The transition temperature of **T(meas**) tabulated value. mple holder were compensated using a For subsequent heating cycles the t The reason for the observed shift in tr BaCO₃ powder during the first heating

Fig. 4

(m#1-3).

intensity at ≈23.4 °2θ

to trigonal(II) phase at T(Lit)=129±1°C ms at 120 ℃ before conversion back to phase(II) phase(III) phase(I) phase(II) #2

displayed sample temperature (deg C)

reen curve shows the transition of

(III) for a second cooling cycle.

ase(II).

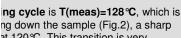
e changes when the phase transition is

t Pro diffractometer and Anton Paar

se formation during cooling, starting with

Fig. 6

100 110 120 130



at 120 °C. This transition is very) proceeds slowly and is not complete. e at which the transition (I)⇒(II) takes ၀ melting point 3 T(literature) 2 1 -1 -2 -3 100 200 500 displayed sample temperature (de

Calibration curve with phase transitions

Calibration of the temperature measure

for an XRK900 heating attachment.

Atmosphere in the chamber: air, 1 bar

shows the calibration curve obtained f phase transition

Transition of BaCO₃ from phase(I)⇒phas samples (s#1,s#2) for repeated heating of

The Graph shows the decrease of the phas

Because of the shown change in trans and its destructive effect on platinum

Temperature calibration w

As an example for a calibration with p

XRD temperature calibration.