Biblioteca Simplificada de Gás Ideal

Biblioteca básica de gás ideal sob hipóteses:

- Substâncias puras;
- Função cp(T) polinomial cúbica;
- Função cp(T) válida para Tmin ≤ T ≤ Tmax;
- sref = s°(Tref) conhecida.

```
    using PlutoUI
    using Formatting
    using DataFrames
    using BrowseTables
```

Setup Mínimo

```
    # Whether the molar base is the default one

• const MOLR = true;
• # Universal gas constant, with an optional precision argument
• \bar{R}(p=Float64) = p(8.314472); # \pm 0.000015 # kJ/kmol \cdot K

    # Standard Tref - for the internal energy

• Tref(p=Float64) = p(298.15); # K
# Standard Pref - for the entropy
• Pref(p=Float64) = p(100.00); # kPa
• # IG (Ideal Gas) structure: values for each gas instance
struct IG
                            # Molecular "Weight", kg/kmol
      MW
      CP::NTuple{4} # Exactly 4 \bar{c}p(T) coefficients

Tmin # T_{-}min, K

Tmax # T_{-}max, K

sref # \bar{s}°ref, kJ/kmol\cdot K
      sref
                            # s̄°ref, kJ/kmol·K
end;
```

Biblioteca Externa

Valores tabelados em Çengel, Y. A., Termodinâmica 7a Ed. ISBN 978-85-8055-200-3 (Tabelas A-1, A-2 e A-26), foram transportados para uma planilha (OpenOffice Spreadsheet) e exportados no formato CSV [Comma Separated Values], os quais podem ser lidos em Julia:

```
Name
               Formula
                                     ср_а
                                                 cp_b
                                                            cp_c
                                                                       cp_d
                                                                                  Tmin
                                                                                            Tmax
                                                                                                      sref
   Nitrogen
                          28.013
                                    +28.900
                                             -1.571E-03 +8.081E-06 -2.873E-09
                                                                                           1800.0
                                                                                                     191.61
                 02
                          31.999
                                    +25.480
                                             +1.520E-02 -7.155E-06 +1.312E-09
                                                                                  273.0
                                                                                           1800.0
                                                                                                     205.04
   0xygen
                          2.016
                 H2
                                    +29.110 -1.916E-03 +4.003E-06 -8.704E-10
                                                                                                     130.68
                                                                                  273.0
                                                                                           1800.0
   Hydrogen
Carbon Monoxide
                 co
                          28.011
                                    +28.160 +1.675E-03 +5.372E-06 -2.222E-09
                                                                                  273.0
                                                                                           1800.0
                                                                                                     197.65
                 C02
                          44.010
                                    +22.260 +5.981E-02 -3.501E-05 +7.469E-09
                                                                                           1800.0
Carbon Dioxide
                                                                                  273.0
                                                                                                     213.80
 Water Vapor
                                    +32.240 +1.923E-03 +1.055E-05 -3.595E-09
                                                                                           1800.0
                 H20
                          18.015
                                                                                  273.0
                                                                                                     188.83
                 CH4
                                                                                           1500.0
   Methane
                          16.043
                                    +19.890 +5.024E-02 +1.269E-05 -1.101E-08
                                                                                  273.0
                                                                                                     186.16
   Ethane
                 C2H6
                          30.070
                                    +6.900
                                             +1.727E-01 -6.406E-05 +7.285E-09
                                                                                  273.0
                                                                                           1500.0
                                                                                                     229.49
   Propane
                 C3H8
                          44.097
                                    -4.040
                                             +3.048E-01 -1.572E-04 +3.174E-08
                                                                                  273.0
                                                                                           1500.0
                                                                                                     269.91
                                             +3.715E-01 -1.834E-04 +3.500E-08
   N-Butane
                                                                                           1500.0
                C4H10
                          58.124
                                    +3.960
                                                                                  273.0
                                                                                                     310.12
   Methanol 

                 CH40
                          32.042
                                    +19.000 +9.152E-02 -1.220E-05 -8.039E-09
                                                                                  273.0
                                                                                           1000.0
                                                                                                     239.70
                                             +2.096E-01 -1.038E-04 +2.005E-08
   Ethanol
                C2H60
                          46.070
                                    +19.900
                                                                                           1500.0
                                                                                                     282.59
```

```
using CSV
```

```
Name
                        Formula
                                    MW
                                                        cp_b
                                                                     cp_c
                                                                                  cp_d
                                              cp_a
                        "N2"
    "Nitrogen"
                                  28.013
                                            28.9
                                                      -0.001571
                                                                 8.081e-6
                                                                               -2.873e-9
1
                        "02"
    "Oxygen"
                                  31.999
                                                      0.0152
                                            25.48
                                                                  -7.155e-6
                                                                               1.312e-9
2
                        "H2"
    "Hydrogen"
                                  2.016
                                            29.11
                                                      -0.001916
                                                                 4.003e-6
                                                                               -8.704e-10
3
    "Carbon Monoxide"
                        "CO"
                                  28.011
                                            28.16
                                                      0.001675
                                                                  5.372e-6
                                                                               -2.222e-9
    "Carbon Dioxide"
                        "CO2"
                                  44.01
                                            22.26
                                                      0.05981
                                                                  -3.501e-5
                                                                              7.469e-9
    "Water Vapor"
                        "H20"
                                  18.015
                                            32.24
                                                      0.001923
                                                                 1.055e-5
                                                                               -3.595e-9
    "Methane"
                        "CH4"
                                  16.043
                                                                  1.269e-5
                                                                               -1.101e-8
                                            19.89
                                                      0.05024
                        "C2H6"
                                  30.07
    "Ethane"
                                            6.9
                                                      0.1727
                                                                  -6.406e-5
                                                                              7.285e-9
    "Propane"
                        "C3H8"
                                  44.097
                                            -4.04
                                                      0.3048
                                                                  -0.0001572 3.174e-8
    "N-Butane"
                        "C4H10"
                                  58.124
                                            3.96
                                                      0.3715
                                                                  -0.0001834
                                                                              3.5e-8
10
    "Methanol"
                        "CH40"
                                  32.042
                                            19.0
                                                      0.09152
                                                                  -1.22e-5
                                                                               -8.039e-9
 gasRaw = CSV.File("IGTable.csv", normalizenames=true)
```

```
# Transforms a row (from the CSV file) into an IG instance
function rowToIG(row)
IG( row.MW, (row.cp_a, row.cp_b, row.cp_c, row.cp_d),
row.Tmin, row.Tmax, row.sref)
end;
```

Gás Padrão

gasRaw =

Escolha do gás padrão para testes, abaixo:

```
Nitrogen

- @bind gas_choice Select([row.Formula => row.Name for row in gasRaw])

stdGas = ▶ IG(28.013, (28.9, -0.001571, 8.081e-6, -2.873e-9), 273.0, 1800.0, 191.61)

- # Standard test gas
- stdGas = gasLib[Symbol(gas_choice)]
```

Funcionalidade - Verificações

```
function inbounds(gas::IG, T)
p = typeof(AbstractFloat(T))
minT, maxT = map(p, (gas.Tmin, gas.Tmax))
if !(minT <= T <= maxT)
throw(DomainError(T, "out of bounds $(minT) ≤ T ≤ $(maxT)."))
end
true
end;</pre>
```



✗ A temperatura de o K está FORA dos limites para o N2!

Funcionalidade - Constantes

```
R (generic function with 3 methods)

- # "R" can be typed by \bfR<tab>
- R(gas::IG, molr=MOLR, p=Float64) = molr ? R(p) : R(p) / p(gas.MW)
```

#	gas	R	M	s°	
1	N2	0.29681	28.013	191.61	
2	02	0.25984	31.999	205.04	
3	H2	4.12424	02.016	130.68	
4	CO	0.29683	28.011	197.65	
5	CO2	0.18892	44.010	213.80	
6	H20	0.46153	18.015	188.83	
7	CH4	0.51826	16.043	186.16	
8	C2H6	0.27650	30.070	229.49	
9	C3H8	0.18855	44.097	269.91	
10	C4H10	0.14305	58.124	310.12	
11	CH40	0.25949	32.042	239.70	
12	C2H60	0.18047	46.070	282.59	
#	gas	R	M	s°	

Funcionalidade – Comportamento P-T-v

```
prTy (generic function with 1 method)
 • # Auxiliary function of promoted types (float types relate to precision bits)!
 prTy(A...) = promote_type(map(typeof, AbstractFloat.(A))...)
P (generic function with 2 methods)
• # "P" can be typed by \bfP<tab>
 P(gas::IG, molr=true; T, v) = begin
      p = prTy(T, v)
       R(gas, molr, p) * T / v
 end
T (generic function with 2 methods)
• # "T" can be typed by \bfT<tab>
 T(gas::IG, molr=true; P, v) = begin
     p = prTy(P, v)
       P * v / R(gas, molr, p)
 end
v (generic function with 2 methods)
 • # "v" can be typed by \bfv<tab>
 v(gas::IG, molr=true; P, T) = begin
       p = prTy(P, T)
       R(gas, molr, p) * T / P
 end
<sup>—</sup> 300 kPa
                     <sup>→</sup> 300 K
Molar base? □
v = 0.2968 \text{ m}^3/\text{kg}; P = 300.0000 \text{ kPa}; T = 300.0000 \text{ K}.
```

Funcionalidade - Comportamento Calórico

Transformação de coeficientes:

Coeficientes, de $c_p(T)$ ou $c_v(T)$, são retornados como uma matriz-linha.

```
# If functions accound for integration factor, then only :cp, :cv are needed here
function coef(gas::IG, kind::Symbol = :cp, molr=MOLR, p=Float64)
if kind == :cp  # No coef. transformation
    ret = hcat(p.(gas.CP)...)
elseif kind == :cv  # Translates first coef.
    ret = hcat(p(gas.CP[1]) - R(p), p.(gas.CP[2:end])...)
end
molr ? ret : ret ./ M(gas, p)
end;
```

Funções dos coeficientes por propriedade:

Estas são as funções para serem aplicadas à temperatura, em três casos distintos. A função apply abaixo, faz a aplicação das funções à temperatura, retornando uma *matriz-coluna*.

```
# Generic f(T) function by Symbol key
function apply(p::Symbol, T, rel=false)

p = typeof(T)
rel ?
apply(p, T, false) - apply(p, Tref(p), false):
vcat((f(T) for f in propF[p])...)
end;
```

Propriedades Calóricas Diretas – $c_{p,v}(T)$, u(T), etc:

Estas funções selecionam as matrizes linha e coluna pertinentes, multiplicando-as, extraindo e retornando o único valor da matrix 1x1 resultante, com verificação de limites e somas de eventuais termos constantes.

As funções de uma única letra ASCII tem os nomes em letras negritas (bold-face) para não conflitarem com variáveis de nomes \mathbf{u} e \mathbf{h} , por exemplo. Por isso nomes como: \mathbf{u} e \mathbf{h} , por exemplo, porém não \mathbf{v} , por não ser ASCII.

```
cp (generic function with 2 methods)

cp(gas::IG, molr=MOLR; T) = begin

p = typeof(T)

inbounds(gas, T) ?

(coef(gas, :cp, molr, p) * apply(:c, T))[1] :

zero(p)

end
```

```
γ (generic function with 1 method)
• # "γ" can be typed by \gamma<tab>
• γ(gas::IG; T) = cp(gas, true, T=T) / cv(gas, true, T=T)
```

```
u (generic function with 2 methods)

# "u" can be typed by \bfu<tab>
u(gas::IG, molr=MOLR; T) = begin
p = typeof(T)
inbounds(gas, T) ?
(coef(gas, :cv, molr, p) * apply(:h, T, true))[1] :
zero(p)
end
```

```
h (generic function with 2 methods)

* # "h" can be typed by \bfh<tab>
h (gas::IG, molr=MOLR; T) = begin

p = typeof(T)
inbounds(gas, T) ?

(coef(gas, :cp, molr, p) * apply(:h, T, true))[1] +

R(gas, molr, p) * Tref(p) :

zero(p)
end
```

```
s° (generic function with 2 methods)

* ""o" can be typed by \degree<tab>
* "Partial" ideal gas entropy

s° (gas::IG, molr=MOLR; T) = begin

p = typeof(T)
inbounds(gas, T) ?
(coef(gas, :cp, molr, p) * apply(:s, T, true))[1] + (
molr ? sref(gas, p) : sref(gas, p) / M(gas, p)

) :
zero(p)
end
```

```
Pr (generic function with 1 method)

- Pr(gas::IG; T) = begin

- p = typeof(T)

- exp(s°(gas, true, T=T) / R̄(p)) / exp(sref(gas, p) / R̄(p))

- end
```

```
vr (generic function with 1 method)
   vr(gas::IG; T) = T / Pr(gas, T=T)
```

```
s (generic function with 2 methods)

* # "s" can be typed by \bfs<tab>
* s(gas::IG, molr=MOLR; T, P) = begin

* p = prTy(P, T)
* inbounds(gas, T) ?

* s°(gas, molr, T=T) - R(gas, molr, p) * log(P / Pref(p)) :

* zero(p)

* end
```

#	Т	h	Pr	u	vr	s°	ср	CV	Υ
1	300.0	90.41	1.02	1.37	293.58	6.85	1.04	0.74	1.4
2	400.0	194.73	2.81	76.0	142.45	7.15	1.05	0.75	1.39
3	500.0	300.29	6.21	151.88	80.54	7.38	1.06	0.77	1.39
4	600.0	407.4	11.99	229.32	50.06	7.58	1.08	0.78	1.38
5	700.0	516.3	21.1	308.53	33.18	7.75	1.1	0.8	1.37
6	800.0	627.17	34.74	389.72	23.03	7.89	1.12	0.82	1.36
7	900.0	740.11	54.38	472.98	16.55	8.03	1.14	0.84	1.35
8	1000.0	855.19	81.81	558.38	12.22	8.15	1.16	0.86	1.34
9	1100.0	972.4	119.19	645.91	9.23	8.26	1.18	0.89	1.34
10	1200.0	1091.66	169.06	735.49	7.1	8.36	1.2	0.91	1.33
11	1300.0	1212.85	234.41	827.0	5.55	8.46	1.22	0.92	1.32
12	1400.0	1335.78	318.61	920.25	4.39	8.55	1.24	0.94	1.32
13	1500.0	1460.18	425.45	1014.97	3.53	8.64	1.25	0.95	1.31
14	1600.0	1585.76	559.01	1110.86	2.86	8.72	1.26	0.96	1.31
15	1700.0	1712.12	723.62	1207.54	2.35	8.79	1.27	0.97	1.31
16	1800.0	1838.83	923.6	1304.57	1.95	8.87	1.27	0.97	1.31
#	Т	h	Pr	Ш	vr	°°	cn	CV	V

```
using Roots, ForwardDiff
```

Funcionalidade – Funções inversas

Métodos numéricos para T(u), T(h), T(pr), etc.

Definição de Tipos

Como todas as funções inversas acima — T(u), T(h), etc. — possuem o mesmo *nome*, a diferenciação entre elas se dará via Multiple Dispatch, e assim, cada função T será especializada com base nos tipos de seus argumentos.

O objetivo de saber se o argumento é uma energia interna ou entalpia, etc., é para que se saiba (i) sua forma funcional **e** (ii) a forma funcional de sua derivada, a fim de ajustar o método numérico.

Para tanto, é necessário a criação de novos **tipos**, que **rotulem** seus valores como "energia interna", "entalpia", etc.:

```
    # A Thermodynamic abstract type to hook all concrete property value types under it
    abstract type THERM end
```

```
begin
    # A type to LABEL values as internal energy ones:
    struct uType <: THERM
         val
    end
    # Functor to extract the stored value 'val'...
    # ... thus avoiding further implementing the type:
    (_u::uType)() = _u.val
end</pre>
```

```
begin
struct hType <: THERM; val; end
(_h::hType)() = _h.val
end</pre>
```

```
begin
struct prType <: THERM; val; end
(_p::prType)() = _p.val
end</pre>
```

```
begin
struct vrType <: THERM; val; end
(_v::vrType)() = _v.val
end</pre>
```

⊳ Ilustração do conceito:

```
▶["ū = 314.15 kJ/kmol", "h = 314.15 kJ/kg", true]

begin

# First METHOD definition for the function "example":
function example(x::uType, molr=MOLR)

molr?

"ū = $(x()) kJ/kmol":
"u = $(x()) kJ/kg"

end

# Second METHOD definition for the function "example":
function example(x::hType, molr=MOLR)

molr?

"h = $(x()) kJ/kmol":
"h = $(x()) kJ/kg"

end

# Same function name "example" called: specialize based on argument(s) TYPE(s):
vcat(
example(uType(314.15)), # uType argument
example(hType(314.15), false), # htype argument
uType(3.14)() == hType(3.14)() # Their _values_ are the same!
)
end
```

Implementação

```
T (generic function with 8 methods)
 begin
       #-----#

T(u) inverse #
       # "T" can be typed by \bfT<tab>
       function T(
               gas::IG, uVal::uType, molr=true;
               maxIt::Integer=0, epsTol::Integer=4
           # Auxiliary function of whether to break due to iterations
           breakIt(i) = maxIt > 0 ? i >= maxIt || i >= 128 : false
           # Set functions f(x) and g(x) \equiv df/dx
          f = x \rightarrow u(gas, molr, T=x)
           g = x \rightarrow cv(gas, molr, T=x)
           thef, symb = (uVal)(), "u'
           \varepsilon, p = eps(thef), typeof(thef)
           # Get f bounds and check
           TMin, TMax = Tmin(gas, p), Tmax(gas, p)
           fMin, fMax = f(TMin), f(TMax)
           if !(fMin <= thef <= fMax)</pre>
               throw(DomainError(thef, "out of bounds $(fMin) ≤ $(symb) ≤ $(fMax)."))
           end
           # Linear initial estimate and initializations
           r = (thef - fMin) / (fMax - fMin)
T = [ TMin + r * (TMax - TMin) ] # Iterations are length(T)-1
           f = [f(T[end])]
           why = :because
           # Main loop
           while true
               append!(T, T[end] + (thef - f[end]) / g(T[end]))
append!(f, f(T[end]))
if breakIt(length(T)-1)
                   why = :it; break
               elseif abs(f[end] - thef) <= epsTol * ε</pre>
                   why = :\Delta f; break
          end
          return Dict(
               :sol => T[end],
               :why => why,
               :it => length(T)-1,
               :\Delta f => f - thef,
               :Ts => T,
               :fs => f
       end
                  T(h) inverse #
       # "T" can be typed by \bfT<tab>
       function T(
               gas::IG, hVal::hType, molr=true;
               maxIt::Integer=0, epsTol::Integer=4
           # Auxiliary function of whether to break due to iterations
           breakIt(i) = maxIt > 0 ? i >= maxIt || i >= 128 : false
           # Set functions f(x) and g(x) \equiv df/dx
           f = x \rightarrow h(gas, molr, T=x)
           g = x \rightarrow cp(gas, molr, T=x)
           thef, symb = (hVal)(), "h"
           \varepsilon, p = eps(thef), typeof(thef)
           # Get f bounds and check
           TMin, TMax = Tmin(gas, p), Tmax(gas, p)
           fMin, fMax = f(TMin), f(TMax)
if !(fMin <= thef <= fMax)
               throw(DomainError(thef, "out of bounds $(fMin) ≤ $(symb) ≤ $(fMax)."))
           end
           # Linear initial estimate and initializations
           r = (thef - fMin) / (fMax - fMin)
           T = [TMin + r * (TMax - TMin)]' # Iterations are length(T)-1

f = [f(T[end])]
           why = :because
           # Main loop
           while true
               append!(T, T[end] + (thef - f[end]) / g(T[end]))
               append!(f, f(T[end]))
if breakIt(length(T)-1)
                   why = :it; break
               elseif abs(f[end] - thef) <= epsTol * ε</pre>
                   why = :\Delta f; break
           end
           return Dict(
               :sol => T[end],
               :why => why,
               :it => length(T)-1,
               :Δf => f .- thef,
               :Ts => T,
               :fs => f
       end
                         T(pr) inverse
       #-----#
       # "T" can be typed by \bfT<tab>
       function T(
```

```
gas::IG, pVal::prType;
        maxIt::Integer=0, epsTol::Integer=4
    # Auxiliary function of whether to break due to iterations
    breakIt(i) = maxIt > 0 ? i >= maxIt || i >= 128 : false
    # Set functions f(x) and g(x) \equiv df/dx
    f = x \rightarrow Pr(gas, T=x)
    g = x -> ForwardDiff.derivative(f,float(x))
    thef, symb = (pVal)(), "Pr"
    \varepsilon, p = eps(thef), typeof(thef)
    # Get f bounds and check
    TMin, TMax = Tmin(gas, p), Tmax(gas, p)
    fMin, fMax = f(TMin), f(TMax)
    if !(fMin <= thef <= fMax)</pre>
        throw(DomainError(thef, "out of bounds $(fMin) ≤ $(symb) ≤ $(fMax)."))
    end
    # Linear initial estimate and initializations
    r = (thef - fMin) / (fMax - fMin)
    T = [TMin + r * (TMax - TMin)] # Iterations are length(T)-1
    f = [f(T[end])]
    why = :because
    # Main loop
    while true
        append!(T, T[end] + (thef - f[end]) / g(T[end]))
append!(f, f(T[end]))
if breakIt(length(T)-1)
            why = :it; break
        elseif abs(f[end] - thef) <= epsTol * ε</pre>
            why = :\Delta f; break
        end
    end
    return Dict(
        :sol => T[end],
        :why => why,
        :it => length(T)-1,
        :\Delta f => f - thef,
        :Ts => T,
        :fs => f
end
              T(vr) inverse
# "T" can be typed by \bfT<tab>
function T(
        gas::IG, vVal::vrType;
        maxIt::Integer=0, epsTol::Integer=4
    # Auxiliary function of whether to break due to iterations
    breakIt(i) = maxIt > 0 ? i >= maxIt || i >= 128 : false
    # Set f(x) function
    thef, symb = (vVal)(), "vr"
    f = x \rightarrow vr(gas, T=x) - thef
    \varepsilon, p = eps(thef), typeof(thef)
    # Get f bounds and check
    TMin, TMax = Tmin(gas, p), Tmax(gas, p)
    fMin, fMax = f(TMax), f(TMin)
    if !(fMin <= zero(p) <= fMax)</pre>
        throw(
            DomainError(
                 "out of bounds \{(fMin+thef) \leq \{(symb) \leq \{(fMax+thef)."\}\}
    # Bisection method initializations
    TB = [ TMin, TMax ] # T bounds
   FB = map(f, TB) # f bounds

T = p[] # Iterations are length(T)
    f = p[
    s = map(signbit, FB)
    why = :unbracketed
    while !reduce(==, s)
        # Main loop
        append!(T, reduce(+, TB) / 2)
        append! (f, f(T[end]))
        sMid = signbit(f[end])
        if sMid == s[1]
            TB[1], FB[1] = T[end], f[end]
             TB[2], FB[2] = T[end], f[end]
        end
        if breakIt(length(T))
            why = :it; break
        elseif abs(f[end]) <= epsTol * ε</pre>
             why = :\Delta f; break
        end
    end
    return Dict(
        :sol => T[end],
        :why => why,
        :it => length(T),
        :\Delta f => f,
        :Ts => T,
        :fs => f .+ thef,
        :TB => TB,
        :FB => FB
```

```
end
T (generic function with 8 methods)
 • T
Tu =
▶ Dict(:it \Rightarrow 2, :\(\Delta f \Rightarrow [-2.91995, 0.000685624, 3.82154e-11], :sol \Rightarrow 300.0, :\(Ts \Rightarrow [296.06])
 - Tu = T(
        stdGas,
        uType(
            u(
                 stdGas,
                 false,
                 T = 300.0
        false,
        epsTol=2^26 # 2<sup>26</sup> = 67108864: don't care about the last 26 bits
        #epsTol=2^16 # 2<sup>16</sup> = 65536: don't care about the last 16 bits
 • )
▶ ["296.0597138486446", "300.0009249874004", "300.0000000000516"]
 collect(sprintf1("%.$(16-3)f", i) for i in Tu[:Ts])
Tu_{32} =
▶ Dict(:it \Rightarrow 2, :∆f \Rightarrow [-2.91994, 0.000678539, 0.0], :sol \Rightarrow 300.0, :Ts \Rightarrow [296.06, 300.001
  Tu_{32} = T(
        stdGas,
        uType(
            u(
                 stdGas,
                 false,
                 T=300.0f0 # literal floats with "f0" are 32-bit, single-precision
        false,
        epsTol=1 # 2° = 1: care about all bits
 • )
▶ ["296.0597", "300.0009", "300.0000"]
 collect(sprintf1("%.$(7-3)f", i) for i in Tu<sub>32</sub>[:Ts])
▶ Dict(:it \Rightarrow 5, :\(\Delta f \Rightarrow [-3.04622, 0.000381454, 6.03033e-12, 1.50709e-27, 9.41309e-59, 0.0],
 - Th = T(
        stdGas,
        hType(
            h(
                 stdGas,
                 false,
                 T=BigFloat(300.0)
        ),
false
▶ ["297.065020802418806975208246766375870391668061284181485246454206601924027496952", "300.
 - collect(sprintf1("%.$(78-3)f", i) for i in Th[:Ts])
```

localhost:1234/edit?id=03b3efb2-ea63-11eb-3580-8bff33aa4e91

```
¶ IGlib.jl — Pluto.jl

Tp =
▶ Dict(:it \Rightarrow 6, :∆f \Rightarrow [-0.282358, 0.0410276, 0.000568156, 1.13878e-7, 3.55271e-15, 3.55271
  \cdot   Tp = T(
        stdGas,
        prType(
            Pr(
                 stdGas,
                 T=300.0
        epsTol=1
 - )
▶["+273.4745933292123", "+303.3952611692405", "+300.0476837891530", "+300.0000095593841", '
 collect(sprintf1("%+.$(16-3)f", i) for i in Tp[:Ts])
▶["-2.823582329270165e-01", "+4.102760579279985e-02", "+5.681556327807868e-04", "+1.138778
 - collect(sprintf1("%+.$(16-1)e", i) for i in Tp[:Δf])
Tv =
▶ Dict(:TB \Rightarrow [300.0, 300.0], :it \Rightarrow 51, :\triangle f \Rightarrow [-282.574, -253.876, -195.909, -118.238, -45
  Tv = T(
        stdGas,
        vrType(
            vr(
                 stdGas,
                 T=300.0
        epsTol=1
 • )
```