namd-l: Bug advisory and workaround: alchemical FEP with IDWS can give wrong comparison energies  Inbox x

# Latest Results (still running, preliminary)

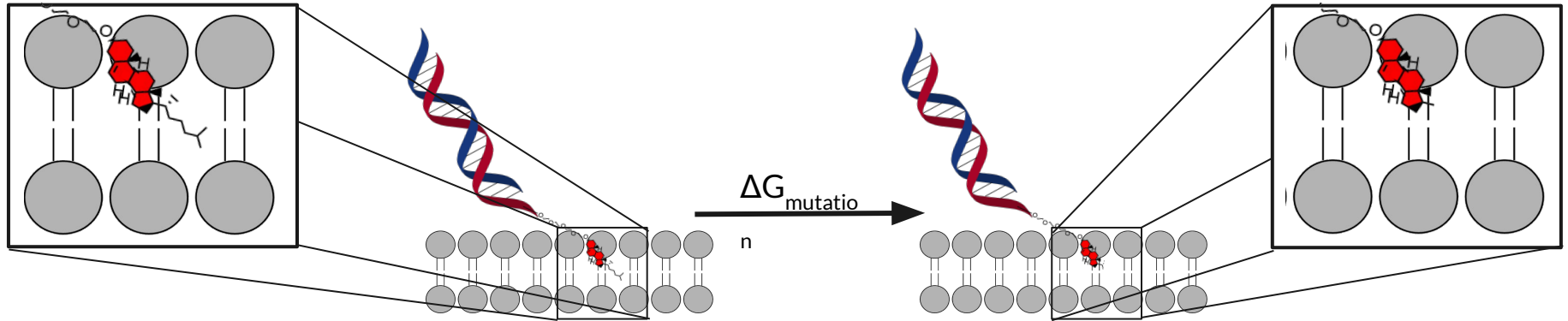


Issues:

1. Differences are smaller still
2. Sign switch near 0.5
3. Stubborn difference at 1

# Sterol is more stable in the membrane

$$\Delta G_{\text{mutation}} = -1.6 \pm 0.6 \text{ kcal/mol}$$



# Summary



## Applicability

We have demonstrated that free energy perturbation calculations converge (work) for these systems

## Cholesterol vs Sterol

We now know that a sterol-TEG is marginally more stable than chol-TEG in the membrane

Need to calculate  $\Delta G_{\text{solution}}$

## Improvements

Calculations still take several days. This can be improved.

## More Prosthetic Groups

Selection of more prosthetic groups is already underway

# Acknowledgements



## Advisors:

- Dr. Grace Brannigan      Primary Advisor
- Dr. Jinglin Fu      Secondary Advisor
- Dr. Julie Gripenburg      Collaborator
- Dr. Jerome Henin      Collaborator
- Dr. Tom Joseph      Collaborator

## Lab Members:

- Jesse Sandberg, MS
- Connor Pitman
- Jahmal Ennis

## Resources:

- Compute Resources: Office of Advanced Research Computing (OARC)
- Funding: Department of Defence (DoD)

**How do biological systems  
deliver metabolites across the  
body?**

—





**The biological approach (e.g. Oxygen Transport):**

dE

dE

dE

dE

dE

Electrostatics

Electrostatics

Electrostatics

Electrostatics

Electrostatics

Lorem 1

Lorem 2

Lorem 3

Lorem 4

Electrostatics

Electrostatics

Lorem 3

Lorem 4

Lorem 5

# Outline



Introduction:

Big-picture - moving and manipulating small groups of molecules

Biological inspiration - vesicles and associated proteins

Our toolbox - DNA, chemical modifications, and synthetic vesicles

Problem statement: optimizing prosthetic groups

Approach: Collect and organize candidates

Compare the free energies of solvation for various prosthetic groups

Energy components

Methods: Free energy perturbation (compare and contrast with LogP)

Results: Convergence and hysteresis?

# Applicability of FEP:

		Proposed	Tested	Published	
	0 Equations of state	✓	✓	1954	RW Zwanzig
	1 Calculation of partition coefficients ( $\Delta G_{\text{solvation}}$ )	✓	✓	✓	
★	2 Ligand Binding in Dilute Aqueous Solution ( $\Delta G_{\text{binding}}$ )	✓	✓	✓	★
	3 Ligand Binding in Heterogeneous, Non-Dilute Systems	✓	✓	2018	Salari et al.
	4 Patching on Membranes	✓	This work	✗	
	5 Insertion of Transmembrane Macromolecules	✓	Upcoming	✗	

# Ensuring Convergence and Reducing Error



Our “Best Practices” so far:

1. Sample frequently
  - a. ~every 16-20 fs
2. Use double-wide sampling
  - a. Simulate intermediate state N while sampling N-1 and N+1
3. Equilibrate thoroughly
  - a. Initially ~100ns
  - b. ~10ps for each window
4. Use long enough sampling times (~2ns)
5. Use small windows to ensure good overlap (~100 divisions between A and B)



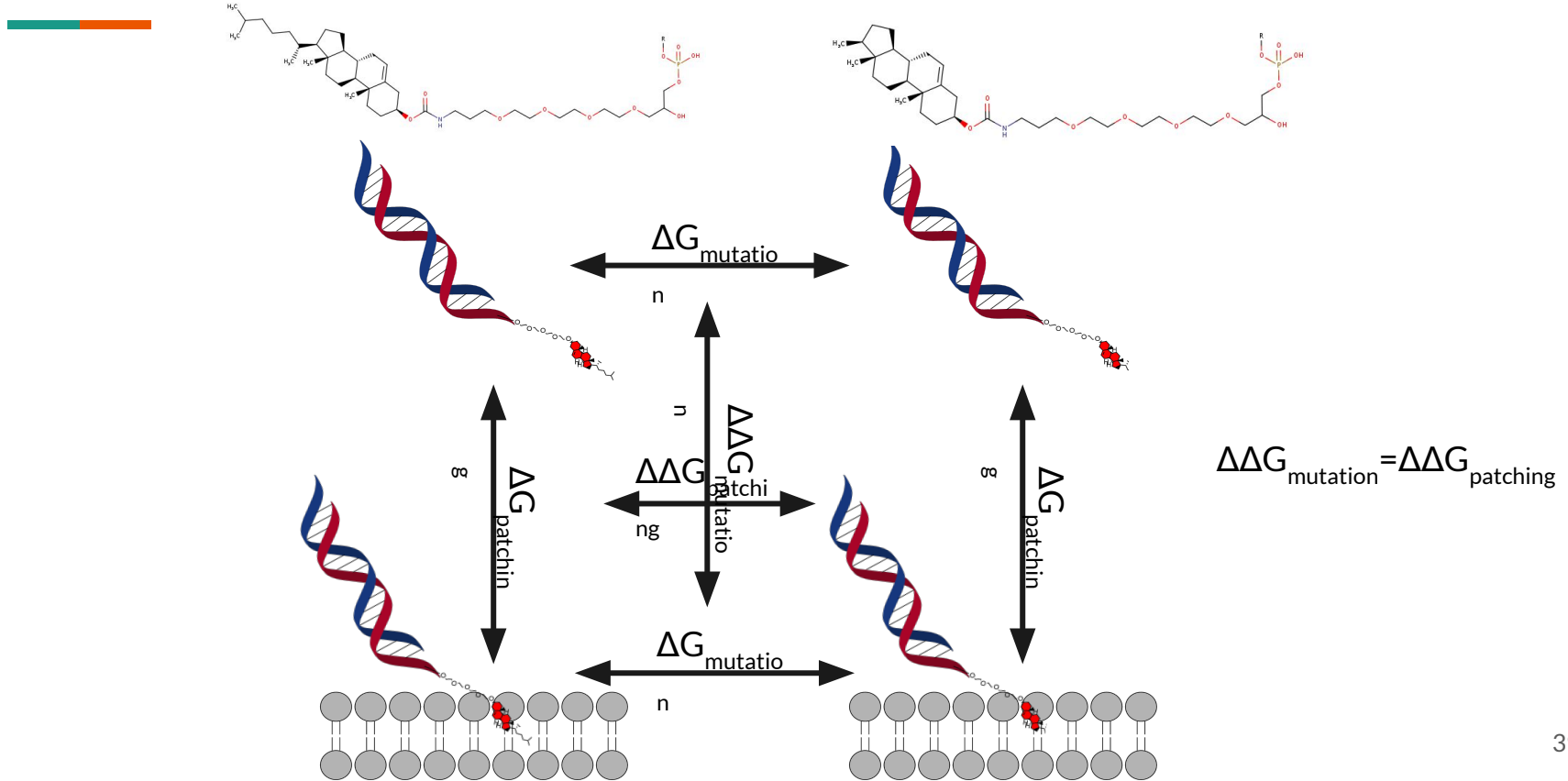
~~Add thermodynamic cycle~~

Add movie of mutation

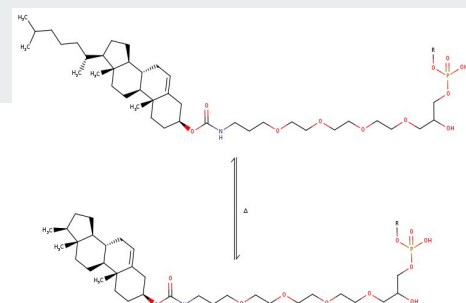
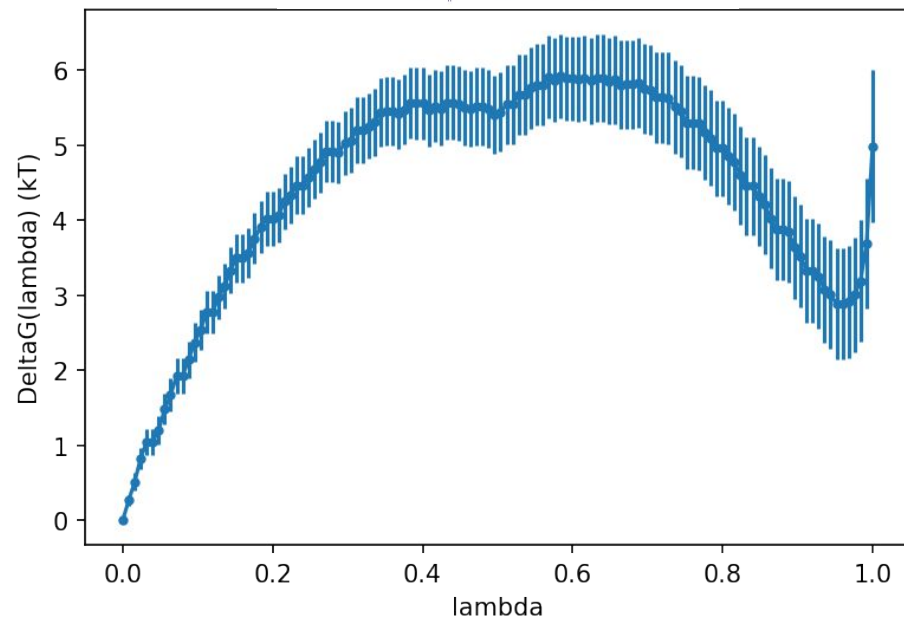
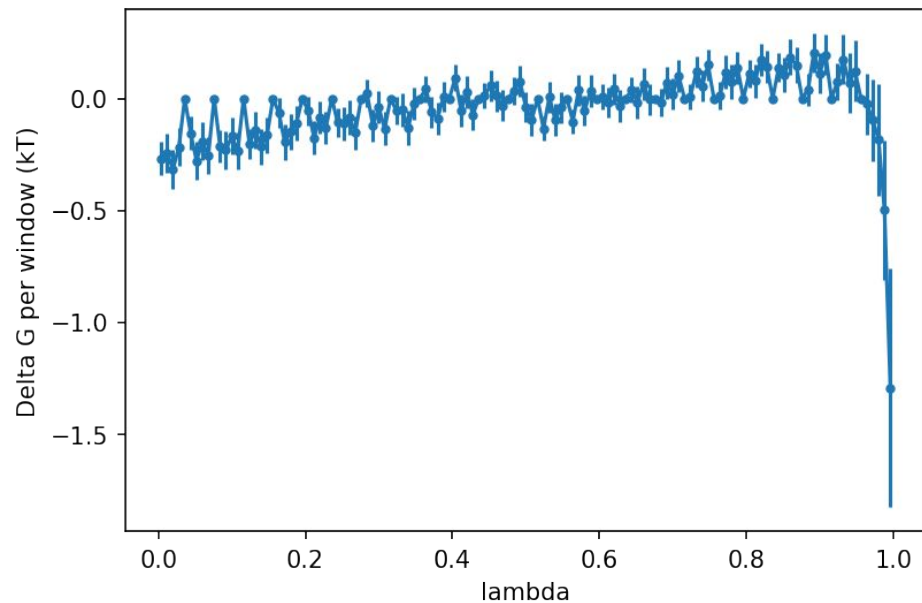
~~Add slide numbers~~

~~Define convergence~~

# Free Energy Comparisons

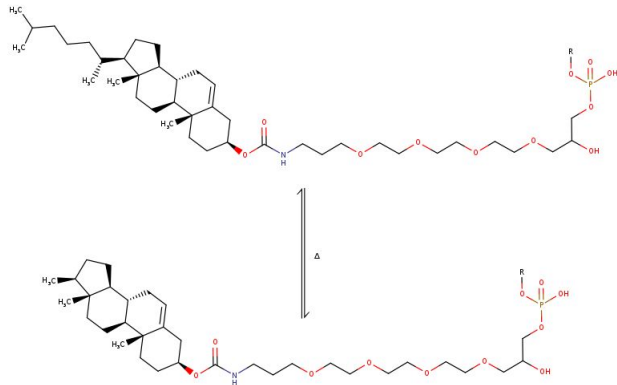


# Best estimate so far:





# Combining DNA and Membranes



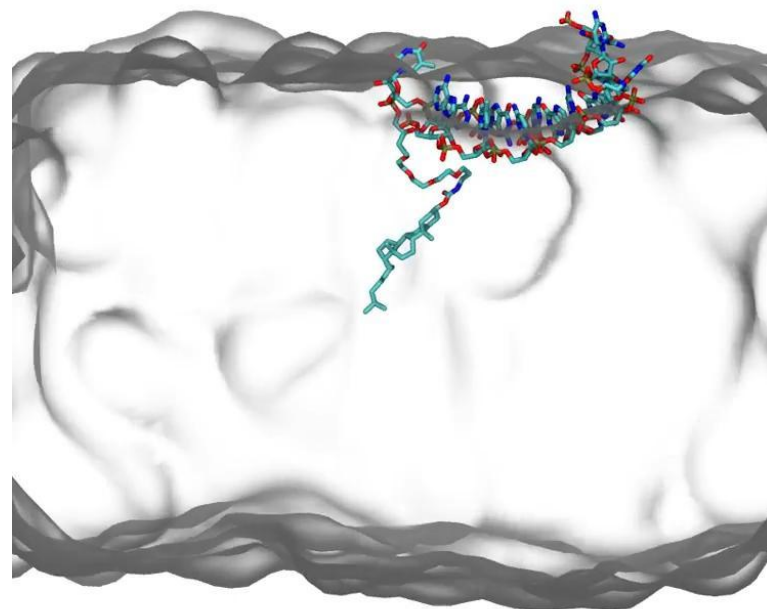
- Simulate in state A
- Periodically calculate the internal energy ( $E$ ) as if we were in state B
- Use the Boltzmann distribution to estimate  $dF$

$$\Delta F(\mathbf{A} \rightarrow \mathbf{B}) = F_{\mathbf{B}} - F_{\mathbf{A}}$$

$$\Delta F_{a \rightarrow b} = \sum_{\lambda_i=0}^1 \ln \langle \exp [-\beta [E_{\lambda_{i+1}} - E_{\lambda_i}]] \rangle_{\lambda}$$

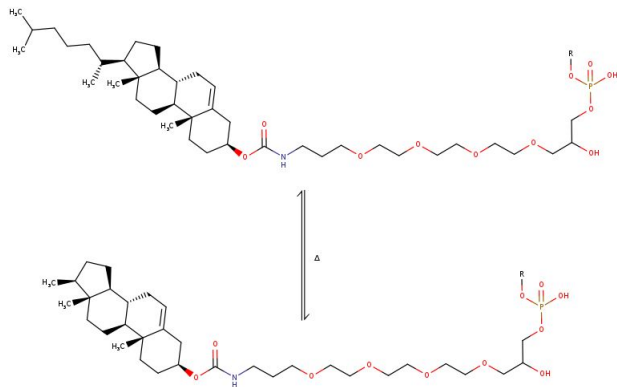
Refinement: decompose the tv  
intermediates:  $\{\lambda_0, \lambda_1, \lambda_2 \dots \lambda_N\}$

- Nitrogen
- Oxygen
- Carbon
- Phosphorous
- Lipids



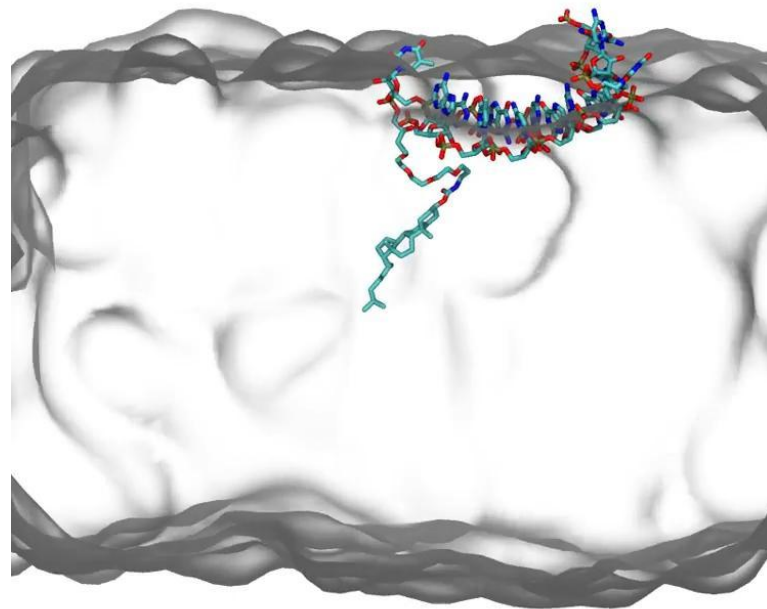
- Nitrogen
- Oxygen
- Carbon
- Phosphorous
- Lipids

## AFEP: a closer look



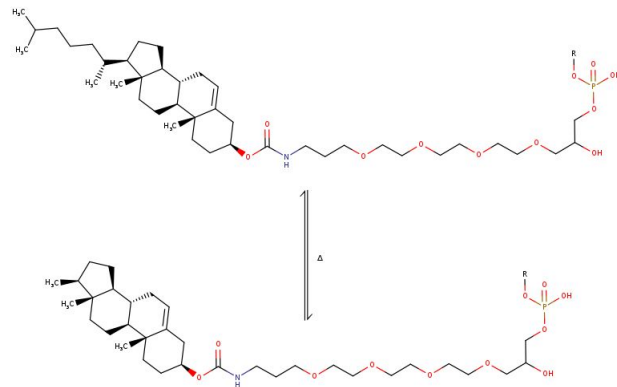
$$\Delta F(\mathbf{A} \rightarrow \mathbf{B}) = F_{\mathbf{B}} - F_{\mathbf{A}}$$

$$\Delta F_{a \rightarrow b} = \sum_{\lambda_i=0}^1 \ln \langle \exp [-\beta [E_{\lambda_{i+1}} - E_{\lambda_i}]] \rangle_{\lambda}$$



# Organization Paradigms

A) by size Maximum Common Structure:



B) by Tanimoto Distance:

Basic cheminformatics:

$$J(A, B) = \frac{|A \cap B|}{|A \cup B|} = \frac{|A \cap B|}{|A| + |B| - |A \cap B|}.$$

C) by MCS with tuning by CHARMM parameter distances:

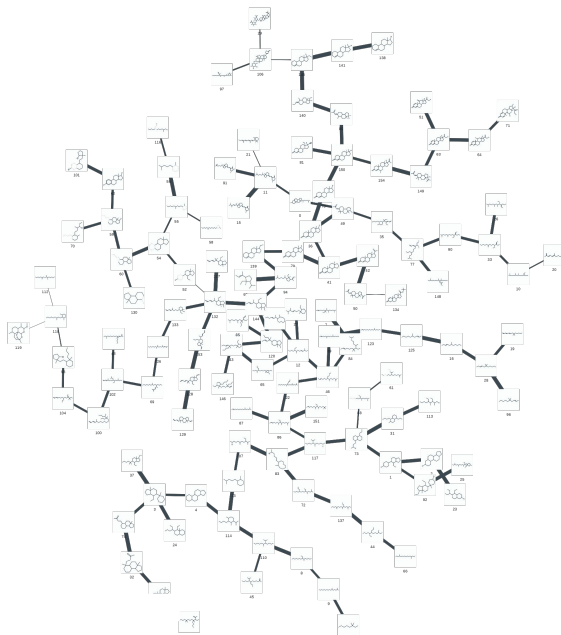
A, but including CHARMM parameters as additional data

# Approach

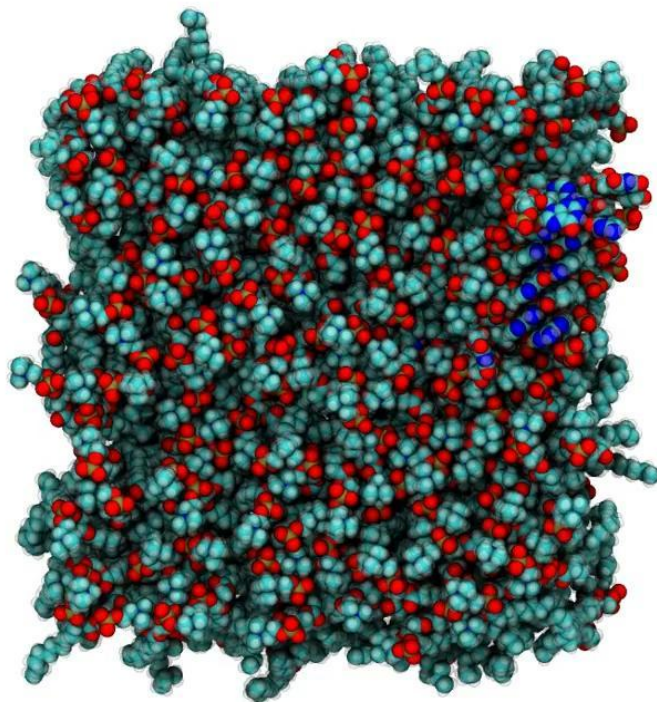
Organize candidate prosthetic groups



Compute the relative free energies of solvation/insertion for each



# Methods: Simulation



- Nitrogen
- Oxygen
- Carbon
- Phosphorous
- Lipids