

CCSI Steady State MEA Model (MEA ssm)

User Manual

Version 3.0.0 August 2019













Copyright (c) 2012 - 2019

Copyright Notice

MEA Steady State Model was produced under the DOE Carbon Capture Simulation Initiative (CCSI), and is copyright (c) 2012 - 2019 by the software owners: Oak Ridge Institute for Science and Education (ORISE), TRIAD National Security, LLC., Lawrence Livermore National Security, LLC., The Regents of the University of California, through Lawrence Berkeley National Laboratory, Battelle Memorial Institute, Pacific Northwest Division through Pacific Northwest National Laboratory, Carnegie Mellon University, West Virginia University, Boston University, the Trustees of Princeton University, The University of Texas at Austin, URS Energy & Construction, Inc., et al.. All rights reserved.

NOTICE. This Software was developed under funding from the U.S. Department of Energy and the U.S. Government consequently retains certain rights. As such, the U.S. Government has been granted for itself and others acting on its behalf a paid-up, nonexclusive, irrevocable, worldwide license in the Software to reproduce, distribute copies to the public, prepare derivative works, and perform publicly and display publicly, and to permit other to do so.

License Agreement

MEA Steady State Model Copyright (c) 2012 - 2019, by the software owners: Oak Ridge Institute for Science and Education (ORISE), TRIAD National Security, LLC., Lawrence Livermore National Security, LLC., The Regents of the University of California, through Lawrence Berkeley National Laboratory, Battelle Memorial Institute, Pacific Northwest Division through Pacific Northwest National Laboratory, Carnegie Mellon University, West Virginia University, Boston University, the Trustees of Princeton University, The University of Texas at Austin, URS Energy & Construction, Inc., et al. All rights reserved.

Redistribution and use in source and binary forms, with or without modification, are permitted provided that the following conditions are met:

- 1. Redistributions of source code must retain the above copyright notice, this list of conditions and the following disclaimer.
- Redistributions in binary form must reproduce the above copyright notice, this list of conditions and the following disclaimer in the documentation and/or other materials provided with the distribution.
- 3. Neither the name of the Carbon Capture Simulation Initiative, U.S. Dept. of Energy, the National Energy Technology Laboratory, Oak Ridge Institute for Science and Education

(ORISE), TRIAD National Security, LLC., Lawrence Livermore National Security, LLC., the University of California, Lawrence Berkeley National Laboratory, Battelle Memorial Institute, Pacific Northwest National Laboratory, Carnegie Mellon University, West Virginia University, Boston University, the Trustees of Princeton University, the University of Texas at Austin, URS Energy & Construction, Inc., nor the names of its contributors may be used to endorse or promote products derived from this software without specific prior written permission.

THIS SOFTWARE IS PROVIDED BY THE COPYRIGHT HOLDERS AND CONTRIBUTORS "AS IS" AND ANY EXPRESS OR IMPLIED WARRANTIES, INCLUDING, BUT NOT LIMITED TO, THE IMPLIED WARRANTIES OF MERCHANTABILITY AND FITNESS FOR A PARTICULAR PURPOSE ARE DISCLAIMED. IN NO EVENT SHALL THE COPYRIGHT OWNER OR CONTRIBUTORS BE LIABLE FOR ANY DIRECT, INDIRECT, INCIDENTAL, SPECIAL, EXEMPLARY, OR CONSEQUENTIAL DAMAGES (INCLUDING, BUT NOT LIMITED TO, PROCUREMENT OF SUBSTITUTE GOODS OR SERVICES; LOSS OF USE, DATA, OR PROFITS; OR BUSINESS INTERRUPTION) HOWEVER CAUSED AND ON ANY THEORY OF LIABILITY, WHETHER IN CONTRACT, STRICT LIABILITY, OR TORT (INCLUDING NEGLIGENCE OR OTHERWISE) ARISING IN ANY WAY OUT OF THE USE OF THIS SOFTWARE, EVEN IF ADVISED OF THE POSSIBILITY OF SUCH DAMAGE.

You are under no obligation whatsoever to provide any bug fixes, patches, or upgrades to the features, functionality or performance of the source code ("Enhancements") to anyone; however, if you choose to make your Enhancements available either publicly, or directly to Lawrence Berkeley National Laboratory, without imposing a separate written license agreement for such Enhancements, then you hereby grant the following license: a non-exclusive, royalty-free perpetual license to install, use, modify, prepare derivative works, incorporate into other computer software, distribute, and sublicense such enhancements or derivative works thereof, in binary and source code form. This material was produced under the DOE Carbon Capture Simulation

Table of Contents

1.0	Reporting Issues	3
2.0	Version Log	3
MEA St	teady State Model	4
1.0	Model Development	4
	1.1 Model Background	4 5 5
2.0	Tutorial	
3.0	2.1 Creating Fortran Subroutines 2.1.1 Viscosity Model 2.1.2 Molar Volume Model 2.1.3 Surface Tension Model 2.1.4 Liquid Diffusivity Model 2.1.5 Reaction Kinetics Model 2.1.6 Interfacial Area Model 2.1.7 Holdup Model 2.1.8 Creation of dll and opt files 2.2 Predicting System VLE 2.3 CO ₂ Capture Process Simulation Base Case Setup 2.4 CO ₂ Capture Process Simulation Example Usage Information	
	3.1 Environment/Prerequisites	
4.0	References	
Figure 1	Figures : CO ₂ partial pressure as a function of loading and temperature (30 wt% MEA) : Results of the "FLOW" sensitivity block for the case study	
	: Absorber temperature profile for the case study	
Ū	•	
List of	: Regenerator temperature profile for the case study. **Tables* Suggested Ranges for Variables in Simulation	
	Results of PCO ₂ Sensitivity Block	
- 4010 2.	22000100 01 2 004 Demontrately Dioentinininininininininininininininininini	

CCSI Process Models	User Manua

Table 3: Variables for Base Case Simulation	. 19
Table 4: Selected Stream Table Results	. 20

To obtain support for the products within this package, please send an e-mail to $\underline{ccsi\text{-support@acceleratecarboncapture.org}}.$

CCSI Process Models

User Manual

1.0 REPORTING ISSUES

To report a problem, make a suggestion or ask a question, please either open an issue at our GitHub repository at: https://github.com/CCSI-Toolset/MEA_ssm/issues or alternatively send an e-mail to our support list: ccsi-support@acceleratecarboncapture.org.

2.0 VERSION LOG

Product	Version Number	Release Date	Description
Steady State MEA Model	3.0.0	8/31/2019	New version of model created for compatibility with Aspen Plus V10. Additional new features include a more rigorous flowsheet and instructions for creating FORTRAN user subroutines needed for the model.
Steady State MEA Model	2.0.0	3/31/2018	Initial Open Source release
Steady State MEA Model	2015.10.0	10/16/2015	

MEA Steady State Model

1.0 MODEL DEVELOPMENT

1.1 Model Background

This document describes a solvent-based CO₂ capture system using aqueous monoethanolamine (MEA). The model consists of the "CCSI_MEAModel.bkp" file with supporting files "ccsi.opt" and "ccsi10.dll," which contain FORTRAN user models associated with the simulation. This model was developed to be compatible with Aspen Plus[®] V10.

This model represents the first version of the "gold standard" model for the MEA capture system. It is composed of individually developed submodels for physical properties of CO₂-loaded aqueous MEA solutions and hydraulic and mass transfer models for the system of interest. Each submodel is developed and calibrated with relevant data over the full range of process conditions of interest (e.g., temperature, composition). For each submodel, existing models are considered as candidates and are modified to better fit experimental data over the conditions of interest.

1.2 Physical Property Models

Physical property models developed in this work include standalone models and an integrated thermodynamic framework. Standalone models for viscosity, density, and surface tension of the system have been developed, with uncertainty quantification, as described in Morgan et al., and are implemented as FORTRAN user models. The thermodynamic framework of this system is developed using UT Austin's Phoenix model thermodynamic framework as a precursor. Here, the solution thermodynamics are represented by the ELECNRTL method in Aspen Plus, which uses the Redlich-Kwong equation of state to calculate the vapor phase fugacity coefficients and the electrolyte non-random two liquid (e-NRTL) model to calculate the activity coefficients in the liquid phase. Model parameters are calibrated by fitting data for VLE, heat capacity, and heat of absorption for the ternary MEA-H₂O-CO₂ system and VLE data for the binary MEA-H₂O system. The kinetic model used in this work is taken from the Phoenix model, in which the overall ionic speciation of the system is simplified into two equilibrium reactions:

$$2MEA + CO_2 \leftrightarrow MEA^+ + MEACOO^- \tag{1}$$

$$MEA + CO_2 + H_2O \leftrightarrow MEA^+ + HCO_3^- \tag{2}$$

The forward reaction rate constants are taken from the Phoenix model, and the overall reaction rate is written in terms of the equilibrium constants which are also calculated as part of the thermodynamic framework of the system. This follows the methodology presented in Mathias and Gilmartin³, and is implemented to ensure that the reaction kinetics are consistent with the thermodynamic framework.

1.3 Mass Transfer and Hydraulic Models

The hydrodynamic models developed in this work include models for pressure drop and hold-up. The Billet and Schultes⁴ correlation is regressed with data from Tsai⁵ for MellapakPlus™ 250Y packing, which is similar to MellapakPlus 252Y packing, which is considered in this work. In this work, a novel and integrated methodology to obtain the mass transfer model is proposed. In this integrated mass transfer model, parameters of the interfacial area, mass transfer coefficients, and diffusivity models are regressed using wetted wall column data from Dugas⁶ and pilot plant data from Tobiesen et al.⁶ This required simultaneous regression of process model and property parameters, which was accomplished using the CCSI software Framework for Optimization and Quantification of Uncertainty and Sensitivity (FOQUS).

1.4 Development of Process Model

The aforementioned submodels are integrated into this steady state process model, which is representative of the configuration of the National Carbon Capture Center (NCCC) in Wilsonville, Alabama, for which data have been obtained for validation of this model. No parameters are tuned to improve the fit to the fit to the pilot plant data. The model includes both the absorber and stripper columns, although the recycle of the lean solvent from the regenerator outlet to the absorber inlet is not modeled. The columns are modeled as rate based columns using RateSepTM.

The various submodels are implemented in Aspen Plus either as built-in models (e.g., ELECNRTL thermodynamic framework) or FORTRAN user models, in cases where built-in models with the appropriate model form are not available. The user models are combined into a dynamic library ("ccsi10.dll" for this model) and a dynamic linking options (DLOPT) file ("ccsi.opt") is also provided, which has already been specified in the Aspen Plus file for this model. The various user models contained in the linked library include physical property models for viscosity, density, surface tension, and diffusivity, the hydraulics model, the interfacial area model, and the reaction kinetics model.

1.5 Model Features

The "CCSI_MEAModel.bkp" file included is representative of a typical operating case at NCCC and some adjustment of operating variables is possible. Table 1 includes some of these variables and suggested ranges for which the model is expected to work, based on the ranges considered in the testing at NCCC.

Variable	Range
Lean Solvent Amine Concentration (g MEA/g MEA+H ₂ O)	0.25 - 0.35
Lean Solvent CO ₂ Loading (mol CO ₂ /mol MEA)	0.05 - 0.50
Lean Solvent Flowrate (kg/hr)	3000 – 12000
Flue Gas Flowrate (kg/hr)	1250 – 3000
Regenerator Reboiler Duty (kW)	150 – 700

Table 1: Suggested Ranges for Variables in Simulation

Table 1 includes the major variables that dictate the performance of the process, although the list is not exhaustive. Other variables, including operating temperature and pressure of the equipment, are set at typical values for the MEA-based CO_2 capture process, and slight variation of these variables is allowable. As the lean solvent flowrate is decreased, the intercooler flow rates should be adjusted accordingly. **Note:** The apparent mole fractions of molecular species may be calculated from the amine concentration (γ) and CO_2 loading (α) using the equations:

$$X_{MEA} = \left(1 + \alpha + \left(\frac{MW_{MEA}}{MW_{H_2O}}\right)\left(\frac{1}{\gamma} - 1\right)\right)^{-1}$$
(3)

$$X_{CO_2} = \alpha X_{MEA} \tag{4}$$

$$X_{H_2O} = 1 - X_{MEA} - X_{CO_2} (5)$$

2.0 TUTORIAL

2.1 Creating Fortran Subroutines

This is an optional tutorial for those users who wish to directly develop the Fortran subroutines used in this model and compile them as a dll file. Otherwise, the user may use the provided 'ccsi.opt' and 'ccsi10.dll' files and skip to the tutorial in section 2.2. In order to start working on the Fortran files, ensure that a compiler is installed in the machine. The compiler used by the authors of this manual is: "Intel Visual Fortran Composer XE 2011 Update 12 for Windows". This is an external application that requires a license file (to be obtained by the user). Double click the application file to begin the compiler installation, and complete it by following the instructions on the window.

Once the installation is complete, open the Aspen Application "Set Compiler for V10", and check the status of the sections. At least one of them must display the status "OK" which would confirm the successful installation of the compiler.

To obtain FORTRAN template .f files distributed with Aspen Tech software, navigate to the following folder:

C:\Program Files (x86)\AspenTech\Aspen Plus V10.0\Engine\User

For other versions of AspenTech software, the template files may be found in the folder corresponding to the specific version. The user is now required to make changes to the template files as directed in the subsections.

2.1.1 Viscosity Model

For the liquid viscosity model, open the file (*mul2u2.f*). In the section of the code titled 'DECLARE ARGUMENTS', add the following code for declaring additional defined variables that are not included in the template. The existing code in this section of the template should not be deleted, as it is needed to declare the major input and output variables of the subroutine.

```
INTEGER DMS_KCCIDC,I
INTEGER IH20,IMEA,IMEACOO,ICO2,IMEAH,IHCO3
REAL*8 XX(100),SUM,DSUM,DPSUM
REAL*8 A,B,C,D,E,F,G
REAL*8 MUW,XCO2T,XMEAT,XH2OT,LDG,WTMEA,MUBLEND
```

In the 'BEGIN EXECUTABLE CODE' section, remove the template code that has been provided. Note that the final section of the template code, in which defines the final liquid viscosity (MUMX), its temperature derivative (DMUMX), and its pressure derivative (DPMUMX), must not be deleted. Insert the following code under the 'BEGIN EXECUTABLE CODE':

```
DO I=1,N
    IF (IDX(I). EQ. IH20) XX(IH20) = Z(I)
    IF (IDX(I), EQ. IMEA) XX(IMEA) = Z(I)
    IF (IDX(I). EQ. IMEACOO) XX(IMEACOO) = Z(I)
    IF (IDX(I). EQ. ICO2) XX(ICO2) = Z(I)
    IF (IDX(I). EQ. IMEAH) XX(IMEAH) = Z(I)
    IF (IDX(I). EQ. IHCO3) XX(IHCO3) = Z(I)
END DO
A = MULU2A(1, IMEA)
B = MULU2A(2, IMEA)
C = MULU2A(3, IMEA)
D = MULU2A(4, IMEA)
E = MULU2A(5, IMEA)
F = MULU2A(1,IH20)
G = MULU2A(2,IH20)
MUW = 1.002
MUW = MUW * 10 * * (1.3272 * (293.15 - T - 0.001053 * (T - 293.15) * * 2) / (T - 168.15))
XCO2T = XX(IMEACOO) + XX(IHCO3) + XX(ICO2)
XMEAT = XX(IMEACOO) + XX(IMEAH) + XX(IMEA)
XH2OT = XX(IHCO3) + XX(IH2O)
LDG = XCO2T/XMEAT
WTMEA = XMEAT*XMW(IMEA) + XH2OT*XMW(IH2O)
WTMEA = 100*((XMEAT*XMW(IMEA))/WTMEA)
MUBLEND=(A*WTMEA+B)*T+(C*WTMEA+D)
MUBLEND=MUBLEND*(LDG*(E*WTMEA+F*T+G)+1)*WTMEA
MUBLEND=DEXP(MUBLEND/T**2)
IF (XMEAT.EQ.0) THEN
     SUM=MUI(IH20)
ELSE IF (XH2OT.EQ.0) THEN
    SUM=DEXP(-102.07+7992.1/T+13.724*LOG(T))/1000
ELSE
    SUM=MUBLEND*MUW/1000
END IF
```

The existing RETURN & END statements at the end of the code must be retained. Ensure that the inserted code lines do not get commented.

2.1.2 Molar Volume Model

For the liquid molar volume model, the process is analogous to that used for the viscosity model. In the folder that contains the Fortran templates, select 'vl2u2.f'. The following code should be added to the 'DECLARE ARGUMENTS' section without deleting the existing code:

```
INTEGER DMS_KCCIDC,I
INTEGER IH2O,IMEA,IMEACOO,ICO2,IMEAH,IHCO3
REAL*8 XX(100),SUM,DSUM,DPSUM
REAL*8 A,B,C,D,E
REAL*8 AM,BM,CM,AW,BW,CW
REAL*8 VH2O,VMEA
REAL*8 XCO2T,XMEAT,XH2OT,XTOT
REAL*8 XCO2,XMEA,XH2O
```

In the section marked 'BEGIN EXECUTABLE CODE', remove the template code and replace with the code given below. Note that the final section of the template code, in which defines the final liquid molar volume (VMX), its temperature derivative (DVMX), and its pressure derivative (DPVMX), must not be deleted.

```
IH20 = DMS KCCIDC('H20')
IMEACOO = DMS KCCIDC('MEACOO-')
ICO2 = DMS KCCIDC('CO2')
IMEAH = DMS KCCIDC('MEA+')
IHCO3 = DMS KCCIDC('HCO3-')
IMEA = DMS KCCIDC('MEA')
DO I=1,100
    XX(I) = 0
END DO
DO I=1.N
    IF (IDX(I). EQ. IH20) XX(IH20) = Z(I)
    IF (IDX(I). EQ. IMEA) XX(IMEA) = Z(I)
    IF (IDX(I). EQ. IMEACOO) XX(IMEACOO) = Z(I)
    IF (IDX(I). EQ. ICO2) XX(ICO2) = Z(I)
    IF (IDX(I). EQ. IMEAH) XX(IMEAH) = Z(I)
    IF (IDX(I). EQ. IHCO3) XX(IHCO3) = Z(I)
END DO
A = VL2U2A(1,IMEA)
B = VL2U2A(2, IMEA)
C = VL2U2A(3, IMEA)
D = VL2U2A(4,IMEA)
E = VL2U2A(5, IMEA)
AM=-0.000000535162
BM=-0.000451417
CM=1.19451
AW=-0.00000324839
BW=0.00165311
CW=0.793041
VH2O = XMW(IH2O)/(AW*T**2+BW*T+CW)
VMEA = XMW(IMEA)/(AM*T**2+BM*T+CM)
XCO2T = XX(IMEACOO) + XX(IHCO3) + XX(ICO2)
XMEAT = XX(IMEACOO) + XX(IMEAH) + XX(IMEA)
XH2OT = XX(IHCO3) + XX(IH2O)
XTOT = XCO2T+XMEAT+XH2OT
XCO2 = XCO2T/XTOT
XMEA = XMEAT/XTOT
XH20 = XH20T/XT0T
SUM = XMEA*VMEA + XH2O*VH2O + XCO2*A + XMEA*XH2O*(B+C*XMEA)
SUM = SUM + XMEA * XCO2 * (D + E * XMEA)
IF (XMEA.EQ.0) THEN
     SUM=VI(IH20)
ELSE IF (XH20.EQ.0) THEN
    SUM=VMEA/1000
```

```
ELSE
SUM=SUM/1000
END IF

DSUM=0D0
DPSUM=0D0
```

The existing RETURN & END statements at the end of the code must be retained.

2.1.3 Surface Tension Model

The process for creating the surface tension model is very similar to the process used for the viscosity and molar volume models. In the folder containing the Fortran templates, select 'sig2u2.f'. The following code should be added to the 'DECLARE ARGUMENTS' section without deleting the existing code:

```
INTEGER DMS_KCCIDC,I
INTEGER IH20,IMEA,IMEACOO,ICO2,IMEAH,IHCO3
REAL*8 XX(100),SUM,DSUM,DPSUM
REAL*8 A,B,C,D,E,F,G,H,K,J
REAL*8 S1,S2,S3,S4,S5,S6
REAL*8 C1W,C1M,C2W,C2M,C3W,C3M,C4W,C4M,TCW,TCM
REAL*8 XMEAT,XCO2T,XH2OT
REAL*8 XMEA,XCO2,XH2O,LDG,WTMEA
REAL*8 FXNF,FXNG,SIGCO2,SIGH2O,SIGMEA
```

In the 'BEGIN EXECUTABLE CODE' section, remove the template code and replace with the code given below. Note that the final section of the template code, in which defines the final liquid surface tension (STMX), its temperature derivative (DSTMX), and its pressure derivative (DPSTMX), must not be deleted.

```
IH20 = DMS \ KCCIDC('H20')
IMEA = DMS KCCIDC('MEA')
IMEACOO = DMS KCCIDC('MEACOO-')
ICO2 = DMS KCCIDC('CO2')
IMEAH = DMS KCCIDC('MEA+')
IHCO3 = DMS KCCIDC('HCO3-')
DO I=1,100
    XX(I) = 0
END DO
DO I=1,N
    IF (IDX(I). EQ. IH20) XX(IH20) = Z(I)
    IF (IDX(I). EQ. IMEA) XX(IMEA) = Z(I)
    IF (IDX(I). EQ. IMEACOO) XX(IMEACOO) = Z(I)
    IF (IDX(I). EQ. ICO2) XX(ICO2) = Z(I)
    IF (IDX(I). EQ. IMEAH) XX(IMEAH) = Z(I)
    IF (IDX(I). EQ. IHCO3) XX(IHCO3) = Z(I)
END DO
A=SIGU2A(1, IMEA)
B=SIGU2A(2, IMEA)
C=SIGU2A(3, IMEA)
D=SIGU2A(4, IMEA)
E=SIGU2A(5, IMEA)
F=SIGU2A(1, IH20)
G=SIGU2A(2,IH2O)
```

```
H=SIGU2A(3,IH2O)
K=SIGU2A(4, IH2O)
J=SIGU2A(5,IH2O)
S1 = -5.987
S2=3.7699
S3 = -0.43164
S4=0.018155
S5=-0.01207
S6=0.002119
C1W=0.18548
C1M=0.09945
C2W=2.717
C2M=1.067
C3W = -3.554
C3M=0
C4W = 2.047
C4M=0
TCW=647.13
TCM=614.45
XCO2T=XX(IMEACOO)+XX(IHCO3)+XX(ICO2)
XMEAT=XX(IMEACOO)+XX(IMEAH)+XX(IMEA)
XH20T=XX(IH20)+XX(IHC03)
WTMEA=(XMW(IMEA)*XMEAT)/(XMW(IMEA)*XMEAT+XMW(IH2O)*XH2OT)
LDG=XCO2T/XMEAT
XMEA=(1+LDG+(XMW(IMEA)/XMW(IH2O))*(1-WTMEA)/WTMEA)**(-1)
XCO2=XMEA*LDG
XH20=1-XMEA-XC02
FXNF=A+B*LDG+C*LDG**2+D*WTMEA+E*WTMEA**2
FXNG=F+G*LDG+H*LDG**2+K*WTMEA+J*WTMEA**2
SIGCO2=S1*WTMEA**2+S2*WTMEA+S3+T*(S4*WTMEA**2+S5*WTMEA+S6)
SIGH20=C1W*(1-T/TCW)**(C2W+C3W*(T/TCW)+C4W*(T/TCW)**2)
SIGMEA = C1M*(1-T/TCM)**(C2M+C3M*(T/TCM)+C4M*(T/TCM)**2)
SUM=SIGH2O+(SIGCO2-SIGH2O)*FXNF*XCO2+(SIGMEA-SIGH2O)*FXNG*XMEA
IF (XMEAT.EQ.0) THEN
    SUM=STI(IH20)
ELSE IF (XH2OT.EQ.0) THEN
    SUM=SIGMEA
ELSE
    SUM=SUM
END IF
DSUM=0D0
DPSUM=0D0
```

The existing RETURN & END statements at the end of the code must be retained.

2.1.4 Liquid Diffusivity Model

Select the template 'dl0u.f' and add the following statement, required for accessing component data stored in the labeled common DMS_PLEX, to the end of the 'DECLARE VARIABLES USED IN DIMENSIONING' section.

```
#include "dms_plex.cmn"
```

Ensure that the other # include statements are retained.

The following code should be added to the 'DECLARE ARGUMENTS' section of the subroutine without deleting the existing code:

```
INTEGER DMS_KCCIDC,DMS_IFCMNC,NBOPST(6),NAME(2)
INTEGER IH20,IMEA,IMEACOO,ICO2,IMEAH,IHCO3,IN2,IO2
REAL*8 VISC,MUMX
REAL*8 E,MU0,THET,A,BB,C,R,HG,MUW
REAL*8 B(1)
EQUIVALENCE (B(1),IB(1))
INTEGER DFACT_IDX,EFACT_IDX
REAL*8 DFACTCO2,DFACTMEA,EFACT,CO2DW,CO2D,MEAD
```

Remove all code given in the template's 'BEGIN EXECUTABLE CODE' section, leaving only the final 'END' statement. Replace this code with the following:

```
IH20 = DMS_KCCIDC('H20')
IMEA = DMS KCCIDC('MEA')
IMEACOO = DMS KCCIDC('MEACOO-')
ICO2 = DMS_KCCIDC('CO2')
IMEAH = DMS_KCCIDC('MEA+')
IHCO3 = DMS_KCCIDC('HCO3-')
IN2 = DMS_KCCIDC('N2')
IO2 = DMS_KCCIDC('O2')
CALL PPUTL GOPSET(NBOPST,NAME)
CALL PPMON VISCL (T, P, X, N, IDX, NBOPST, KDIAG, VISC, KER)
MUMX = VISC
E = 4.753D0
MU0 = 0.000024055D0
THET = 139.7D0
A = 0.000442D0
BB = 0.0009565D0
C = 0.0124D0
 R = 0.008314D0
 P = P / 10000000
HG = A * P + ((E - BB * P)/(R * (T - THET - C * P)))
MUW = (MU0 * EXP(HG))
DFACT IDX = DMS IFCMNC('DFACT1')
EFACT IDX = DMS IFCMNC('EFACT')
DFACTCO2 = B(DFACT IDX+IDX(ICO2))
DFACTMEA = B(DFACT IDX+IDX(IMEA))
EFACT = B(EFACT IDX+IDX(ICO2))
```

```
CO2DW = 0.00000235D0*EXP(-2119D0/T)
    CO2D = CO2DW * (MUW / MUMX)**(0.8D0)*((T/313.15)**(EFACT))
    CO2D = CO2D * DFACTCO2
   CO2D = ((DFACTCO2)**2)/DFACTMEA * (MUW/MUMX)**0.8
    CO2D = CO2D*(T/313.15)**(EFACT)
   MEAD = (1/((MUMX/MUW)**0.8D0))*((T/313.15)**(EFACT))
   MEAD = MEAD * DFACTMEA
   DO 200 I = 1, N
      DO 100 J = 1, N
        IF (I.EQ.J) THEN
          OBIN(I,J) = 0D0
        ELSE
          QBIN(I,J) = MEAD
          IF (I.EO.ICO2)OBIN(I.J) = CO2D
          IF (J.EQ.ICO2)QBIN(I,J) = CO2D
          IF (I.EQ.IN2)QBIN(I,J) = CO2D
          IF (J.EQ.IN2)QBIN(I,J) = CO2D
        END IF
      CONTINUE
100
200 CONTINUE
```

2.1.5 Reaction Kinetics Model

The template to be used for the reaction kinetics model is titled 'usrknt.f', which is designed specifically for use with reaction kinetics in rate-based columns (REACT-DIST type reaction). The following code should be placed at the end of the 'DECLARE VARIABLES USED IN DIMENSIONING' section, after the code lines EQUIVALENCE (RMISS, USER RUMISS) & EQUIVALENCE (IMISS, USER IUMISS):

```
#include "dms_rglob.cmn"
#include "dms_lclist.cmn"
#include "pputl_ppglob.cmn"
#include "dms_ipoff3.cmn"
#include "dms_plex.cmn"
EQUIVALENCE(IB(1),B(1))
```

The following code should be placed in the 'DECLARE ARGUMENTS' section without deleting the existing code:

```
INTEGER I,K,FN,L_GAMMA,L_GAMUS,GAM,US,DMS_KFORMC,KPHI,KER
INTEGER DMS_ALIPOFF3,IHELGK
REAL*8 B(1)
REAL*8 N_H2O,N_CO2,N_MEA,N_MEAH,N_MEAC,N_HCO3
REAL*8 PHI(100),DPHI(100),GAMMA(100),COEFFCO2,COEFFMEA
REAL*8 ACCO2,ACMEA,ACH2O,ACMEAH,ACMEAC,ACHCO3,R,STOI(100),LNRKO
REAL*8 DUM,KEQ1,KEQ2,RXNRATES(100)
```

The following code should be placed in the 'BEGIN EXECUTABLE CODE' section:

```
FN(I) = I+LCLIST\_LBLCLIST

L\_GAMMA(I) = FN(GAM) + I

L\_GAMUS(I) = FN(US) + I
```

```
= DMS_KFORMC('H2O')
 N H20
N CO2 = DMS KFORMC('CO2')
N_MEA = DMS_KFORMC('C2H7NO')
N_MEAH = DMS_KFORMC('C2H8NO+')
N_MEAC = DMS_KFORMC('C3H6NO3-')
N HCO3 = DMS KFORMC('HCO3-')
 T = TLIO
KPHI = 1
CALL PPMON FUGLY(T,P,X,Y,NCOMP,IDX,NBOPST,KDIAG,KPHI,PHI,DPHI,KER)
GAM = DMS ALIPOFF3(24)
DO I=1, NCOMP
GAMMA(I)=1.D0
IF (INT(1).EQ.1) GAMMA(I) = DEXP(B(L GAMMA(I)))
 END DO
US = DMS ALIPOFF3(29)
COEFFCO2 = DEXP(B(L GAMUS(N CO2)))
 COEFFMEA = DEXP(B(L_GAMUS(N_MEA)))
ACCO2 = COEFFCO2*X(N CO2,1)
ACMEA = COEFFMEA*X(N MEA,1)
ACH2O = GAMMA(N H2O)*X(N H2O,1)
ACMEAH = GAMMA(N_MEAH)*X(N_MEAH,1)
ACMEAC = GAMMA(N_MEAC)*X(N_MEAC, 1)
ACHCO3 = GAMMA(N_HCO3)*X(N_HCO3,1)
R = PPGLOB RGAS/1000
DO I=1,100
     STOI(I) = 0D0
 END DO
DO I=1, NCOMP
     IF (IDX(I).EQ.N_MEA) STOI(I)=-2D0
     IF (IDX(I).EQ.N CO2) STOI(I)=-1D0
     IF (IDX(I).EQ.N_MEAH) STOI(I)=1D0
     IF (IDX(I).EQ.N_MEAC) STOI(I)=1D0
 END DO
 LNRKO = RGLOB_RMISS
CALL PPELC_ZKEQ(T,1,1,0,STOI,0D0,NCOMP,IDX,0,1,1,NBOPST,KDIAG,
2 LNRKO,P,IHELGK,DUM,0,0,0)
 KEQ1 = DEXP(LNRKO)
DO I=1,100
     STOI(I) = 0D0
 END DO
DO I=1, NCOMP
     IF (IDX(I).EQ.N_MEA) STOI(I)=-1D0
     IF (IDX(I).EQ.N_CO2) STOI(I)=-1D0
```

```
IF (IDX(I).EQ.N_H20) STOI(I)=-1D0
     IF (IDX(I).EQ.N_MEAH) STOI(I)=1D0
     IF (IDX(I).EQ.N HCO3) STOI(I)=1D0
 ENDDO
 LNRKO = RGLOB RMISS
 CALL PPELC ZKEO(T,1,1,0,STOI,0D0,NCOMP,IDX,0,1,1,NBOPST,KDIAG,
2 LNRKO,P,IHELGK,DUM,0,0,0)
 KEO2 = DEXP(LNRKO)
RXNRATES(1) = REAL(1)*DEXP(-REAL(3)/R*(1/TLIO-1/298.15))*
2 (ACMEA**2*ACCO2-ACMEAC*ACMEAH/KEQ1)
 RXNRATES(2)=REAL(2)*DEXP(-REAL(4)/R*(1/TLIQ-1/298.15))*
2 (ACMEA*ACCO2-ACMEAH*ACHCO3/(KEQ2*ACH2O))
 DO K=1,NRL(1)
     RXNRATES(K) = RXNRATES(K)*HLDLIQ
     RATEL(K) = RXNRATES(K)
 END DO
DO I=1, NCOMP
     RATES(I)=0.D0
 END DO
DO K=1,NRL(1)
     DO I=1, NCOMP
         IF (DABS(STOIC(I,K)).GE.RGLOB_RMIN) RATES(I) = RATES(I) +
             STOIC(I,K)*RXNRATES(K)
     END DO
 END DO
```

The existing RETURN & END statements at the end of the code must be retained.

2.1.6 Interfacial Area Model

The template to be used for the interfacial area model is titled 'usrintfa.f'. The following should be added to the section stating 'Declare local variables used in the user correlations':

```
REAL*8 Aa, Bb
```

Remove the equations defining the variable 'dTemp' and replace with the following:

```
Aa = REAL(2)
Bb = REAL(3)
dTemp = Aa*((WeL*FrL**(-1/3))**Bb)
```

The existing RETURN & END statements at the end of the code must be retained.

2.1.7 Holdup Model

The template to be used for the liquid and vapor holdup in the RateSep routine is titled 'usrhldup.f'. No additional variable names need to be declared. Remove the code between the statements {IF (COLTYP .EQ. 1) THEN} and {ELSE IF (COLTYP .EQ. 2) THEN}. Insert the following replacement code:

```
IF (USRCOR .EQ. 1) THEN
RHOL = AVMWLI*DENMXL
```

```
UL = FRATEL/DENMXL/TWRARA

HT=REAL(1)*(3.185966*(VISCML/RHOL)**0.3333*(UL))

**REAL(2)

LHLDUP = HT * TWRARA * HTPACK
    VHLDUP = (1D0 - HT - VOIDFR) * TWRARA * HTPACK
END IF
```

The existing RETURN & END statements at the end of the code must be retained.

2.1.8 Creation of dll and opt files

Once the updated Fortran subroutines are ready to be implemented in Aspen, open "Customize Aspen Plus V10". Within the simulation window, navigate to the directory containing all the updated of files. Enter aspcomp *.f which creates a .obj file for each of file in the current directory. An obj file is a more compiled version of of files that Aspen can use. Once the obj files are created, enter asplink ccsi10, which will create a dll file named 'ccsi10.dll' in the current directory. The 'ccsi10.dll' file is called within the 'ccsi.opt' file distributed with the model. The opt file may be created as a text file by entering the name of the dll file that it points to, and changing the file extension to opt. The opt file is specified within the Aspen model for accessing the Fortran subroutines. For users who choose not to create the dll file, a version will be provided with the release notes in the GitHub repository.

Note: In case of a compilation error, the Aspen Plus Linker Diagnostics (.ld) file created within the current directory needs to be opened using Notepad, and the link message displayed towards the end of the file needs to be analyzed, followed by an appropriate code correction in the .f files. The process of creating a new dll needs to be repeated thereafter.

2.2 Predicting System VLE

- 1. Place the "CCSI_MEAModel.bkp" file and the supporting files "ccsi.opt" and "ccsi10.dll" in the same directory. Open the "CCSI_MEAModel.bkp" file. When prompted with the "Column Sizing/Rating Detected" box, select the "Use Legacy Hydraulics" option. If the Model Palette is not visible, it may be selected from the "View" tab at the top of the window. In the Model Palette, navigate to the "Manipulators" tab and then select "Mult" to create a multiplier block, which will be referred to by its default name "B1." Double-click "B1" and then set the multiplication factor to "1." Add an inlet stream to the block by clicking "Material" in the Model Palette, the red arrow on the inlet of B1, and then elsewhere in the flowsheet. Repeat the procedure for the outlet stream of B1. Name the inlet and outlet streams as "IN" and "OUT," respectively. **Note:** The streams may be renamed by double clicking the default name and typing the new name.
- 2. Double-click "IN" and configure it as follows:
 - a. Select "Temperature" and "Vapor Fraction" as the "Flash Type" specifications.
 - b. Temperature: 40°C.
 - c. Vapor Fraction: 0.0001.
 - d. Select "Mass-flow" in "gm/hr" as the composition basis. Set the values for "H₂O" and "MEA" as "7" and "3," respectively.

"component," and "mol/hr" as the "units." Under "Manipulated variable limits," specify "0.0005" and "0.03" as the "lower" and "upper limits," respectively, and "10" as the "number of points." Navigate to the "Define" tab and then create a new measured variable named "PCO2." Under "Edit selected variable," select "Streams" as the "category," "Stream-Prop" as the "type," "IN" as the "stream," and "PPCO2" as the "prop set". Change the units to "kPa". Navigate to the "Tabulate" tab and then click "Fill Variables." Navigate to the "Options" tab and select the "Do not execute base case," option under "Execution options."

4. Run the simulation by clicking the "Run" arrow or pressing "F5." The results of the "PCO2" sensitivity block should be consistent with what is shown in Table 2. **Note:** All of the warnings that appear in the "Control Panel" while running the simulation may be ignored.

Row/ Case	Status	CO2 MOLEFLOW (MOL/HR)	PCO2 (KPA)
1	OK	0.0005	2.24E-5
2	OK	0.003778	0.00097
3	OK	0.007056	0.00363
4	OK	0.010333	0.00955
5	OK	0.013611	0.02339
6	OK	0.016889	0.06171
7	OK	0.020167	0.21295
8	OK	0.023444	1.47244
9	OK	0.026722	18.5729
10	OK	0.03	103.162

Table 2: Results of PCO₂ Sensitivity Block

5. From this example, the vapor-liquid equilibrium (VLE) of the ternary MEA-H₂O-CO₂ system as a function of temperature and CO₂ loading may be determined for 30 wt% MEA. The CO₂ loading (mol CO₂/mol MEA) may be calculated by multiplying the CO₂ molar flow by the molecular weight of MEA and dividing by the mass flow of MEA. For example,

$$\frac{0.0005\,mol\,CO_2}{hr} \times \frac{61.08308\,g\,MEA}{mol\,MEA} \times \frac{hr}{3\,g\,MEA} \approx 0.0102\,mol\,CO_2/mol\,MEA \tag{6}$$

Following this procedure and evaluating the sensitivity block for temperatures of 80 and 120°C, by changing the temperature of the stream "IN" and re-running the simulation, a plot similar to Figure 1 may be generated.

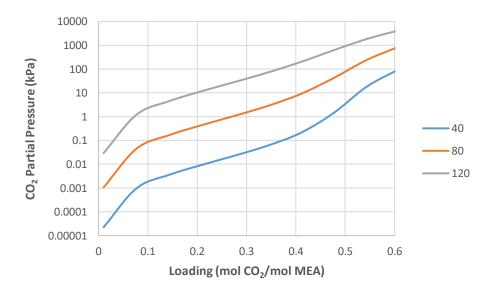


Figure 1: CO₂ partial pressure as a function of loading and temperature (30 wt% MEA).

2.3 CO₂ Capture Process Simulation Base Case Setup

The base case model that is set up in the file "CCSI_MEAModel.bkp" has operating variables and equipment configurations as specified in Table 3.

Table 3: Variables for Base Case Simulation

Variable	Value		
ABSLEAN Stream (Absorber Solvent Inlet)			
Temperature (°C)	40.97		
Pressure (kPa)	245.94		
Mass Flow (kg/hr)	6803.7		
Component Mole Fractions			
H ₂ O	0.87457		
CO ₂	0.01585		
MEA	0.10958		
GASIN Stream (Absorber Gas	Inlet)		
Temperature (°C)	42.48		
Pressure (kPa)	108.82		
Mass Flow (kg/hr)	2266.1		
Component Mass Fractions			
H₂O	0.04623		
CO ₂	0.17314		
N ₂	0.71165		
O ₂	0.06898		
Absorber			
Intercooler #1 Flowrate (kg/hr)	7364.83		
Intercooler #1 Return Temperature (°C)	40.13		
Intercooler #2 Flowrate (kg/hr)	7421.57		
Intercooler #2 Flowrate (°C)	43.32		
Absorber Top Pressure (kPa)	108.82		
Absorber Packing Diameter (m)	0.64135		
Absorber Packing Height (ft)	60.7184		
Regenerator			
Inlet Temperature (°C)	104.81		
Inlet Pressure (kPa)	183.87		
Top Pressure (kPa)	183.7		
Reboiler Duty (kW)	430.61		
Packing Diameter (in)	23.25		
Packing Height (ft)	39.6837		

The variables described in Table 3 may be varied within reason, although abrupt changes in certain variables may results in failure of the simulation to converge. In the simulation provided in the example file, the variables for the "ABSLEAN" and "GASIN" streams can be located by double-clicking the respective streams. The variables for the absorber intercoolers can be located from the navigation pane by selecting "Blocks" \rightarrow "ABSORBER" \rightarrow "Configuration" \rightarrow "Pumparounds," and the first and second intercoolers are referred to as "P-1" and "P-2," respectively. The top pressure of the absorber and regenerator can be located by double-clicking the "ABSORBER" and "REGEN" blocks and selecting the "Pressure" tab. Moreover, the reboiler duty for "REGEN" is located under the "Configuration" tab. The column packing diameters and height can be located by selecting "Blocks" \rightarrow "ABOSRBER" or "REGEN" \rightarrow "Sizing and Rating" \rightarrow "Packing Rating" \rightarrow "1" \rightarrow "Setup." The values of the regenerator inlet pressure and temperature are specified in the "PUMP" and "EXCHANGE" blocks, respectively.

Note: A sensitivity block, referred to as "FLOW" in the simulation, is used to set the flowrate of the inlet solvent stream, as the simulation will not automatically converge for such a low flow rate.

2.4 CO₂ Capture Process Simulation Example

In this example, the CO₂ capture process, which includes the absorber and regenerator columns, is evaluated for two sets of operating conditions.

- 1. Open the "CCSI_MEAModel.bkp" file. In the navigation pane, right-click "Blocks," select "Activate," right-click "Streams," and then select "Activate." Run the simulation. **Note:** All streams and blocks have been deactivated to reduce the time required to obtain the results for the test in Section 2.2 Predicting System VLE. If block "B1" and streams "IN" and "OUT" have already been created in the same file, they need to be deactivated by right-clicking them and selecting "Deactivate" before activating all streams with the aforementioned procedure.
- 2. In the flowsheet, right-click stream "ABSRICH," select "Results," and then select "STRIPOUT" from the drop-down arrow at the top of the right column. Ensure that the results obtained match those given in Table 4, noting that only selected rows are included in the table.

Mole Flow mol/hr	ABSRICH	STRIPOUT
H2O	259990	256374
CO2	0.342210	0.976323
MEA	8702.45	26273.15
MEA+	12183.51	3270.13
MEACOO-	11816.96	3152.56
HCO3-	366.55	117.57
N2	33.63	1.64E-20
O2	5.49	8.18E-17
Temperature C	52.01	120.94
Pressure kPa	108.82	183.7
Enthalpy J/kmol	-301827235	-281379164

Table 4: Selected Stream Table Results

- 3. Reinitialize the simulation by clicking "Reset" or pressing "Shift+F5," and then selecting "OK." In the navigation pane, navigate to "Blocks" → "Absorber" → "Configuration" → "Pumparounds" → "P-1," and then change the "flow rate" to "3000 kg/hr." Navigate to "P-2" and then change the "flow rate" to the same value.
- 4. Navigate to "Model Analysis Tools" and activate the "FLOW" sensitivity block, which is used to determine the CO₂ capture percentage in the absorber and the required reboiler duty for the stripper as a function of the lean solvent flowrate. Execute the model, navigate to the results of the sensitivity block, and verify that the results are similar to those shown in Figure 2.

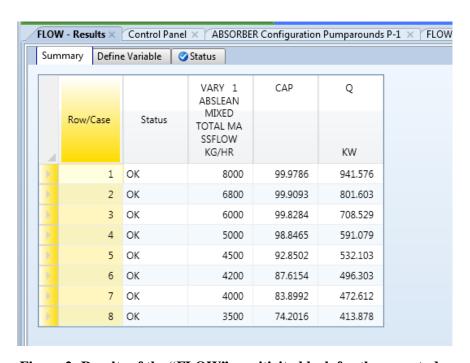


Figure 2: Results of the "FLOW" sensitivity block for the case study.

5. Navigate to "Blocks" → "Absorber" → "Profiles" and then highlight the columns labeled "Vapor Temperature" and "Liquid Temperature." Under "Plot" on the "Home" tab, select "Custom," and then verify that the resulting plot resembles Figure 3. **Note:** These temperature profiles correspond to the last simulation executed (Case 8).

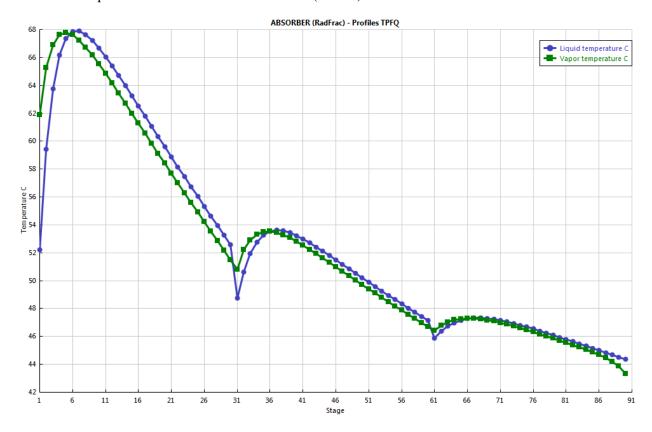


Figure 3: Absorber temperature profile for the case study.

6. Navigate to "Blocks" → "Regen" → "Profiles" and then repeat the procedure described in Step 5. Verify that the temperature profile resembles what is shown in Figure 4.

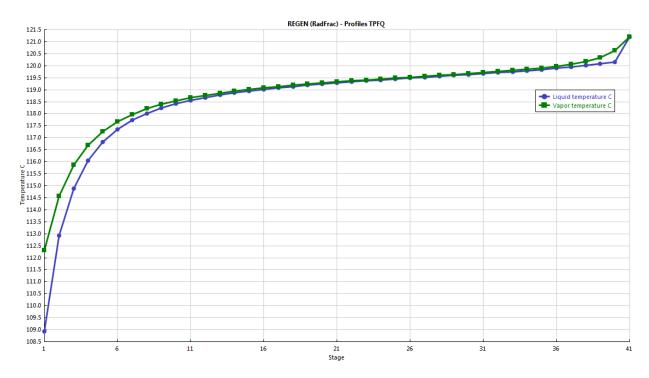


Figure 4: Regenerator temperature profile for the case study.

3.0 USAGE INFORMATION

3.1 Environment/Prerequisites

This product requires Aspen Plus V10 or newer with an Aspen Rate-Based Distillation license.

3.2 Support

Support can be obtained from the email support list ccsi-support@acceleratecarboncapture.org or by opening an issue at our GitHub repository: https://github.com/CCSI-Toolset/MEA_ssm/issues

4.0 REFERENCES

- 1. Morgan, J.C.; Bhattacharyya, D.; Tong, C.; Miller, D.C., Uncertainty Quantification of Property Models: Methodology and its Application to CO₂-Loaded Aqueous MEA Solutions. *AIChE Journal* 2015, 61, (6), 1822-1839.
- 2. Plaza, J.M. Modeling of Carbon Dioxide Absorption Using Aqueous Monoethanolamine, Piperazine, and Promoted Potassium Carbonate. The University of Texas at Austin, 2012.
- 3. Mathias, P.M.; Gilmartin, J.P., Quantitative Evaluation of the Effect of Uncertainty in Property Models on the Simulated Performance of Solvent-Based CO₂ Capture. *Energy Procedia* 2014, 63, 1171-1185.
- 4. Billet, R., Schultes, M., Prediction of mass transfer columns with dumped and arranged packings: updated summary of the calculation method of Billet and Schultes. *Chem Eng Res Des* 1999, 77, 498-504.
- 5. Tsai, R.E. Mass transfer area of structured packing. The University of Texas at Austin, 2010.
- 6. Dugas, R.E. Carbon Dioxide Absorption, Desorption, and Diffusion in Aqueous Piperazine and Monoethanolamine. The University of Texas at Austin, 2009.
- 7. Tobiesen, F.A.; Svendsen, H. F.; Juliussen, O., Experimental validation of a rigorous absorber model for CO₂ postcombustion capture. *AIChE Journal* 2007, 53, 846-865.