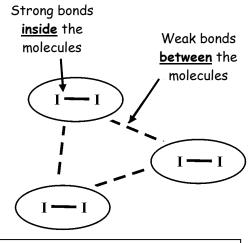
Summary 3.1a

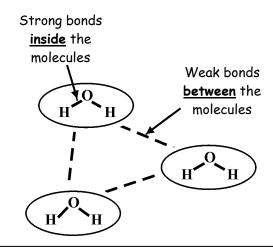
Section 3 Summary

3.1 Molecules & Ions

- a) There are about 100 different elements.
 - Every element is made up of very small particles called atoms.
- b) Atoms of different elements are different from each other.
- c) Molecules are made up of two or more atoms joined together. (can be same element or different elements)
- d) Molecules are held together inside by strong bonds
 - However bonds between different molecules are weak bonds.
- e) Compounds are made up of two of more different elements chemically joined together
- f) Substances made of molecules tend to have
 - low melting points
 - low boiling points
- g) Substances made of molecules do not conduct electricity.



This substance can be called both a Element: Only one kind of atom present Molecule: 2 or more atoms joined together



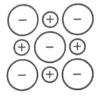
This substance can be called both a

Molecule: 2 or more atoms joined together Compound: 2 or more different atoms joined together

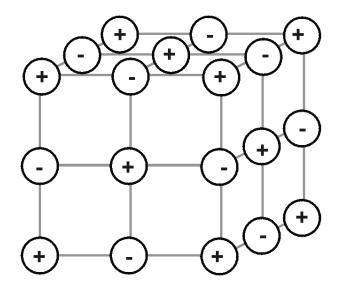
Section 3 Summary

3.1 Molecules & Ions (cont.)

- h) Ions are charged atoms
- i) Ions can be positively charged or negatively charged
- j) Ionic compounds are compounds (2 or more different elements) made up of oppositely charged ions



- k) Bonds between ions are strong
- 1) Substances made of ions have
 - high **melting** points
 - high boiling points
- m) Substances made of ions conduct electricity when:
 - dissolved in water
 - molten

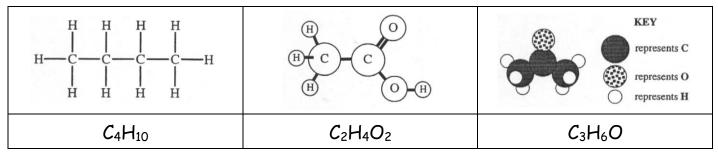


Summary 3.2&3.3

Section 3 Summary

3.2 Formulae: Using Models

a) Be able to work out formulae from diagrams and pictoral representations.



Remember: drop number 1 in a formula

$$C_3H_6O_1 \longrightarrow C_3H_6O$$

3.3 Formulae: Using Prefixes

Prefix	Meaning	Example	Chemical Formula
mono	1	carbon mono xide	СО
di	2	di nitrogen oxide	N₂O
tri	3	phosphorus tri hydride	PH₃
tetra	4	carbon tetra fluoride	CF ₄

Remember:

drop number 1 in a formula

$$C_1O_1 \longrightarrow CO$$

If an element has no prefix then it has the number 1

sulphur dioxide
$$\rightarrow$$
 $S_1O_2 \xrightarrow{\text{Drop number 1}} SO_2$