Hartree-Fock Theory 1

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 - We built our electronic Hamiltonian from logically thinking about the most relevant physical interactions in our system

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 of electrons requires antisymmetric wavefunction under exchange. So, we make an
 antisymmetric product (Slater determinant) wavefunction.

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 - We built our electronic Hamiltonian from logically thinking about the most relevant physical interactions in our system
 - We reasoned that our wavefunction for a many electron system is best approximated by a product of one electron wavefunctions (spin-orbitals). But, this is not enough; half-integer spin of electrons requires antisymmetric wavefunction under exchange. So, we make an antisymmetric product (Slater determinant) wavefunction.
 - Applying this wavefunction and Hamiltonian to the time-independent Schrodinger equation gives us an expression for the electronic energy, known as the first Slater-Condon rule:

$$E = \sum_i \langle \psi^i_i | \hat{h}(i) | \psi^i_i
angle + \sum_{i < j} \langle \psi^i_i \psi^j_j | \hat{g}(i,j) | \psi^i_i \psi^j_j
angle - \langle \psi^i_i \psi^j_j | \hat{g}(i,j) | \psi^i_j \psi^j_i
angle$$

Know how to derive this equation. There isn't just one way to reason through it, come up with a way that you're comfortable with

Closed-shell case

$$E = \sum_i \langle \psi^i_i | \hat{h}(i) | \psi^i_i
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angle$$

Above is a general expression, but if we assume a closed-shell Slater determinant, the sum over N electrons can be converted to sum over the number of doubly-occupied (N/2) molecular orbitals by integrating out the spin functions (derivation in lecture notes):

$$E=2\sum_{i}^{N/2}\langle\phi_{i}|\hat{h}(i)|\phi_{i}
angle+\sum_{i}^{N/2}\sum_{j}^{N/2}2\langle\phi_{i}\phi_{j}|\hat{g}(i,j)|\phi_{i}\phi_{j}
angle-\langle\phi_{i}\phi_{j}|\hat{g}(i,j)|\phi_{j}\phi_{i}
angle$$

Restricted Hartree-Fock

Our restricted energy expression

$$E=2\sum_{i}^{N/2}\langle\phi_{i}|\hat{h}(i)|\phi_{i}
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angle-\langle\phi_{i}\phi_{j}|\hat{g}(i,j)|\phi_{j}\phi_{i}
angle$$

- We have not specified what our spatial orbitals are
- Using any random set of orbitals will give us some energy, but it probably won't be very good
- What is the best set of orbitals which give the best single Slater
 determinant wavefunction so that our energy is as accurate as possible?

Restricted Hartree-Fock

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 According to the Variational Principle, any trial wavefunction will overestimate the true ground state energy. So, varying our wavefunction's parameters (the orbitals) such that the energy is lowest will give the best wavefunction and the best energy

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$$\delta E = 2\sum_{i}^{N/2} \langle \delta \phi_i | \hat{h}(i) | \delta \phi_i
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angle - \langle \delta \phi_i \delta \phi_j | \hat{g}(i,j) | \delta \phi_j \delta \phi_i
angle$$

We cannot just vary the orbitals however we want. This energy expression
was derived under the assumption the orbitals are orthonormal. So they need
to be kept that way!

$$\delta E = 2\sum_{i}^{N/2} \langle \delta \phi_i | \hat{h}(i) | \delta \phi_i
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• So, the problem is simple: **minimize** the energy under the **constraint** that orbitals remain orthonormal

- How might we do that?
 - We could minimize the energy brute-force; compute the energy over and over again, using basic optimization algorithms like Newton conjugate gradient, BFGS, ...
 - By using the method of Lagrange multipliers which can give a general set of equations for finding extrema of a particular function subject to some constraint

$$\delta E = 2\sum_{i}^{N/2} \langle \delta \phi_i | \hat{h}(i) | \delta \phi_i
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angle$$

Minimizing the energy expression under the constraint that the spin orbitals remain orthonormal (using Lagrange's method) gives us a set of equations ^{1,2,3} describing the conditions the *best possible* spin orbitals satisfy. These *best possible* spin orbitals give the *best possible* single Slater determinant wavefunction and *best possible* energy

$$\hat{f}\,\phi_i=\epsilon_i\phi_i$$

- 1. C. C. J. Roothaan, "New Developments in Molecular Orbital Theory", 1951
- 2. Dr. Andreas Copan's Notes (Github \rightarrow CCQC \rightarrow chem-8950 \rightarrow 2017 \rightarrow 1q-1h-hartree-fock.pdf
- 3. Szabo and Ostlund "Modern Quantum Chemistry"

$$\hat{f}\,\phi_i=\epsilon_i\phi_i$$

$$\left|\hat{h} + \sum_i (2\hat{J}_j - \hat{K}_j)
ight| \phi_i = \epsilon_i \phi_i$$

$$\hat{J}_{j}|\phi^{\mu}
angle = \langle\phi_{i}^{
u}|\hat{g}(\mu,
u)|\phi_{i}^{
u}\phi^{\mu}
angle$$

 $|\hat{K}_j|\phi^\mu
angle = \langle\phi_j^
u|\hat{g}(\mu,
u)|\phi^
u\phi_j^\mu
angle$ Exchange operator

Coulomb operator

$$\hat{f}\,\phi_i=\epsilon_i\phi_i$$

 For an N-electron system, the Hartree-Fock equations are a set of N-coupled equations for N spin-orbitals (or N/2 equations for N/2 spatial orbitals in the restricted case)

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- "Solving" the HF equations means finding a set of spin-orbitals for which the HF equations are *stationary*, or "self-consistent": the set of orbitals used to construct the Fock operator are no different from the eigenfunctions of the Fock operator

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 HF equations are *stationary*, or "self-consistent": the set of orbitals used to
 construct the Fock operator are no different from the eigenfunctions of the
 Fock operator
- Trial and error: Assume a set of ϕ_i , calculate f, solve the above equation for the N/2 lowest eigenvalues and corresponding eigenfunctions, compare the eigenfunctions to the original set, update the orbitals, repeat until consistency

Solving the Hartree-Fock equations is really hard

$$\hat{f} \phi_i = \epsilon_i \phi_i$$

- What is the form of ϕ_i ?
- How to apply the Fock operator? Solve for new ϕ_i ?

Hartree-Fock has really only ever been done for single atoms, or very simple diatomics, but even then some approximations are made

We need to approximate the best MO's given by Hartree-Fock. The most common approximation is by a *linear combination of atomic orbitals* (LCAO) to form the *approximate* Hartree-Fock MO's (LCAO-MO)

$$\hat{f}\,\phi_i=\epsilon_i\phi_i$$

$$\phi_i = \sum_q \chi_q C_{qi}$$

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$$\hat{f}\,\phi_i = \epsilon_i\phi_i \qquad \qquad \phi_i = \sum_q \chi_q C_{qi}$$

$$\hat{f} \, \sum_{q} \chi_q C_{qi} = \epsilon_i \sum_{q} \chi_q C_{qi}$$

Left-multiplying by some other AO χ_p^* and integrating both sides:

$$\sum_{q} \langle \chi_p \mid \hat{f} \mid \chi_q
angle C_{qi} = \epsilon_i \sum_{q} \langle \chi_p \mid \chi_q
angle C_{qi}$$

$$egin{aligned} \hat{f} \, \phi_i &= \epsilon_i \phi_i \ \ \hat{f} \, \sum_q \chi_q C_{qi} &= \epsilon_i \sum_q \chi_q C_{qi} \ \ \sum_q \langle \chi_p \mid \hat{f} \mid \chi_q
angle C_{qi} &= \epsilon_i \sum_q \langle \chi_p \mid \chi_q
angle C_{qi} \end{aligned}$$

Defining the integrals above $\langle \chi_p \mid \hat{f} \mid \chi_q
angle$ and $\langle \chi_p \mid \chi_q
angle$ as F_{pq} and S_{pq} ,

$$\sum_{q}F_{pq}C_{qi}=\epsilon_{i}\sum_{q}S_{pq}C_{qi}$$
 ————— $oldsymbol{FC}=oldsymbol{SC\epsilon}$

True for a single ϕ_i

True for all ϕ_i

Roothaan-Hall equations:

$$FC = SC\epsilon$$

For m AO basis functions, c is an m x m matrix with each column containing the expansion coefficients for some LCAO-MO ϕ_i

Roothaan-Hall equations:

$$FC = SC\epsilon$$

$$m{F}_{pq} = egin{pmatrix} \langle \chi_1 \mid \hat{f} \mid \chi_1
angle & \ldots & \langle \chi_1 \mid \hat{f} \mid \chi_m
angle \ dots & \ddots & dots \ \langle \chi_m \mid \hat{f} \mid \chi_1
angle & \ldots & \langle \chi_m \mid \hat{f} \mid \chi_m
angle \end{pmatrix} \ m{S}_{pq} = egin{pmatrix} \langle \chi_1 \mid \chi_1
angle & \ldots & \langle \chi_1 \mid \chi_m
angle \ dots & \ddots & dots \ \langle \chi_m \mid \chi_1
angle & \ldots & \langle \chi_m \mid \chi_m
angle \end{pmatrix}$$

Applying the definition of the Fock operator, and Coulomb and Exchange

operators:
$$\hat{m{f}} = \left[\hat{h} + \sum_{j} (2\hat{J}_{j} - \hat{K}_{j})
ight] egin{array}{c} \langle \chi_{p} \mid \hat{J}_{j} \mid \chi_{q}
angle = \langle \chi_{p} \phi_{j} \mid \hat{g} \mid \chi_{q} \phi_{j}
angle \ \langle \chi_{p} \mid \hat{K}_{j} \mid \chi_{q}
angle = \langle \chi_{p} \phi_{j} \mid \hat{g} \mid \phi_{j} \chi_{q}
angle \end{array}$$

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angle = \langle \chi_p \phi_j \mid \hat{g} \mid \chi_q \phi_j
angle \ \langle \chi_p \mid \hat{K}_j \mid \chi_q
angle = \langle \chi_p \phi_j \mid \hat{g} \mid \phi_j \chi_q
angle \end{pmatrix}$$

$$\langle \chi_p \mid \hat{f} \mid \chi_q
angle = \langle \chi_p \mid \hat{h} \mid \chi_q
angle + \sum_j^{N/2} 2 \langle \chi_p \phi_j \mid \hat{g} \mid \chi_q \phi_j
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angle - \langle \chi_p \phi_j \mid \hat{g} \mid \phi_j \chi_q
angle$$

Expand:
$$\langle \phi_j \mid = \sum_r \langle \chi_r \mid C^*_{rj} \qquad \mid \phi_j \rangle = \sum_s \mid \chi_s \rangle C_{sj}$$

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angle = \sum \mid \chi_s
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$$egin{aligned} \langle \chi_p \mid f \mid \chi_q
angle = \langle \chi_p \mid \hat{h} \mid \chi_q
angle + \sum_j \sum_s C_{rj} C_{sj} \left[2 \langle \chi_p \chi_r \mid g \mid \chi_q \chi_s
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ight] \ \langle \chi_p \mid \hat{f} \mid \chi_q
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ight] \end{aligned}$$

$$\langle \chi_p \mid \hat{f} \mid \chi_q
angle = \langle \chi_p \mid \hat{h} \mid \chi_q
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angle - \langle \chi_p \chi_r \mid \hat{g} \mid \chi_s \chi_q
angle
ight]$$

Each element of the Fock matrix is a simple sum of:

- A single electron-kinetic-energy integral over two AO basis functions
- A single electron-nuclear attraction integral over two AO basis functions
- A sum over weighted two-electron integrals

The whole Fock matrix is built from matrices which hold all possible kinetic and potential integrals over the set of AO basis functions, and an array of all possible two electron integrals 'contracted' with the density matrix

$$FC = SC\epsilon$$

ullet If $oldsymbol{S}$ is an identity matrix, this equation is easy to solve by 'diagonalizing' $oldsymbol{F}$

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- Generally, our AO basis is not orthogonal to begin with
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So, we need to perform a change-of-basis, and this change-of-basis transformation needs to satisfy: $\mathbf{T} T \cdot \mathbf{S} \mathbf{T} T = \mathbf{T}$

$\boldsymbol{U}^T \boldsymbol{S} \boldsymbol{U} = \boldsymbol{I}$

$oldsymbol{U}^Toldsymbol{S}oldsymbol{U}=oldsymbol{I}$

$$oldsymbol{U} = oldsymbol{S}^{-1/2}$$

$oldsymbol{U}^Toldsymbol{S}oldsymbol{U}=oldsymbol{I}$

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$$m{U}^T m{S} m{U} = m{S}^{-1/2} m{S} m{S}^{-1/2} = m{I}$$

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$$oldsymbol{U}^T oldsymbol{S} oldsymbol{U} = oldsymbol{S}^{-1/2} oldsymbol{S} oldsymbol{S}^{-1/2} = oldsymbol{I}$$

Change our basis to an orthogonalized basis:

$$m{FC} = m{SC}m{\epsilon}$$
 $m{S}^{-1/2}m{FS}^{-1/2}$ $m{S}^{-1/2}m{SS}^{-1/2}$

We can't just stuff ${m S}^{-1/2}$ into this equation wherever we want; we have to do some algebraic manipulations

$$FC = SC\epsilon$$

$$m{S}^{-1/2}m{F}m{S}^{-1/2}$$

$$m{S}^{-1/2} m{S} m{S}^{-1/2}$$

$$ilde{m{F}} ilde{m{C}}= ilde{m{C}}m{\epsilon}$$

$$ilde{oldsymbol{F}} = oldsymbol{S}^{-1/2} oldsymbol{F} oldsymbol{S}^{-1/2}$$

$$ilde{m{C}} = m{S}^{1/2}m{C}$$

RHF Algorithm

- 1. Collect all one and two-electron integrals in a matrix, form the orthogonalizer $m{S}^{-1/2}$
- 2. Guess $oldsymbol{D}=oldsymbol{0}$
- 3. Build $m{F}$
- 4. Compute the energy
- 5. Diagonalize $ilde{m{F}} = m{S}^{-1/2} m{F} m{S}^{-1/2}$ to get $ilde{m{C}}$ and $m{\epsilon}$
- 6. Backtransform to unorthogonalized AO basis $oldsymbol{C} = oldsymbol{S}^{-1/2} ilde{oldsymbol{C}}$
- 7. Compute the new density matrix $m{D} = \sum_{j}^{N/2} C_{rj}^* C_{sj}$
- 8. If new $oldsymbol{D}$ and old $oldsymbol{D}$ differ by too much, return to step 3.