1. Use

$$\vec{B}_{\rm int} = \frac{1}{4\pi\epsilon_0} \frac{e}{m_e c^2 r^3} \vec{L},$$

where e is the elementary charge, to estimate the internal magnetic field in a hydrogen atom. This value characterizes the boundary between the strong and weak field limit.

The magnitude of the internal magnetic field is given by

$$\vec{B}_{\rm int} \cdot \vec{B}_{\rm int} = \left| \vec{B}_{\rm int} \right| = \left(\frac{1}{4\pi\epsilon_0} \frac{e}{m_e c^2} \right)^2 \left(\frac{1}{r^3} \right)^2 \left| \vec{L}^2 \right|.$$

Let β be defined as the pre-factors,

$$\beta \equiv \frac{1}{4\pi\epsilon_0} \frac{e}{m_e c^2}.$$

Then, the magnitude of the internal magnetic field may be represented in terms of operators

$$\left| \vec{B}_{\text{int}} \right| = \hat{B}_{\text{int}} = \beta^2 \frac{1}{\hat{r}^3} \frac{1}{\hat{r}^3} \hat{L}^2.$$

The average internal magnetic field can then be found as

$$\langle \Psi | \hat{B}_{\rm int} | \Psi \rangle = \beta^2 \langle \Psi | \left(\frac{1}{\hat{r}^3} \frac{1}{\hat{r}^3} \hat{L}^2 \right) | \Psi \rangle = \beta^2 \left\langle \frac{1}{r^3} \frac{1}{r^3} L^2 \right\rangle.$$

We may insert the identity matrix between each of the three terms of the expectation value. The identity matrix may be expressed as the outer-product of wavefunctions. This allows us to express the expectation value of the internal magnetic field as

$$\langle B_{\rm int} \rangle = \beta^2 \left\langle \frac{1}{r^3} \right\rangle \left\langle \frac{1}{r^3} \right\rangle \left\langle L^2 \right\rangle.$$

These expectation values are known for our chosen basis:

$$\left\langle \frac{1}{r^3} \right\rangle = \frac{1}{l\left(1 + \frac{1}{2}\right)(l+1)n^3a^3}, \quad \left\langle L^2 \right\rangle = \hbar^2 l(l+1).$$

Then,

$$\langle B_{\rm int} \rangle = \beta^2 \frac{\hbar^2 l(l+1)}{\left[l\left(l+\frac{1}{2}\right)(l+1)n^3a^3\right]^2} = \beta^2 \frac{\hbar^2}{l(l+1)\left[\left(l+\frac{1}{2}\right)n^3a^3\right]^2}.$$

This implies that the strength of the internal magnetic field is strongest at low values of n and l, which makes sense by the classical picture. Let n=1. Then, $l \in \{0,1\}$. While l=0 appears problematic, we recall that for l=0 $\langle L^2 \rangle = 0$ and thus $\langle B_{\rm int} \rangle = 0$. Therefore, let l=1. Then, substituting known values,

$$\langle B_{\rm int} \rangle \approx 34 \, \mathrm{T}.$$

Therefore, $|B_{\rm ext}| \ll 30\,{\rm T}$ is a weak field while $|B_{\rm ext}| \gg 30\,{\rm T}$ is a strong field.

2. Consider the eight n=2 states for the hydrogen atom, $\langle 2,l,j,m_j|$. Determine the energy of each state under weak-field Zeeman splitting and construct a diagram like the one in Figure 6.11 of Griffiths to show how the energies evolve as a function of $B_{\rm ext}$. Label each line clearly and indicate the slope of each line on the graph.