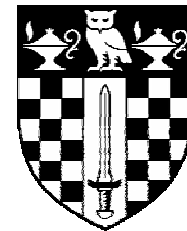




Supplementary Lecture at
10th Chemical Crystallography Group School,
Trevelyan College, University of Durham,
April 2005



An Introduction to Neutrons & Neutron Diffraction Experiments

(with a slight bias towards powder diffraction ☺)

Jeremy Karl Cockcroft

Industrial Materials Group

Department of Chemistry, University College London
and School of Crystallography, Birkbeck College,
University of London

Overview

- Properties of the Neutron
- Reactors Neutron Sources
 - Generation of Neutrons
 - Neutron Instrumentation
- Pulsed Neutron Sources
 - Generation of Neutrons
 - Neutron Instrumentation
- Neutron Experiments

Properties of the Neutron

- Particle with:

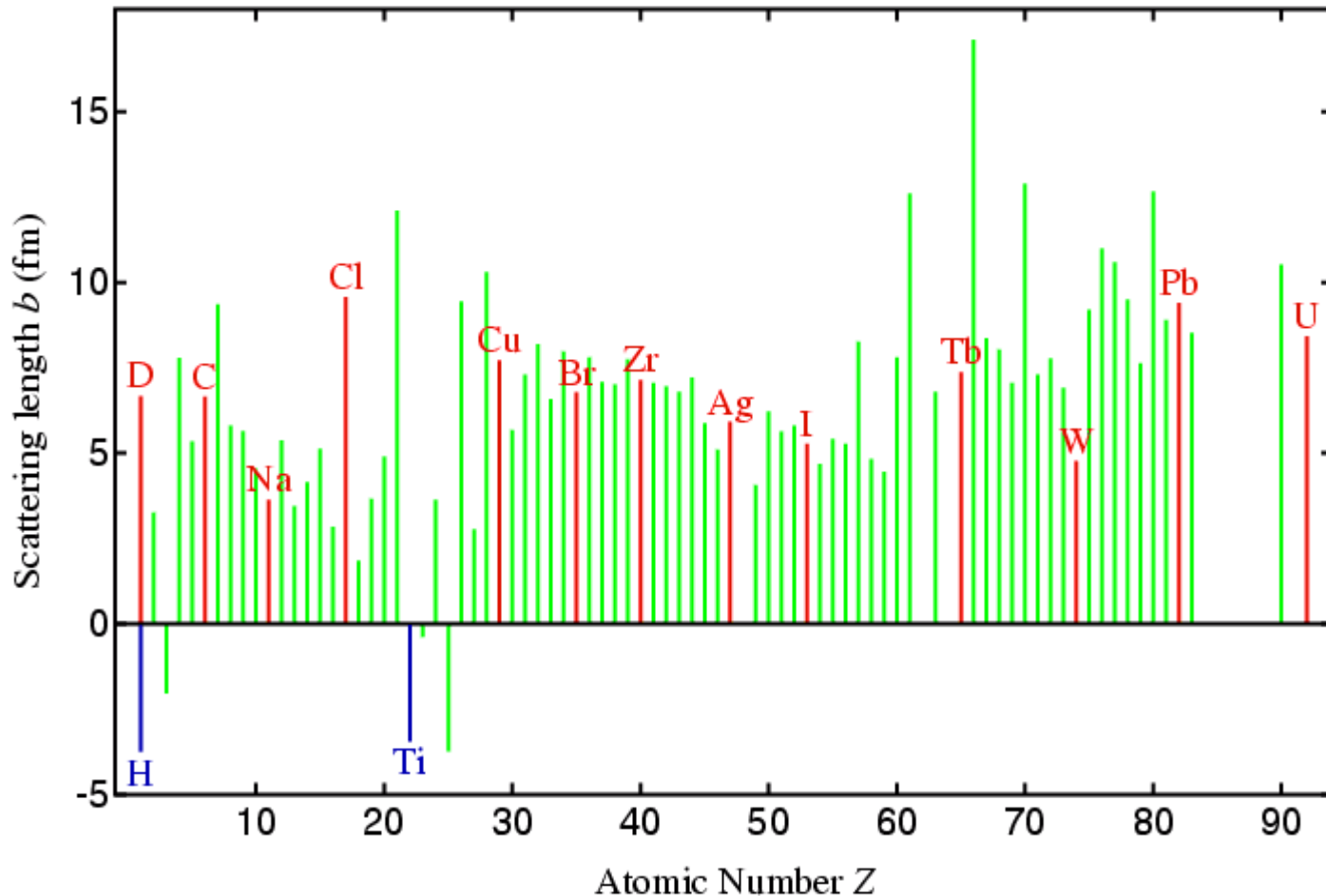
- Mass 1, Charge 0, Spin $\frac{1}{2}$
 - Strong interaction with nucleus
 - Weaker interaction with unpaired electrons
 - » Magnetic crystallography
- Kinetic Energy given by $E = \frac{1}{2}mv^2 = k_B T$
 - k_B is Boltzmann constant = $1.381 \times 10^{-23} \text{ JK}^{-1}$
 - RT = 20°C (293 K) $\rightarrow v = 2200 \text{ ms}^{-1}$
 - de Broglie relationship $\lambda = h / mv \rightarrow \lambda = 1.8 \text{ Å}$
 - » Thermal neutrons
 - » Properties often quoted for this wavelength

Scattering Processes

- Coherent Scattering Cross-Section, σ_c
 - Diffraction
 - Scattering length, b , where $\sigma_c = 4\pi b^2$
 - Units: b in fm (10^{-13} cm), σ in barns (10^{-24} cm²)
- Incoherent Scattering, σ_i
 - Background
- Absorption, σ_a
 - Loss of intensity
 - Radioactive samples & γ background

Neutron Scattering-Lengths

- Dependant on the Nucleus Type



Neutron Scattering-Lengths

Element	H	D	U	Fe	Co
X-ray $f(0^\circ)$ (Z)	1	1	92	26	27
Neutron b (fm)	-3.74	6.67	8.42	9.45	2.78

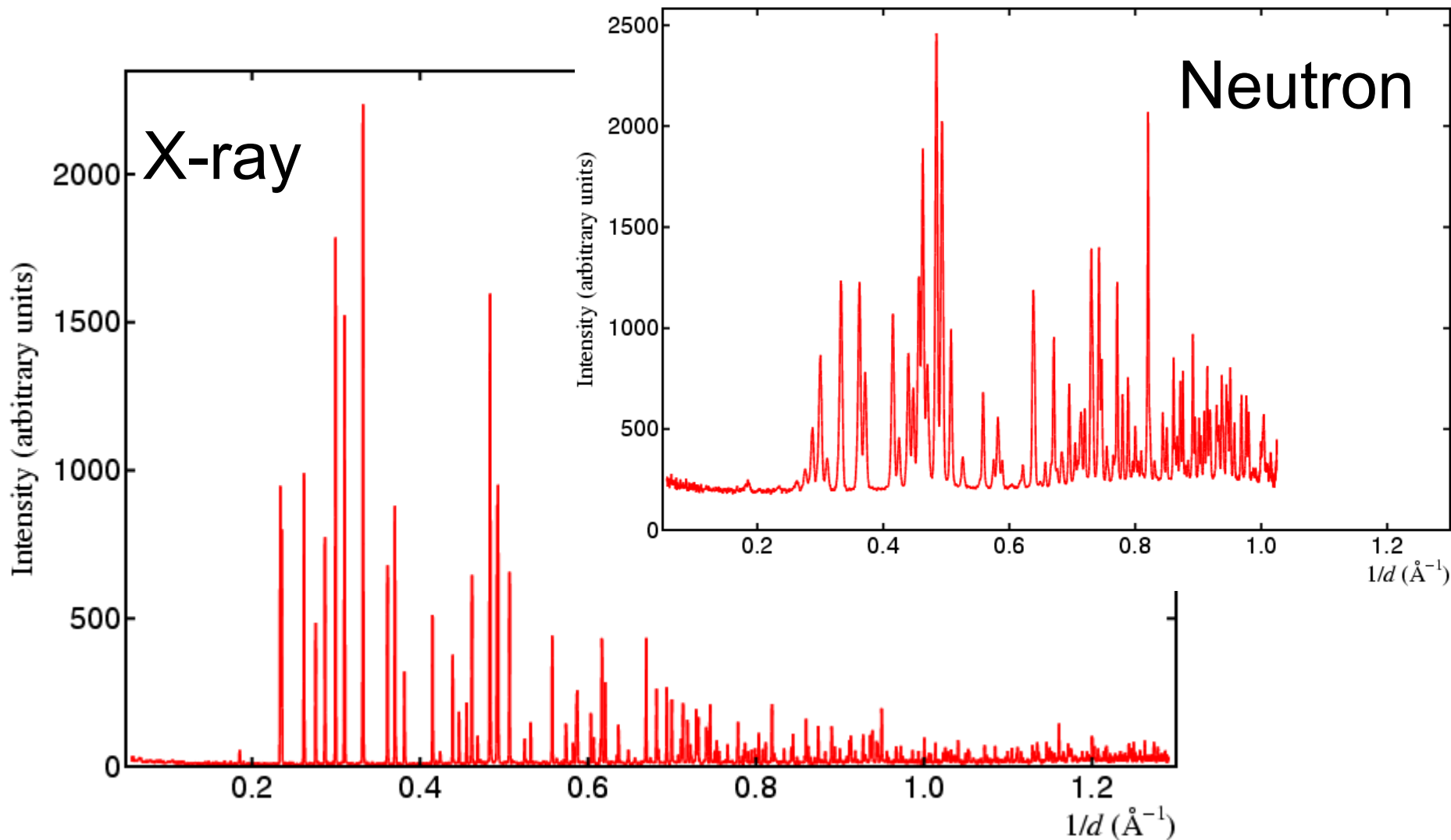
Element	Ba	O	V	Ti	Zr
X-ray $f(0^\circ)$ (Z)	56	8	23	22	40
Neutron b (fm)	5.28	5.81	-0.38	-3.44	5.28

Absolute comparison: $f(0^\circ) \times r_e$
where r_e classical electron radius = 2.818 fm

Scattering as a Function of 2θ

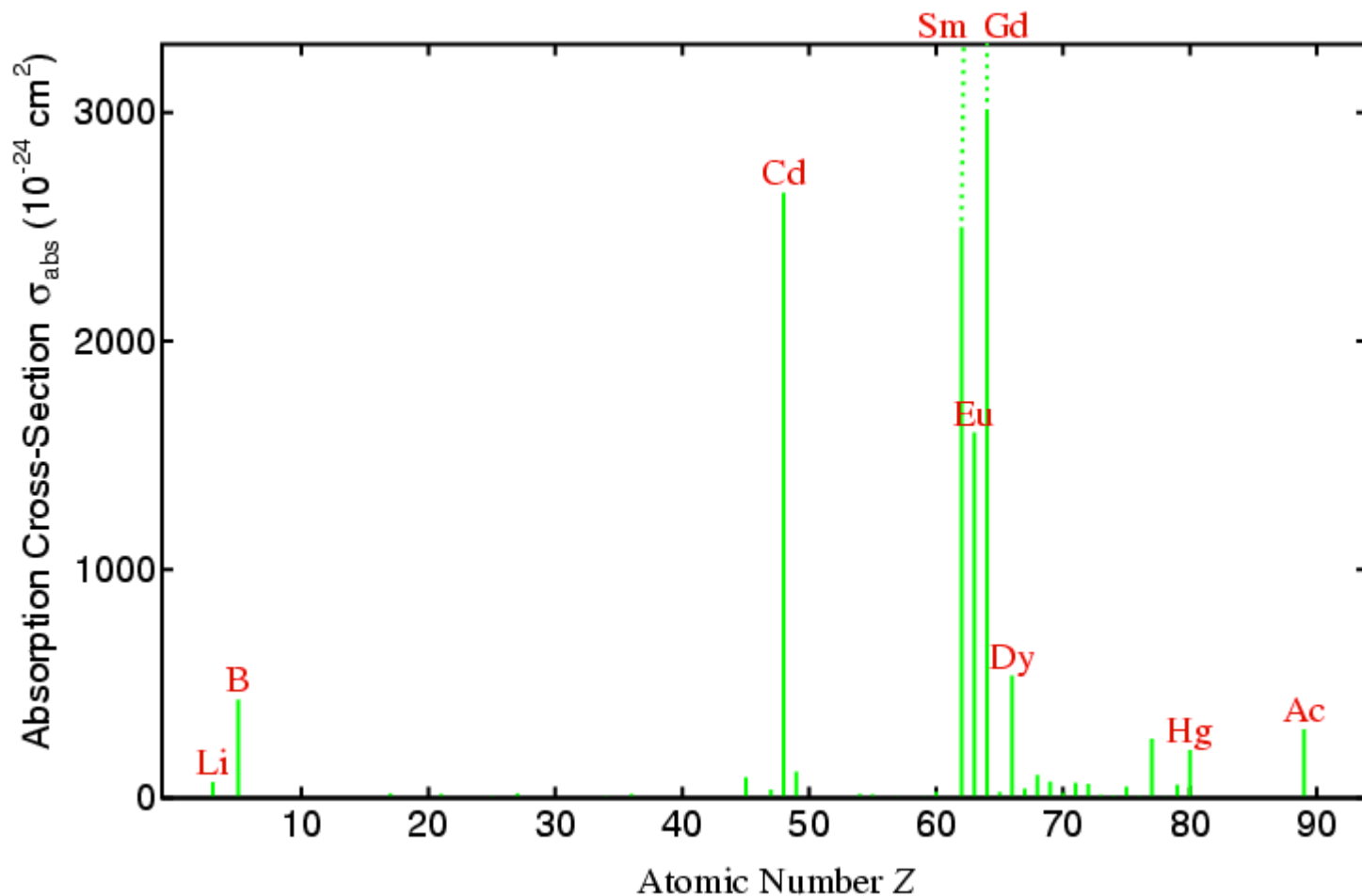
- X-ray interacts with electron cloud
 - Similar size to wavelength
 - Fourier transform of the electron cloud is the X-ray form factor (F.T. Gaussian \rightarrow Gaussian)
- Neutron interacts with nucleus
 - Very small (point scatterer) compared to λ
 - Fourier transform of a delta function is a horizontal line \rightarrow no neutron form factor
 - Interaction with unpaired electron(s)
 - Magnetic form factor

Scattering as a Function of 2θ



Absorption Cross-Section

- Relevant for Li, B, Cd plus some Lanthanides

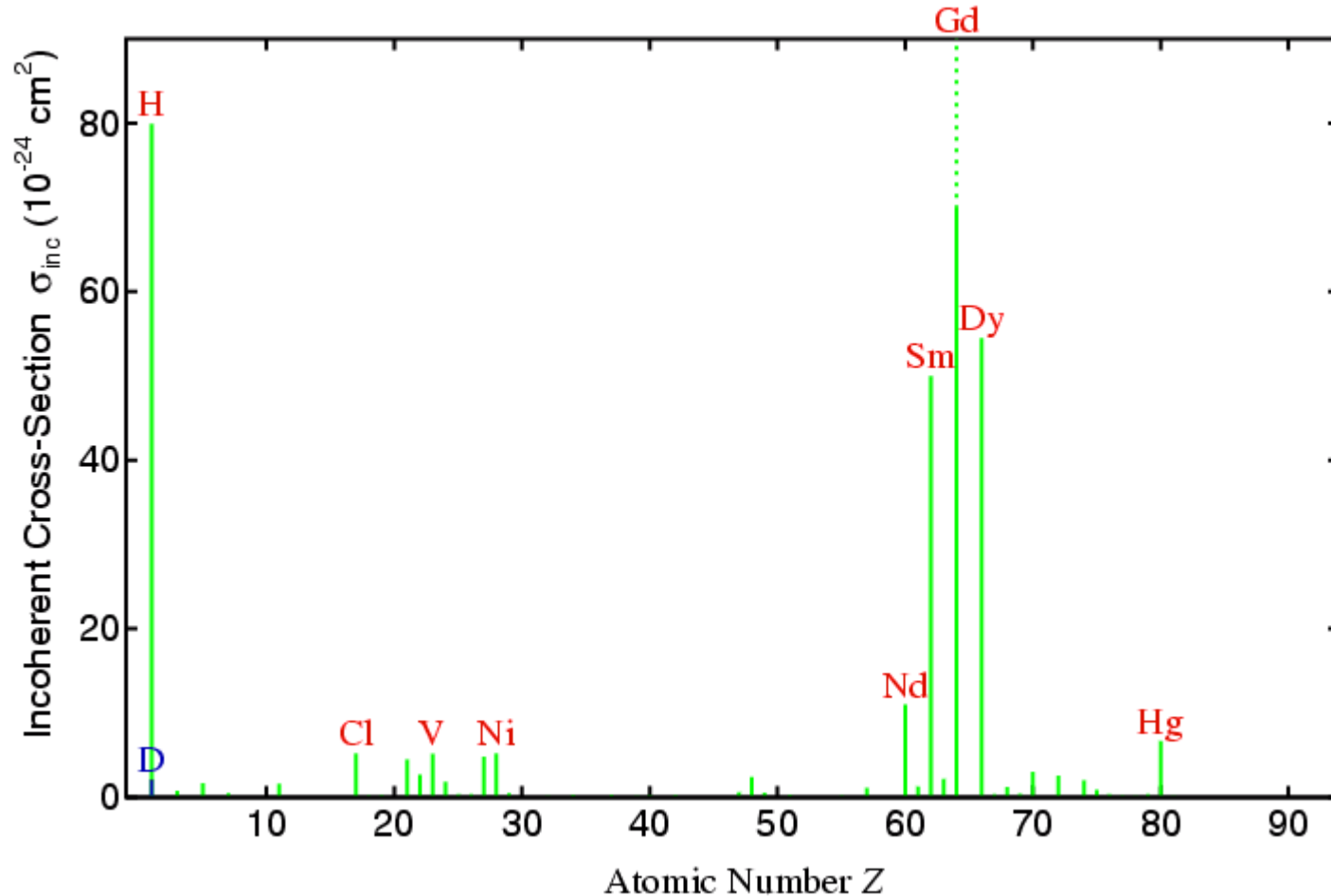


Neutron Shielding

- X-rays
 - Use Pb or thick Steel
- Neutrons
 - Use B₄C doped rubber or plastics
 - Use Cd sheet metal
 - Pliable at 1mm thickness
 - Use Gd₂O₃ paints (Soller collimators)
 - ⁶Li now rarely used
 - Atomic strategic material required for military use

Incoherent Cross-Section

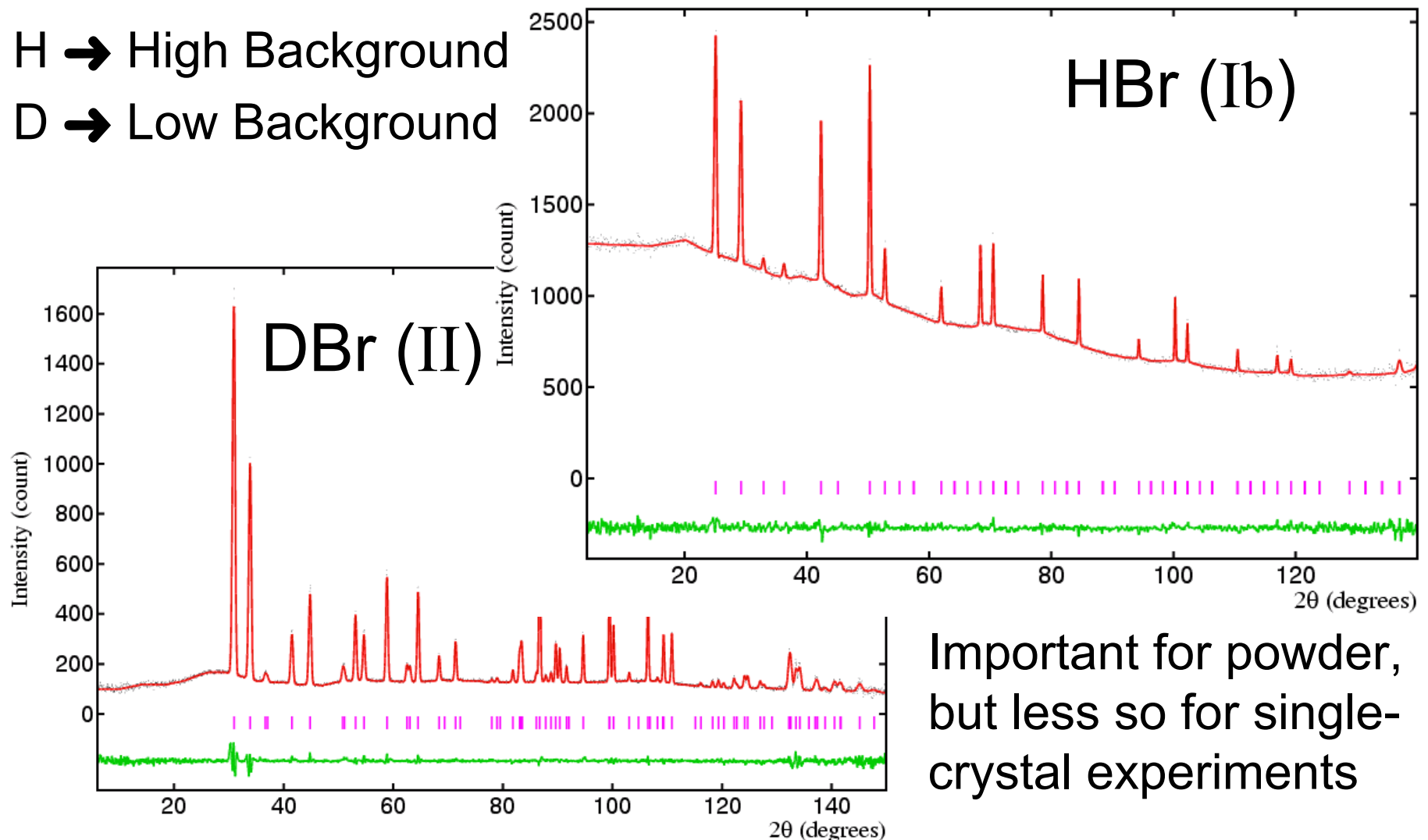
- Highly relevant for H plus some Lanthanides



Effect of Incoherent X-Section

H → High Background

D → Low Background



Important for powder,
but less so for single-
crystal experiments

σ Comparisons

Element	σ_c	σ_i	σ_a
H	1.76	80.3	0.333
D	5.59	2.05	0.000519
V	0.0184	5.08	2.80*
Gd	29.3	150	49700
Al	1.495	0.0082	0.231

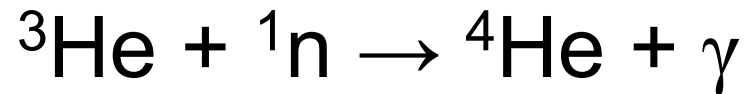
No need for special windows as aluminium can be used

Neutron Detectors

- Neutrons have no charge & non-ionizing
 - Less easy to detect than X-rays
- Detection relies on absorption by atom nucleus with simultaneous emission of γ -ray
 - Referred to as (n,γ) reaction
- Most common are proportional gas type
 - Absorbing material must absorb neutrons and be capable of existing in gaseous form
 - Limited choice of materials

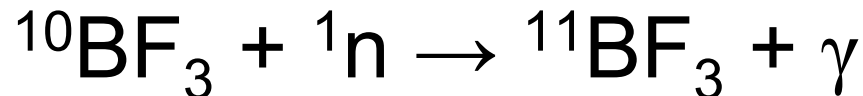
Neutron Detectors

- Most common material is ^3He gas



- Very expensive gas!

- Alternative is BF_3



- Better (lower background)

- Toxic and disliked by physicists

- Now rarely used

- Gas pressurised (5-10 bar)

- ➔ improves efficiency

Neutron Detectors

- Solid State Detectors

- Use (n, γ) reaction, but γ -ray converted to visible photon
 - Measure with photomultiplier tube
 - Early versions very sensitive to γ background
- Use lithium salts or rare-earth glasses
 - e.g. $\text{Gd}_2\text{O}_3 \cdot \text{Gd}_2\text{S}_3$
- Advantage is high efficiency
 - Very thin
 - ➔ important for Time-of-Flight measurements

Neutron Monitors

- Required to Monitor Incident Flux
 - No X-ray laboratory equivalent
 - X-ray tube assumed to produce constant flux
- Inefficient Detectors
 - e.g. 99.9% ^4He with 0.1% ^3He
 - Dilution optimised for particular wavelength & flux

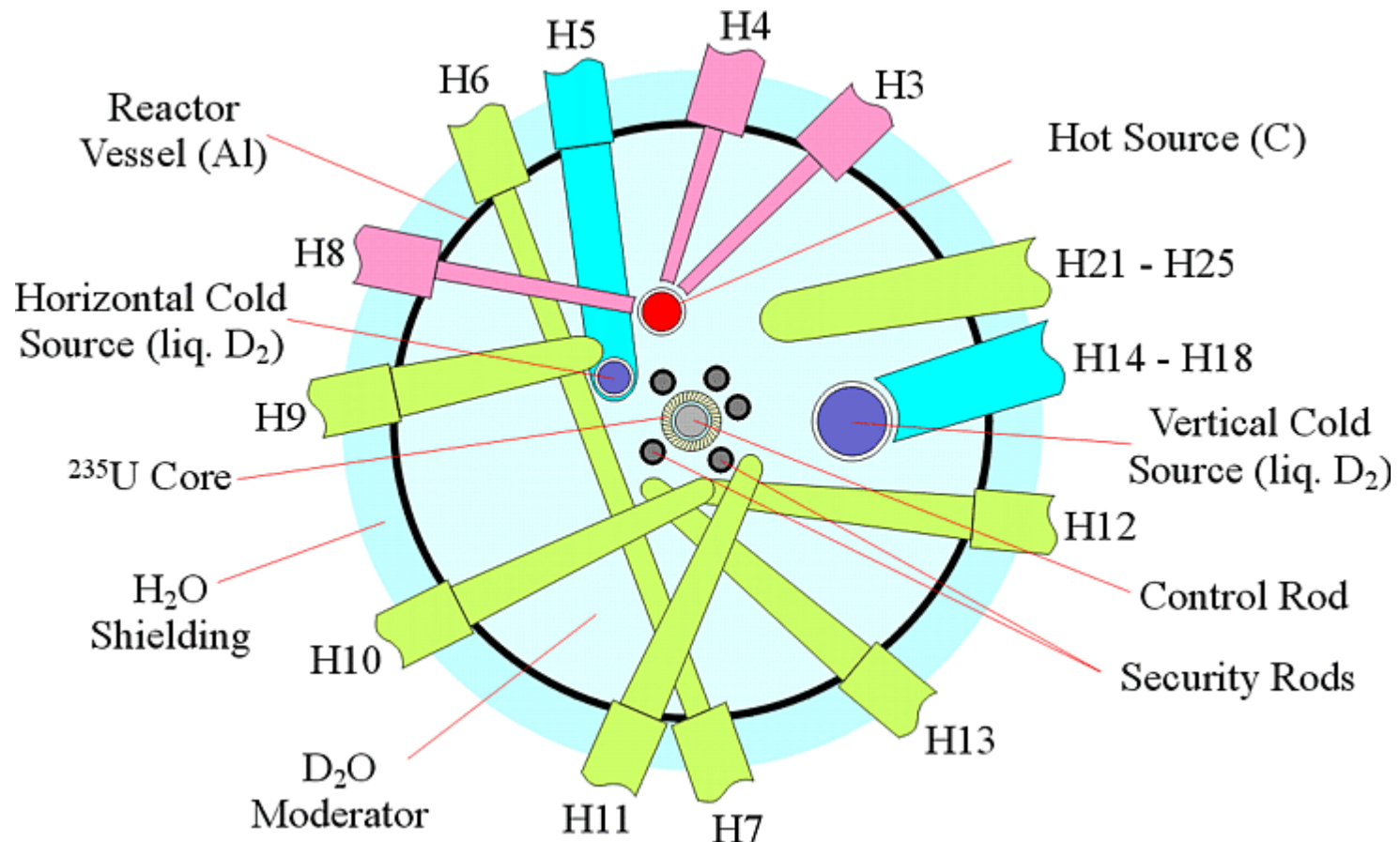
Reactor Neutron Sources

- ILL, Grenoble
 - Still World's Highest Flux Neutron Source
 - Other European sources are much lower flux



Centre of the Reactor

- 10 kg of ^{235}U heavy water cooled at 35°C



Light Water Biological Shield

- Cerenkov Radiation



(Photo supplied by ILL Publicity Office)

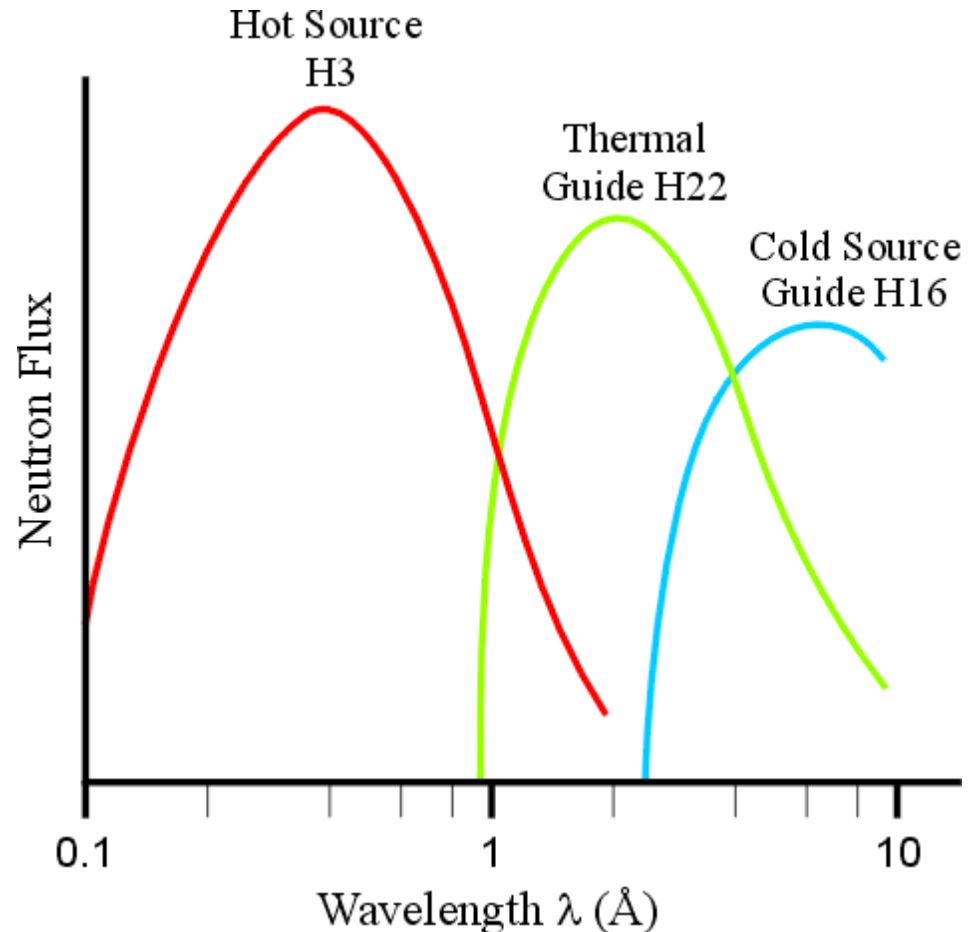
Hot and Cold Neutrons

- Cold Source

- Liquid H₂ fridge
- Long λ neutrons
 - SAS

- Hot Source

- Solid graphite
 - Self-heated
- Short λ neutrons
 - Small d spacings



Neutron Guide Halls

- Limited Space around Reactor Core
- Transmit Neutrons along curved Guides
 - Total internal reflection (especially long λ)
 - Removes γ -radiation (straight lines)



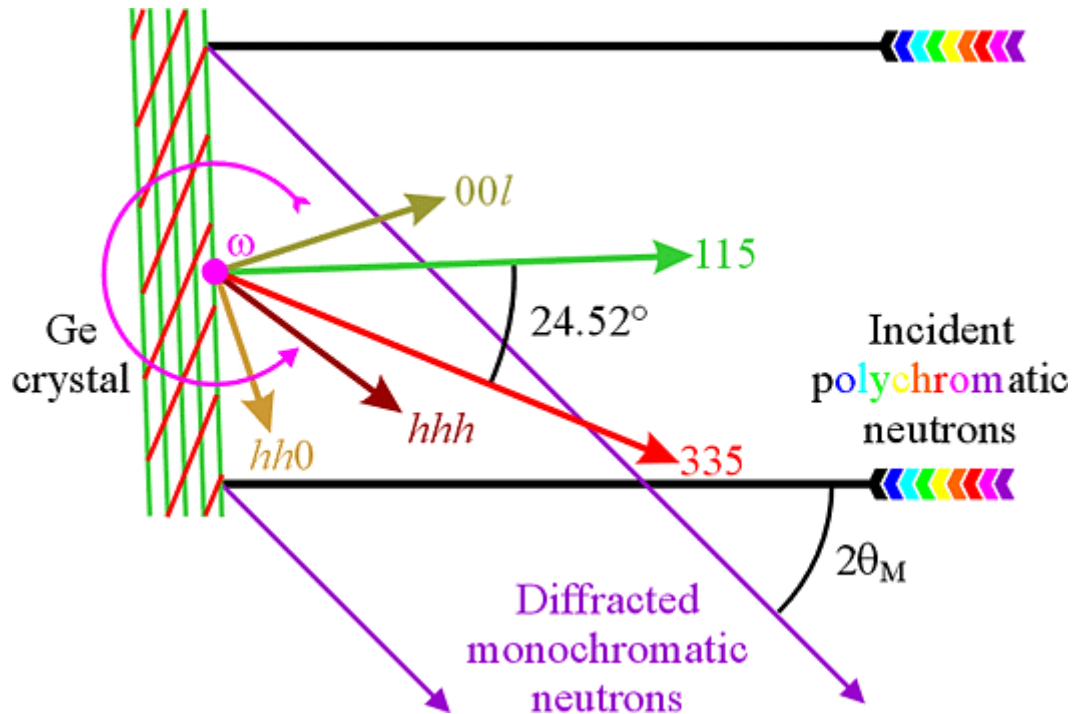
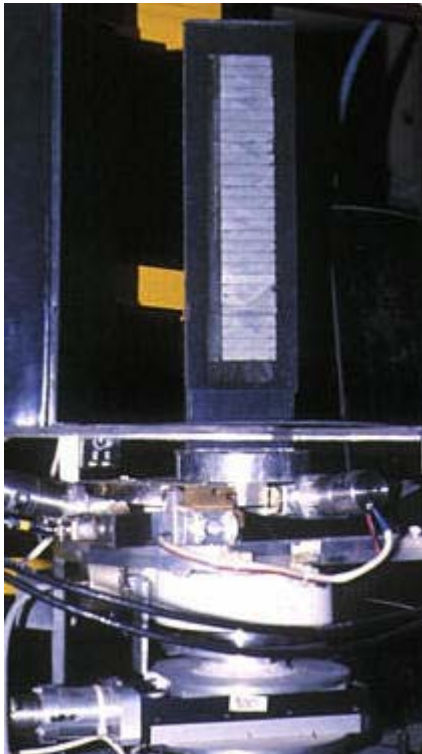
Neutron Guides

- Nickel coated glass
 - Isotopic ^{58}Ni ($b = 14.4$ fm) improves reflectivity
 - OK for **thermal** neutrons
 - Excellent for **cold** neutrons
 - Unsuitable for **hot** neutrons



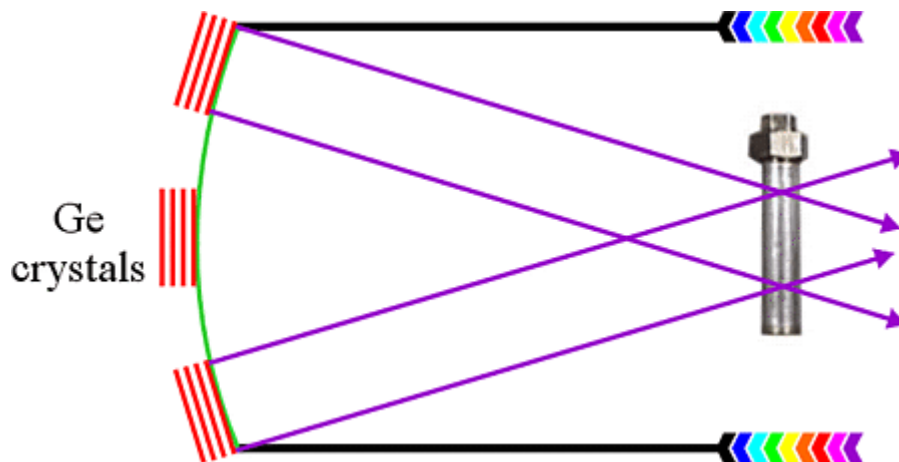
Monochromators

- White Beam with no Time Structure
 - Require monochromators (Ge, C, Cu) for single λ
 - Single bounce – Rotate to obtain different d spacings



Focussing

- Relatively Low Flux at Sample
 - Use vertically-focussing monochromators
 - Increase angular divergence
 - Lower resolution



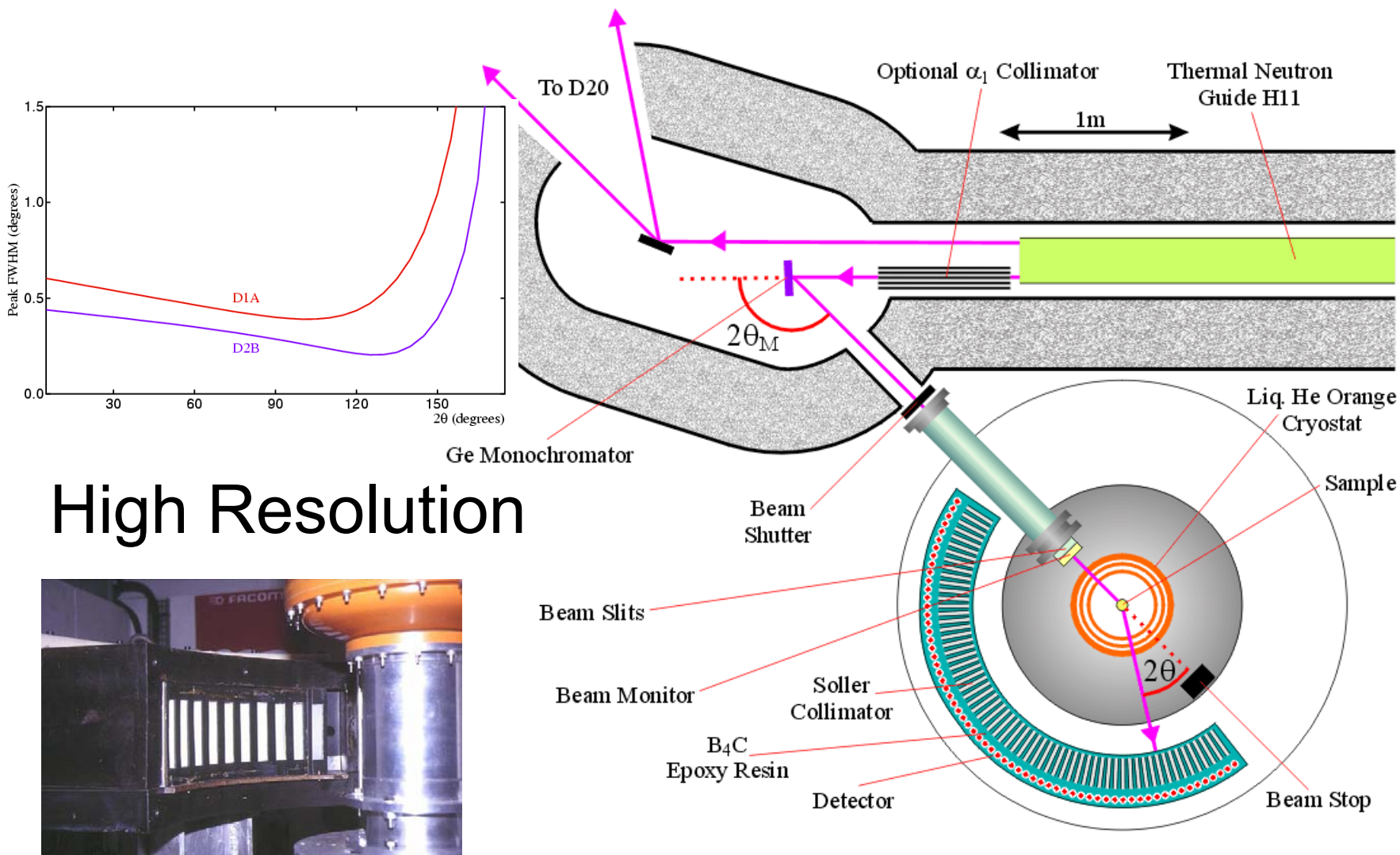
Instrumentation

- Single Crystal Diffraction at ILL
 - D9 (**Hot**)
 - D19 & Vivaldi (**Thermal**)
 - D10 & D15 (**Thermal**, plus additional features)
 - D16 (**Cold**, plus powder diffraction)
- Powder Diffraction at ILL
 - D2B, D1A (**Thermal**, high resolution)
 - D20, D1B (**Thermal**, high flux)

Single-Crystal Neutron

- Comparison with X-rays
 - Relative large crystals ($10\text{s mm}^3 - 1\text{ mm}^3$)
 - Use χ -circles
 - Modernised instruments have big detectors
 - PSD or image plate
 - Older ones have point detectors
 - Some experiments require manual setting crystal orientation
 - Understanding *UB* orientation matrices helps
 - Some data sets still collected one reflection at a time
 - Data acquisition times are long

Powder Diffraction



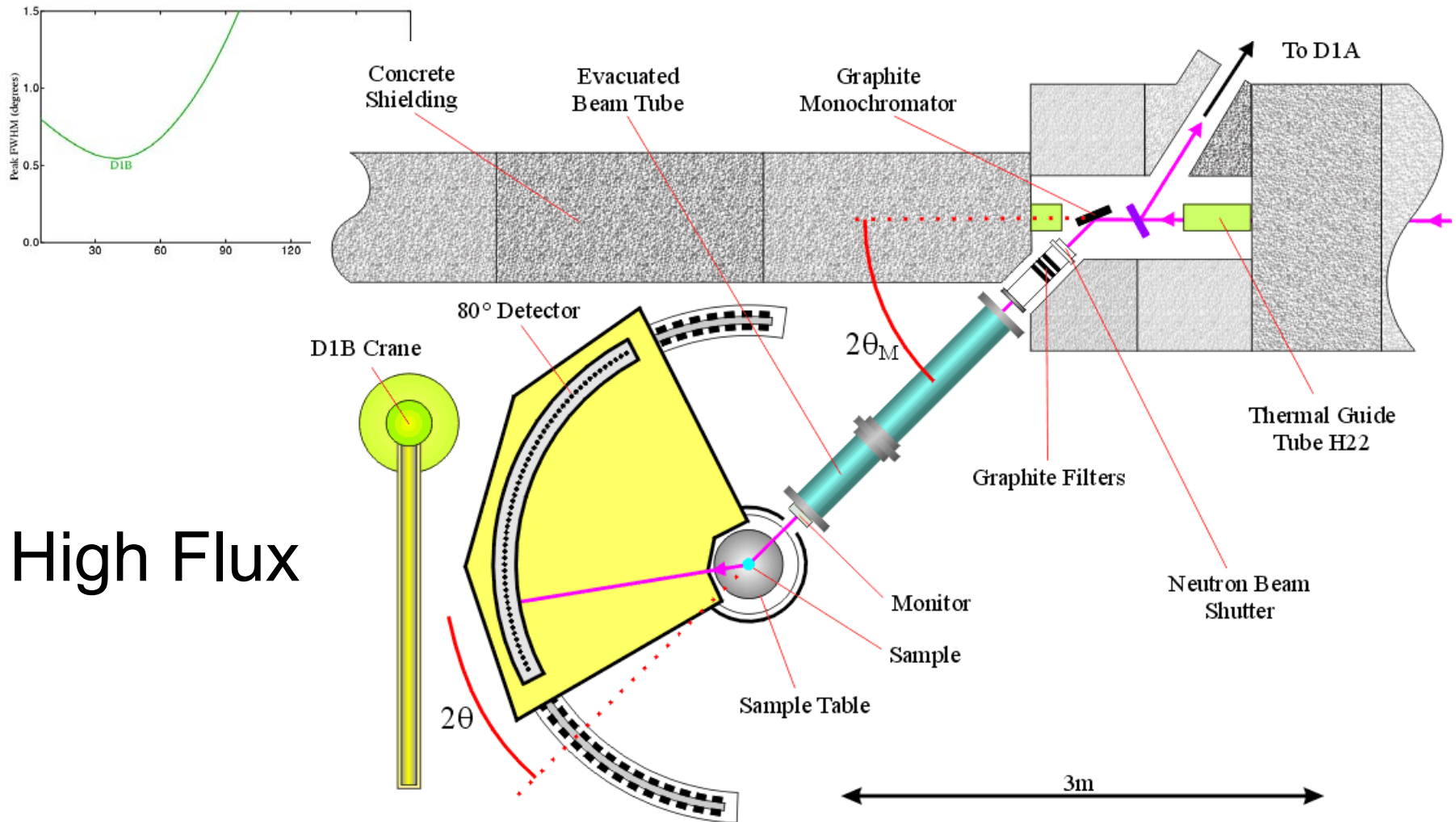
High Resolution

Improving Statistics

- 1st Generation
 - (1960s to early 1970s)
 - Single Point Detectors
- 2nd Generation
 - (Late 1970s to early 1990s)
 - Multiple Detectors
- 3rd Generation
 - (Late 1990s to present day)
 - Sophisticated PSDs
 - Delay line ^3He tubes



Powder Diffraction



High Flux

View of Old High Flux P.D.

- Instrument with the ILL's best publication record these last 5 years!
 - Now a CRG instrument



Sample Holders

- Vanadium Cans or Quartz (not Glass) Tubes
- Relatively Large Quantities of Powder



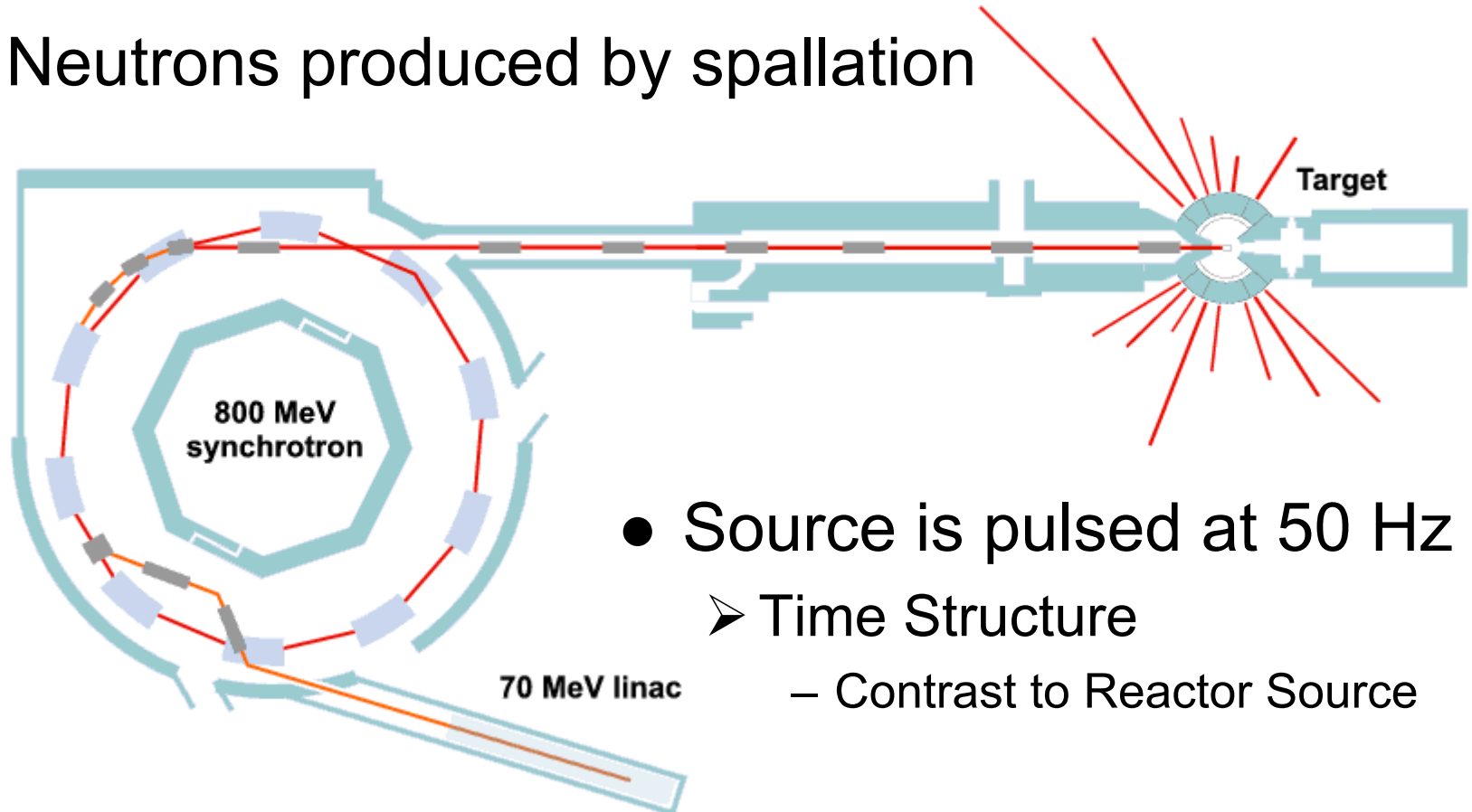
Pulsed Neutron Sources

- Pulsed Neutron Sources share some similarities to Synchrotron Sources
 - Both accelerate particles in a ring
- Use Charged Heavy Particles e.g. H^- ions
 - Accelerate in a linear accelerator (linac)
- Strip Electrons to produce H^+ beam
- Accelerate to High Energy in Storage Ring
- Collide with Heavy Metal Target (water cooled)
 - Ta, U, W
 - Produce smaller nuclei plus neutrons

Pulsed Neutron Sources

- ISIS World's Most Intense PNS

- Neutrons produced by spallation



- Source is pulsed at 50 Hz

- Time Structure

- Contrast to Reactor Source

TOF Concepts

- Essential to Use Time Structure of Pulsed Neutrons
 - Time-averaged flux is relatively low
- de Broglie equation $\lambda = h / mv$
 - Planck's constant $h = 6.626 \times 10^{-34}$ Js
 - Neutron mass $m = 1.675 \times 10^{-27}$ kg
 - Velocity, v , is inversely proportional to λ
 - In 1 ms a neutron will travel a few metres
 - Time over a fixed path length, L , to obtain λ

TOF Concepts

- Combine de Broglie and Bragg Equations:

$$\lambda = ht / mL = 2d \sin\theta$$

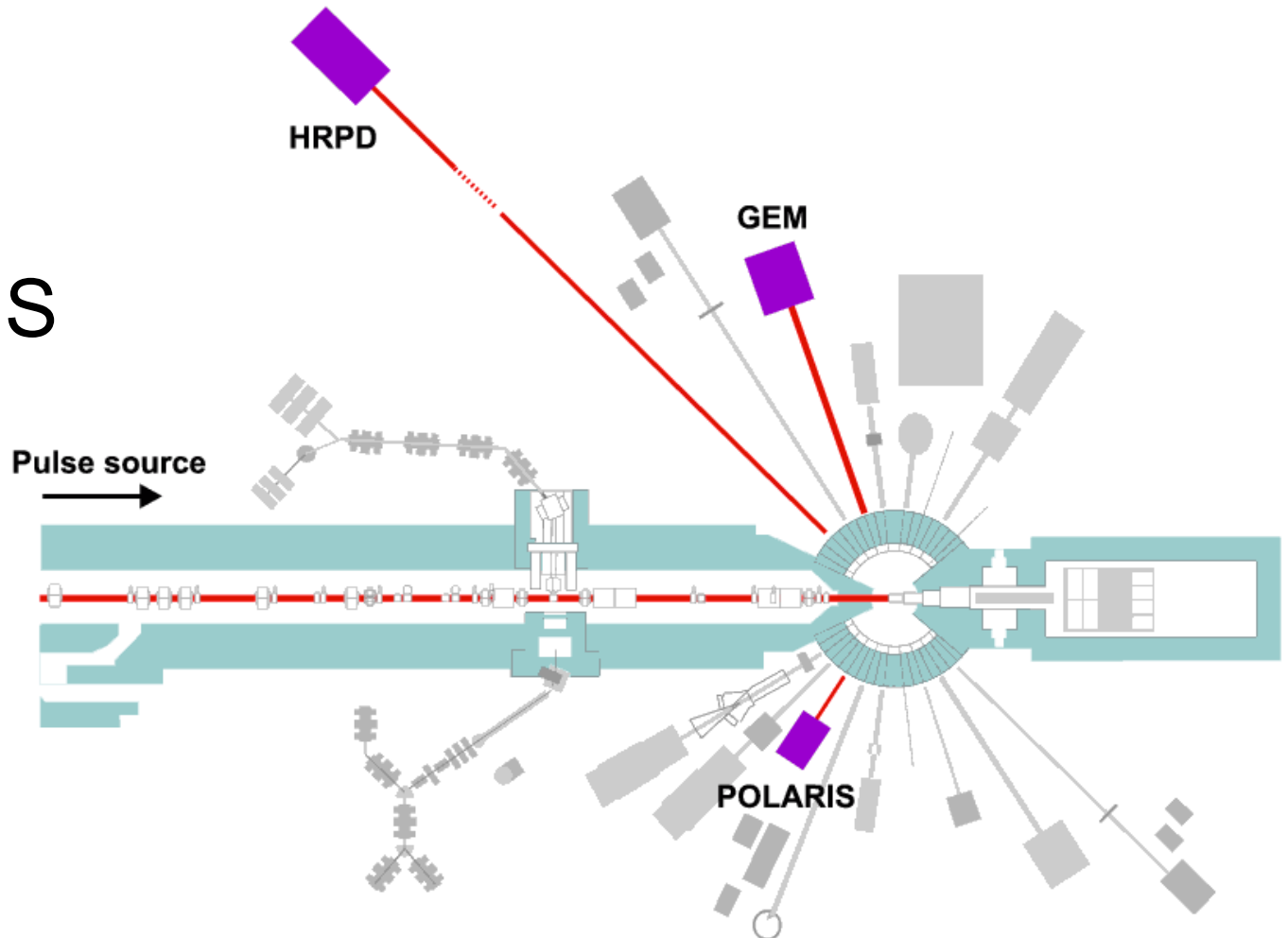
- d spacing is proportional to time, t
- Timing is between start of the pulse and arrival time on the detector
 - “Time-of-Flight” (TOF)
 - Need thin detectors to reduce uncertainty in path length

TOF Single Crystal

- SXD
 - Measure all of reciprocal space
 - Bragg peaks *plus* space between peaks
 - Useful for some experiments
e.g. incommensurate phase transitions

TOF Powder Instruments

- HRPD
- GEM
- POLARIS



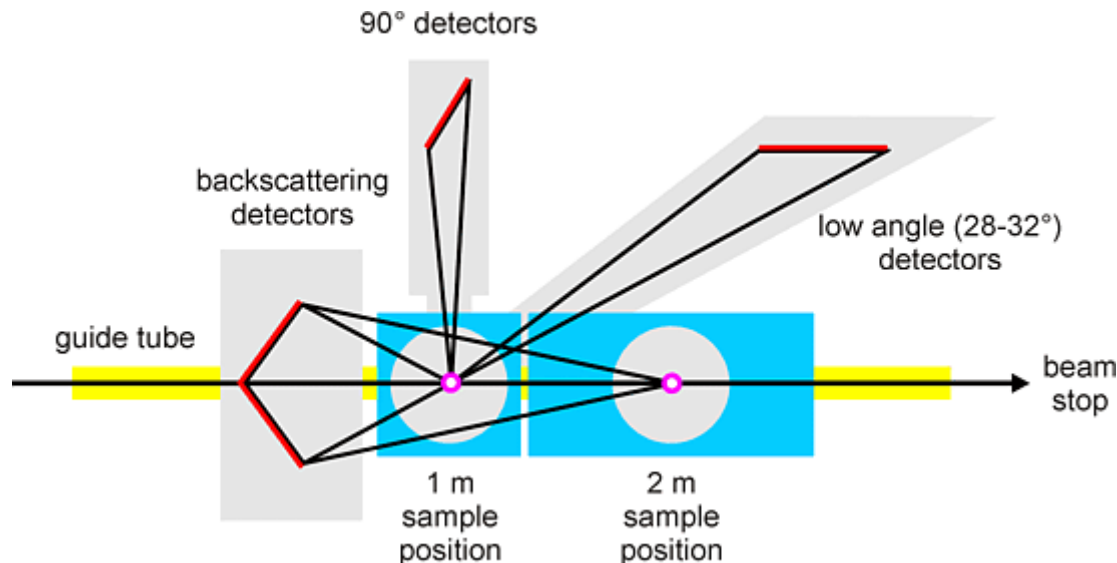
Inside ISIS Instrument Hall

- Require Heavy Shielding – Less to See

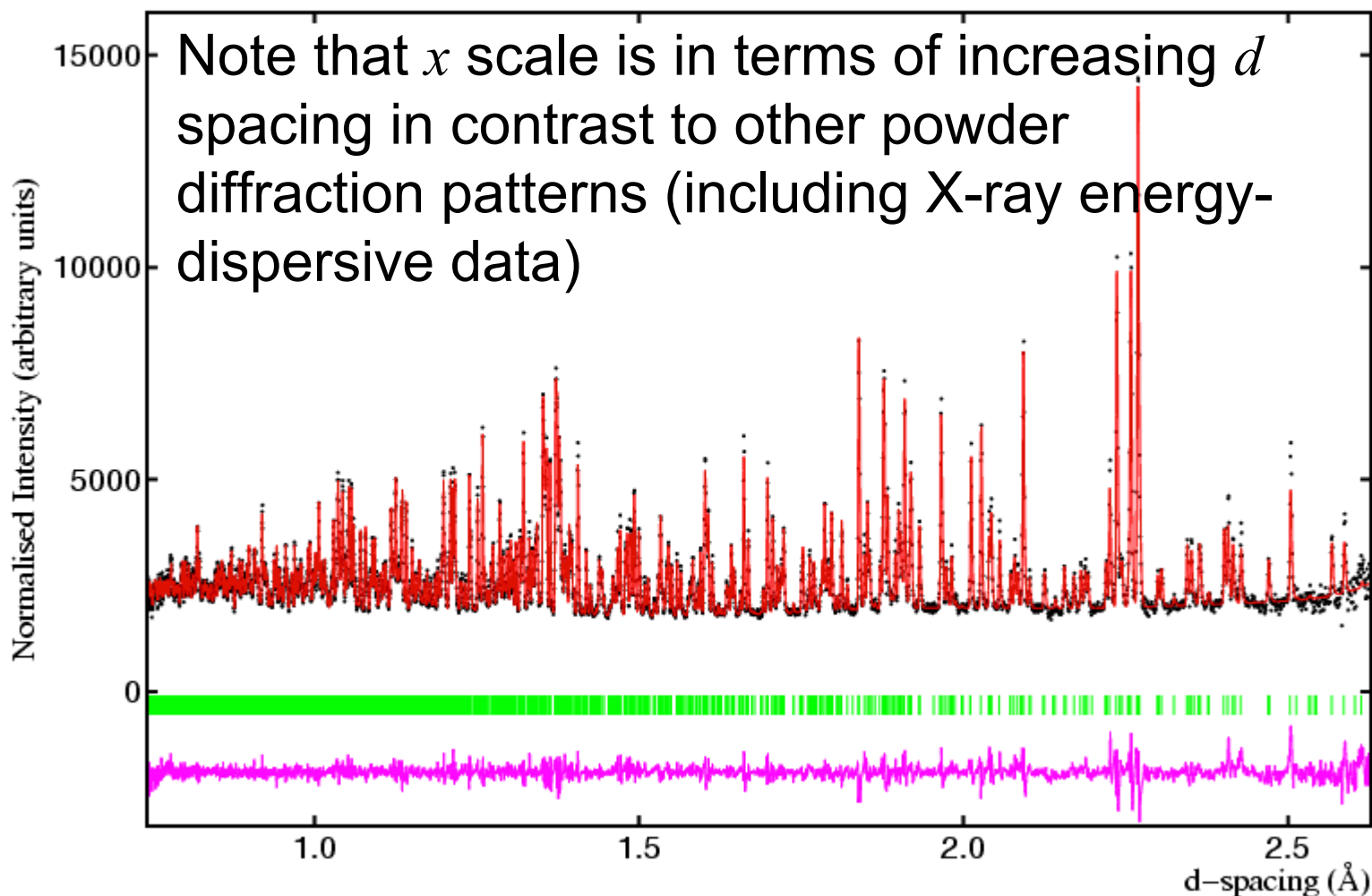


Instrument Design

- Resolution versus Flux
 - Flight Path & Detector Angle
 - Massive Solid Area Coverage – GEM
- Resolution determined by errors in t , L , 2θ



Typical TOF Powder Data



Typical Experiments

- Single-crystal

- Precise location of H
 - Especially in organometallics
 - Hydrogen-bonding studies
- Precise location of nucleus
 - Needed for X – N electron density difference maps
- High-accuracy thermal motion studies
 - No form factors
- High-accuracy disorder studies
 - No form factors
- Magnetic Crystal Structures

Typical Experiments

- Powders

- Precise location of atoms including anisotropic B-values by Rietveld refinement
 - Metal oxides, e.g. ceramics, high-temperature superconductors
 - Hydrogen bonding (but with deuterated samples)
- Solid-state phase transition studies
 - Non-magnetic and magnetic
- Solid-state reactions
- Texture measurements (bulk)

Typical Experiments

- Historically
 - Many experiments requiring non-ambient conditions
 - Expensive
 - Often developed by physicists
- Non-ambient now available in many chemistry laboratories
- Better done with hard synchrotron X-rays?

Question Time / Dinner

