# THERMODYNAMICS AND STATISTICAL PHYSICS

### Formulas and constants

Thermodynamic functions and relations

$$H=E+pV$$
  $F=E-TS$   $G=E-TS+pV$ 

$$\left(\frac{\partial E}{\partial S}\right)_{V} = T \qquad \left(\frac{\partial E}{\partial V}\right)_{S} = -p \qquad \left(\frac{\partial H}{\partial S}\right)_{p} = T \qquad \left(\frac{\partial H}{\partial p}\right)_{S} = V$$

$$\left(\frac{\partial F}{\partial T}\right)_{V} = -S \qquad \left(\frac{\partial F}{\partial V}\right)_{T} = -p \qquad \left(\frac{\partial G}{\partial T}\right)_{p} = -S \qquad \left(\frac{\partial G}{\partial p}\right)_{T} = V$$

Maxwell relations

$$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V \qquad \left(\frac{\partial T}{\partial p}\right)_S = \left(\frac{\partial V}{\partial S}\right)_p \qquad \left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial p}{\partial T}\right)_V \qquad \left(\frac{\partial S}{\partial p}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_p$$

Specific heat

$$c_V = rac{1}{
u} \left(rac{dQ}{dT}
ight)_V \qquad \qquad c_p = rac{1}{
u} \left(rac{dQ}{dT}
ight)_p$$

Entropy

$$S = k \ln \Omega$$
  $S = -k \sum_{r} P_{r} \ln P_{r}$   $S = k (\ln Z + \beta \overline{E})$ 

Partition functions

$$Z = \sum_{r} e^{-\beta E_{r}}$$
  $Z = \sum_{r} e^{-\beta E_{r} - \alpha N_{r}}$   $\ln Z = \alpha N \pm \sum_{r} \ln \left( 1 \pm e^{-\beta \epsilon_{r} - \alpha} \right)$ 

Clausius-Clapeyron equation

$$rac{dp}{dT} = rac{\Delta S}{\Delta V} \qquad \qquad rac{dp}{dT} = rac{L_{12}}{T \Delta V}$$

Fermi energy ( $\mu = -kT\alpha$ )

$$\mu_j = -T \left( \frac{\partial S}{\partial N_j} \right)_{E,V,N} \qquad \mu_j = \left( \frac{\partial E}{\partial N_j} \right)_{S,V,N} \qquad \mu_j = \left( \frac{\partial F}{\partial N_j} \right)_{T,V,N} \qquad \mu_j = -\left( \frac{\partial G}{\partial N_j} \right)_{T,p,N}$$

Stefan-Boltzmann law

$$\mathcal{P}=a\sigma T^4=arac{\pi^2k^4}{60c^2\hbar^3}T^4$$

Stirlings formula

$$\ln N! = N \ln N - N + rac{1}{2} \ln(2\pi N) + ...$$

#### Integrals

$$\int_0^\infty x^n e^{-ax} dx = rac{n!}{a^{n+1}}$$
  $\int_0^\infty x^{2n+1} e^{-ax^2} dx = rac{n!}{2a^{n+1}}$   $\int_0^\infty x^{2n} e^{-ax^2} dx = rac{(2n-1)!!}{2(2a)^n} \sqrt{rac{\pi}{a}}$ 

### The gamma function

$$\Gamma(t) = \int_0^\infty x^{t-1} e^x dx$$
  $\Gamma(t+1) = t\Gamma(t)$ 

### Value of some integrals

# Physical constants

$$c = 2.997925 \cdot 10^8 \text{ m/s}$$

$$e = 1.6022 \cdot 10^{-19} \text{ C}$$

$$h = 6.6262 \cdot 10^{-34} \text{ Js} = 4.1357 \cdot 10^{-15} \text{ eVs}$$

$$\hbar = 1.0546 \cdot 10^{-34} \text{ Js} = 0.65821 \cdot ^{-15} \text{ eVs}$$

$$m_e = 0.91094 \cdot 10^{-30} \text{ kg} = 0.51100 \text{ MeV/c}^2$$

$$m_p = 1.6726 \cdot 10^{-27} \text{ kg} = 938.27 \text{ MeV/c}^2$$

$$m_n = 1.6605 \cdot 10^{-27} \text{ kg} = 931.48 \text{ MeV/c}^2$$

$$N_A = 6.0221 \cdot 10^{23} \text{ mole}^{-1}$$

$$R = 8.314 \text{ JK}^{-1} \text{ mole}^{-1}$$

$$k_B = 1.38066 \cdot 10^{-23} \text{ J/K} = 8.61739 \cdot 10^{-5} \text{ eV/K}$$

$$\sigma = 5.6697 \cdot 10^{-8} \text{ Jm}^{-2} \text{s}^{-1} \text{K}^{-4}$$