



DE LA RECHERCHE À L'INDUSTRIE

# Computing transport coefficients from the atomic scale

Long-time multi-scale simulations of activated events: from theory to practice

T. Schuler

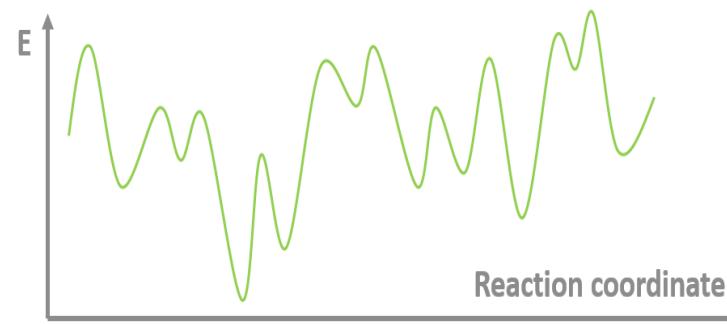
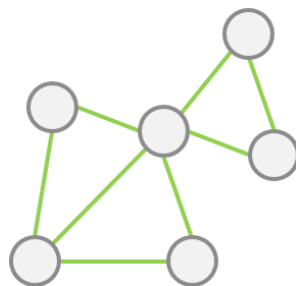
Assume we know our transitions and the associated rates :

$$\omega_{nm} = \nu_{nm} \exp \left( -\frac{G^{mig}(n \rightarrow m)}{k_B T} \right)$$



How do I get physical insight at a higher scale from these transitions ?  
e.g. diffusion coefficients

$$D_\alpha = D_\alpha^0 \exp \left( -\frac{Q_\alpha}{k_B T} \right)$$

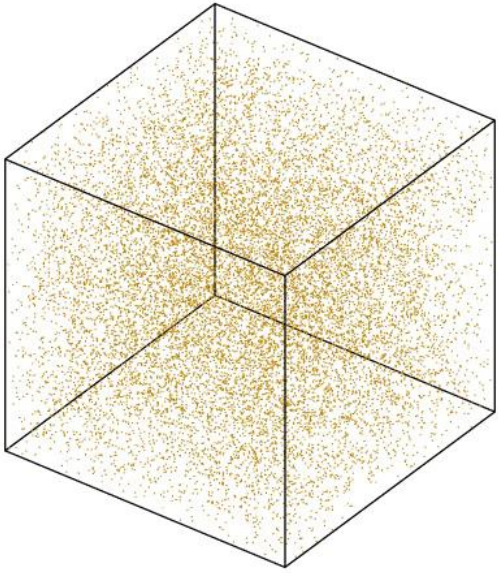


Such representation is not enough for diffusion coefficient.

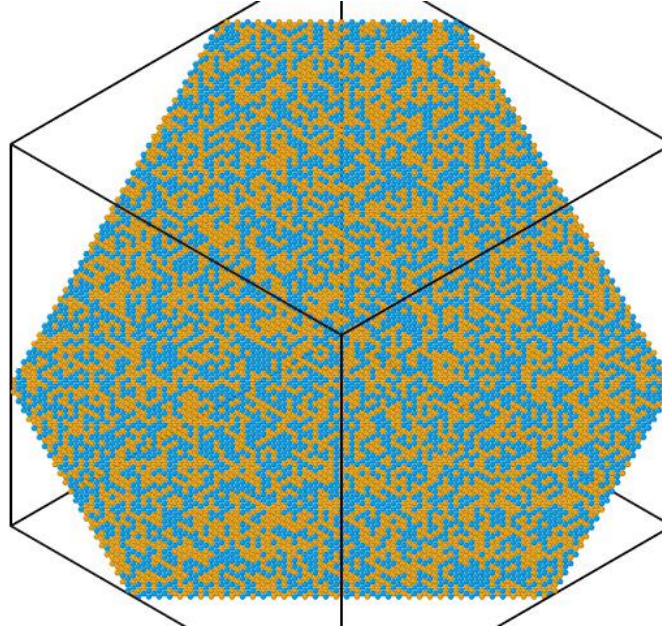
➤ We need to add information about how atoms move in (real) space during each transition.

We will focus on crystals :

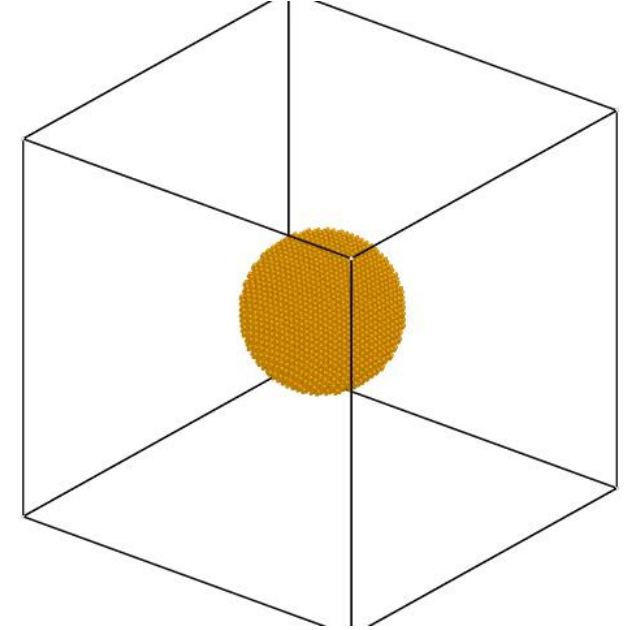
- easier to discretize (manageable number of configurations )
- defect mediated diffusion with rather small number defect concentrations (local events)
- yet this introduces the so-called « kinetic correlations »



*Nucleation-Growth-Coalescence  
(Classical theory of nucleation)*



*Spinodal decomposition  
(Cahn-Hilliard equations)*



*Driven alloys  
(Ballistic mixing under irradiation)*

Other coarse-grain simulations where atom redistribution is of interest:

- Object Kinetic Monte Carlo
- Cluster Dynamics
- Phase Field
- Finite Elements

**Material science needs  
diffusion/transport  
coefficients**

Difficult to measure experimentally

- Long experiments
- Use of radioactive tracers

Diffusion coefficients are not enough

=> Onsager (or transport coefficients)

- Matrix of coefficients
- Only measured in a small number of systems (Anthony)

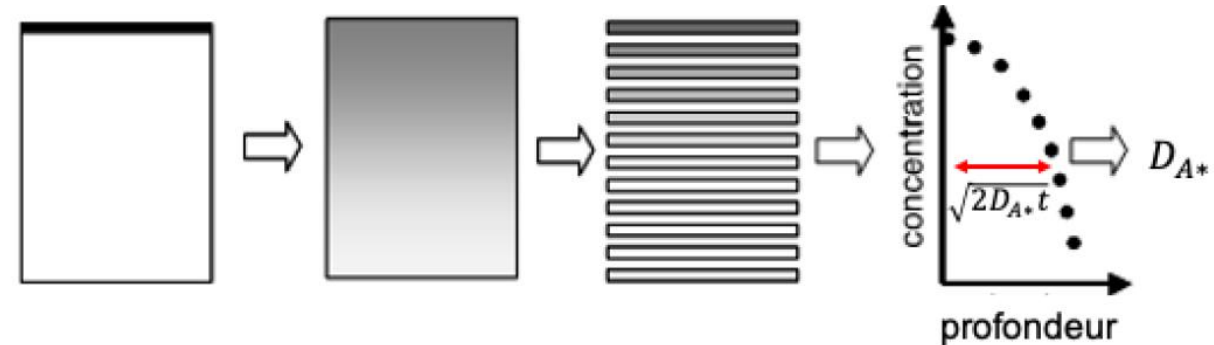
*Onsager equation*

$$\vec{J}_i = - \sum_j L_{ij} \vec{\nabla} \mu_j$$

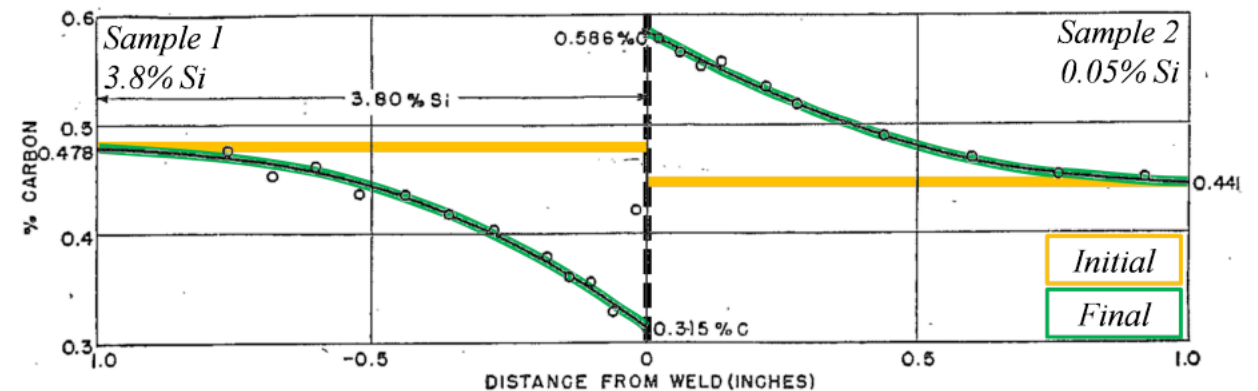
(matrix formulation)

$$\begin{pmatrix} J_A \\ J_B \\ J_C \end{pmatrix} = - \begin{pmatrix} L_{AA} & L_{AB} & L_{AC} \\ L_{BA} & L_{BB} & L_{BC} \\ L_{CA} & L_{CB} & L_{CC} \end{pmatrix} \begin{pmatrix} \nabla \mu_A \\ \nabla \mu_B \\ \nabla \mu_C \end{pmatrix}$$

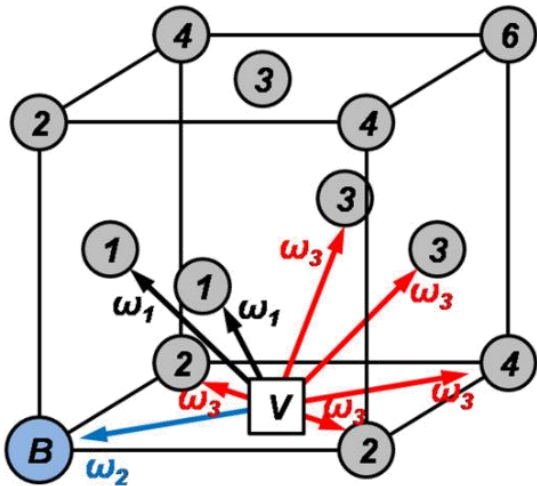
*Tracer diffusion experiment*



*Darken's experiment*



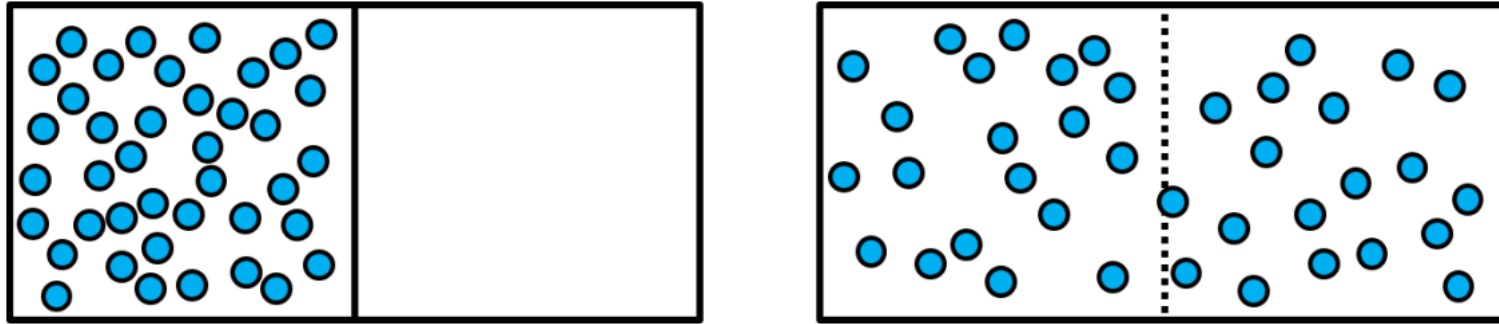
$$\vec{J}_i = - \sum_j L_{ij} \vec{\nabla} \mu_j$$



Jump	$E_m(if)$	$v_{if}^d$
$\omega_0^I$	0.128	2.35
$\omega_0^V$	0.717	3.47
$\omega_{ex}^V$	0.473	7.26
$\omega_{11}^V$	0.945	4.68
$\omega_{12}^V$	0.830	5.10
$\omega_{13}^V$	0.805	5.32
$\omega_{14}^V$	0.747	5.22
$\omega_{23}^V$	0.793	5.16
$\omega_{25}^V$	0.673	5.03
$\omega_{33}^V$	0.699	4.86
$\omega_{34}^V$	0.776	4.83
$\omega_{35}^V$	0.676	4.96
$\omega_{36}^V$	0.737	5.01
$\omega_{37}^V$	0.717	5.01
$\omega_{45}^V$	0.665	4.88
$\omega_{47}^V$	0.746	4.95
$\omega_{48}^V$	0.709	4.94

- ☐ Phenomenological approach : Fick's law
- ☐ 1-D random walk model
- ☐ Diffusion coefficient  
vs. equilibrium fluctuations of atomic positions
- ☐ Kinetic correlation
- ☐ Self-Consistent Mean-Field (SCMF) theory





*Particles perform a random walk in the available volume*

*Fick's first law*

$$\vec{J}_\alpha = -D_\alpha \vec{\nabla} C_\alpha$$

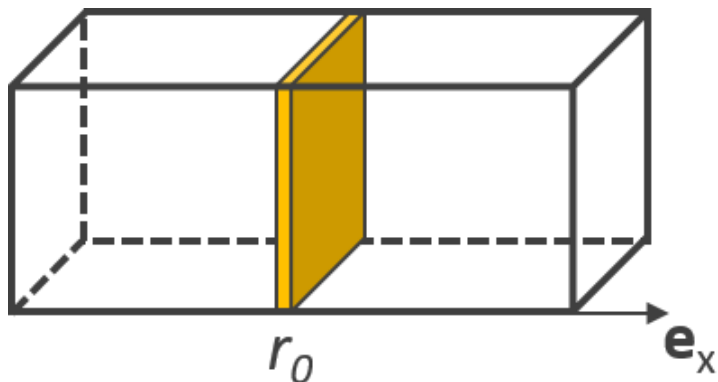
At the macroscopic scale, diffusion is an irreversible phenomenon.  
At the microscopic scale, diffusion is reversible.

Matter conservation law  
(continuity equation)

$$\frac{\partial C_A(\mathbf{r}, t)}{\partial t} + \nabla \cdot \mathbf{J}_A = 0$$

Diffusion equation  
(2<sup>nd</sup> Fick law)

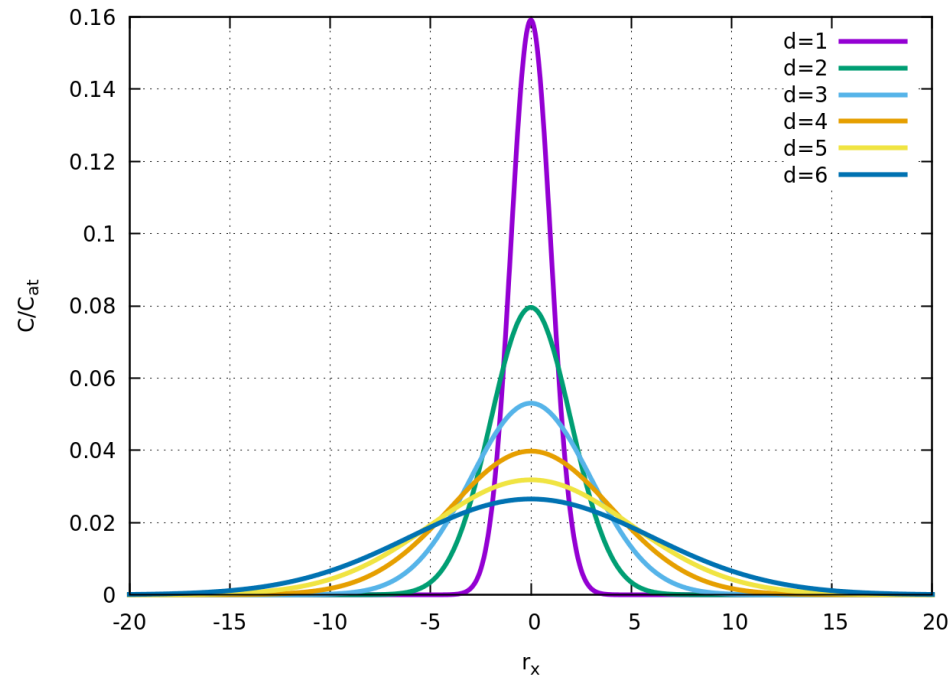
$$\frac{\partial C_A(\mathbf{r}, t)}{\partial t} = D_A \Delta C_A(\mathbf{r}, t)$$



Solution of the diffusion equation:

$$C_A(r_x, t) = \frac{C_{at}}{\sqrt{2\pi}\delta_x} \exp\left(\frac{-(r_x - r_0)^2}{2\delta_x^2}\right)$$

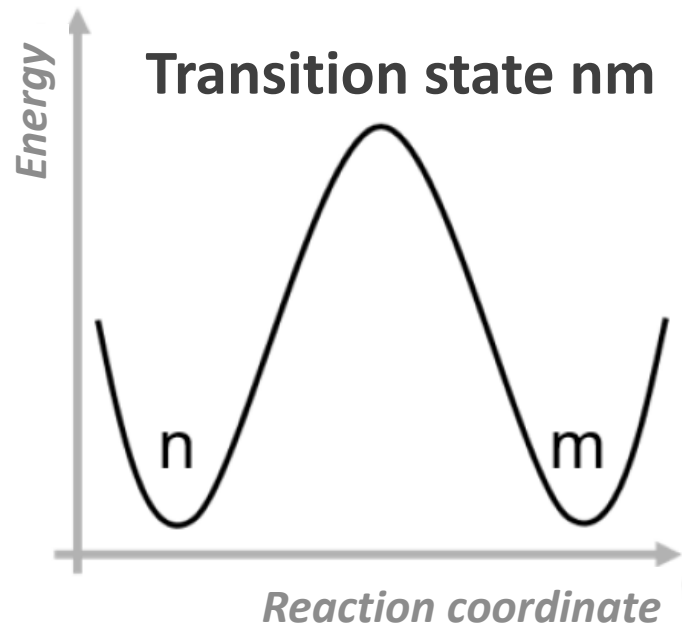
$$\delta_x = \sqrt{2D_A^x t}$$



*Various interpretations of  $C_A$*

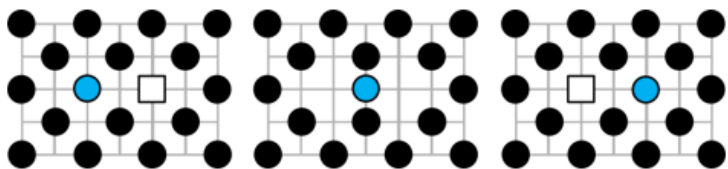
- Time average
- Ensemble average
- Average over space

$$\omega_{nm} = \nu_{nm} \exp \left( -\frac{G^{mig}(n \rightarrow m)}{k_B T} \right) = \nu_{nm} \exp \left( -\frac{G_{nm}^{TS} - G_n^{TS}}{k_B T} \right)$$

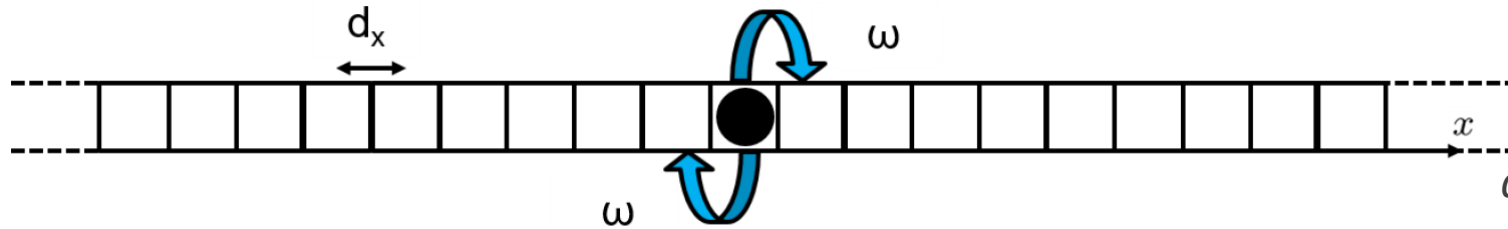


*Additional assumption :*

If migration energies are large enough compared with  $k_B T$ , the system has **no memory** in the sense that the probability to perform a given jump will not depend on the preceding jumps. Such process is called a **Markov process**.







$x$  : diffusion direction

$d_x$  : distance between two sites

$r_x$  : position of the particle

$\omega$  : jump frequency to a neighboring site

$n$  : total number of jumps performed during time  $t$

$p$  : probability to jump to the right

$p_n(k)$  : probability of performing  $k$  jumps to the right for a total of  $n$  jumps

Link between our problem and the classical binomial problem

$$\begin{cases} n = 2\omega t \\ r_x = d_x(2k - n) \\ p = 1/2 \end{cases}$$

$$\langle r_x \rangle = 0 \quad \langle r_x^2 \rangle = 2d_x^2\omega t$$

$$\delta_x = \sqrt{\langle r_x^2 \rangle - \langle r_x \rangle^2} = \sqrt{2d_x^2\omega t}$$

Reminder: characteristic diffusion length

$$\delta_x = \sqrt{2D_\alpha^x t}$$

$$D_\alpha^x = d_x^2\omega$$

### Results for the binomial law

$$p_n(k) = \binom{n}{k} p^k (1-p)^{n-k}$$

$$\langle k^m \rangle = \sum_{k=0}^n k^m p_n(k)$$

$$\langle k \rangle = np$$

$$\langle k^2 \rangle = n^2 p^2 + np(1-p)$$

## The Einstein-Smoluchowski formula for 3D systems

$$D_{\alpha} = \frac{\langle \vec{R}^2 \rangle}{6\tau}$$

For a random walk where all transitions have equivalent jump length  
( $z$  is the coordination number of the lattice)

$$\langle \vec{R}^2 \rangle = n_{\tau} \times \vec{d}^2 = z\omega\tau\vec{d}^2$$

For a cubic lattice (SC, BCC, FCC) where atoms jump by first nearest neighbor jumps, we always have the following relation ( $a$  is the lattice parameter) :

$$\frac{z\vec{d}^2}{6} = a^2 \quad \Rightarrow \quad D_{\alpha} = a^2\omega$$

Tracer diffusion coefficient

Einstein-Smoluchowski formula

$$D_\alpha = \frac{\langle \vec{R}^2 \rangle}{6\tau} = \frac{\left\langle \left( \sum_{i=0}^{\tau/\delta t} \vec{R}_i \right)^2 \right\rangle}{6\tau}$$

We can average the result obtained for a tracer atom on a population of  $N_\alpha$  traced atoms

$$D_\alpha = \frac{1}{N_\alpha} \sum_{k=1}^{N_\alpha} \frac{\langle \vec{R}(k)^2 \rangle}{6\tau} = \frac{\left\langle \sum_{k=1}^{N_\alpha} \left( \sum_{i=0}^{\tau/\delta t} \vec{R}_i(k) \right)^2 \right\rangle}{6\tau N_\alpha}$$

A tracer diffusion coefficient characterizes the mobility of a **single particle**.

Transport coefficient

Allnatt formula

$$L_{\alpha\beta} = \frac{\langle \vec{R}(\alpha) \vec{R}(\beta) \rangle}{6\tau V k_B T}$$

Displacement of the group of atoms belonging to species  $\alpha$

$$\vec{R}(\alpha) = \sum_{k=1}^{N_\alpha} \vec{R}(k)$$

$$L_{\alpha\alpha} = \frac{C_\alpha}{k_B T} \frac{\left\langle \left( \sum_{k=1}^{N_\alpha} \sum_{i=0}^{\tau/\delta t} \vec{R}_i(k) \right)^2 \right\rangle}{6\tau N_\alpha}$$

A transport coefficient characterizes the mobility of a **group of particles** (mobility of the center of gravity of atoms belonging to a given species).

In the dilute limit :

$$D_\alpha = k_B T L_{\alpha\alpha} / C_\alpha$$

The correlation coefficient ( $0 \leq f_\alpha \leq 1$ ) quantifies the deviation from random walk.

A coefficient close to 0 relates to strongly correlated diffusion.

A coefficient equal to 1 relates to uncorrelated or random diffusion.

Correlations effects **always slow down** diffusion.

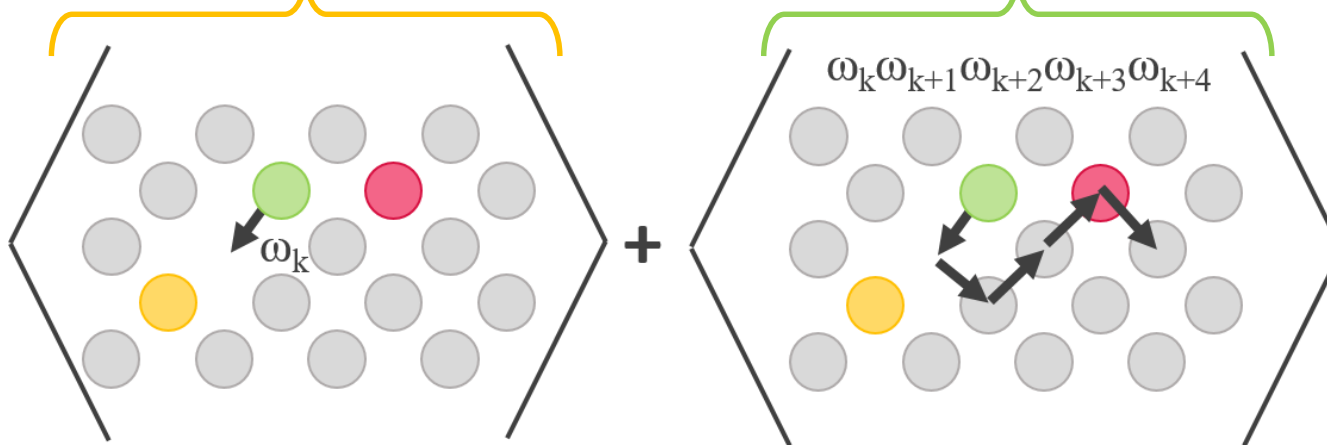
It can be seen as a loss of efficiency in the motion of atoms.

$$f_\alpha = \frac{D_\alpha}{D_\alpha^0}$$

$$D_\alpha = \frac{\langle \vec{R}^2 \rangle}{6\tau} = \frac{\left\langle \left( \sum_i \vec{R}_i \right)^2 \right\rangle}{6\tau} = \frac{\left\langle \sum_i \vec{R}_i^2 + \sum_i \sum_{j \neq i} \vec{R}_i \vec{R}_j \right\rangle}{6\tau} = f_\alpha D_\alpha^0$$

Random walk,  
**Average of individual contributions** from each jump

$$D_\alpha^0 = \frac{\sum_i \langle \vec{R}_i^2 \rangle}{6\tau}$$

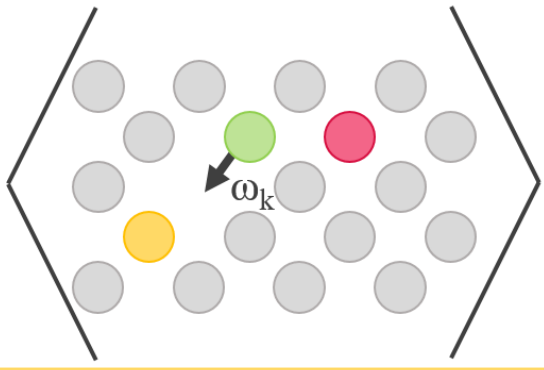


Kinetic correlations accounting for the **probability of each trajectory, a multi-body contribution which is hard to compute**

$$f_\alpha = 1 + 2 \frac{\sum_i \sum_{j>i} \langle \vec{R}_i \vec{R}_j \rangle}{\sum_i \langle \vec{R}_i^2 \rangle}$$

Random walk,  
**Average of individual  
contributions** from  
each jump

$$D_{\alpha}^0 = \frac{\sum_i \langle \vec{R}_i^2 \rangle}{6\tau}$$



$$\frac{\sum_i \langle \vec{R}_i^2 \rangle}{6\tau} = \frac{1}{6\tau} \sum_i P_i \vec{R}_i^2$$

$i$  : trajectory step

$n_i$  : configuration of the system before step  $i$

$m_i$  : configuration of the system after step  $i$

$P_{n_i \rightarrow m_i}$  : probability of jumping from  $n$  to  $m$  (at step  $i$ )

$W_{n_i m_i}$  : jump rate [ $s^{-1}$ ] between configurations  $n$  and  $m$   
(independent of  $i$ , Markov process)

$\theta_n$  : available jumps from configuration  $n$

$\tau_{n_i}$  : average residence time on configuration  $n$

$P_n^{eq}$  : equilibrium configuration probability (Boltzmann statistics)

$G_n$  : Gibbs free energy associated with configuration  $n$

$G_{nm}^{TS}$  : Gibbs free energy at the saddle point between  $n$  and  $m$

$z$  : number of jumps available (coordination number)

$a$  : lattice parameter

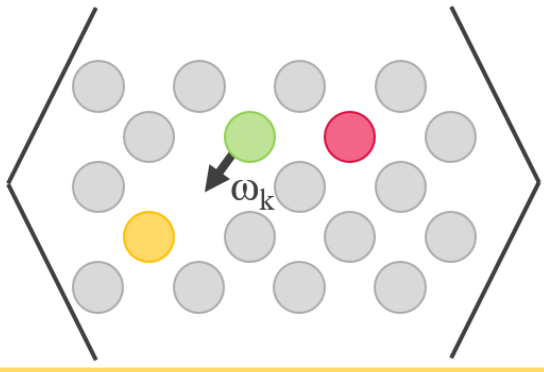
$\mathcal{Z}$  : partition function

$\nu_0$  : attempt frequency [ $s^{-1}$ ]

$$\mathcal{Z} = \sum_n \exp\left(-\frac{G_n}{k_B T}\right)$$

Random walk,  
**Average of individual contributions** from each jump

$$D_{\alpha}^0 = \frac{\sum_i \langle \vec{R}_i^2 \rangle}{6\tau}$$



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$$\mathcal{Z} = \sum_n \exp\left(-\frac{G_n}{k_B T}\right)$$

$$\begin{aligned} \frac{\sum_i \langle \vec{R}_i^2 \rangle}{6\tau} &= \frac{1}{6\tau} \sum_i P_i \vec{R}_i^2 \\ &= \frac{1}{6\tau} \sum_i P_{n_i \rightarrow m_i} \vec{R}_{n_i \rightarrow m_i}^2 \\ &= \frac{1}{6\tau} \sum_i \frac{W_{n_i m_i}}{\sum_{k \in \theta_{n_i}} W_{n_i k}} \vec{R}_{n_i \rightarrow m_i}^2 \end{aligned}$$

$$= \frac{1}{6} \sum_i \frac{\tau_{n_i}}{\tau} W_{n_i m_i} \vec{R}_{n_i \rightarrow m_i}^2$$

$$\xrightarrow{\tau \rightarrow \infty} \frac{1}{6} \sum_n P_n^{eq} \sum_{m \in \theta_n} W_{nm} \vec{R}_{n \rightarrow m}^2$$

**Equilibrium (ergodicity principle)**

$$= \frac{1}{6\mathcal{Z}} \sum_n \exp\left(-\frac{G_n}{k_B T}\right) \sum_{m \in \theta_n} W_{nm} \vec{R}_{n \rightarrow m}^2$$

$$= \frac{1}{6\mathcal{Z}} \sum_n \exp\left(-\frac{G_n}{k_B T}\right) \sum_{m \in \theta_n} \nu_0 \exp\left(-\frac{G_{nm}^{TS} - G_n}{k_B T}\right) \vec{R}_{n \rightarrow m}^2$$

$$= \frac{\nu_0}{6\mathcal{Z}} \sum_n \sum_{m \in \theta_n} \exp\left(-\frac{G_{nm}^{TS}}{k_B T}\right) \vec{R}_{n \rightarrow m}^2$$

$$= \frac{a^2 \nu_0}{\mathcal{Z}} \sum_n \frac{1}{z} \sum_{m \in \theta_n} \exp\left(-\frac{G_{nm}^{TS}}{k_B T}\right)$$

**Cubic system + 1NN jumps**



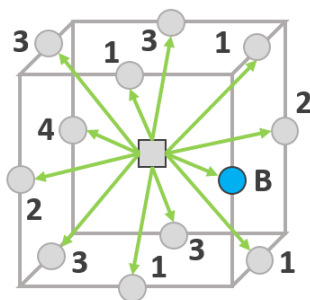
$$D_{\alpha}^0 = \frac{a^2 \nu_0}{z} \frac{\sum_n \sum_{m \in \theta_n} \exp\left(-\frac{G_{nm}^{TS}}{k_B T}\right)}{\sum_n \exp\left(-\frac{G_n}{k_B T}\right)} = \frac{a^2 \nu_0}{z} \frac{\sum_n \sum_{m \in \theta_n} Y_{nm}}{\sum_n Y_n}$$

**Monovacancy :**

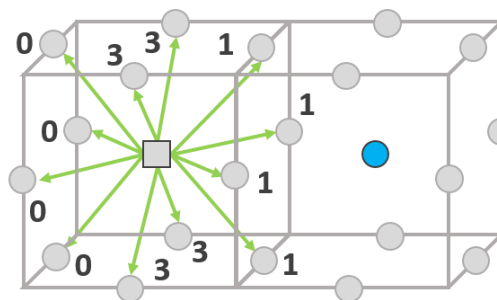
$$D_V^0 = \frac{a^2 \nu_0}{z} \frac{z \exp\left(-\frac{G^{for}(V) + G^{mig}(V)}{k_B T}\right)}{\exp\left(-\frac{G^{for}(V)}{k_B T}\right)} = a^2 \omega_V$$

**Solute vacancy pair**  
**(up to 4nn):**

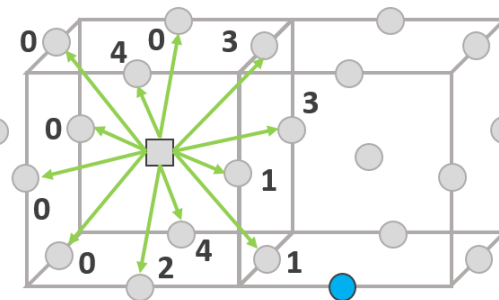
1NN



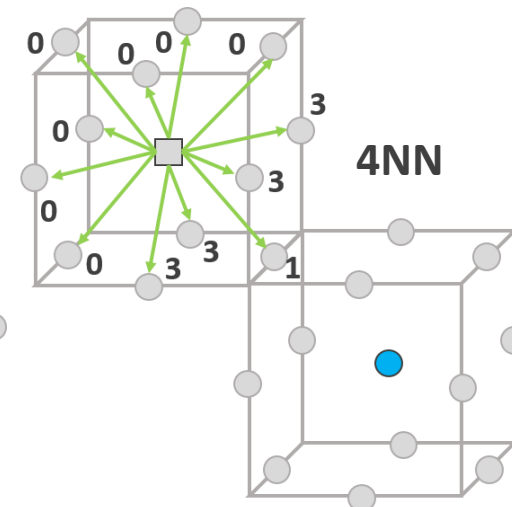
2NN



3NN



4NN



**Denominator**

$$12 \times Y_1$$

$$6 \times Y_2$$

$$24 \times Y_3$$

$$12 \times Y_4$$

**Numerator**

$$12 \times (4Y_{11} + Y_{11B} + 2Y_{12} + 4Y_{13} + Y_{14})$$

$$6 \times (4Y_{21} + 4Y_{23} + 4Y_{20})$$

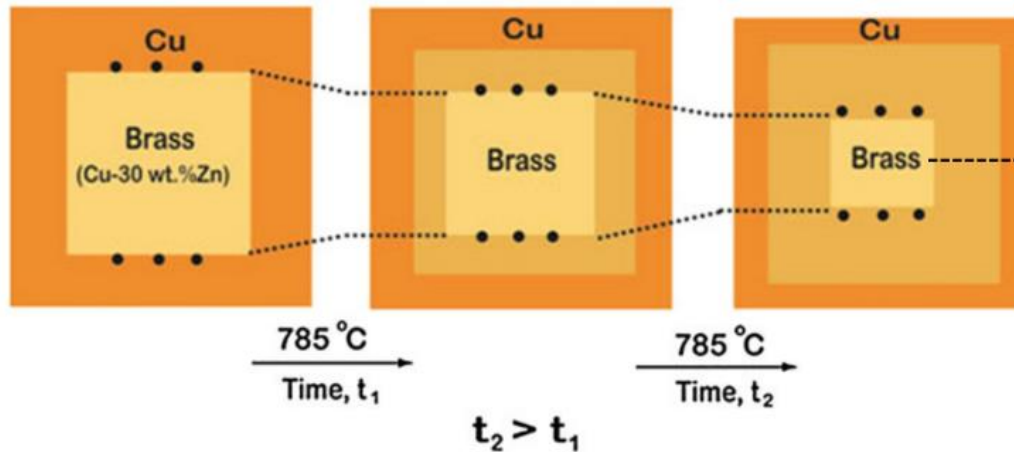
$$24 \times (2Y_{31} + 2Y_{33} + 2Y_{34} + Y_{32} + 5Y_{30})$$

$$12 \times (Y_{41} + 4Y_{43} + 7Y_{40})$$

\*col=Saddle-Point

$$X_V^{eq} = \exp \left( -\frac{G^{for}(V)}{k_B T} \right)$$

### Kirkendall's experiment



From a thermodynamic point of view, equilibrium vacancies result from a competition between the energetic cost of a vacancy (all neighboring atoms lose one of the chemical bonds which ensure the stability of the structure) and the disorder created by vacancies, as one vacancy can be located on any site of the crystal, and this possibility generates configurational entropy. At equilibrium, the site concentration of vacancies is a function of its formation free energy.

Point defect concentrations are monitored at every instant at extended crystal defects (dislocations, joints de grains, surfaces, interfaces).

Point defect concentrations can be increased by placing the system in out-of-equilibrium conditions, for instance under irradiation, after quenching or during severe plastic deformation. Diffusion properties are automatically modified under such conditions.

The correlation coefficient ( $0 \leq f_\alpha \leq 1$ ) quantifies the deviation from random walk.

A coefficient close to 0 relates to strongly correlated diffusion.

A coefficient equal to 1 relates to uncorrelated or random diffusion.

Correlations effects **always slow down** diffusion.

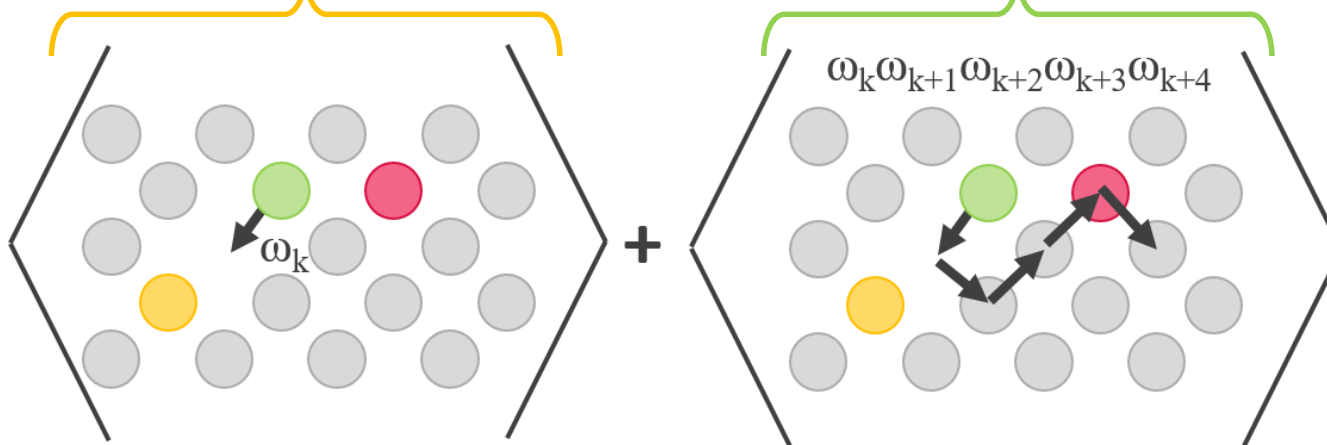
It can be seen as a loss of efficiency in the motion of atoms.

$$f_\alpha = \frac{D_\alpha}{D_\alpha^0}$$

$$D_\alpha = \frac{\langle \vec{R}^2 \rangle}{6\tau} = \frac{\left\langle \left( \sum_i \vec{R}_i \right)^2 \right\rangle}{6\tau} = \frac{\left\langle \sum_i \vec{R}_i^2 + \sum_i \sum_{j \neq i} \vec{R}_i \vec{R}_j \right\rangle}{6\tau} = f_\alpha D_\alpha^0$$

Random walk,  
**Average of individual contributions** from each jump

$$D_\alpha^0 = \frac{\sum_i \langle \vec{R}_i^2 \rangle}{6\tau}$$



Kinetic correlations accounting for the **probability of each trajectory, a multi-body contribution which is hard to compute**

$$f_\alpha = 1 + 2 \frac{\sum_i \sum_{j>i} \langle \vec{R}_i \vec{R}_j \rangle}{\sum_i \langle \vec{R}_i^2 \rangle}$$

The diffusion of an isolated vacancy in a pure metal is non-correlated.

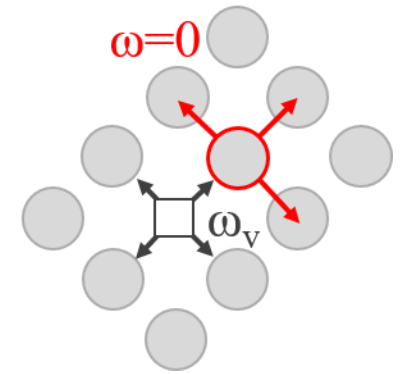
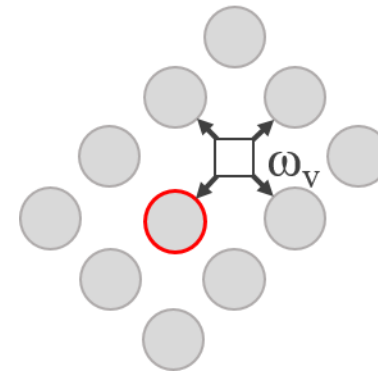
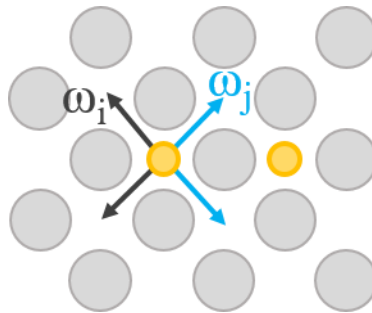
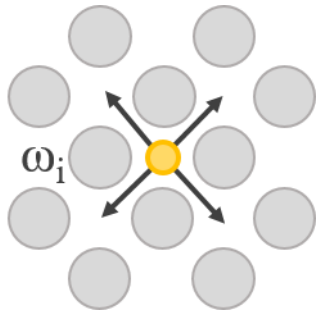
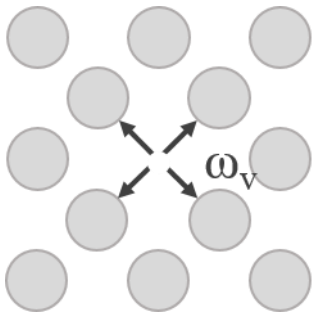
The diffusion of an interstitial atom in a dilute alloy is non-correlated.

The diffusion of a substitutional solute with vacancy-mediated diffusion is correlated.

As soon as there are interactions between atoms and/or defects that are different from the interactions with the matrix atoms, diffusion is generally correlated.

$$f_{\alpha} = 1 + 2 \frac{\sum_i \sum_{j>i} \langle \vec{R}_i \vec{R}_j \rangle}{\sum_i \langle \vec{R}_i^2 \rangle}$$

Let's see if we can understand why...

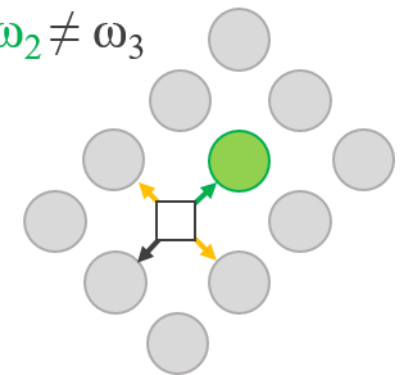
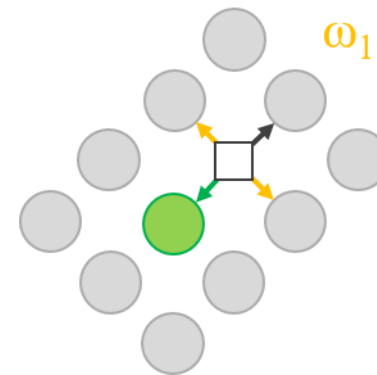


For a tracer atom, the correlation coefficient is purely geometric and it can be estimated to first order as :

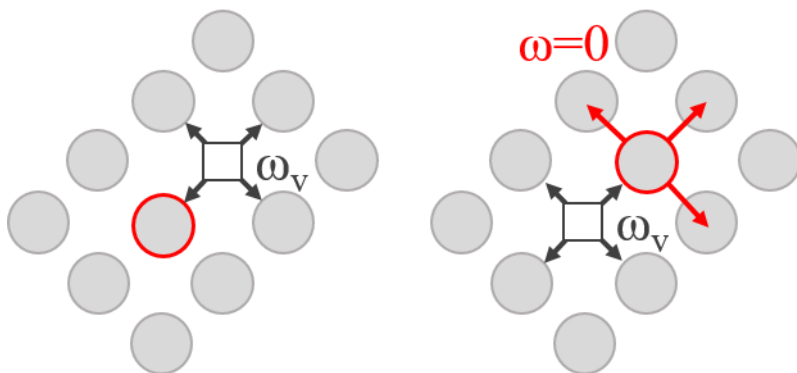
$$f \simeq \frac{z-1}{z+1}$$

#### Lattice

Square	$f_0$ 0.46705
Hexagonal	0.56006
Honeycomb	0.33333
Diamond	0.50000
Simple cubic	0.65310
Body-centered cubic	0.72719
Face-centered cubic	0.78145
Hexagonal compact (basal)	0.78121
Hexagonal compact (non basal)	0.78145

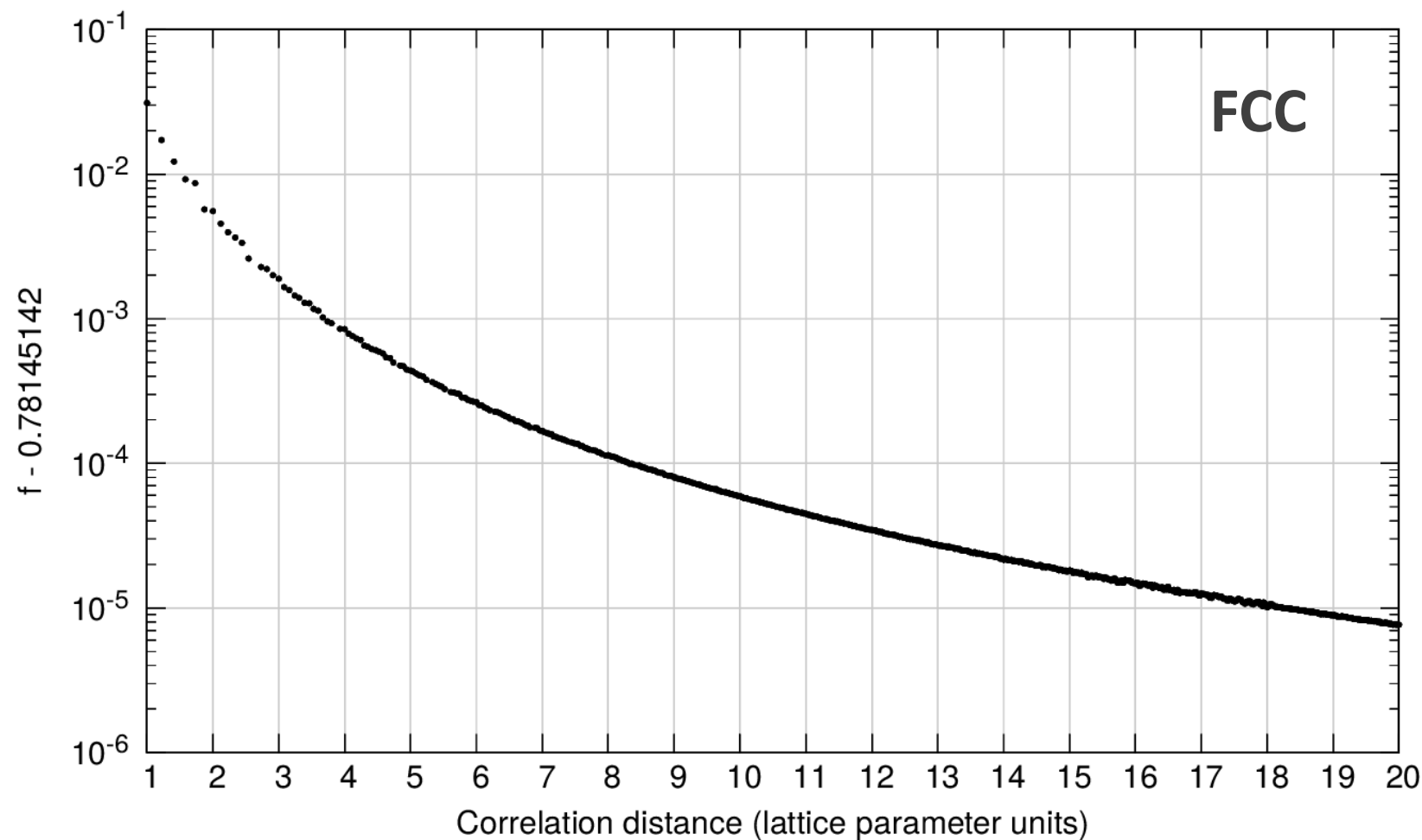


$$f_{\alpha} = 1 + 2 \frac{\sum_i \sum_{j>i} \langle \vec{R}_i \vec{R}_j \rangle}{\sum_i \langle \vec{R}_i^2 \rangle}$$



For a tracer atom, the correlation coefficient is purely geometric and it can be estimated to first order as :

$$f \simeq \frac{z-1}{z+1}$$



Expression for the tracer diffusion coefficient. In this case, we explicitly find a Arrhenius relation

$$D_{\alpha^*} = X_V^{eq} f_0 \frac{zd^2}{6} \omega_0 = \frac{zd^2}{6} f_0 \nu_V \exp \left( -\frac{G^{for}(V) + G^{mig}(V)}{k_B T} \right)$$

Expression for the solute diffusion coefficient

$$D_B = X_V^{eq} f_B \frac{zd^2}{6} \omega_2$$

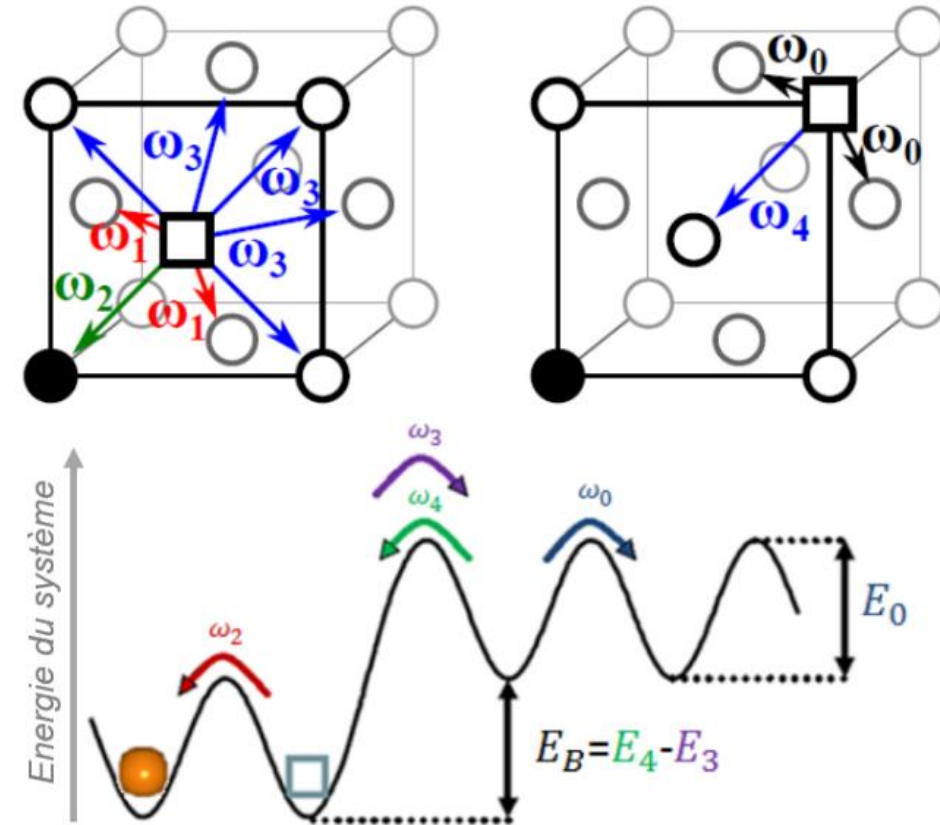
General expression for the solute correlation factor in which appears the competition between the vacancy-solute exchange ( $\omega_2$ ) and the effective vacancy escape frequency  $H$ .

$$f_B = \frac{H}{2\omega_2 + H}$$

In the 5-frequency model for vacancy-solute diffusion in FCC, The escape frequency reads :

$$H = 2\omega_1 + 7F_3(\omega_4/\omega_0)$$

$$7F_3\left(\frac{\omega_4}{\omega_0}\right) = 7 - \frac{10\left(\frac{\omega_4}{\omega_0}\right)^4 + 180.5\left(\frac{\omega_4}{\omega_0}\right)^3 + 927\left(\frac{\omega_4}{\omega_0}\right)^2 + 1341\left(\frac{\omega_4}{\omega_0}\right)}{2\left(\frac{\omega_4}{\omega_0}\right)^4 + 40.2\left(\frac{\omega_4}{\omega_0}\right)^3 + 254\left(\frac{\omega_4}{\omega_0}\right)^2 + 597\left(\frac{\omega_4}{\omega_0}\right) + 436}$$





Out-of-equilibrium probability (weak driving force, hence the first order expansion)

$$P_n = P_n^0 \delta P_n = P_n^0 \exp \left( \frac{\delta \Omega - \nu_n + \sum_{\beta} \delta \mu_n^{\beta}}{k_B T} \right) \simeq P_n^0 \left( 1 + \delta \Omega - \frac{\nu_n}{k_B T} + \sum_{\beta} \frac{\delta \mu_n^{\beta}}{k_B T} \right)$$

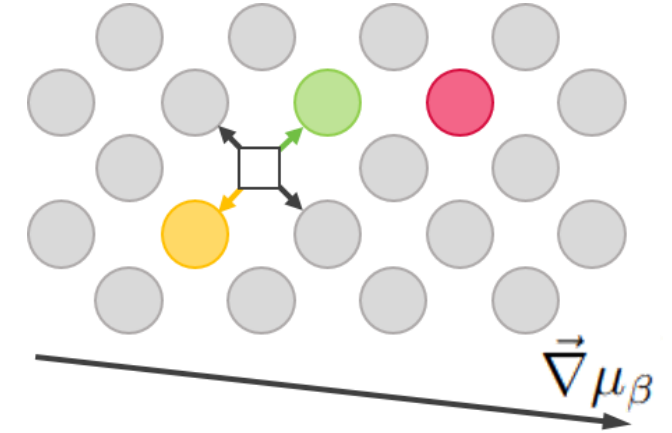
Expression for the out-of-equilibrium flux

$$\vec{J}_{\alpha} = \frac{1}{V} \sum_{n \in V} P_n \sum_m W_{nm} \vec{d}_{nm}^{\alpha} = \frac{1}{V} \sum_{n \in V} P_n^0 \delta P_n \sum_m W_{nm} \vec{d}_{nm}^{\alpha}$$

Master equation

$$\begin{aligned} \forall n, \quad \frac{dP_n}{dt} &= \sum_m P_m W_{mn} - P_n W_{nm} = \sum_m P_m \delta P_m W_{mn} - P_n \delta P_n W_{nm} \\ &= \sum_m P_m^0 W_{mn} (\delta P_m - \delta P_n) = \sum_m P_m^0 W_{mn} \left( \frac{\nu_n - \nu_m + \sum_{\beta} \vec{d}_{nm} \cdot \vec{\nabla} \mu_{\beta}}{k_B T} \right) \end{aligned}$$

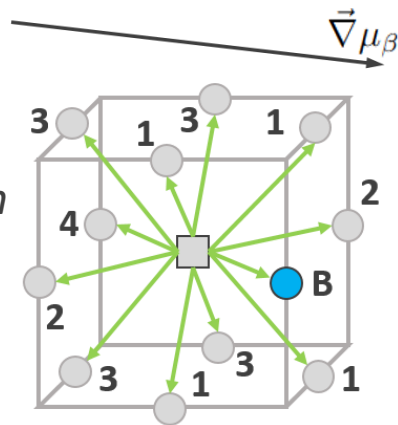
$= 0$  As a consequence of the stationary condition, effective interactions are proportional to driving forces. Therefore the flux is proportional to driving forces which allow us to identify transport coefficients.



$$\forall n, \forall \beta, \sum_{m \in \theta_n} P_m^0 W_{mn} (\nu_n - \nu_m) = - \left( \sum_{m \in \theta_n} P_m^0 W_{mn} \vec{d}_{nm} \right) \cdot \vec{\nabla} \mu_\beta$$

The size of this system of equation we need to solve is a priori gigantic but it can be reduced by using :

- ❑ **Dilute system**, i.e. a small group of traced atoms and/or defects in a homogeneous material
- ❑ **Translation invariance** of the system (homogeneous crystal)
- ❑ **Crystal symmetries** for the out-of-equilibrium system (the chemical potential gradient breaks part of the crystal symmetries)
- ❑ An interaction **cut-off radius** (kinetic radius) which defines how far atoms interact (thermodynamic and « kinetic » interactions)



Vacancy solute  
1NN configuration  
in a FCC crystal;  
CPG along the  
[100] direction

$$2Y_{11}(\nu_1 - 0) + Y_{11B}(\nu_1 + \nu_1) + Y_{12}(\nu_1 - 0) + 2Y_{11}(\nu_1 - \nu_1) + 2Y_{13}(\nu_1 - \nu_3) \\ + 2Y_{13}(\nu_1 - \nu_{3'}) + Y_{14}(\nu_1 - \nu_4) + Y_{12}(\nu_1 - \nu_2) = -\frac{a}{2} (2Y_{11} + Y_{11B} + Y_{12} - Y_{12} - 2Y_{13} - Y_{14}) \nabla \mu_\beta$$

*Then we do the same for all configurations below the cut-off radius, we solve this linear system by matrix inversion and we get the effective interactions which are proportionnal to chemical potential gradients*


Don't want to do this all by hand ?

Use KineCluE !  
(It's free)

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Editor's Choice

### KineCluE: A kinetic cluster expansion code to compute transport coefficients beyond the dilute limit

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**Thank you for your attention**

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