Microscopic Models multiplicity of macrostates Entropy (equally probable fundamental assumption Two-State Paramagnet $\Omega(N) = 2^N$; $\Omega(N,N) = (M)$ L. Laws of probabilities Einstein Solid $\Omega(N,q) = \begin{pmatrix} q t N - 1 \\ q \end{pmatrix}$ $\rightarrow S = k \ln \Omega$ Monatomic Ideal Gas $-\Omega(V,U,V) = f(W)V^{N}U^{\frac{3N}{2}}$ there is an equilibrium state? U Statistical Mechanics
What equilibrium state is. Ω. Ω max S = S(U, V, N) $as = \frac{\partial s}{\partial u} du + \frac{\partial s}{\partial v} dv + \frac{\partial s}{\partial v} dv.$ Rates of Process: transport of particles

Transport Theory / Kinetics.

How long to reach equibibrium. as = - du + PalV - Halv. Governing Variable, TalV - Halv. $\frac{1}{D} = \begin{pmatrix} \frac{\partial 5}{\partial u} \end{pmatrix}_{v,v} \frac{P}{T} = \begin{pmatrix} \frac{\partial 5}{\partial v} \end{pmatrix}_{u,v} \frac{P}{T} = -\begin{pmatrix} \frac{\partial 5}{\partial v} \end{pmatrix}_{u,v}.$ Thermal Volume Partides. Quantity: Energy. What equilibrium state is.

(2) Classical Thermal dynamics. Mechanical Diffusive. Mechanical. Diffusive Type of = Thermal. $T = \left(\frac{\partial S}{\partial u}\right)^{-1} \longrightarrow \mu = -T\left(\frac{\partial S}{\partial v}\right)$ $C_{v} = \left(\frac{\partial u}{\partial T}\right)_{v}$ $S = k \ln \Omega \longrightarrow$ $\left(C_{p} = \left(\frac{\partial H}{\partial T} \right)_{p} = \left(\frac{\partial \left(U + PV \right)}{\partial T} \right)_{p} = \left(\frac{\partial U}{\partial T} \right)_{p} \left(\frac{\partial V}{\partial T} \right)_{p}$ 5(U, V, N.)= S H = U + PV

Microscopic Models I. Ideal Gas (Monatomic) $\Omega\left(N,U,V\right) = \frac{1}{N!} \frac{\sqrt{N}}{\left(h^{3}\right)^{N}} = \frac{1}{N!} \frac{\sqrt{N} A_{3N}(\sqrt{2m}U)}{\left(h^{3}\right)^{N}}$ $A_{3N}(J_{2mu}) = \frac{2\pi^{\frac{3N}{2}}}{\Gamma(\frac{3N}{2})}(J_{2mu})^{3N-1} = \frac{2\pi^{\frac{3N}{2}}}{(\frac{3N}{2}-1)!}(J_{2mu})^{3N-1}$ $Vadius = J_{2mu}:$ $Colimension = 3N: P_{1x}^{2} + P_{1y}^{2} + P_{1z}^{2} + \cdots + P_{nx}^{2} + P_{ny}^{2} + P_{nz}^{2} = 2mM.$ $(1) - \Omega \left(\frac{1}{N!} \frac{1}{N!}$ $\approx \left[\frac{V}{N} \cdot \left(\frac{4\pi mU}{3Nh^2}\right)^{\frac{3}{2}} \cdot e^{\frac{5}{2}}\right]^{N}$ Stirling's approximation (2) $S = k \ln \Omega = Nk \left[\ln \left(\frac{V}{N} \left(\frac{4\pi mu}{3Nk^2} \right)^{\frac{2}{2}} + \frac{5}{2} \right]$ $\star \cdot \Omega(N, U, V) = f(N) \cdot V^{N} \cdot U^{\frac{3N}{2}}$ $S = k \ln \Omega = Nk \ln V + \frac{3}{2} Nk \ln U + k \ln f(N)$ 2. $T = \left(\frac{\partial S}{\partial U}\right)^{-1} = \left(\frac{\frac{3}{2}Nk}{U}\right)^{-1} \Leftrightarrow U = \frac{3}{2}NkT$ 3. $P = T \left(\frac{\partial S}{\partial V} \right) = T \cdot \frac{\partial k}{\partial V} \iff PV = NkT$ $\mu = -T \frac{\partial S}{\partial N} = -Tk \ln \left(\frac{V}{N} \left(\frac{4\pi m U}{3Nh^2} \right)^{\frac{2}{2}} \right) = -Tk \ln \left(\frac{V}{N} \left(\frac{2\pi m k T}{h^2} \right)^{\frac{2}{2}} \right)$ $C_V = \left(\frac{\partial U}{\partial T}\right)_V = \frac{3}{2}N/k$ $C_{V} = \left(\frac{\partial H}{\partial T}\right)_{V} = \frac{1}{2}Nk.$ for ideal gas $C_{P} = \left(\frac{\partial H}{\partial T}\right)_{P} = \left(\frac{\partial H}{\partial T}\right)_{P} + P\left(\frac{\partial V}{\partial T}\right) = C_{V} + Nk. = \frac{5}{2}Nk$

II. Einstein Solid.

1.
$$\Omega$$
 (N,q) = $\binom{N+q-1}{q} = \binom{N+q-1}{q!(N-1)!} \approx \binom{q+N}{q!(N)!}$; $U = q \in \mathbb{R}$

(1) $\binom{q+N!}{q!N!} \approx \frac{\binom{q+N}{q}}{\binom{q+N}{N}} \approx \binom{\binom{q+N}{q}}{\binom{q+N}{N}} = \binom{\binom{q+N}{q}}{\binom{q+N}{q}} = \binom{\binom{q+N}{q}}{\binom{q+N}{q}}{\binom{q+N}{q}} = \binom{\binom{q+N}{q}}{\binom{q+N}{q}} = \binom{\binom{q+N}{q}}{\binom{q+N}{q}} = \binom{\binom{q+N}{q}}{\binom{q+N}{q}} = \binom{q+N}{q}}{\binom{q+N}{q}} = \binom{q+N}{q}$

1. N as a constant. $N = N \uparrow + N \downarrow . T$.

(1) $\Omega (N \uparrow) = (N \uparrow) = N \uparrow . N \downarrow . T$ (2) $\Omega (N \uparrow) \approx N | N \rangle - N \uparrow | N \rangle - N \uparrow | N \rangle$ energy II Two State Paramagnet * W= WB (NJ-M) = WB (N-2NT) M= | MZ| * $(M) = \mathcal{P}(N \uparrow - N \downarrow) = -\frac{\mathcal{L}}{B}$ megnetization For One particle (dipole). ($U = -\mu \cdot B = -\mu_z \cdot B = S = k \ln \Omega (N) = k V \ln N - \left(\frac{N - \frac{U}{\mu B}}{2} \right) \ln \left(\frac{N - \frac{U}{\mu B}}{2} \right) - \left(\frac{N + \frac{U}{\mu B}}{2} \right) \ln \left(\frac{N + \frac{U}{\mu B}}{2} \right)$ S = S(N,U) = S(N,-U)2. $T = \left(\frac{\partial S}{\partial U}\right)^{-1} = \left(\frac{1}{2\mu B} \ln \frac{N - \frac{u}{\mu B}}{N + \frac{u}{\mu B}}\right)^{-1}$ $U = N\mu B \left(\frac{1 - e^{-\frac{2\mu B}{kT}}}{1 + e^{\frac{2\mu B}{kT}}} \right) = -N\mu B \tanh \left(\frac{\mu B}{kT} \right) - \frac{1}{1 + e^{-\frac{2\mu B}{kT}}} = -\frac{1}{1 + e^{-\frac{2\mu B}{kT}}} = -\frac{1}{$ $M = -\frac{M}{B} = N_{\mu tomh} \left(\frac{\mu B}{ET} \right)$

3.
$$C_{R} = \left(\frac{3U}{2T}\right)_{N,R} = Nk \frac{f_{R}^{**}}{cosh(RR)}$$

Viandonness f

(100%)

Number 1

Regnetization \Rightarrow zero.

The sequence of freedom.

I. Monatomic Ideal Gas.

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(400, C_{N})

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2. Einstein Solid.

①
$$\Omega(M) = (N+q-1)$$
② $S = k \ln \Omega(M) = k \left[(q+v) \ln (q+v) - q \ln q - v m N \right]$
 $S = S(N, U)$, $U = QE$
③ $T = \left(\frac{2\omega}{2U} \right)^{-1} = \left[\frac{k}{E} \ln \left(\frac{U+EU}{U} \right) \right]^{-1}$, $U = \frac{NE}{e^{\frac{2\omega}{E}} - 1} = \frac{NE$