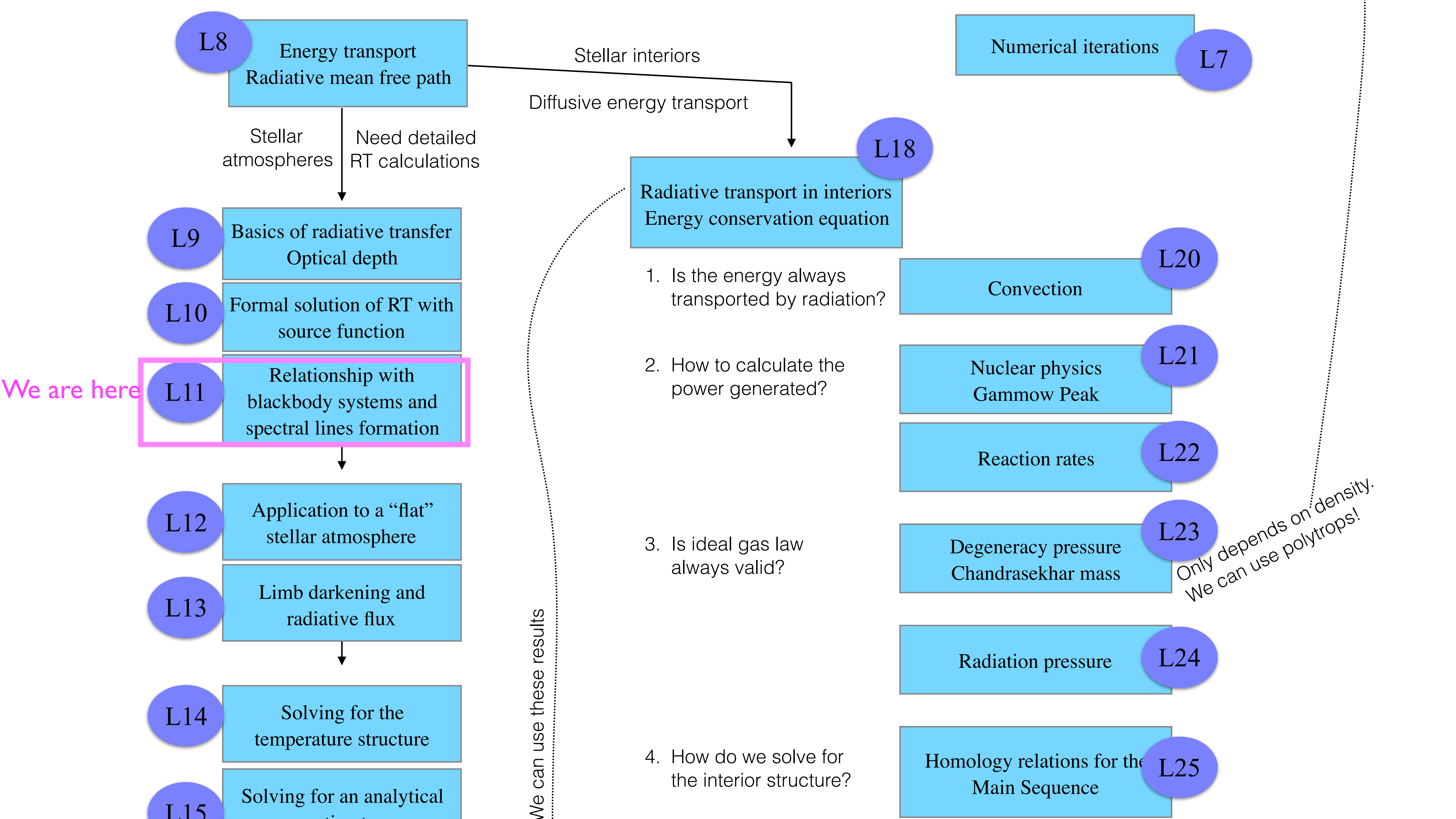


Week 6 Tuesday

L11

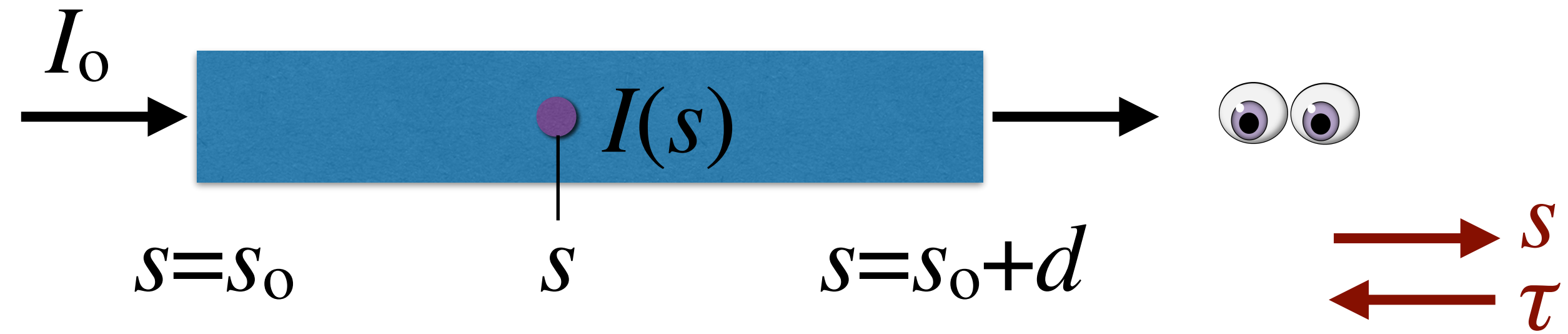


Exam: week after the break -> sign up sheet (google doc?).

Exam questions will be posted on Thursday

Material for exam: up until this lecture inclusively (thermal radiation)

Change in intensity



Absorption + emission

$$-\frac{dI_{\lambda}(s)}{\kappa_{\lambda}(s)\rho(s)ds} = I_{\lambda}(s) - S_{\lambda}(s)$$

$$\frac{dI_{\lambda}(\tau_{\lambda})}{d\tau_{\lambda}} = I_{\lambda}(\tau_{\lambda}) - S_{\lambda}(\tau_{\lambda})$$

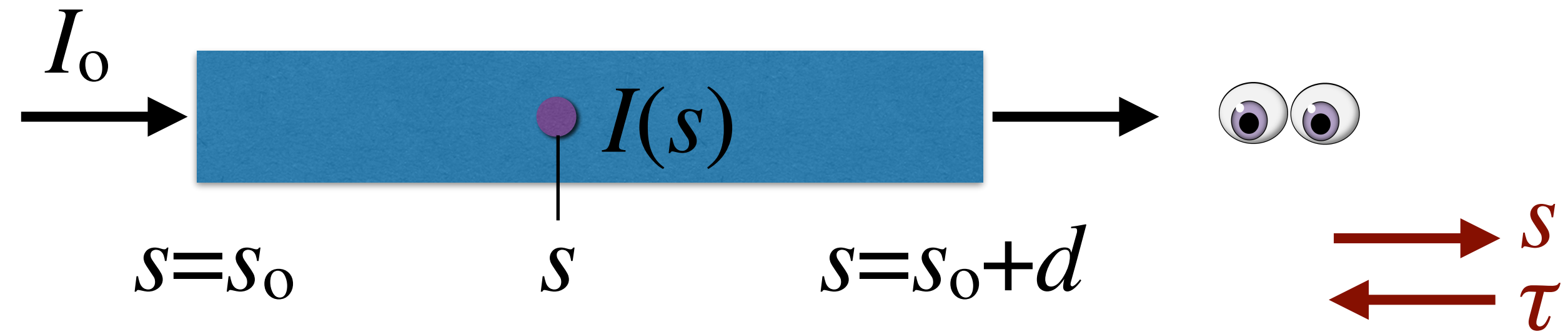
General Solution

$$I(\tau(s)) = I_o e^{\tau(s)-\tau_o} + \int_{\tau'=\tau(s)}^{\tau'=\tau_o} S(\tau') e^{\tau(s)-\tau'} d\tau'$$

Constant source function

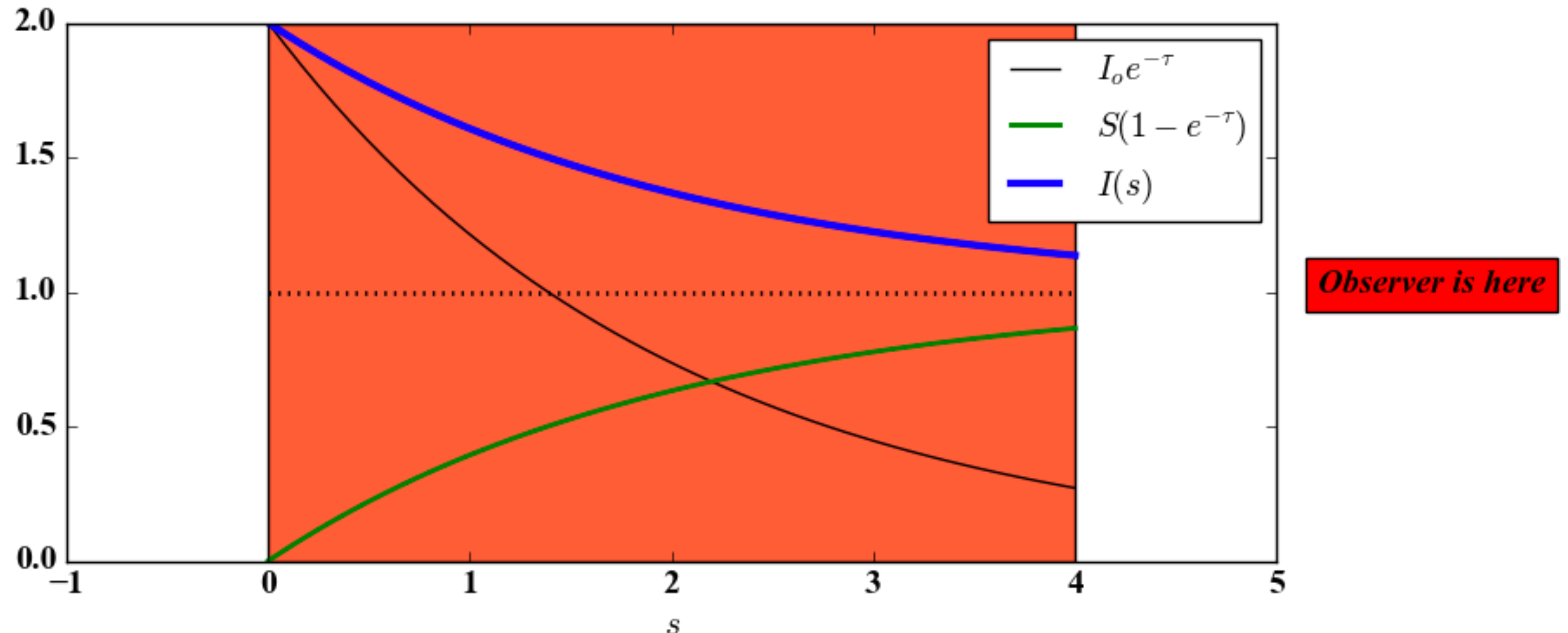
$$I(\tau(s)) = I_o e^{\tau(s)-\tau_o} + S \left[1 - e^{\tau(s)-\tau_o} \right]$$

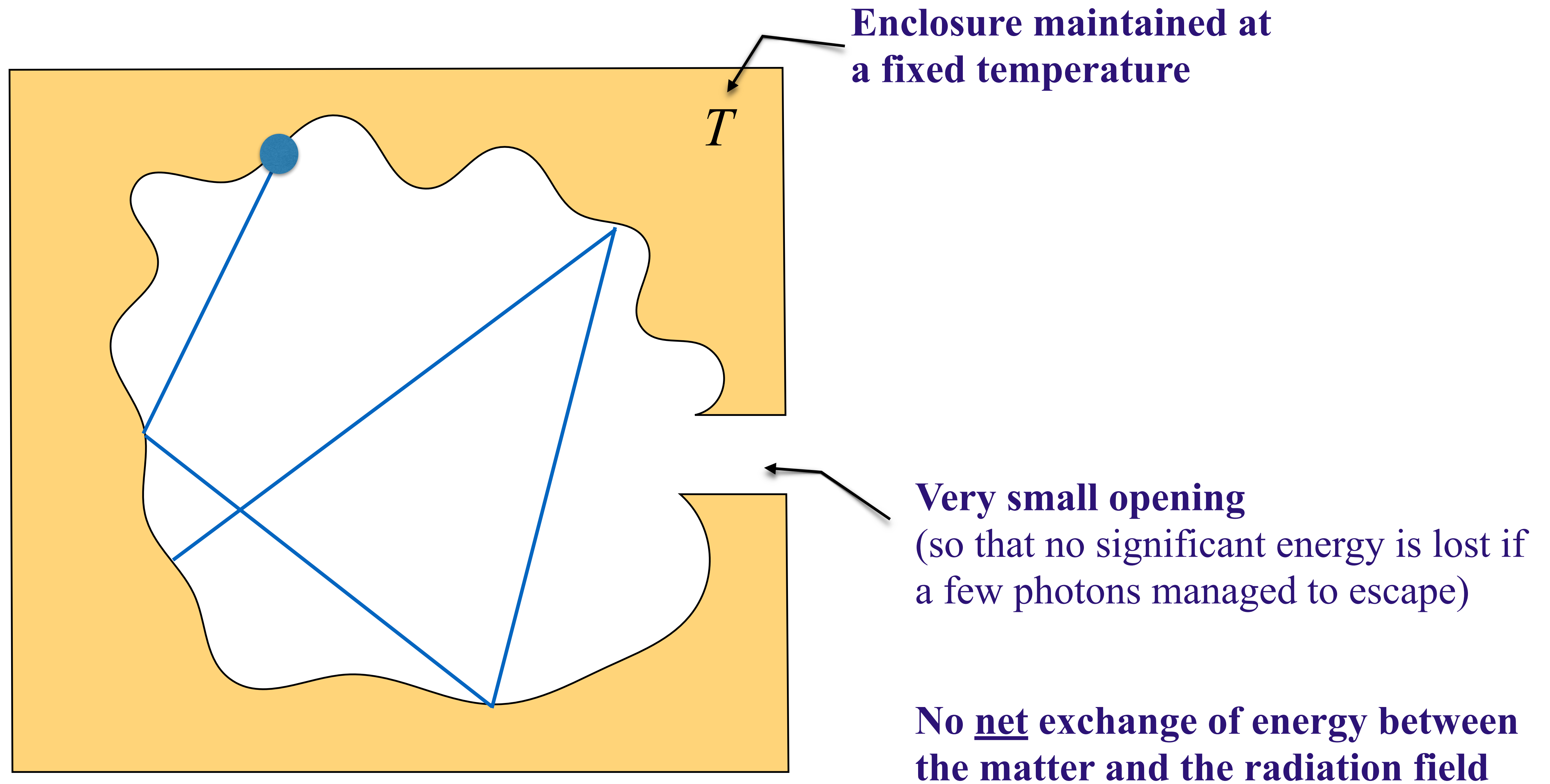
Change in intensity

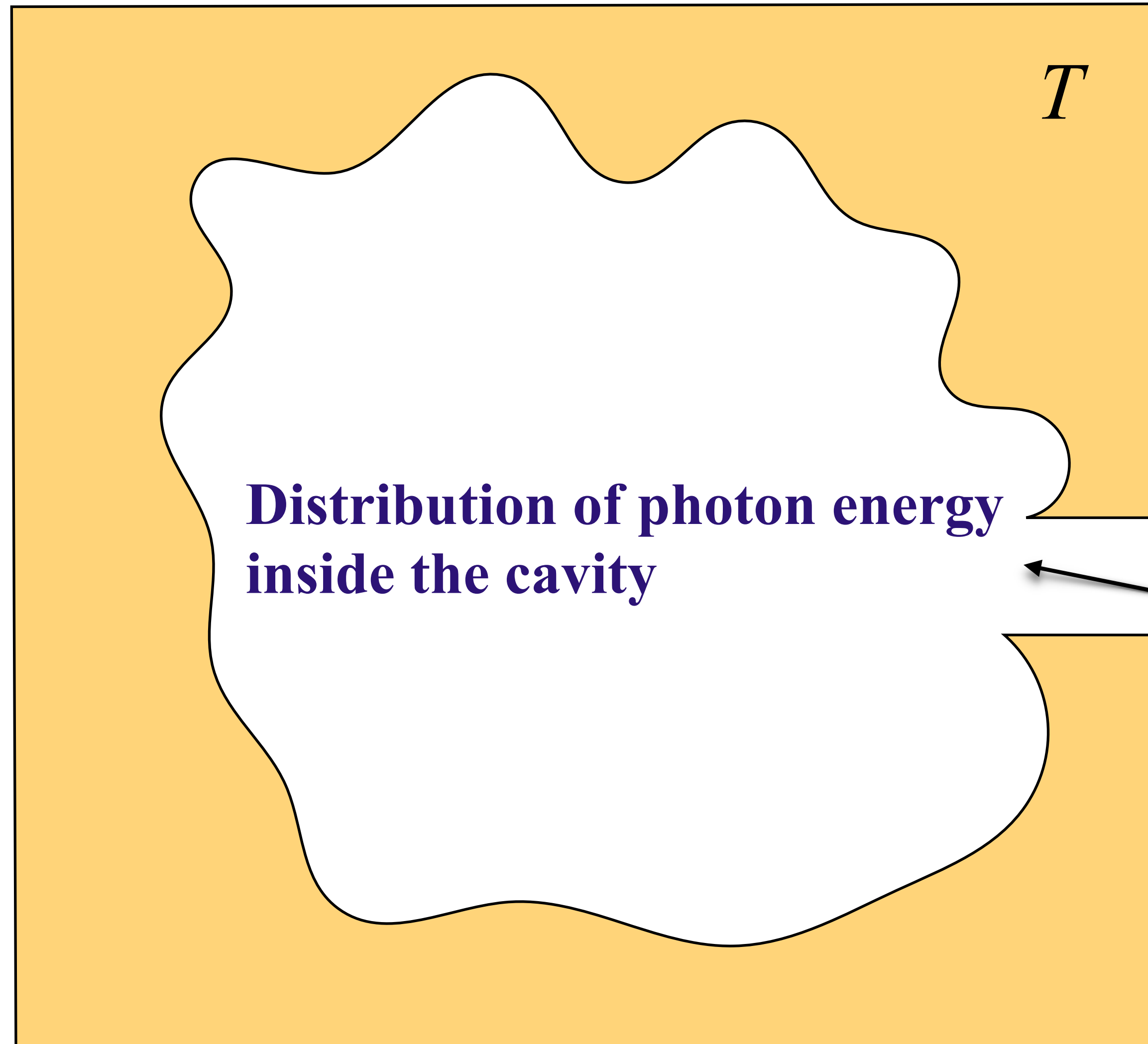


Constant values

$$I(\tau(s)) = I_o e^{\tau(s)-\tau_o} + S \left[1 - e^{\tau(s)-\tau_o} \right]$$





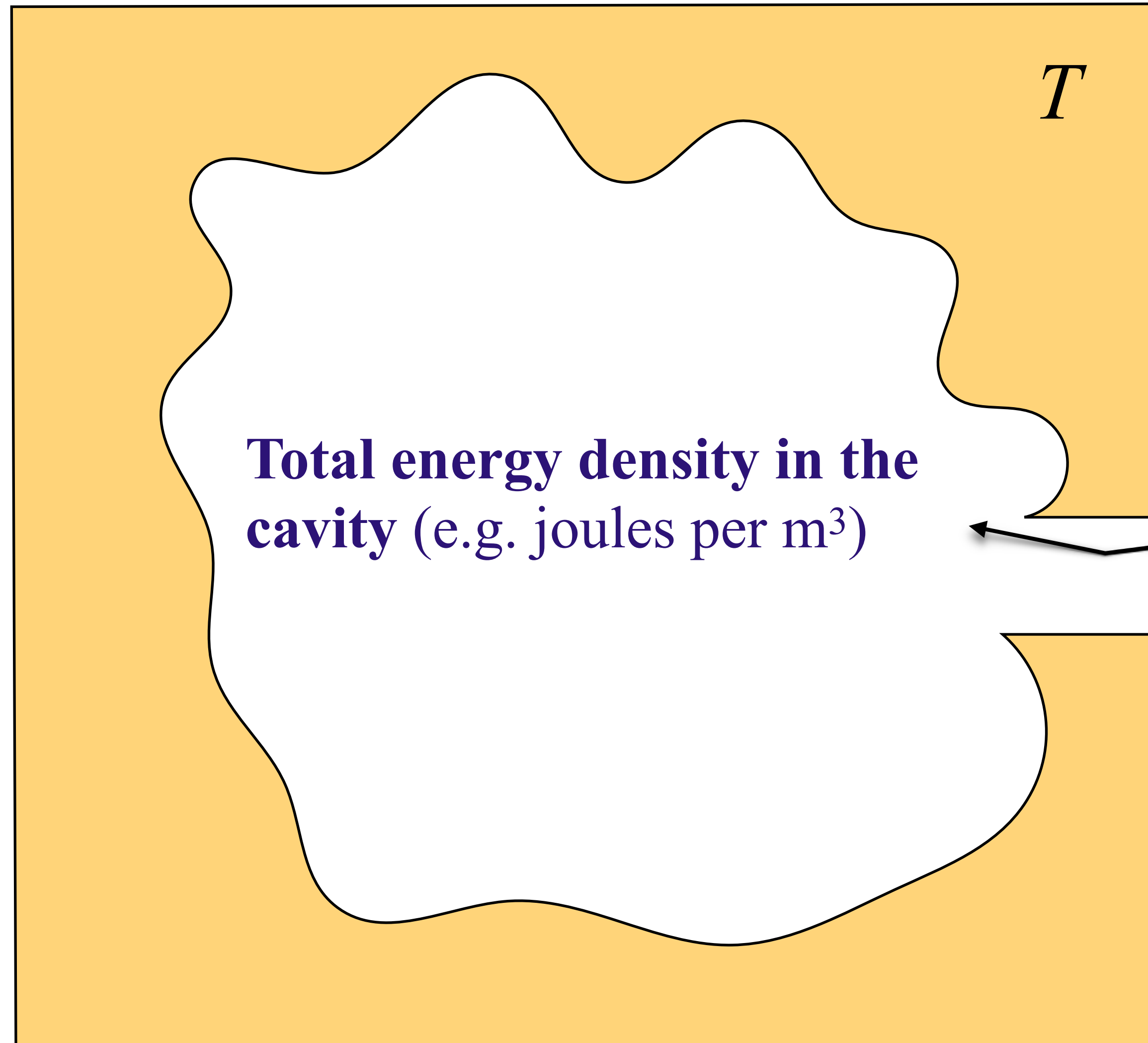


**Energy density (u)
per unit of photon energy (E)**

e.g. joules per m³ per keV

$$u(E) = \frac{8\pi}{(hc)^3} \frac{E^3}{e^{E/kT} - 1}$$

(Undergrad Thermal Physics
textbook by Schroeder, Sec 7.4)



$$U = \frac{8\pi}{(hc)^3} \int_0^\infty \frac{E^3}{e^{E/kT} - 1}$$

$$= \frac{4}{c} \boxed{\frac{2\pi^4 k^4}{15h^3 c^2}} T^4$$

Stefan-Boltzmann constant σ

$$U = \frac{4\sigma}{c} T^4$$

Energy density (u)
per unit of photon energy (E)

e.g. joules per m³ per keV

Energy density (u)
per unit of photon frequency (ν)

e.g. joules per m³ per Hz

Energy density (u)
per unit of photon wavelength (λ)

e.g. joules per m³ per nm

$$\int_0^\infty u(E) \, dE$$

$$= \frac{4\sigma}{c} T^4$$

$$\int_0^\infty u(\nu) \, d\nu$$

$$= \frac{4\sigma}{c} T^4$$

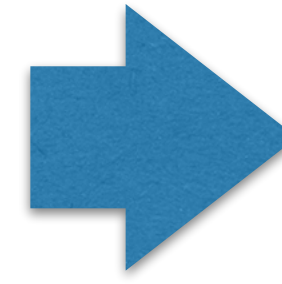
$$\int_\infty^0 u(\lambda) \, d\lambda$$

$$= \frac{4\sigma}{c} T^4$$

$$u(E)dE = u(\nu)d\nu = -u(\lambda)d\lambda$$

**Energy density (u)
per unit of photon energy (E)**

e.g. joules per m³ per keV



**Energy density (u)
per unit of photon frequency (ν)**

e.g. joules per m³ per Hz

$$u(E)dE = u(\nu)d\nu$$

$$E = h\nu \quad dE = h d\nu$$

$$u(\nu) = u(E) \frac{dE}{d\nu}$$

$$\frac{dE}{d\nu} = h$$

$$= \frac{8\pi}{(hc)^3} \frac{(h\nu)^3}{e^{(h\nu)/kT} - 1}$$

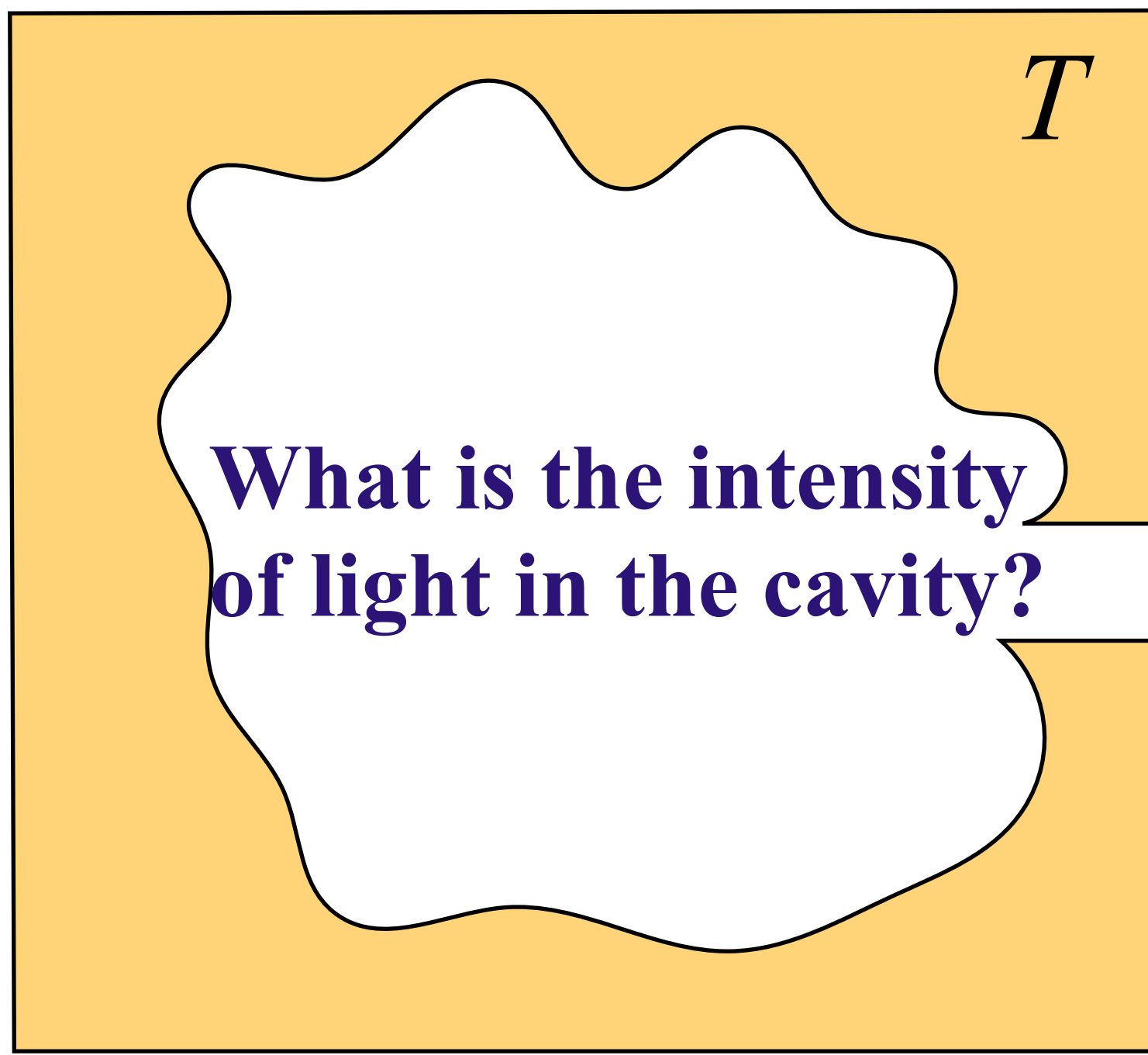


Still has units of energy

$$\frac{dE}{d\nu}$$



Units of energy per frequency



Energy density (u)
per unit of photon frequency (ν)
 e.g. joules per m³ per Hz

$$u(\nu) = \frac{8\pi h\nu^3}{c^3} \frac{1}{e^{h\nu/kT} - 1}$$

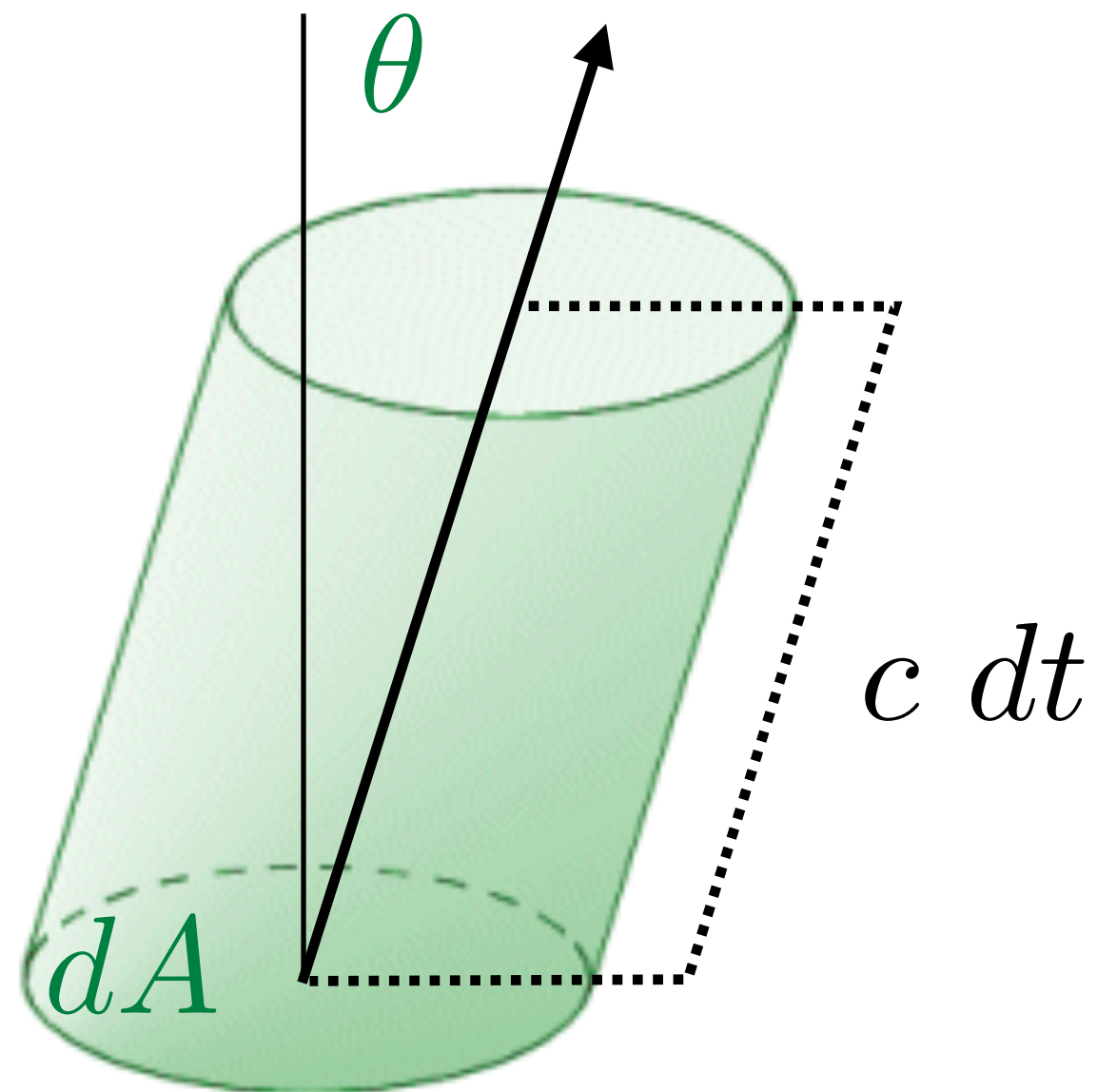
$$E = \int I_\nu \, d\nu \, dt \, dA \cos(\theta) \, d\Omega$$

$$u(\nu) = \frac{E_{\text{Transported by the light rays}}}{d\nu dV}$$

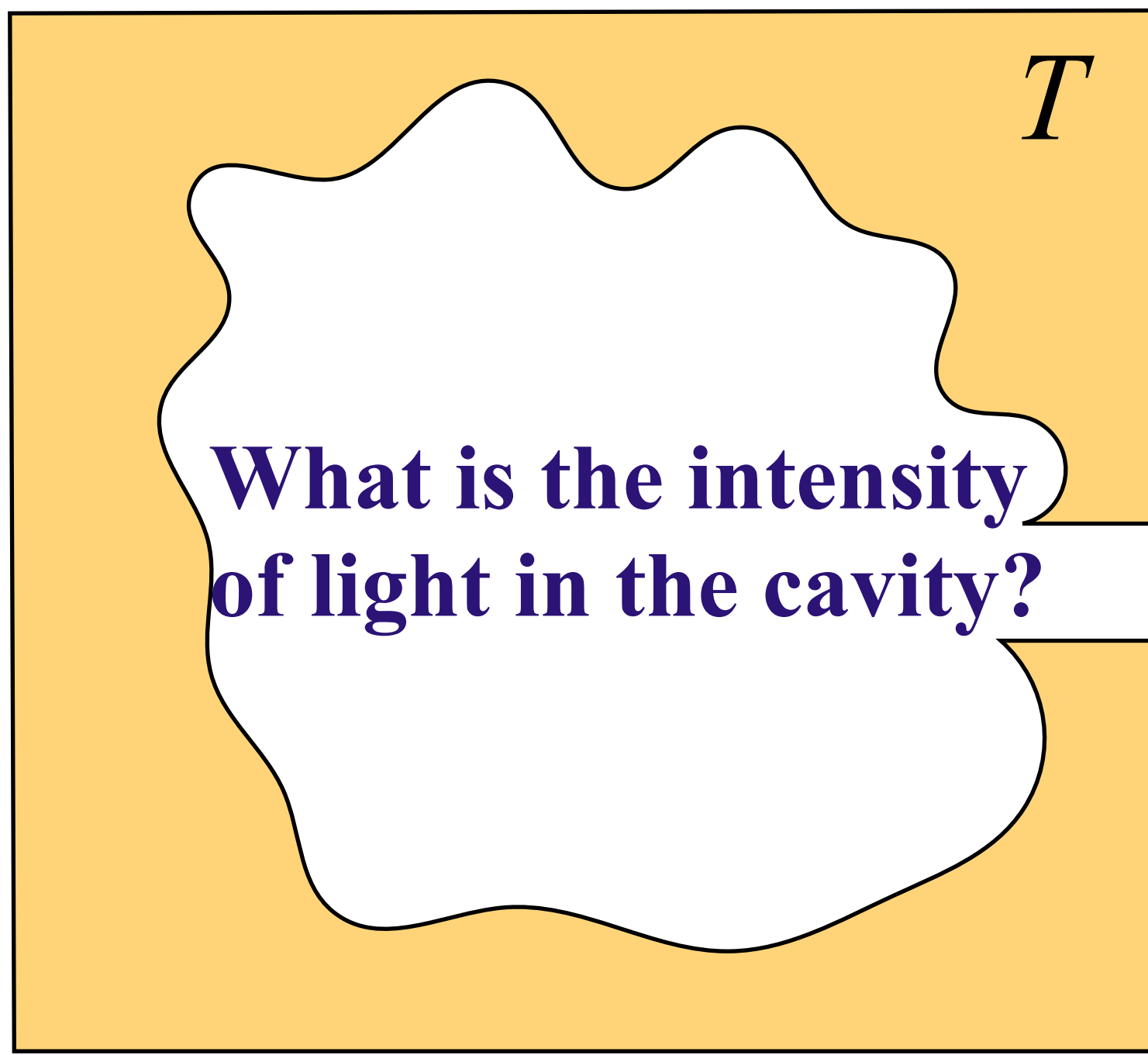
$$= \int I_\nu \frac{d\nu \, dt \, dA \cos(\theta)}{d\nu \, c \, dt \, dA \cos(\theta)} \, d\Omega$$

$$= \frac{1}{c} \int I_\nu \, d\Omega$$

[
4\pi J_\nu
]



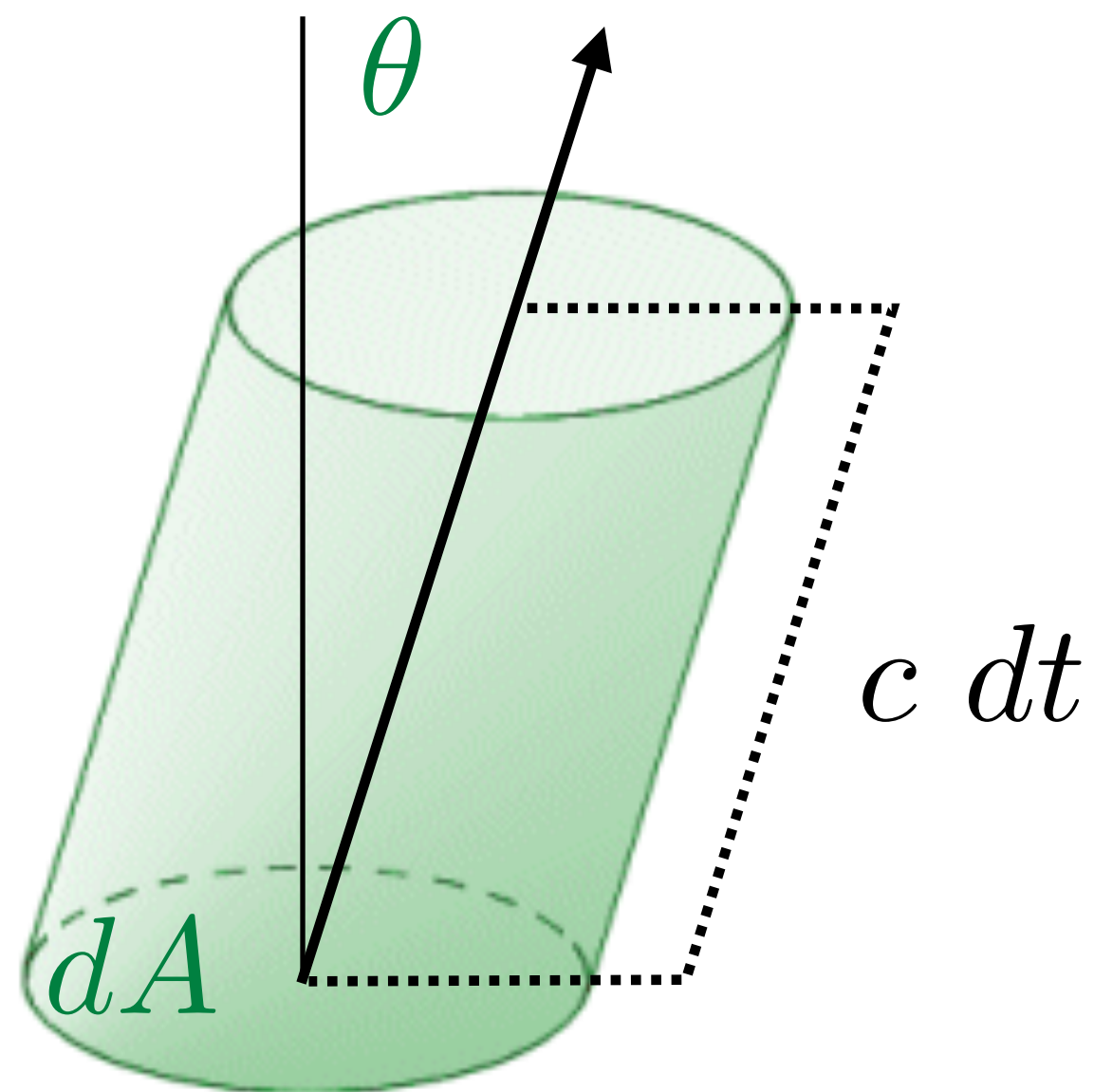
$$dV = c \, dt \, dA \cos(\theta)$$



Energy density (u)
per unit of photon frequency (ν)
 e.g. joules per m³ per Hz

$$u(\nu) = \frac{8\pi h\nu^3}{c^3} \frac{1}{e^{h\nu/kT} - 1} = \frac{4\pi}{c} J_\nu \quad \text{if radiation is isotropic} = \frac{4\pi}{c} I_\nu$$

$$I_{\nu, BB} = B_\nu(T) = \frac{c}{4\pi} u(\nu) = \frac{2h}{c} \frac{\nu^3}{e^{h\nu/kT} - 1}$$

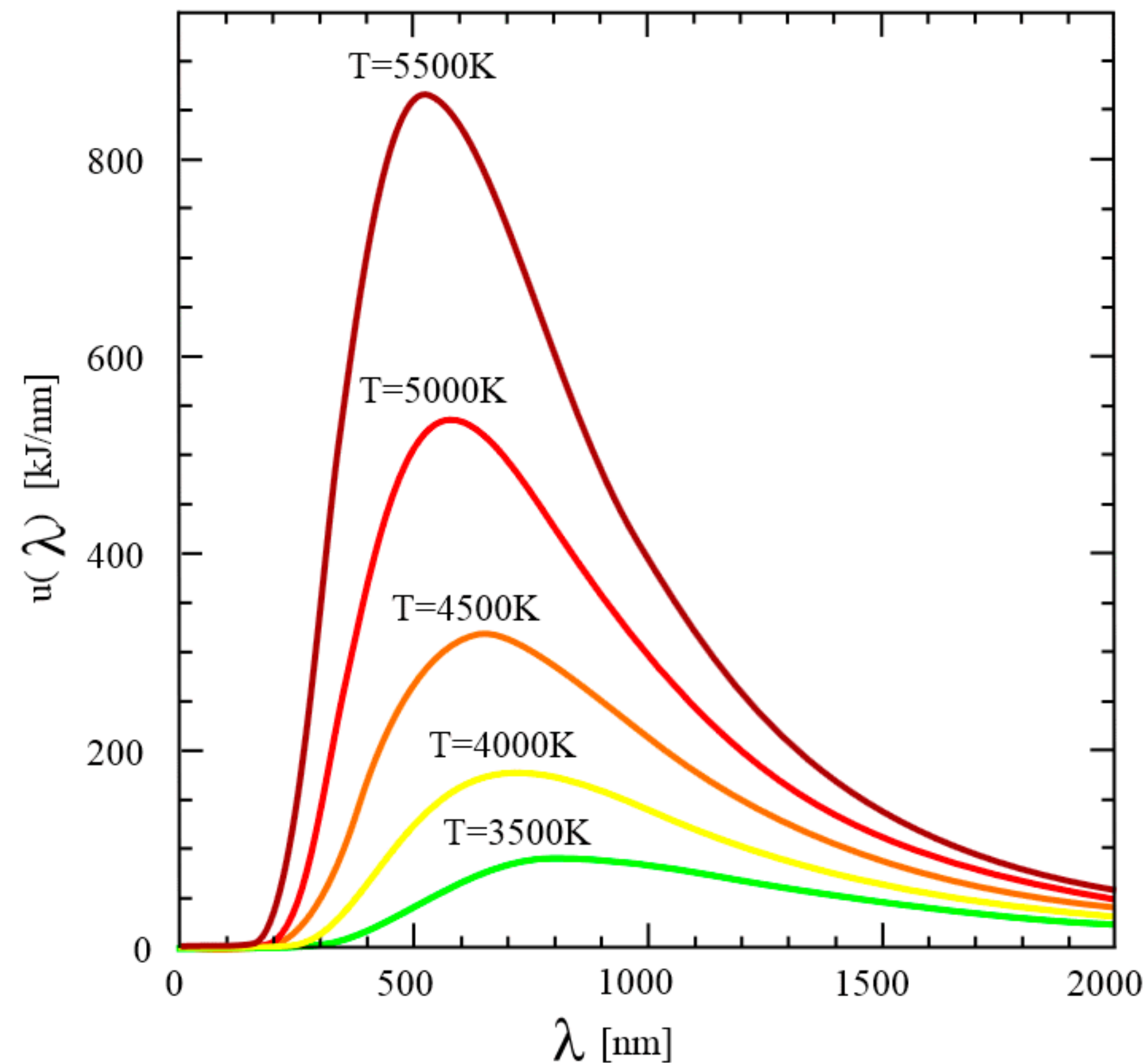


$$dV = c dt dA \cos(\theta)$$

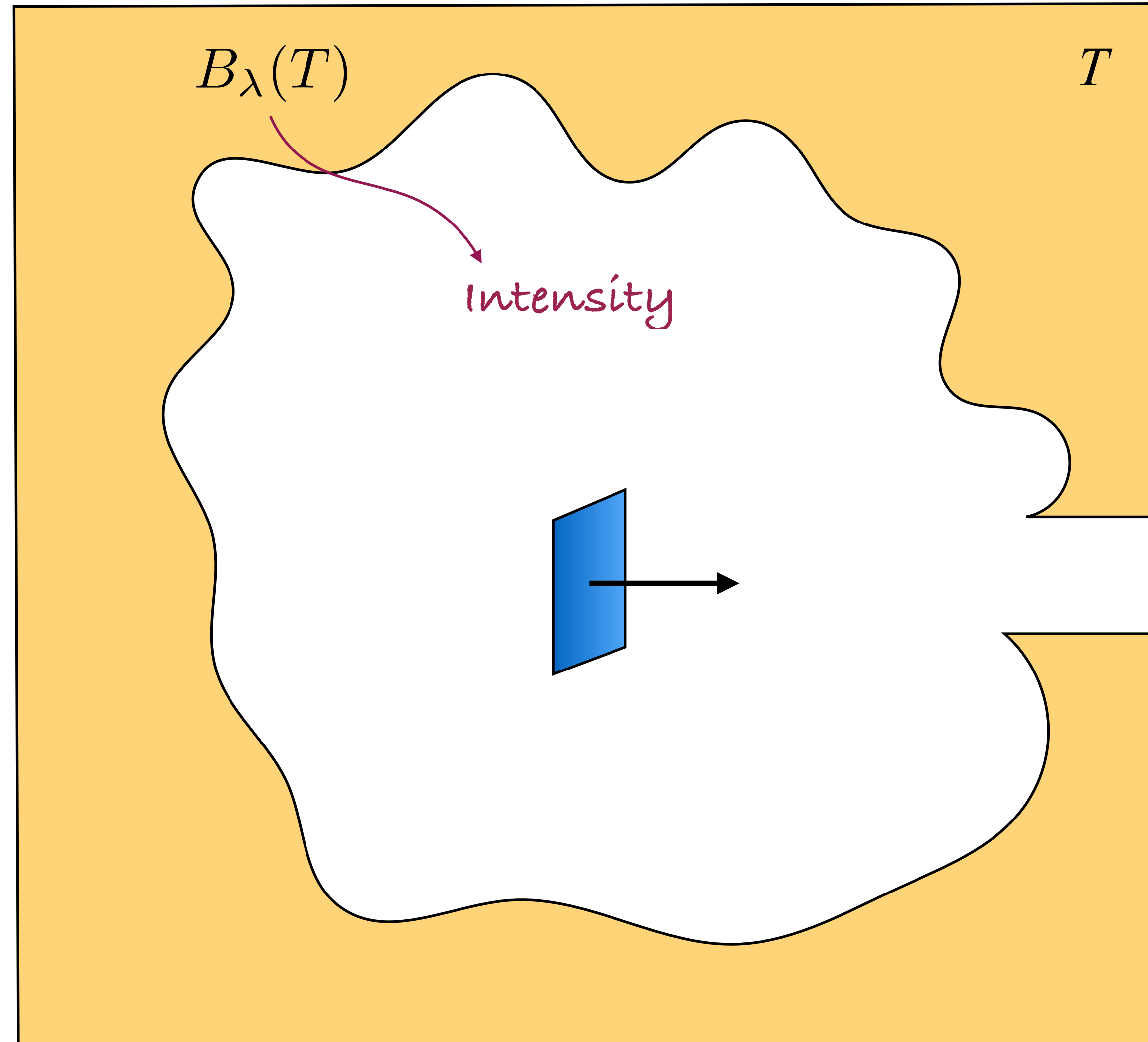
One of the exam question: find B_λ
 (beware: same principle as converting between $u(E)$ and $u(\nu)$)

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(beware: same principle as converting between $u(E)$ and $u(\nu)$)

(And it should look like that :))



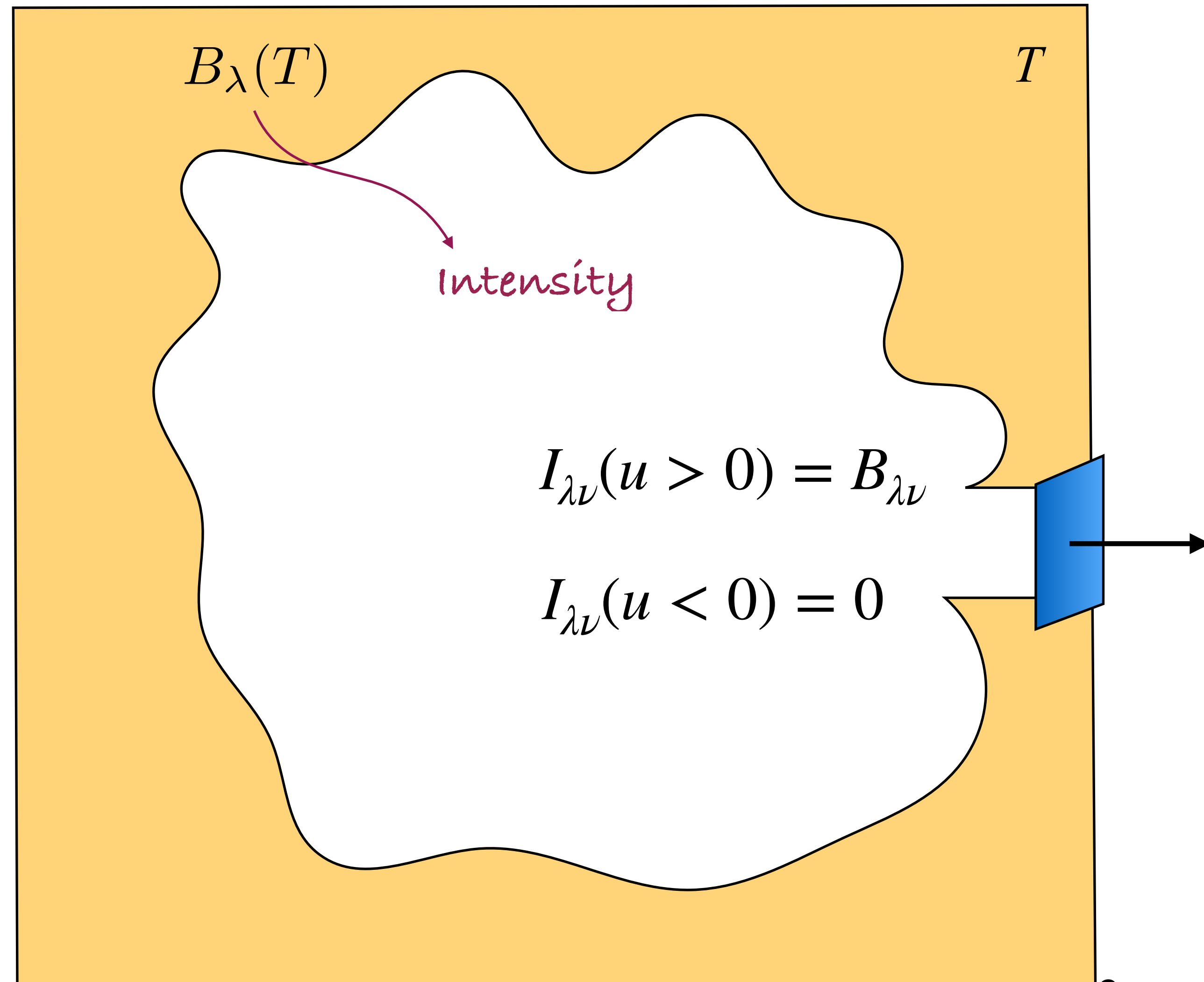
Flux through a surface in a blackbody radiation field



Isotropic radiation:

$$F = 0$$

Flux through a surface in a blackbody radiation field



$$I_{\nu, BB} = B_\nu(T) = \frac{c}{4\pi} u(\nu) = \frac{2h}{c} \frac{\nu^3}{e^{h\nu/kT} - 1}$$

$$\begin{aligned} F_{\lambda\nu} &= 2\pi \int_{-1}^{+1} B_{\lambda\nu} u du \\ &= 2\pi B_{\lambda\nu} \int_0^{+1} u du \\ &= \pi B_{\lambda\nu} \\ F_{\text{tot}} &= \pi \int_0^\infty B_{\lambda\nu} d\nu \\ &= \pi \int_0^\infty \frac{2h}{c^2} \frac{\nu^3}{e^{h\nu/kT} - 1} d\nu \\ &= \frac{2\pi^5 k^4}{15h^3 c^2} T^4 \\ &= \sigma T^4 \end{aligned}$$

Source function of a blackbody radiation field

Thermodynamical equilibrium: each photon absorbed is replaced by an emitted photon of the same wavelength. $dI_\lambda = 0$

On the board: this implies that $S_\lambda = I_\lambda$

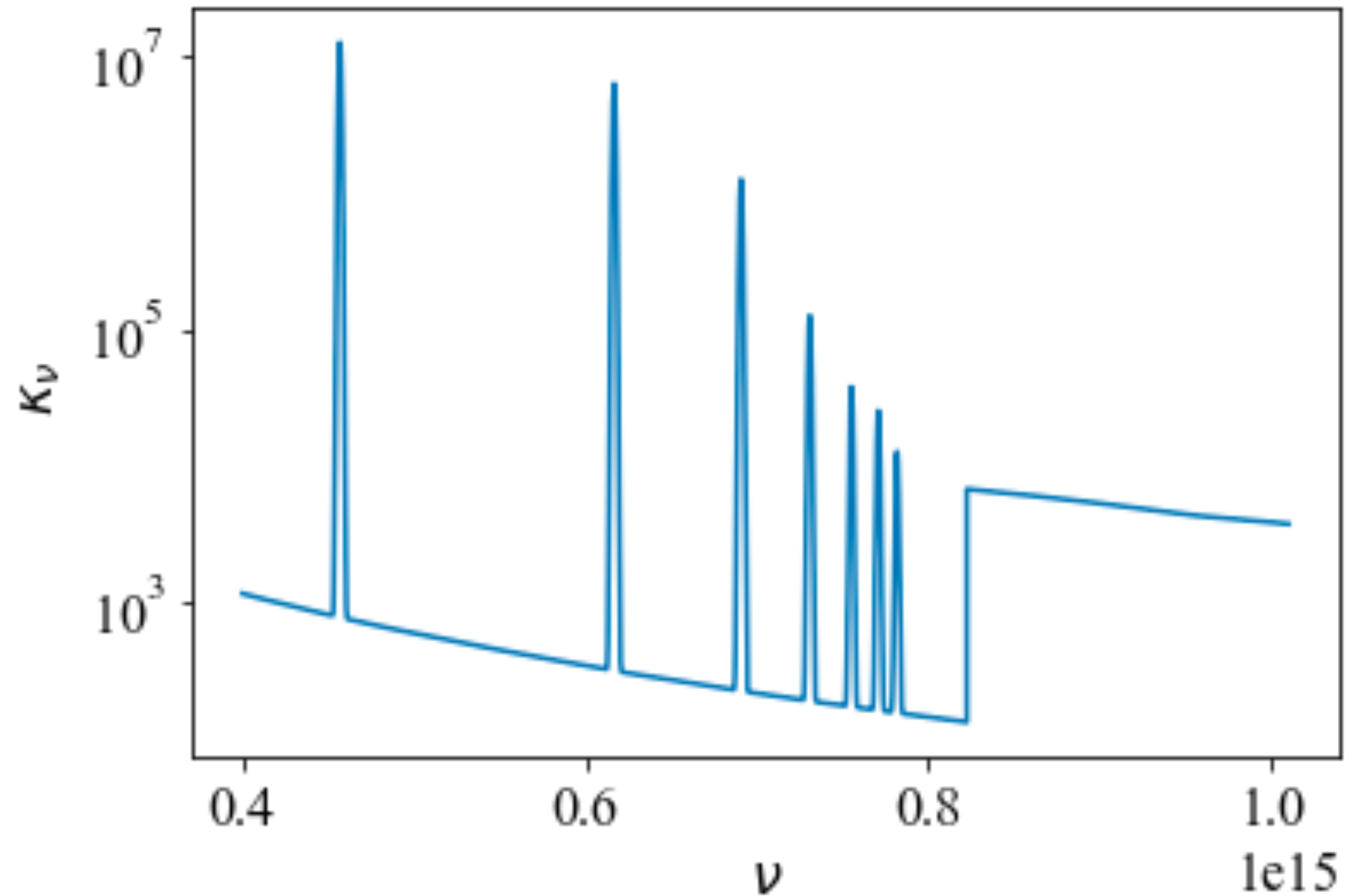
In the more general case where there is both absorption k_λ and scattering σ_λ processes

On the board: this implies that $S_\lambda = \frac{k_\lambda B_\lambda + \sigma_\lambda J_\lambda}{k_\lambda + \sigma_\lambda}$

The opacity wavelength dependance:

Q: what are the spikes?
Q: what is the step?

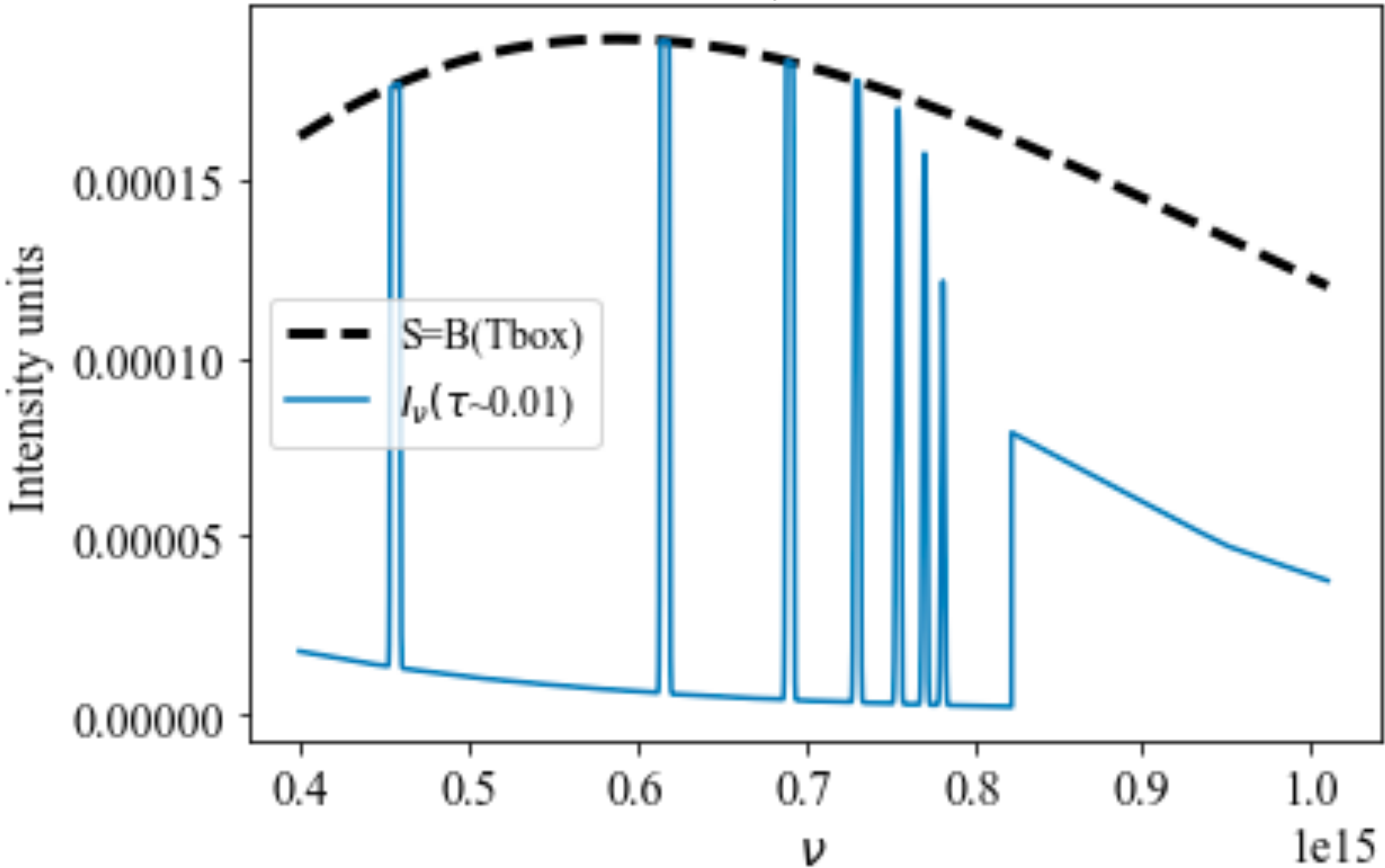
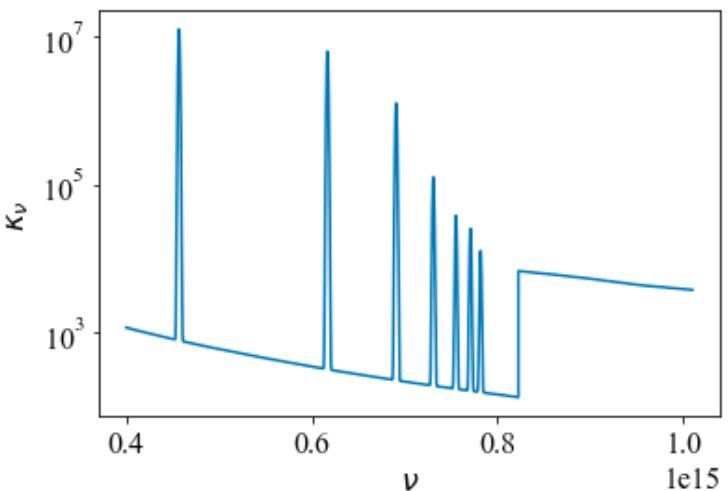
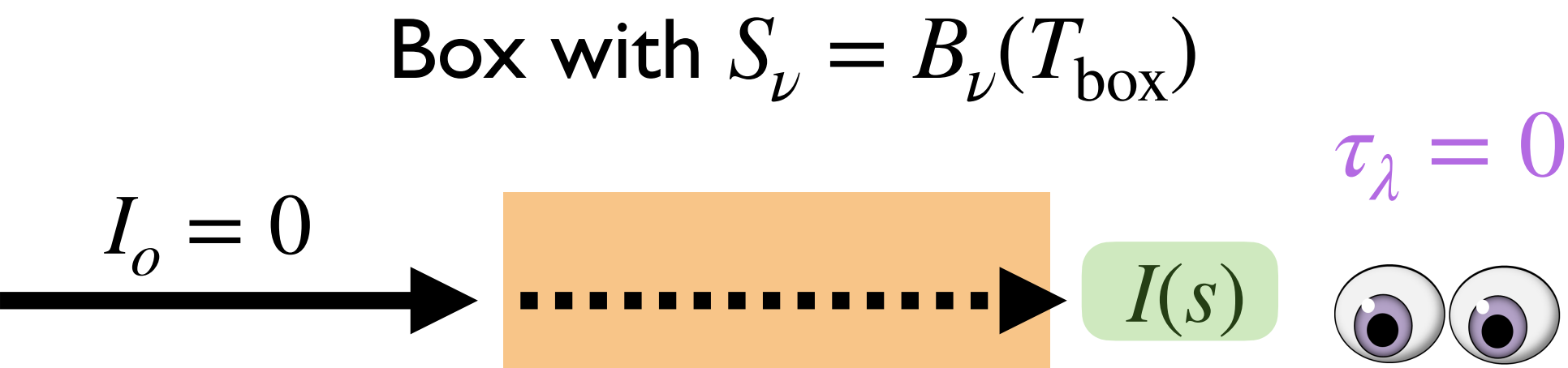
Example of pure hydrogen



Example 1

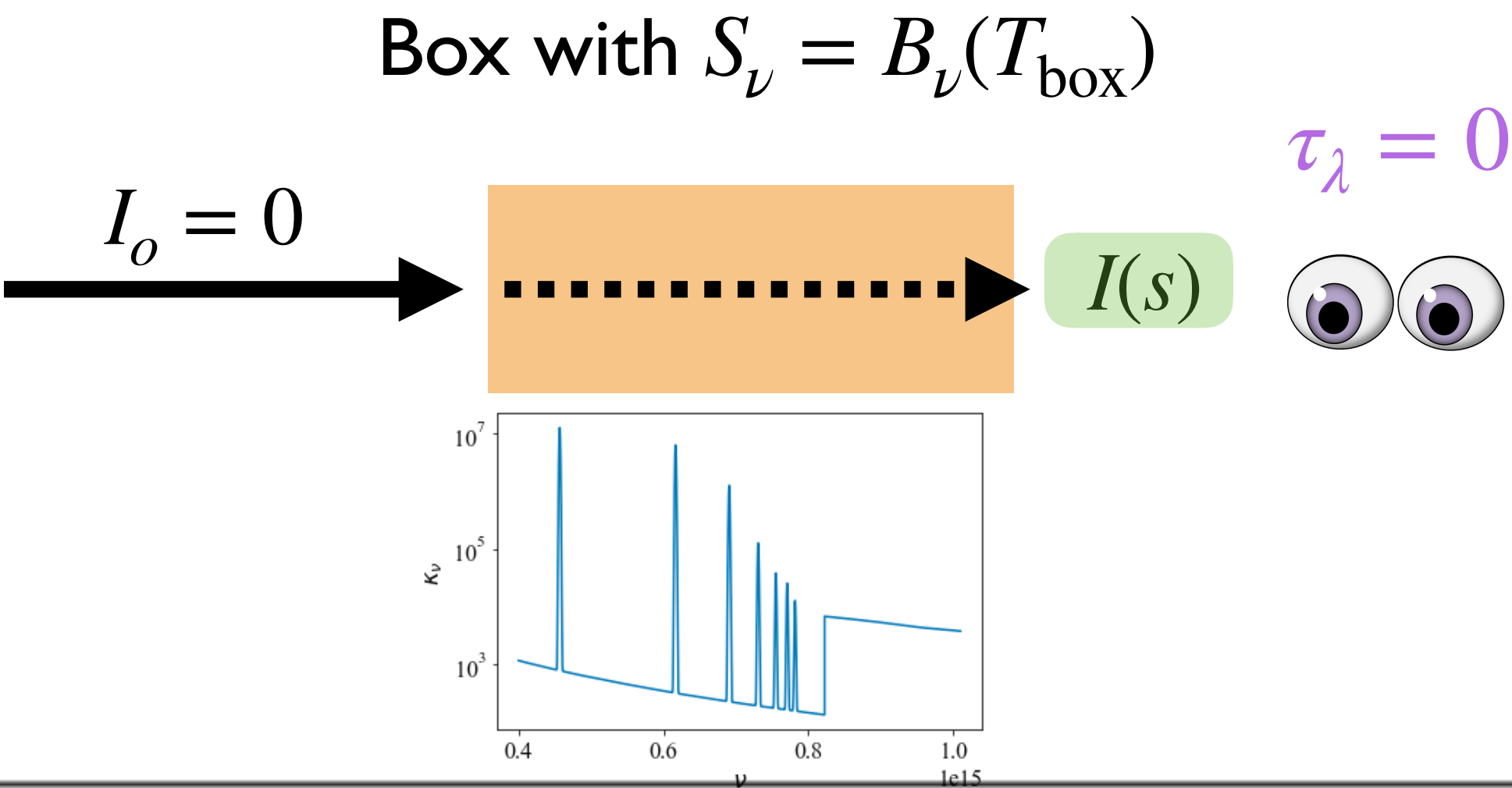
$$I(s) = I_o e^{\tau(s)-\tau_o} + S_o [1 - e^{\tau(s)-\tau_o}]$$
$$I(\text{obs}) = + B_\nu(T_{\text{box}}) [1 - e^{-\tau_o}]$$

a. The optical depth of the box is small

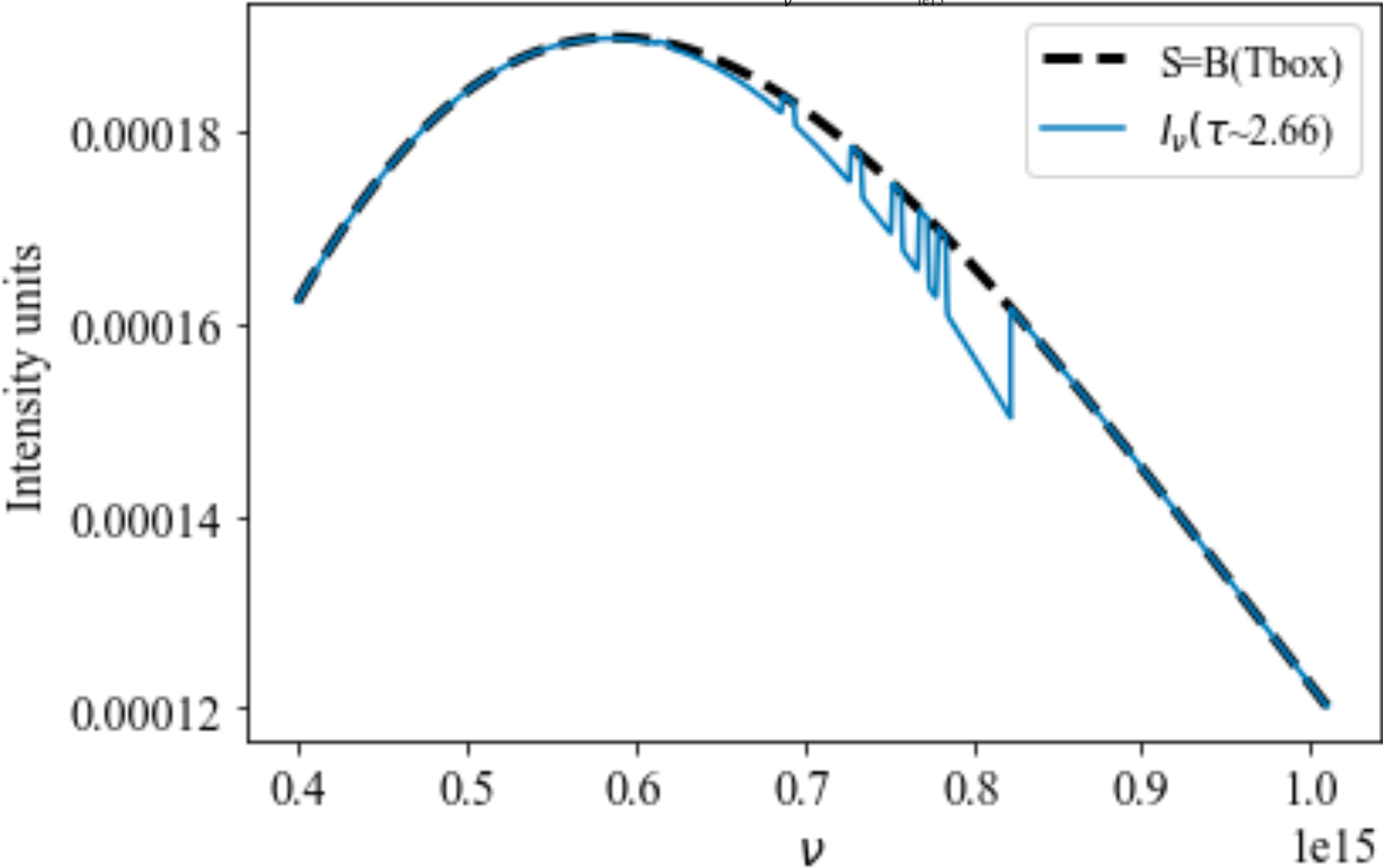


Example 1

$$I(s) = I_o e^{\tau(s)-\tau_o} + S_o \left[1 - e^{\tau(s)-\tau_o} \right]$$
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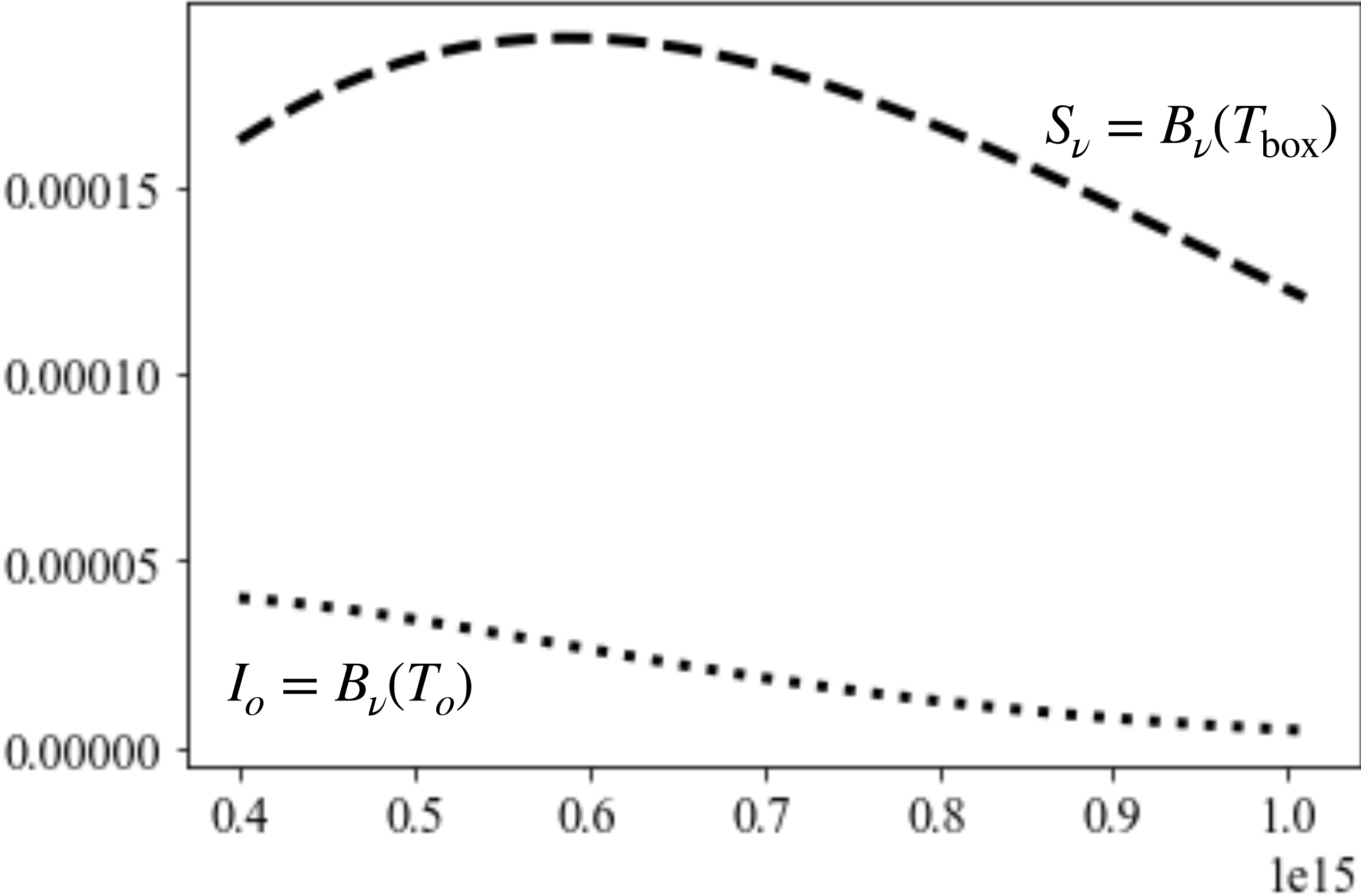
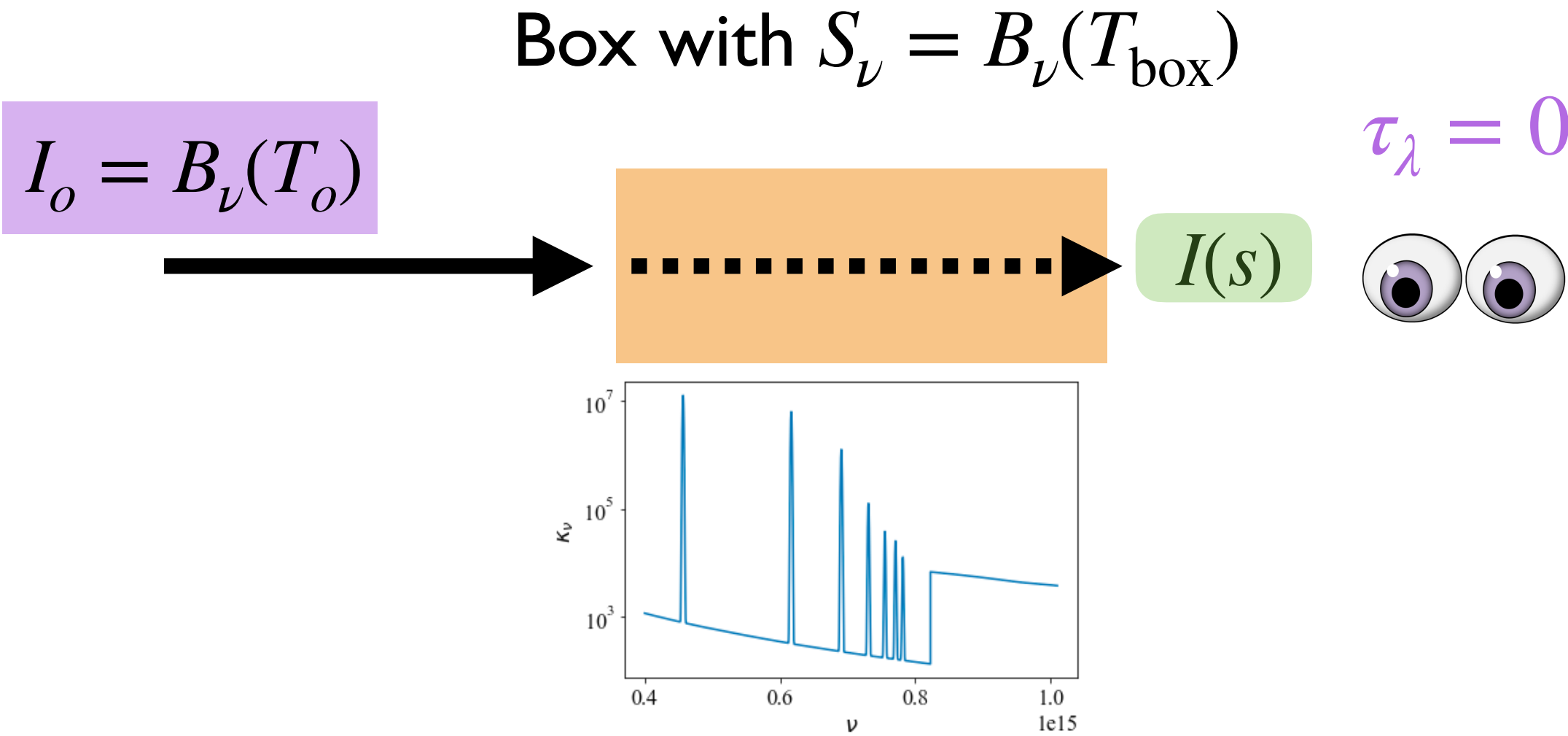


b. The optical depth of the box is large



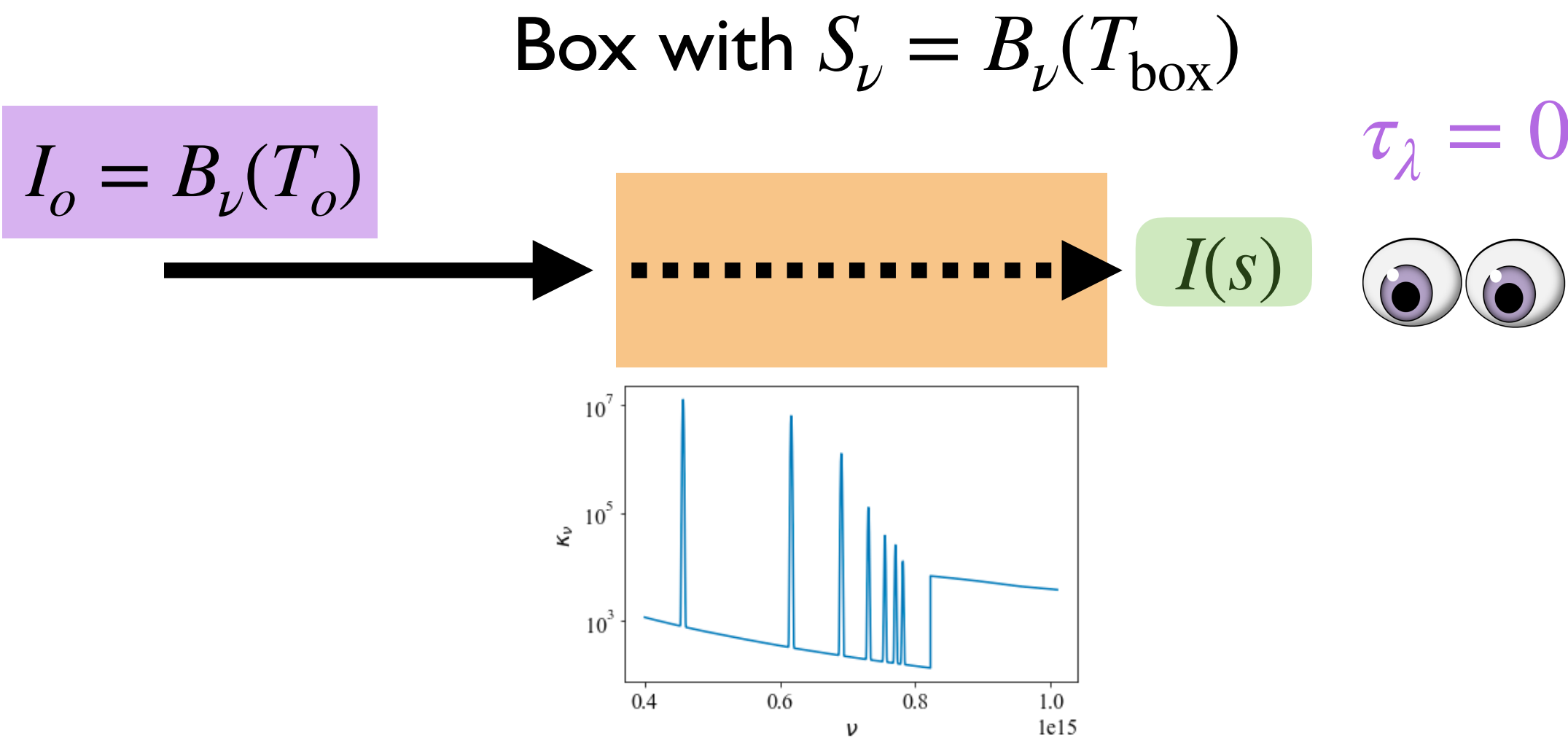
Example 2: $T_{\text{box}} > T_o$

$$I(s) = I_o e^{\tau(s) - \tau_o} + S_o [1 - e^{\tau(s) - \tau_o}]$$
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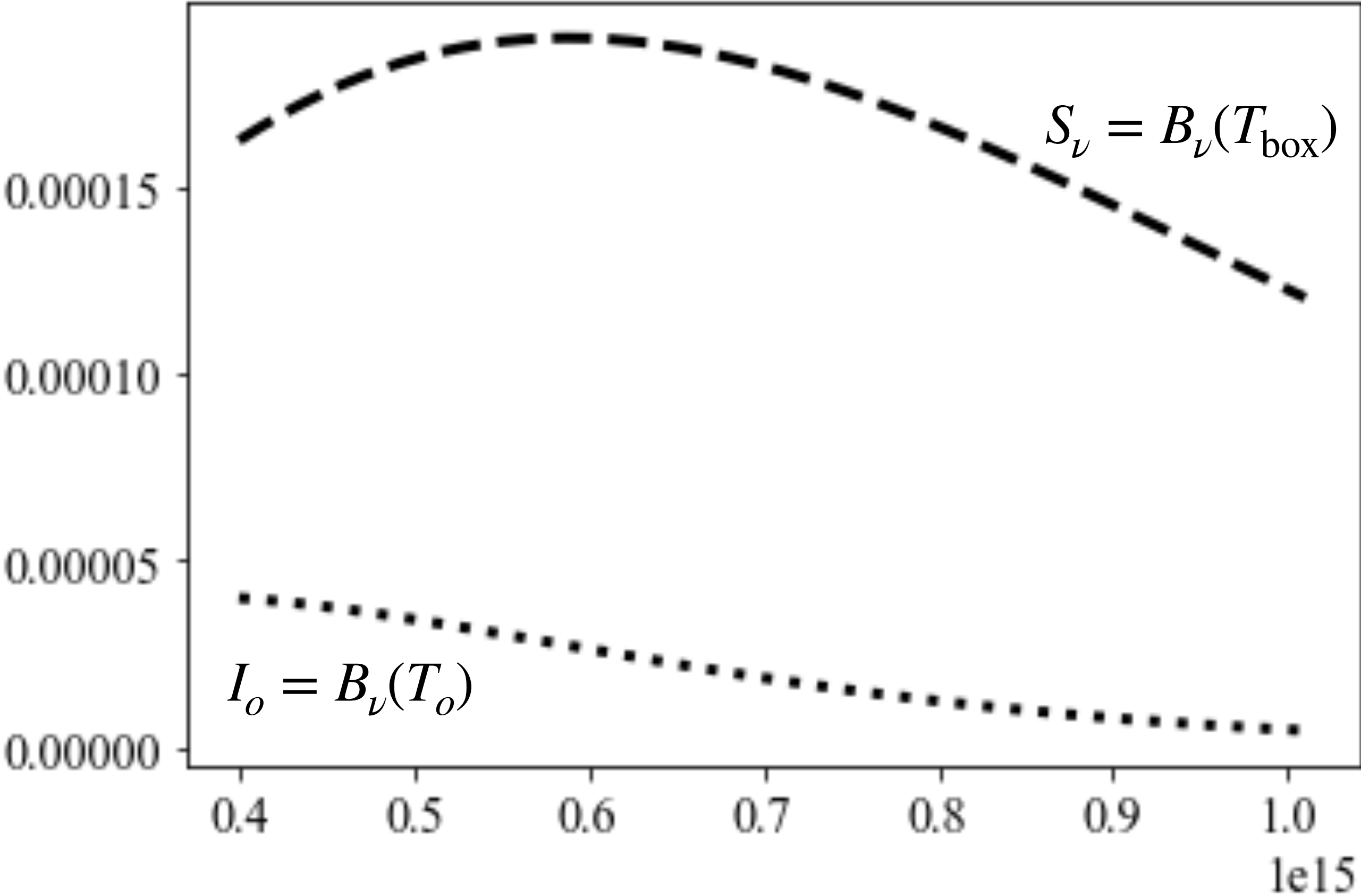


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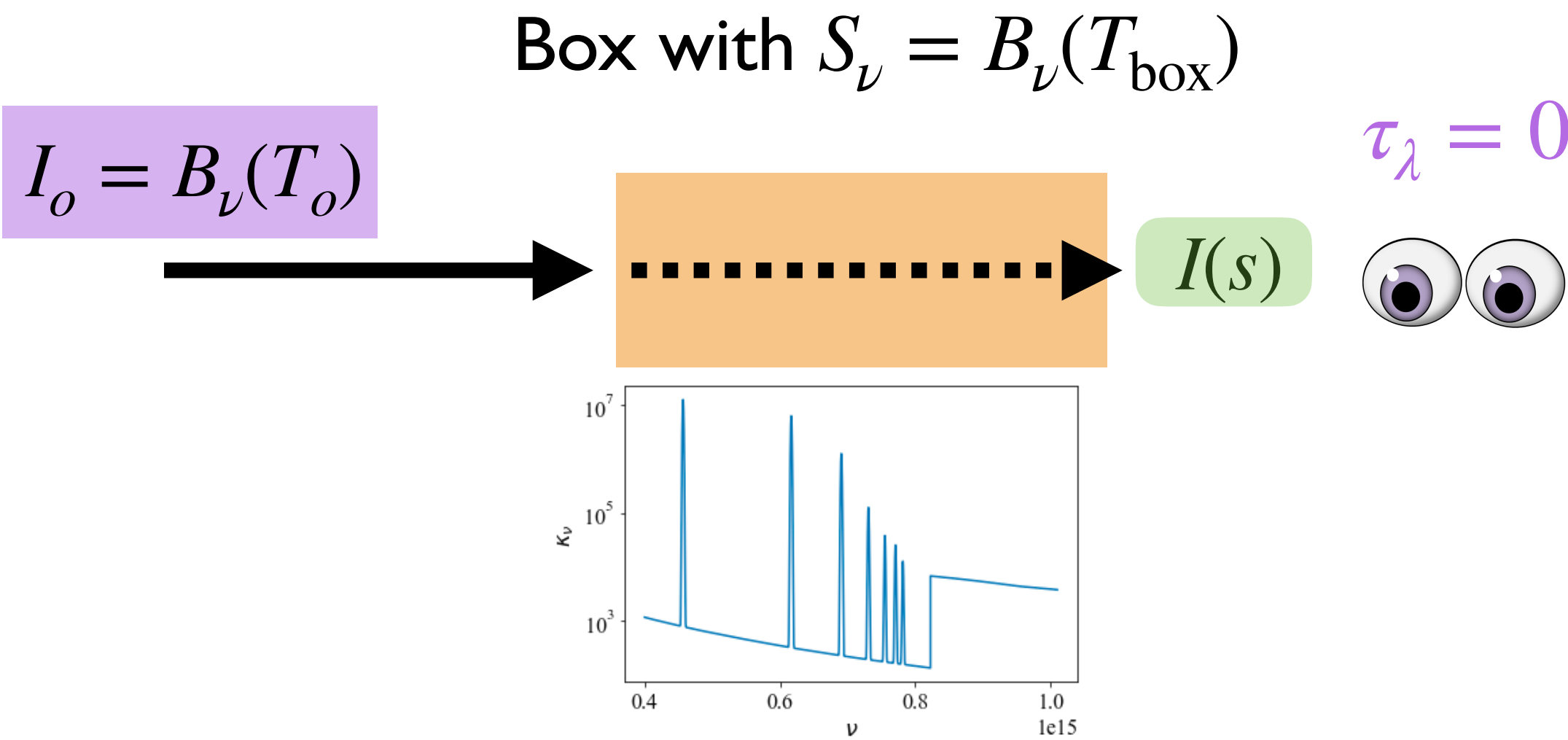


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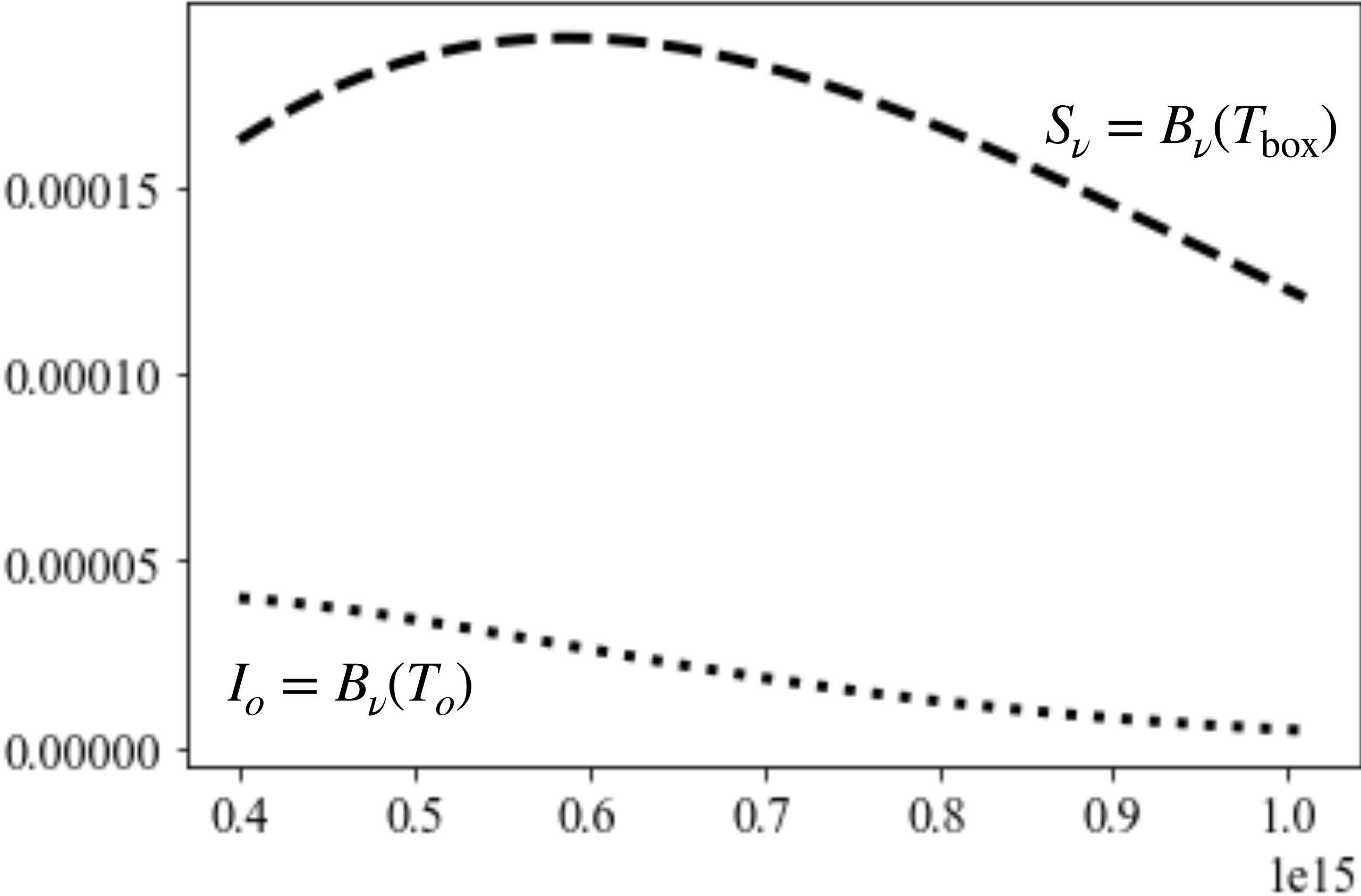


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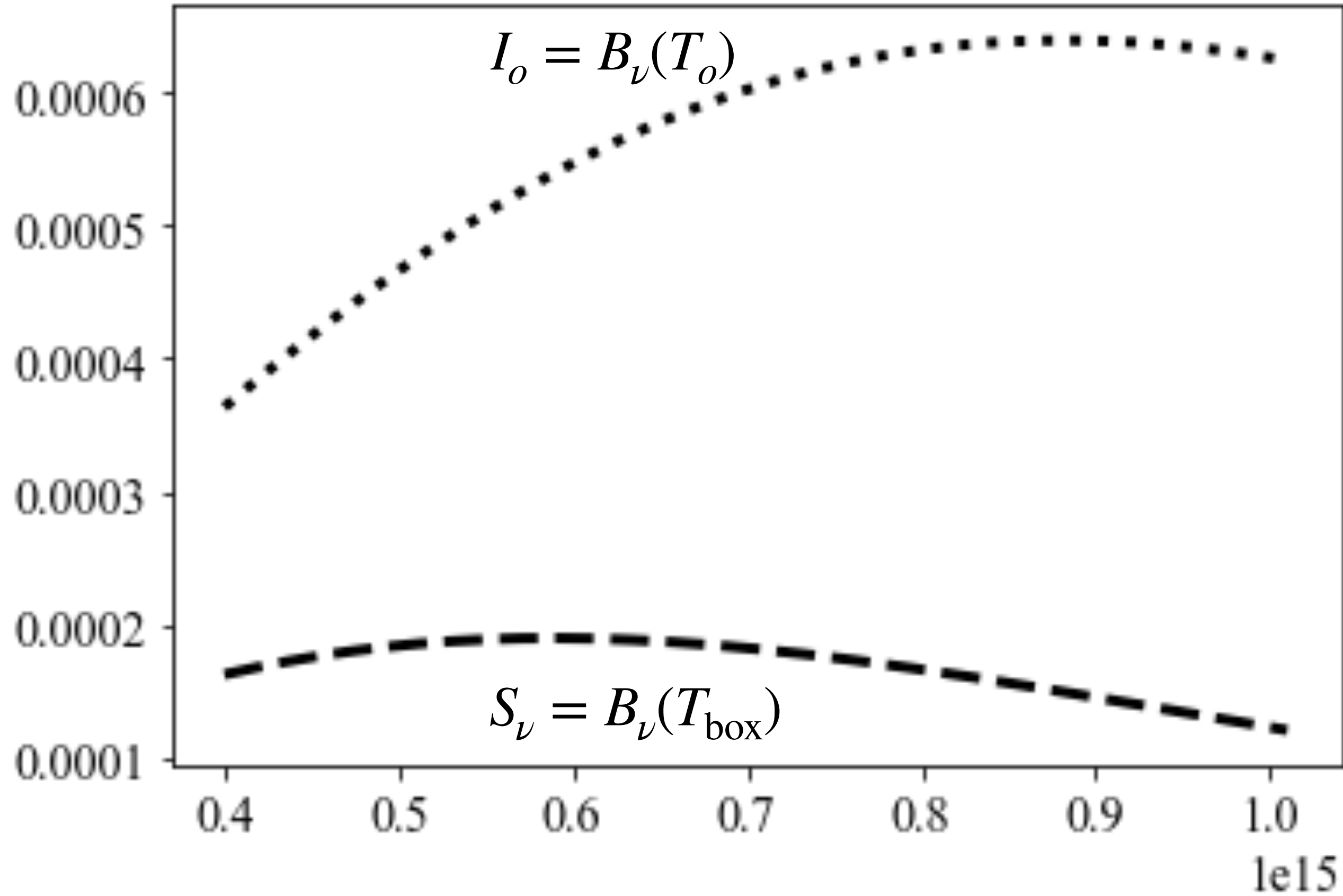
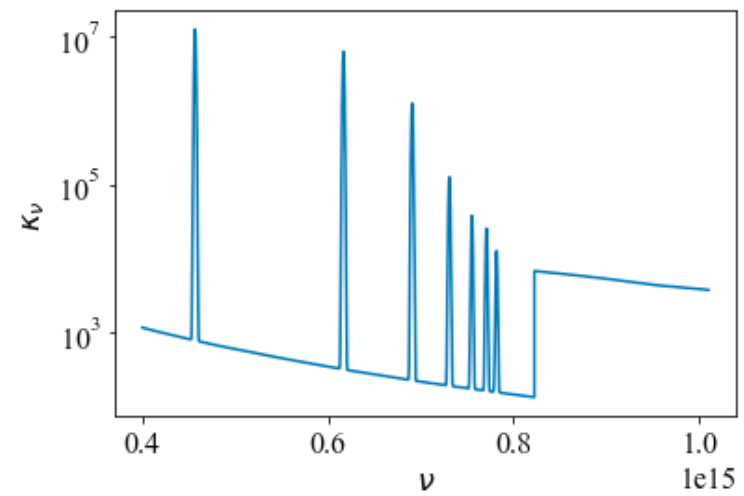
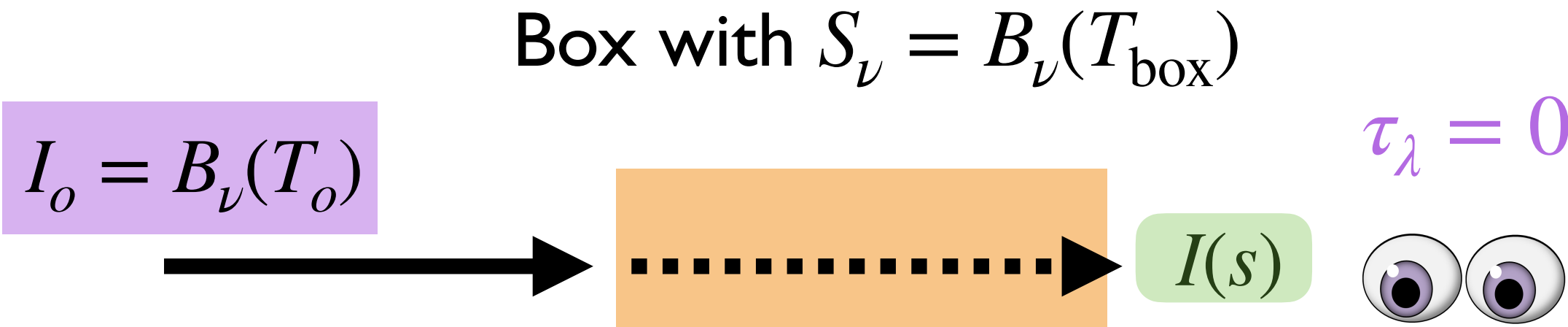


b.The optical depth of the box is large



Example 3: $T_{\text{box}} < T_o$

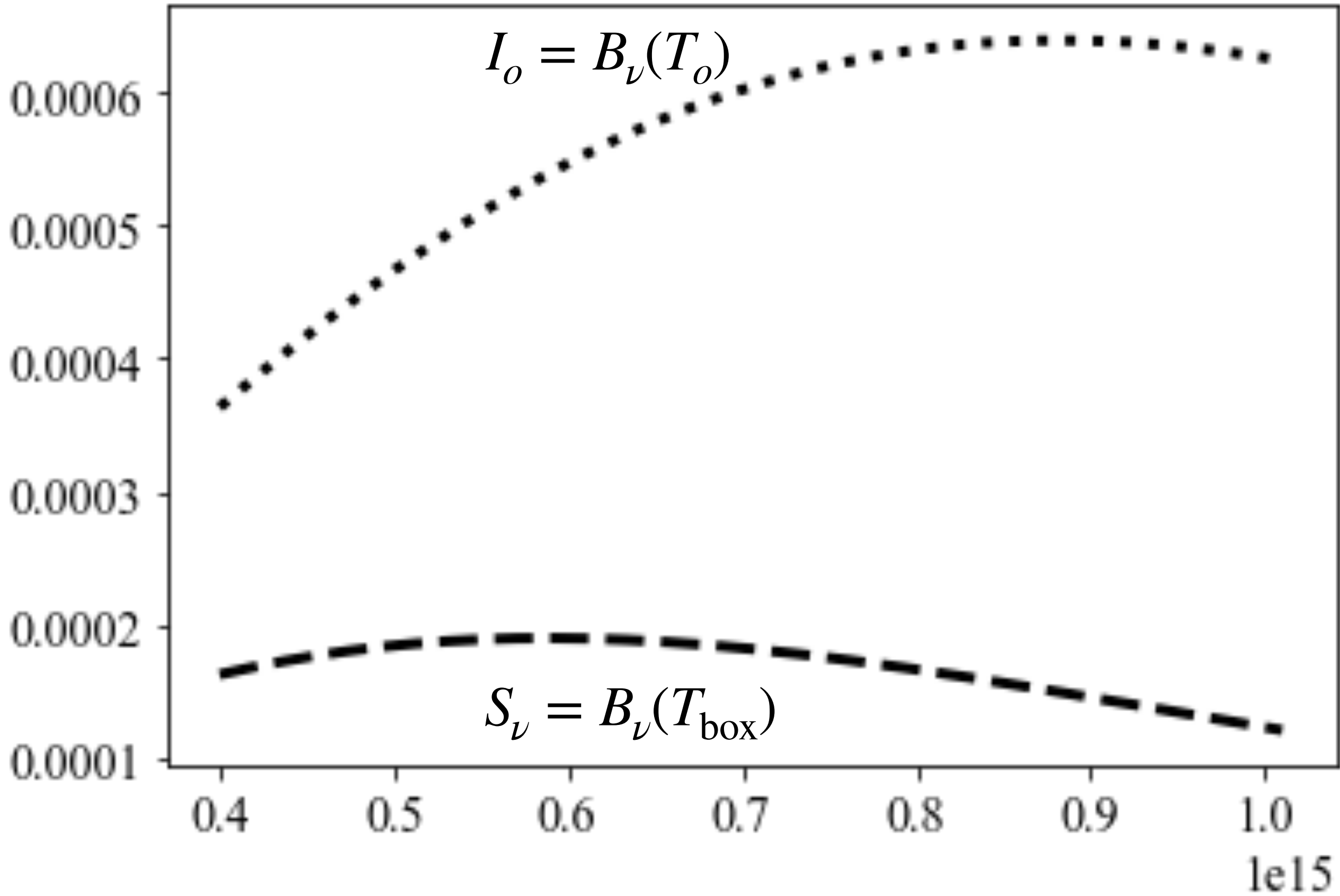
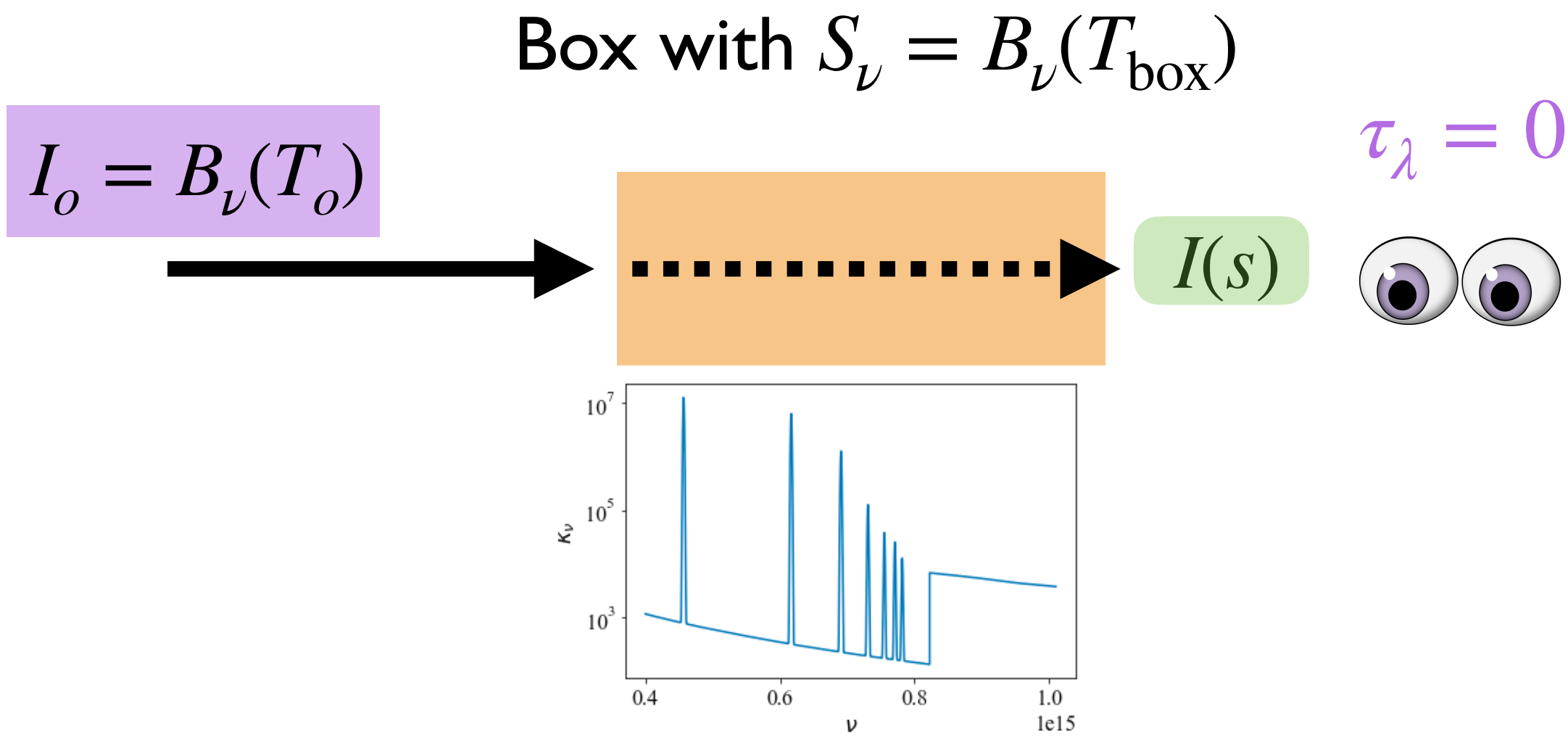
$$I(s) = I_o e^{\tau(s)-\tau_o} + S_o [1 - e^{\tau(s)-\tau_o}]$$
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Example 3: $T_{\text{box}} < T_o$

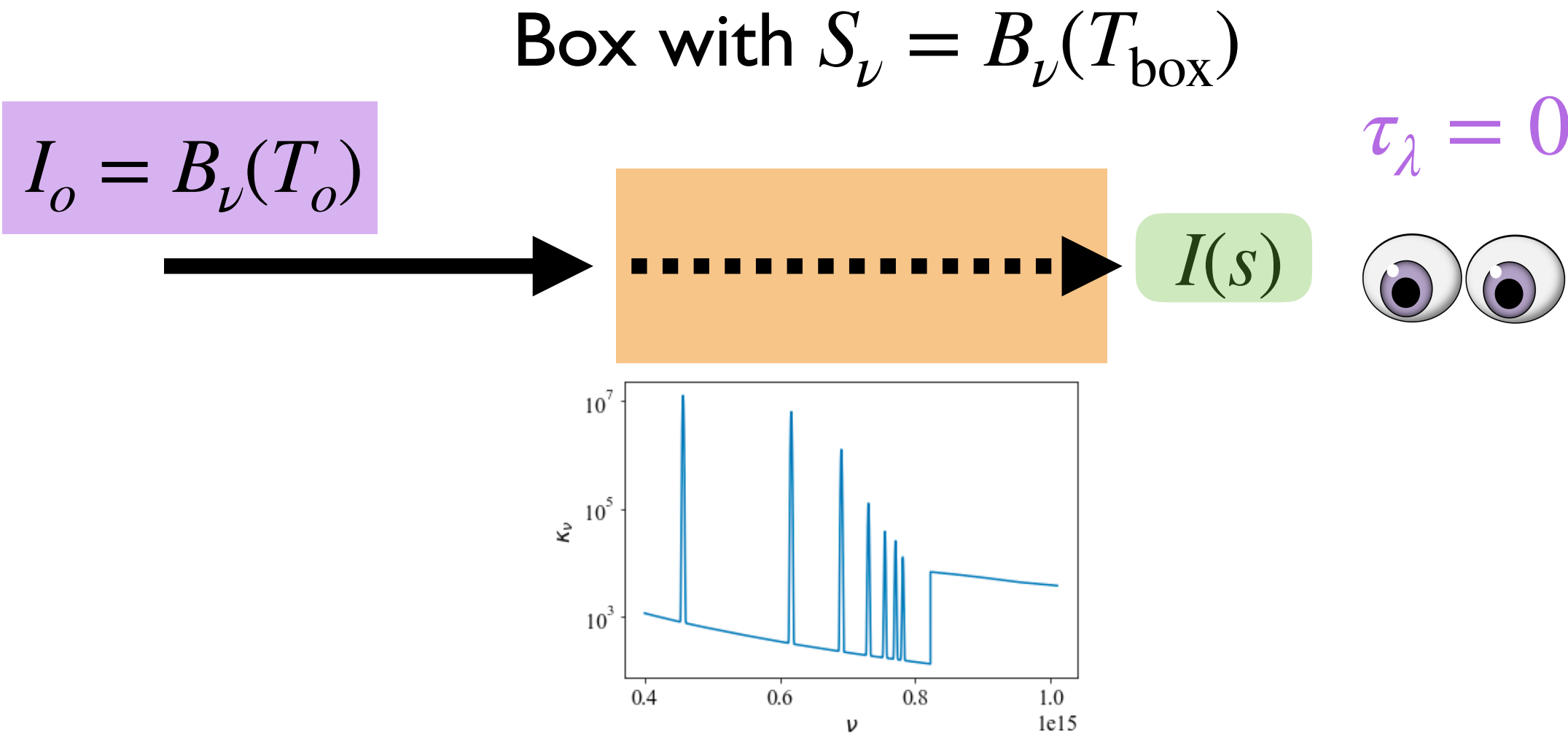
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b.The optical depth of the box is large

