CHEMICAL ARITHMETIC (MOLE CONCEPT)

Significant figures, Units for measurement, Matter and Separation of mixture

1. One fermi is

- o **Answer:** (a) 10⁻¹³ cm
- Reasoning: 1 fermi = 1 femtometer = 10^{-15} m. Converting to cm: 10^{-15} m × 100 cm/m = 10^{-13} cm.

2. A picometre is written as

- Answer: (d) 10⁻¹² m
- **Reasoning:** 1 pm = 10^{-12} m by definition.

3. One atmosphere is equal to

- o Answer: (a) 101.325 kPa
- o **Reasoning:** Standard atmospheric pressure = 101325 Pa = 101.325 kPa.

4. Dimensions of pressure are same as that of

- o **Answer:** (c) Energy per unit volume
- **Reasoning:** Pressure = Force/Area = ML^{-1T} -2; Energy/Volume = $(ML^{2T}-2)/L^3 = ML^{-1T}$ -2.

5. The prefix 10^18 is

- Answer: (d) Exa
- o **Reasoning:** Kilo=10³, Mega=10⁶, Giga=10⁹, Tera=10¹², Peta=10¹⁵, Exa=1018.

6. Significant figures for 161 cm, 0.161 cm, 0.0161 cm

- o Answer: (b) 3, 3, 3
- Reasoning: Leading zeros do not count; all numbers have 3 significant figures.

7. Significant figures in 0.00051

- Answer: (c) 2
- o **Reasoning:** Leading zeros are not counted; digits 5 and 1 are counted.

8. Halogen purified by sublimation

- o Answer: (d) l₂
- o **Reasoning:** lodine sublimes easily; other halogens are gases or liquids.

9. Difference in density is the basis of





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- o Answer: (c) Gravity Separation
- Reasoning: Gravity separation separates components based on density differences.

10. Element indicating life on Earth

- o **Answer:** (c) Carbon
- o Reasoning: Carbon forms organic compounds; essential for life.

11. Compound added to table salt for health

- o Answer: (c) Nal
- o Reasoning: Iodine deficiency is prevented by adding NaI to salt.

12. Contains only one element

- o Answer: (b) Diamond
- Reasoning: Diamond is pure carbon; other options are compounds or mixtures.

13. Maximum number of known elements

- Answer: (a) Metals
- o Reasoning: Most elements in the periodic table are metals.

14. Not an element

- o Answer: (c) Silica
- Reasoning: Silica (SiO₂) is a compound, not an element.

15. Mixture of ZnCl2 and PbCl2 separated by

- Answer: (b) Crystallization
- Reasoning: Different solubilities allow selective crystallization.

16. Mixture of methyl alcohol and acetone separated by

- Answer: (b) Fractional distillation
- o **Reasoning:** Close boiling points require fractional distillation.

17. Significant figures in ((29.2-20.2)(1.79×10⁵))/1.37

- o **Answer:** (b) 2
- Reasoning: Subtraction: 29.2-20.2=9.0 (2 sig. figs), Multiplication/Division
 → least sig. figs = 2.

18. Pure ethyl alcohol in 81.4 g sample containing 0.002 g water

- o **Answer:** (d) 81 g
- o **Reasoning:** $81.4-0.002 = 81.398 \rightarrow \text{rounded to correct sig. figs} = 81 g.$



IIT-JEE CHEMISTRY



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19. Unit J Pa⁻¹ is equivalent to

- o Answer: (a) m³
- Reasoning: $J/Pa = (N \cdot m)/(N/m^2) = m^3$.

20. Mass nearest to milligram

- o **Answer:** (c) 16.428 g
- o **Reasoning:** Milligram = $0.001 \text{ g} \rightarrow \text{closest mass with 3 decimal places} = 16.428 \text{ g}.$

21. Significant figures in 6.02×10^23

- o **Answer:** (b) 3
- o **Reasoning:** Only 6, 0, 2 counted \rightarrow 3 sig. figs.

22. Prefix zepto stands for

- o **Answer:** (d) 10⁻²¹
- \circ **Reasoning:** Zepto = 10^{-21} , femto= 10^{-15} , atto= 10^{-18} .

23. Significant figures in 3400

- o Answer: (a) 2
- o **Reasoning:** No decimal \rightarrow trailing zeros not counted \rightarrow 2 sig. figs.

24. Significant figures in 6.0023

- o **Answer:** (a) 5
- o Reasoning: All digits counted.

25. Significant figures in P=0.0030, Q=2.40, R=3000

- o **Answer:** (b) 2, 3, 4
- o **Reasoning:** $0.0030 \rightarrow 2$ sig. figs, $2.40 \rightarrow 3$ sig. figs, $3000 \rightarrow 4$ sig. figs if decimal written.

26. Significant figures in 60.0001

- o **Answer:** (b) 6
- o Reasoning: All digits counted.

27. Weight reported from 3.929 g and 4.0 g

- o **Answer:** (c) 3.9 g
- o **Reasoning:** Least precise measurement = $4.0 \text{ g} \rightarrow 2 \text{ sig. figs} \rightarrow \text{report}$ weight as 3.9 g.