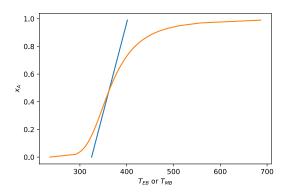
Answers Tut 2

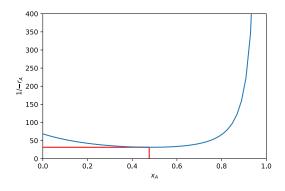
- a) x = 93.3% OR 2.2% OR 42.4% (plot x vs T (mol balance) and x vs T(energy) balance to notice the multiplicity)
 - b) x = 97.7% not multiple steady states any more
 - c) Increase inlet T by about 20 K conversion = 96.2%
 - d) x = 0.3% (almost nothing)
 - e) Tu = 539.6 K (Reactor Temperature = 530 K)

2.2

a) No multiplicity Plot:



- b) Operating T of RX2: $T_2=424$ K, $\textit{V}_{\textit{RX}_2}=68.03$ L (Conversion of A after first reactor: $\textit{x}_A=47.6\%$, $T_1=361.8$
- c) Conversion in on PFR with V=150~L and inlet conditions same as for $CSTR_1$: $x_A=31.03\%$ vs 47.6~% Explanation; Levenspiel!



d) $V_{CSTR}=152.7~L~(x_A=48.5\%)$ followed by PFR with $V_{PFR}=239.6~L~$ Total volume = 392.3~L~