

Week 12 Thursday

L-22

Radiation pressure

$$\frac{dM_r(r)}{dr} = 4\pi r^2 \rho(r)$$

$$\frac{dP(r)}{dr} = -\rho(r) \frac{GM_r(r)}{r^2}$$

$$\frac{dT(r)}{dr} = -\frac{3\kappa_R(r)\rho(r)}{64\pi\sigma} \frac{L_r(r)}{r^2 T(r)^3}$$

$$\frac{dL_r(r)}{dr} = 4\pi r^2 \rho(r) \epsilon(r)$$

$$M_r(r) \quad P(r) \quad L_r(r) \quad T(r)$$

$$\rho(r) \quad \mu(r) \quad \epsilon_{\text{nuc}}(r) \quad \kappa_R(r)$$

$$P(r) = \frac{\rho(r)}{\mu(r)} kT(r)$$

$$\mu(r) = f(\text{comp}, T(r), P(r))$$

$$\kappa_R(r) = f(\text{comp}, T(r), P(r))$$

$$\epsilon_{\text{nuc}}(r) = f(\text{comp}, T(r), P(r))$$

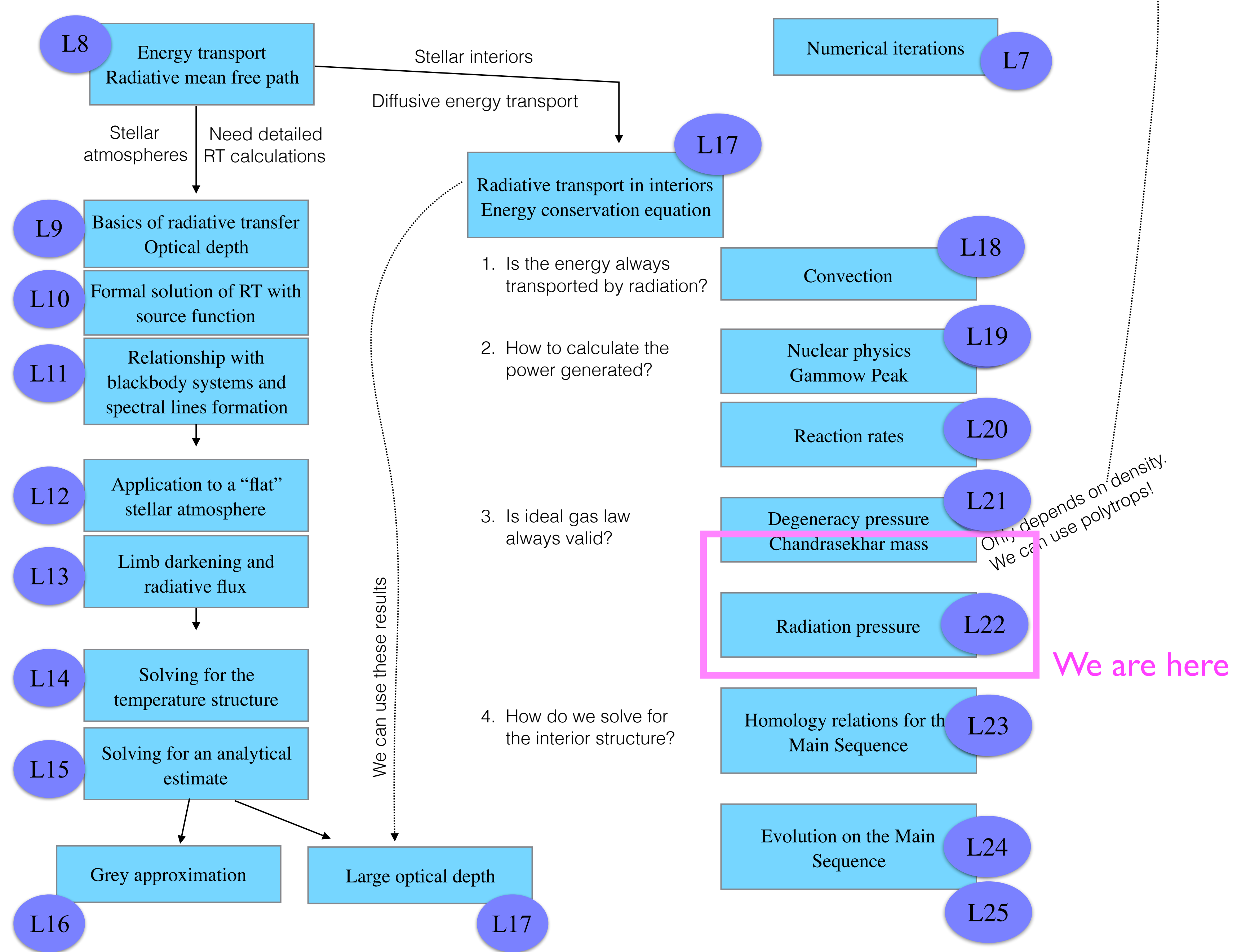
Other energy transport?

Ideal gas always valid?

Nuclear mechanism?

How to solve?

What changes with time?



Pressure is a momentum flux:
intermediate step from last lecture

$$P = \frac{1}{3} \int_0^{p_F} p \, v(p) \, n(p) \, 4\pi p^2 \, dp$$

Adapt this for photons:

$$v(p) = c$$

Not degenerate, no limit
on momentum

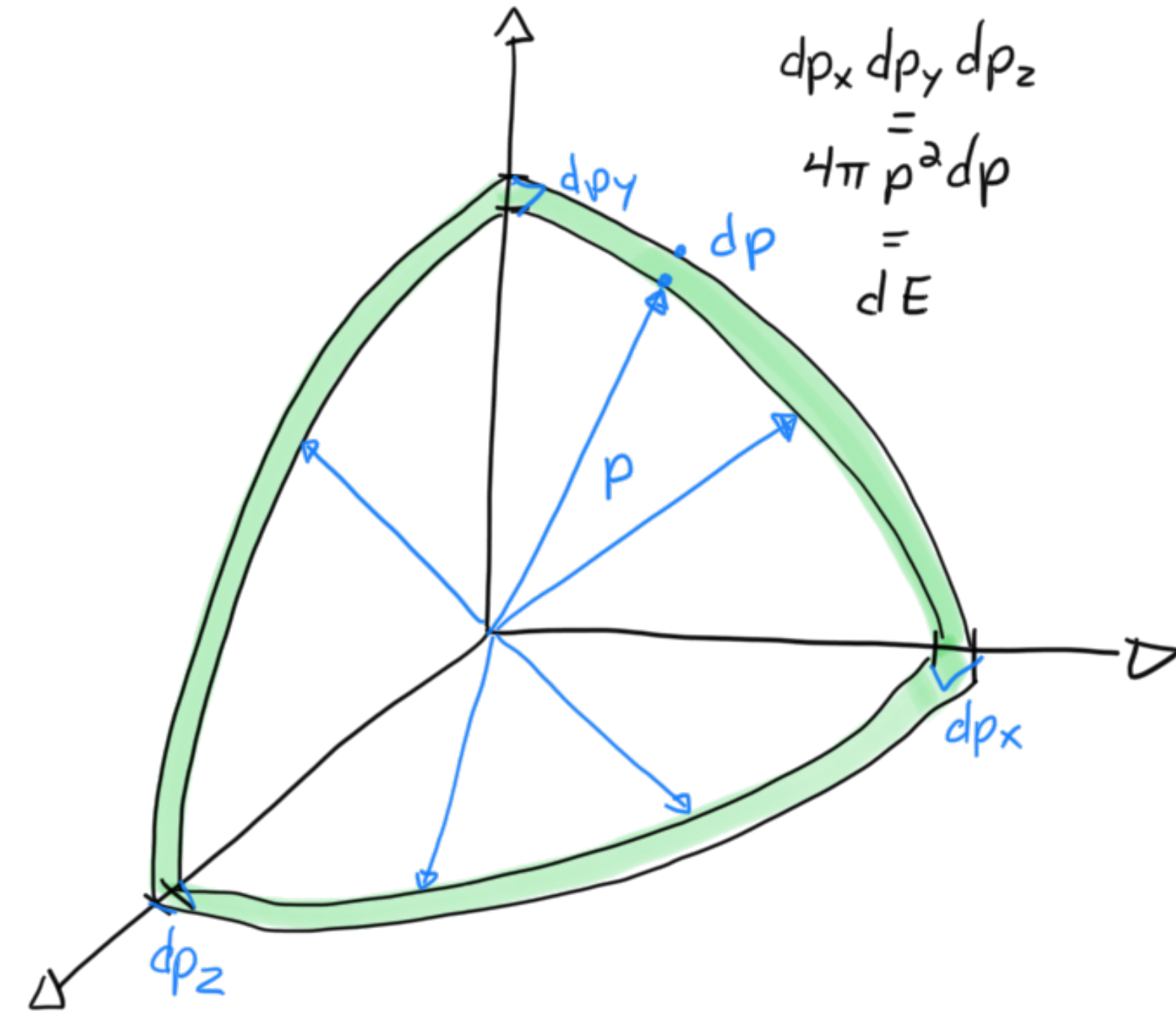
$$P = \frac{1}{3} \int_0^{\infty} p \, c \, n(p) \, 4\pi p^2 \, dp$$

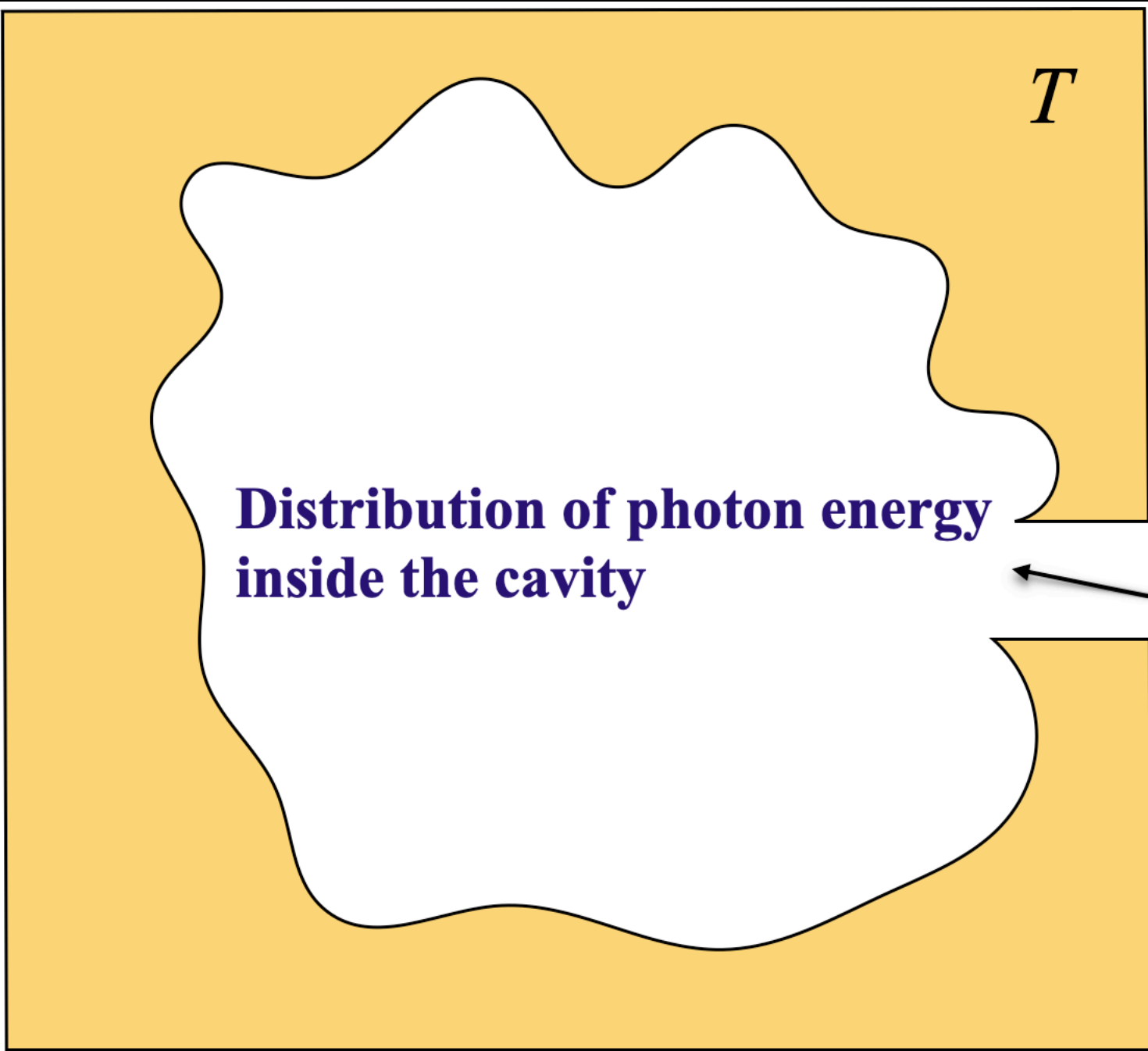
pc is also the energy of a photon.

We can relate this to the energy density of a “gas” of photons

$$P = \frac{1}{3} \int_0^{\infty} E \, n(E) \, dE$$

Remember that $n(p)$ is the number of particle per volume with momentum between p and $p + dp$
So $n(E)$ is the number of particle per volume with energy between E and $E + dE$





**Energy density (u)
per unit of photon energy (E)**

e.g. joules per m³ per keV

$$u(E) = \frac{8\pi}{(hc)^3} \frac{E^3}{e^{E/kT} - 1}$$

(Undergrad Thermal Physics
textbook by Schroeder, Sec 7.4)

But if we were to “add up” the energy of all the photons in the cavity, we could also express the energy density as:

$$U = \int_0^\infty E n(E) dE$$

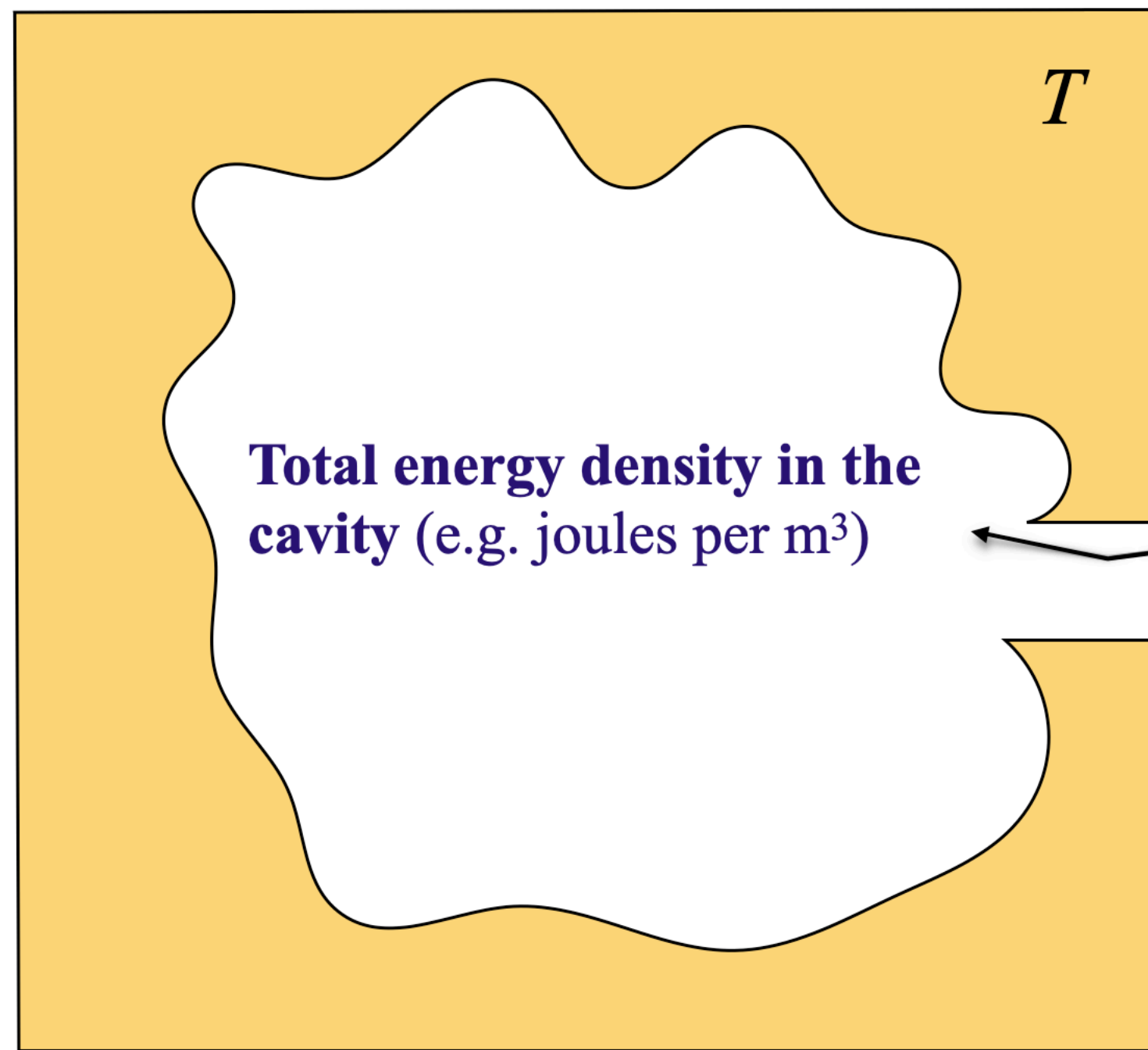
Let’s compare this with our expression for pressure from the previous slide:

$$P = \frac{1}{3} \int_0^\infty E n(E) dE$$

So

$$P = \frac{1}{3} U$$

$$P_{\text{rad}} = \frac{4\sigma}{3c} T^4$$



$$U = \frac{8\pi}{(hc)^3} \int_0^\infty \frac{E^3}{e^{E/kT} - 1}$$

$$= \frac{4}{c} \boxed{\frac{2\pi^4 k^4}{15h^3 c^2}} T^4$$

Stefan-Boltzmann constant σ

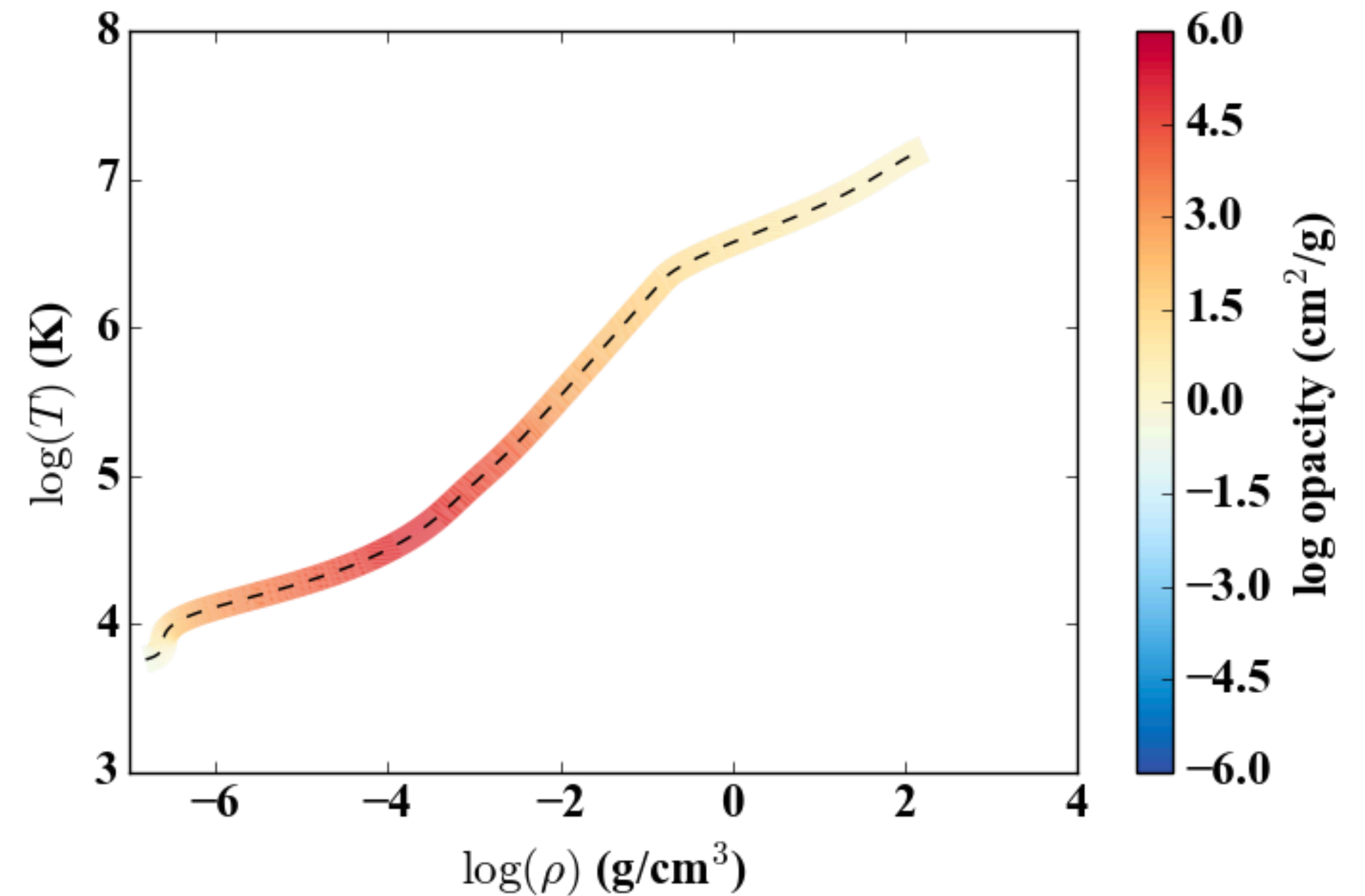
$$U = \frac{4\sigma}{c} T^4$$

At which ρ and T does the radiation pressure becomes important?

1. Transition from ideal gas law to radiation pressure (on the board)

$$\left(\frac{\rho}{\mu}\right)_{\text{trans}} = 3.05 \times 10^{-23} T^3 \text{ [dyn/cm]}$$

Let's add this to our notebook



If there is a mixture of P_{ideal} and P_{rad} : The Eddington model

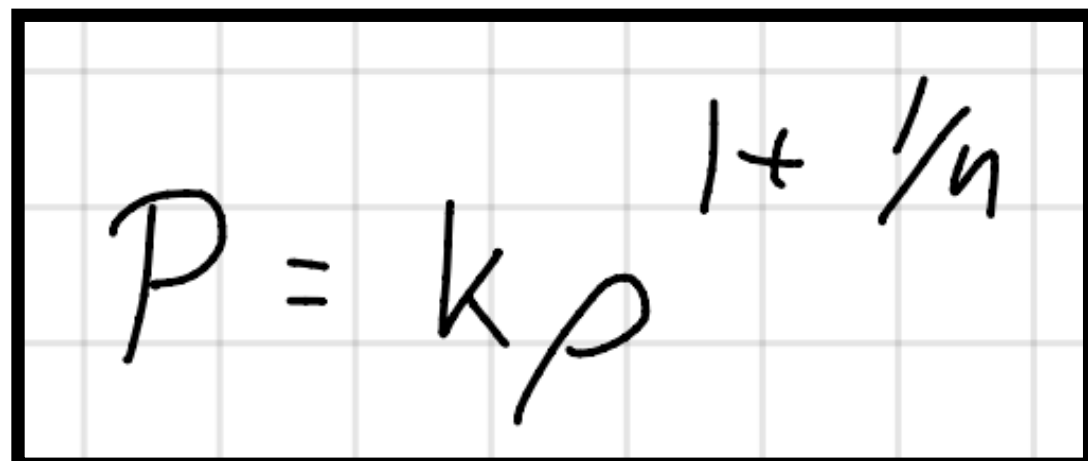
$$P_{\text{tot}} = \frac{\rho k}{\mu m_H} T + \frac{4\sigma}{3c} T^4$$

Let's set $\beta = \frac{P_{\text{ideal}}}{P_{\text{tot}}}$

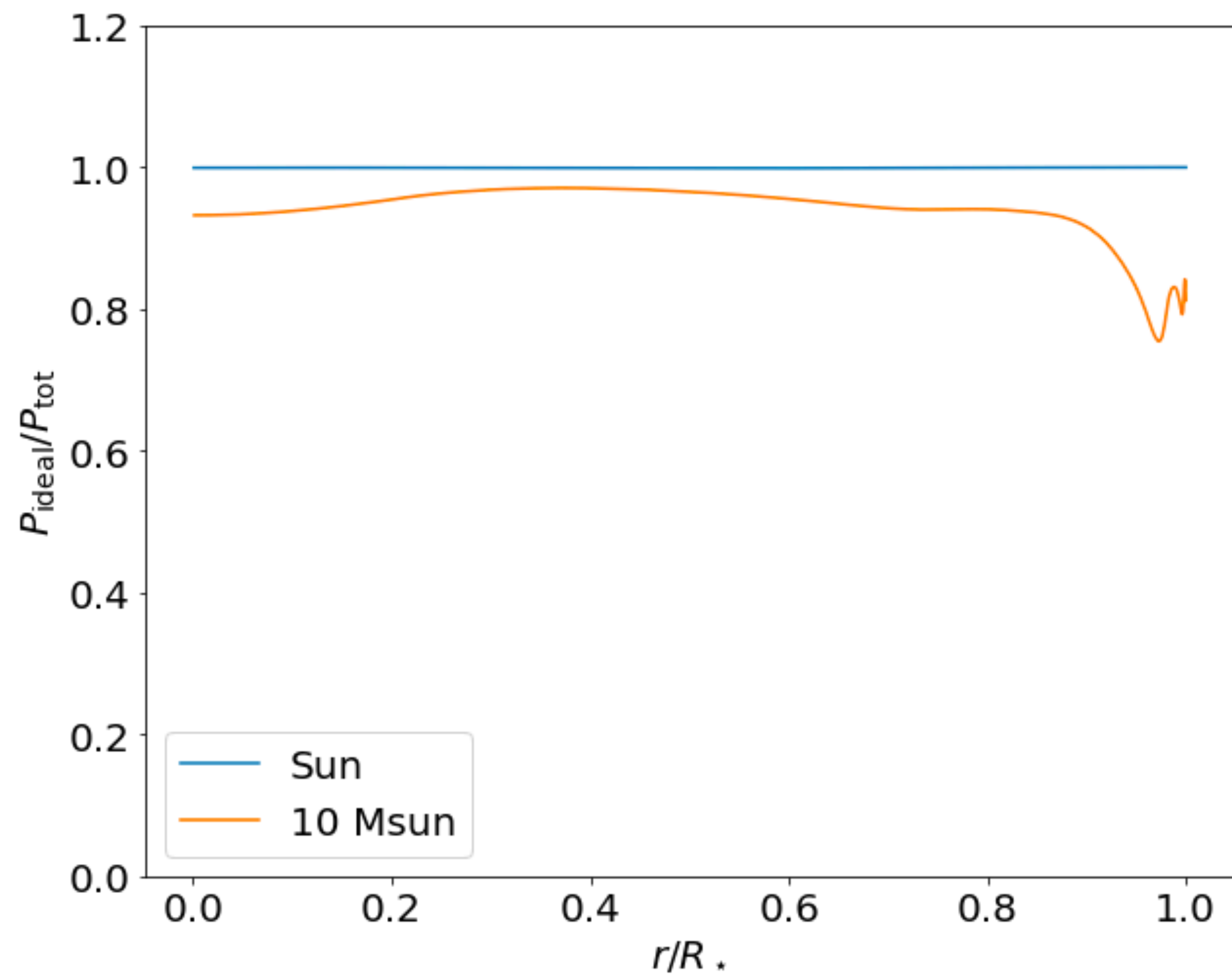
Some algebra later....

$$P_{\text{tot}} = \left[\frac{(1 - \beta)}{\beta^4} \frac{3c}{4\sigma} \left(\frac{k}{\mu m_H} \right)^4 \right]^{1/3} \rho^{4/3}$$

Polytrops


$$P = k \rho^{1 + 1/n}$$

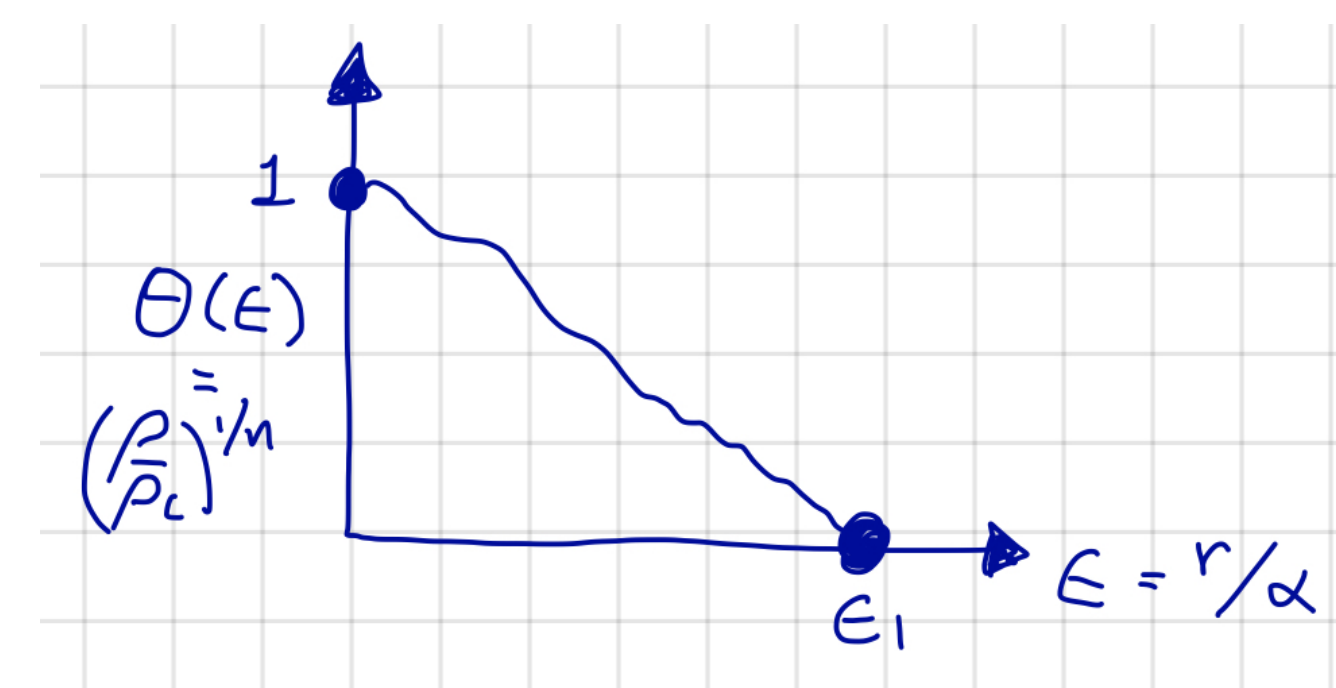
So if β is constant as a function of radius,
this would be a $n = 3$ polytrope....
So, is $P_{\text{gas}}/P_{\text{tot}}$ roughly constant?



$$P_{\text{tot}} = \underbrace{\left[\frac{(1-\beta)}{\beta^4} \frac{3c}{4\sigma} \left(\frac{k}{\mu_{\text{MH}}} \right)^4 \right]^{1/3}}_K \rho^{4/3}$$

Polytrops

$$P = k \rho^{1+1/n}$$



$$1 \quad R_{\star} = \alpha \epsilon_1 = \left(\frac{(n+1)P_c}{4\pi G \rho_c^2} \right)^{1/2} \epsilon_1$$

$$2 \quad P_c = K \rho_c^{(n+1)/n}$$

$$3 \quad M_{\star} = -\frac{1}{(4\pi)^{1/2}} \left(\frac{n+1}{G} \right)^{3/2} \frac{P_c^{3/2}}{\rho_c^2} \epsilon_1^2 \theta'(\epsilon_1)$$

n

P_c ρ_c R_{\star}

then known K
(so does β)

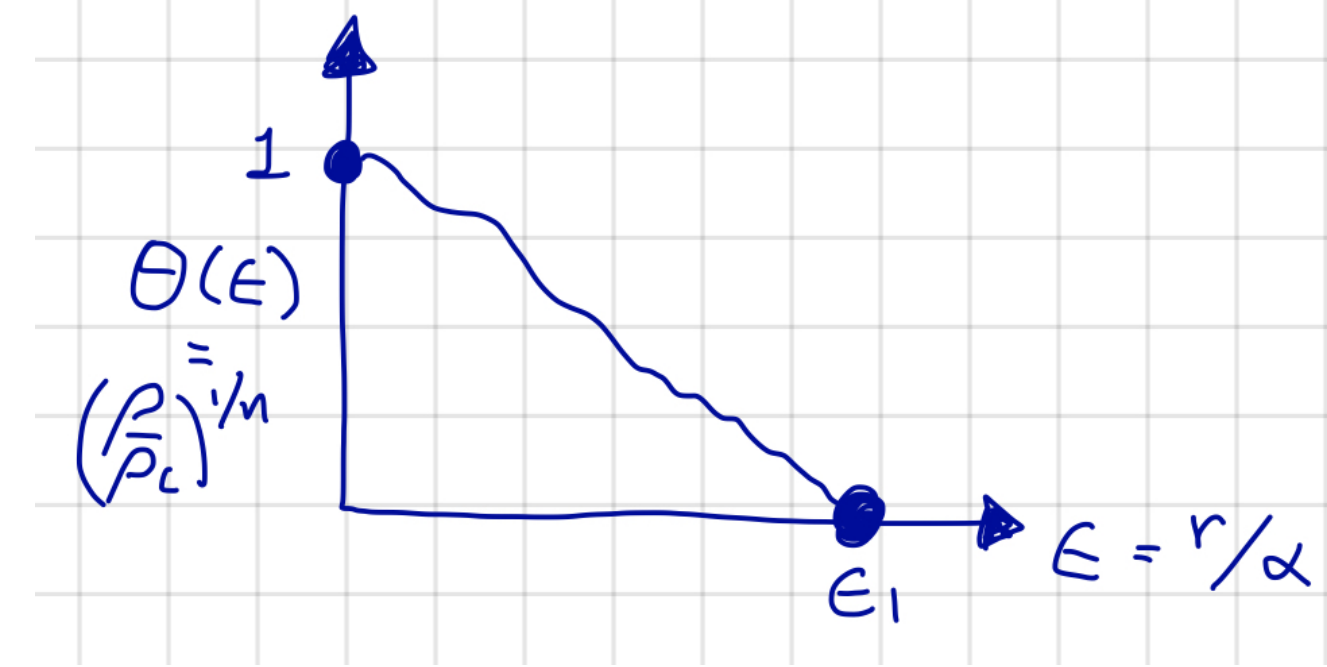
if we pick

M_{\star}

$$P_{\text{tot}} = \underbrace{\left[\frac{(1-\beta)}{\beta^4} \frac{3c}{4\sigma} \left(\frac{k}{\mu_{\text{MH}}} \right)^4 \right]^{1/3}}_K \rho^{4/3}$$

Polytrops

$$P = k \rho^{1+1/n}$$



$$1 \quad R_{\star} = \alpha \epsilon_1 = \left(\frac{(n+1)P_c}{4\pi G \rho_c^2} \right)^{1/2} \epsilon_1$$

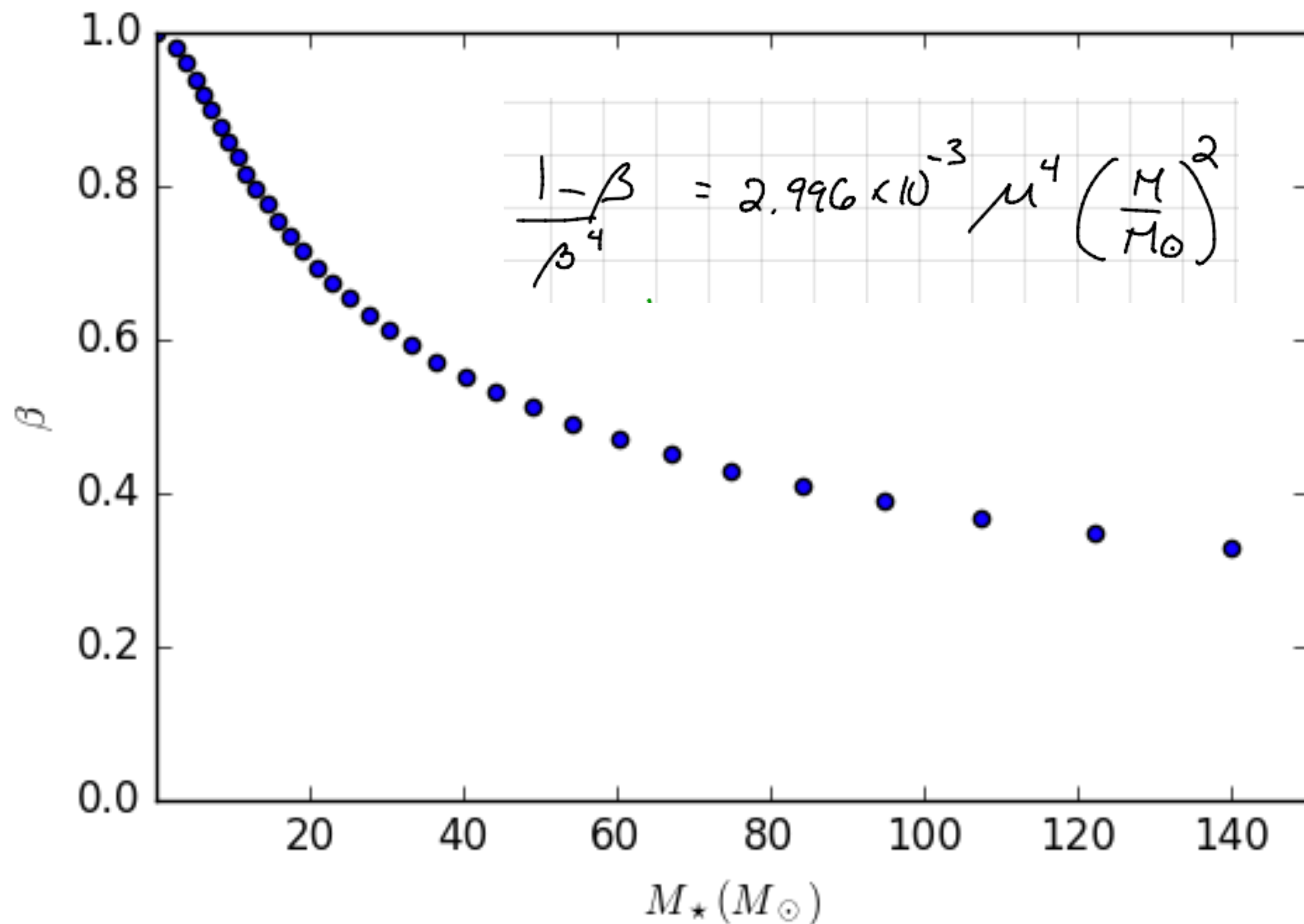
$$2 \quad P_c = K \rho_c^{(n+1)/n} \quad \text{If } n=3, P_c = K \rho_c^{4/3}, \text{ so } K = \frac{P_c}{\rho_c^{4/3}} \text{ and } K^{3/2} = \frac{P_c^{3/2}}{\rho_c^2}$$

$$3 \quad M_{\star} = -\frac{1}{(4\pi)^{1/2}} \left(\frac{n+1}{G} \right)^{3/2} \boxed{\frac{P_c^{3/2}}{\rho_c^2}} \epsilon_1^2 \theta'(\epsilon_1)$$

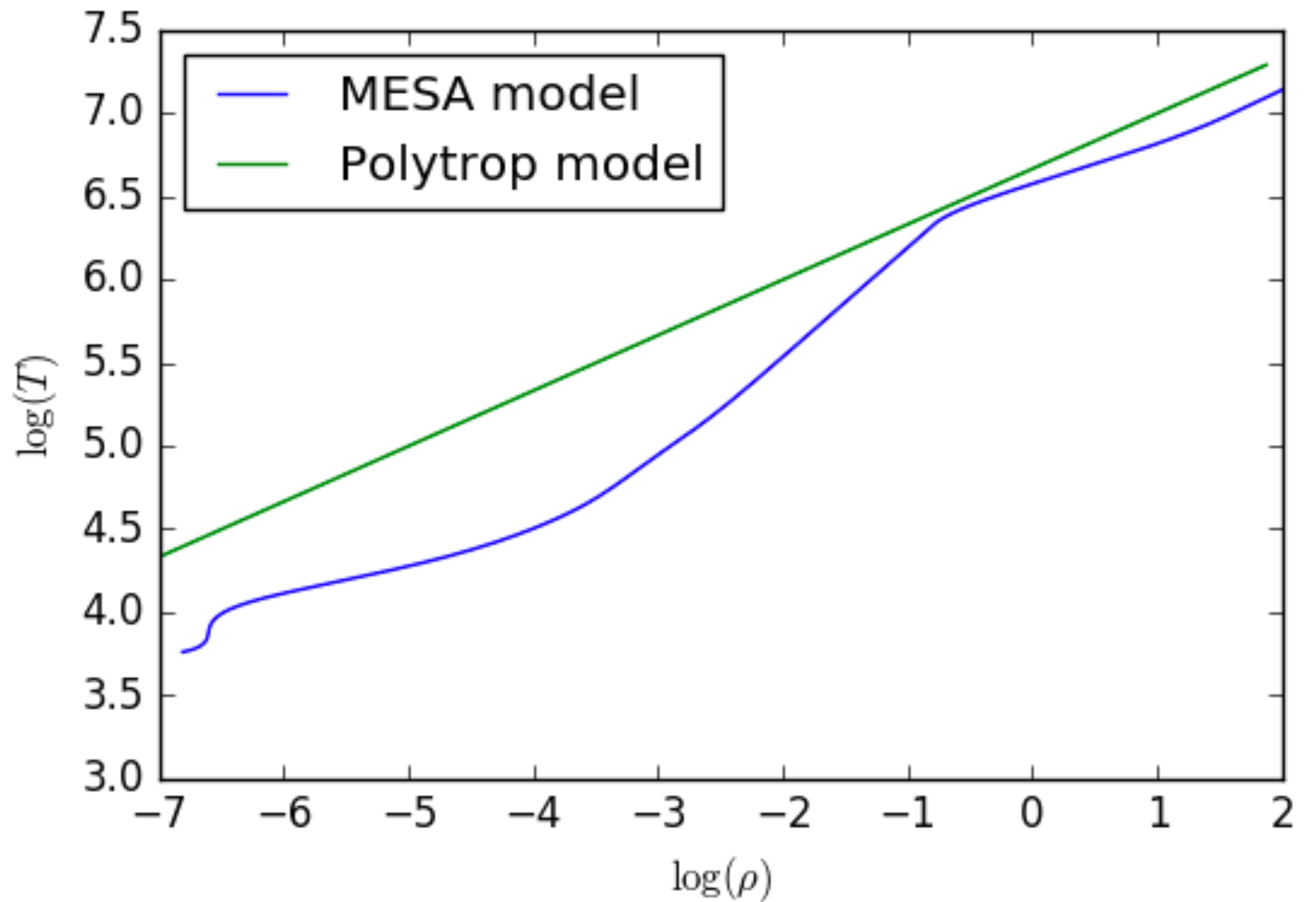
$$K = \frac{(4\pi)^{1/3}}{4} \frac{G M_{\star}^{2/3}}{[\epsilon_1^2 (-\theta'_1)]^{2/3}}$$

$$P_{\text{tot}} = \underbrace{\left[\frac{(1-\beta)}{\beta^4} \frac{3c}{4\sigma} \left(\frac{k}{\mu_{\text{MH}}} \right)^4 \right]^{1/3}}_K \rho^{4/3}$$

$$K = \frac{(4\pi)^{1/3}}{4} \frac{GM_*^{2/3}}{[E_1^2 (-\theta'_{t_1})]^{2/3}}$$



$$T(r) = \left[\frac{(1-\beta)}{\beta} \frac{3c}{4\sigma} \frac{k_B}{\mu m_H} \right]^{1/3} \rho^{1/3}$$



We already know how to calculate
the density for a polytrop