

CENG 102 Project Submission Sheet

Plant for Hydrodealkylation of Toluene to Benzene

Team Name: _____

Names (Last, First)	Student ID #	Project Contribution (General or Specific)

If groups have members from both CENG 100 sections, please clarify in the table above which section (professor) you have!

Problem #		Solution
1	Hydrogen Compression: Work & Final Temperature	Work (kW): Temperature (K):
2	Hydrogen Non-idealities	Residual 1 (J/mol): Residual 2 (J/mol): New Work (kW): Ideal gas? Why or why not?
3	Toluene Heating (kW)	Q =
4	Toluene Compression	Work (kW): Temperature (K)
5	Toluene Vapor Non-Ideality	Residual 1 (J/mol): Residual 2 (J/mol): New Work (kW): Ideal gas? Why or why not?
6	Mixing Temperature (K)	T =
7	Pre-Reactor Compression	Work (kW): Temperature (K)
8	Pre-Reactor Heating (kW)	Q =

9	Reactor Heat (kW)	$Q =$
10	Adiabatic Temperature	$T =$
11	Influence on Equilibrium	<p>a.) $K_1 =$ _____ $K_2 =$ _____</p> <p>b.) $K_{1,\text{eff}} =$ _____ $K_{2,\text{eff}} =$ _____</p> <p>Effect on Rxn 1:</p> <p>Effect on Rxn 2:</p>
12	Pre-Separation Pressure Reduction	<p>a.) $T \text{ (K)} =$ _____</p> <p>b.) $W \text{ (kW)} =$ _____ $T \text{ (K)} =$ _____</p>
13	Flash I (1) = Toluene; (2) = Hydrogen (3) = Benzene; (4) = Methane; (5) = Biphenyl	$V \text{ (mol/s)} =$ _____ $L \text{ (mol/s)} =$ _____ $x_1 =$ _____ $y_1 =$ _____

	R = recovery fraction	$x_2 =$ _____ $y_2 =$ _____ $x_3 =$ _____ $y_3 =$ _____ $x_4 =$ _____ $y_4 =$ _____ $x_5 =$ _____ $y_5 =$ _____ $R_{2,V} =$ _____ $R_{4,V} =$ _____ $R_{1,L} =$ _____ $R_{3,L} =$ _____ $R_{5,L} =$ _____
14	Flash II (1) = Toluene; (2) = Hydrogen (3) = Benzene; (4) = Methane; (5) = Biphenyl	V (mol/s) = _____ L (mol/s) = _____ $x_1 =$ _____ $y_1 =$ _____ $x_2 =$ _____ $y_2 =$ _____ $x_3 =$ _____ $y_3 =$ _____ $x_4 =$ _____ $y_4 =$ _____ $x_5 =$ _____ $y_5 =$ _____ Benzene Purge (mol/s) = _____ Value of Flash II:

15	Flash III (Ideal) (1) = Toluene; (2) = Biphenyl	a.) $x_1 = \underline{\hspace{2cm}}$ $y_1 = \underline{\hspace{2cm}}$ $x_2 = \underline{\hspace{2cm}}$ $y_2 = \underline{\hspace{2cm}}$ b.) V (mol/s) = $\underline{\hspace{2cm}}$ L (mol/s) = $\underline{\hspace{2cm}}$ c.) P (bar) = $\underline{\hspace{2cm}}$ $y_1 = \underline{\hspace{2cm}}$ V (mol/s) = $\underline{\hspace{2cm}}$ L (mol/s) = $\underline{\hspace{2cm}}$
16	Flash III (Non-Ideal) (1) = Toluene; (2) = Biphenyl	a.) $x_1 = \underline{\hspace{2cm}}$ $y_1 = \underline{\hspace{2cm}}$ $x_2 = \underline{\hspace{2cm}}$ $y_2 = \underline{\hspace{2cm}}$ b.) V (mol/s) = $\underline{\hspace{2cm}}$ L (mol/s) = $\underline{\hspace{2cm}}$
17	Final Question: Value of Flash III	

XC I	Extra Credit I: Compression with Inter-Stage Cooling	a.) All in kW $W_1 = \underline{\hspace{2cm}}$ $W_2 = \underline{\hspace{2cm}}$ $W_3 = \underline{\hspace{2cm}}$ $W_4 = \underline{\hspace{2cm}}$ b.) All in K $T_{F1} = \underline{\hspace{2cm}}$ $T_{F2} = \underline{\hspace{2cm}}$ $T_{F3} = \underline{\hspace{2cm}}$ $T_{F4} = \underline{\hspace{2cm}}$ c.) All in kW $Q_1 = \underline{\hspace{2cm}}$ $Q_2 = \underline{\hspace{2cm}}$ $Q_3 = \underline{\hspace{2cm}}$
XC II	Extra Credit II: Mixing Temperature with Residuals	T (K) =
XC III	Extra Credit III: Non-Ideal Flash Pressure	P (bar) = $\underline{\hspace{2cm}}$ $y_1 = \underline{\hspace{2cm}}$ V (mol/s) = $\underline{\hspace{2cm}}$ L (mol/s) = $\underline{\hspace{2cm}}$