

Structural and stereo isomerism

61. $C_6H_5C \equiv N$ and $C_6H_5N \equiv C$ exhibit which type of isomerism
 (a) Position
 (b) Functional
 (c) Dextro isomerism
 (d) Metamerism
62. Which of the following compounds is not chiral
 (a) $DCH_2CH_2CH_2Cl$
 (b) CH_3CH_2CHDCl
 (c) $CH_3CHDCH_2CH_2Cl$
 (d) $CH_2CHClCH_2D$
63. *cis* and *trans* 2-butene are
 (a) Conformational isomers
 (b) Optical isomers
 (c) Position isomers
 (d) Geometrical isomers
64. Which one of the following is the chiral molecule
 (a) CH_3Cl (b) CH_2Cl_2
 (c) $CHBr_3$ (d) $CHClBr$
65. Cyanide and isocyanide are isomers of type
 (a) Positional (b) Functional
 (c) Tautomer (d) Structural
66. Glucose and fructose are
 (a) Optical isomers
 (b) Functional isomers
 (c) Position isomers
 (d) Chain isomers
67. Which of the following compounds which is an optically active compound
 (a) 1-butanol (b) 2-butanol
 (c) 3-butanol (d) 4-heptanol
68. *d*-tartaric acid and *l*-tartaric acid are
 (a) Enantiomers
 (b) Tautomers
 (c) Diastereoisomers
 (d) Structural isomers
69. Minimum resistance in bond rotation will be observed in the compound
 (a) Hexachloroethane
 (b) Ethylene
 (c) Acetylene
 (d) Ethane
70. Which pair show *cis-trans* isomerism
 (a) Maleic-fumaric acid
 (b) Lactic-tartaric acid
 (c) Malonic-succinic acid
 (d) Crotonic-acrylic acid
71. 1, 2-Dichloroethene shows
 (a) Geometrical isomerism
 (b) Optical isomerism



- (c) Ring-chain isomerism
(d) Resonance
72. Which compound is optically active
(a) 4-chloro, 1 hydroxy butane
(b) 3° -butyl alcohol
(c) Secondary butyl amine
(d) *n*-butyl alcohol
73. Choose the pair of chain isomer
(a) CH_3CHBr_2 and $\text{CH}_2\text{BrCH}_2\text{Br}$
(b) 1-propanol and 2-propanol
(c) Neo-pentane and isopentane
(d) Diethyl ether and methyl-*n*-propyl ether
74. Optical isomerism arises due to the presence of
(a) An asymmetric carbon atom
(b) Centre of symmetry
(c) Axis of symmetry
(d) Plane of symmetry
75. Least hindered rotation about carbon-carbon bond is observed in
(a) Ethane
(b) Ethylene
(c) Ethyne
(d) Hexachloroethane
76. Which pair represents chain isomer
(a) CH_3CHCl_2 and $\text{ClCH}_2\text{CH}_2\text{Cl}$
(b) *n*-propyl alcohol and isopropyl alcohol
(c) 2-methyl-1 propanol and 2-Methyl-2 propanol
(d) 2-methyl butane and neopentane
77. Which of the following compounds will exhibit geometrical isomerism
(a) 1-phenyl-2-butene
(b) 3-phenyl-1-butene
(c) 2-phenyl-1-butene
(d) 1, 1-Diphenyl-1-propene
78. On bromination, propionic acid yields two isomeric 2-bromopropionic acids. This pair is an important example of
(a) Chain isomers
(b) Optical isomers
(c) Cis-trans isomers
(d) Position isomers
79. Geometrical isomerism is not possible in
(a) Propene
(b) 3-hexane
(c) Butenedioic acid
(d) Cyclic compound
80. Only two isomeric monochloro derivatives are possible for
(a) 2-methyl propane
(b) *n*-pentane
(c) Benzene



(d) 2, 4-dimethyl pentane

