

AP PHYSICS 2: FLUID MECHANICS & THERMODYNAMICS (Classes 19 & 20)

Directions: Each of the questions or incomplete statements below is followed by five suggested answers or completions. Select the one that is best in each case and place the letter of your choice in the corresponding box on the student answer sheet.

Note: To simplify calculations, you may use $g = 10 \text{ m/s}^2$ in all problems.

- Two blocks of different sizes and masses float in a tray of water. Each block is half submerged, as shown in the figure. Water has a density of 1000 kg/m^3 . What can be concluded about the densities of the two blocks?
 - The two blocks have different densities, both of which are less than 1000 kg/m^3 .
 - The two blocks have the same density of 500 kg/m^3 .
 - The two blocks have the same density, but the density cannot be determined with the information given.
 - The larger block has a greater density than the smaller block, but the densities of the blocks cannot be determined with the information given.
 - The figure shows four cylinders of various diameters filled to different heights with water. A hole in the side of each cylinder is plugged by a cork. All cylinders are open at the top. The corks are removed. Which of the following is the correct ranking of the velocity of the water (v) as it exits each cylinder?
 - $v_A > v_D > v_C > v_B$
 - $v_A = v_D > v_C > v_B$
 - $v_B > v_C > v_A = v_D$
 - $v_C > v_A = v_B = v_D$
- Questions 3 and 4**
Four differently shaped sealed containers are completely filled with alcohol, as shown in the figure. Containers A and B are cylindrical. Containers C and D are truncated conical shapes. The top and bottom diameters of the containers are shown.
- Which of the following is the correct ranking of the pressure (P) at the bottom of the containers?
 - $P_A = P_B = P_C = P_D$
 - $P_A = P_D > P_C = P_B$
 - $P_A > P_D > P_C > P_B$
 - $P_D > P_A > P_C > P_B$
 - The force on the bottom of container A due to the fluid inside the container is F . What is the force on the bottom of container B due to the fluid inside?
 - F
 - $F/4$
 - $F/8$
 - $F/16$
 - Two cylinders filled with a fluid are connected by a pipe so that fluid can pass between the cylinders, as shown in the figure. The cylinder on the right has 4 times the diameter of the cylinder on the left. Both cylinders are fitted with a movable piston and a platform on top. A person stands on the left platform. Which of the following lists the correct number of people that need to stand on the right platform so neither platform moves. Assume that the platform and piston have negligible mass and that all the people have the same mass.
 - 16 people
 - 4 people
 - 1 person
 - It is impossible to balance the system because you need $1/16$ of a person on the right side.
 - A mass (m) is suspended in a fluid of density (ρ) by a thin string, as shown in the figure. The tension in the string is T . Which of the following is an appropriate equation for the buoyancy force? Select two answers.
 - $F_b = mg$
 - $F_b = mg - T$
 - $F_c = a_2 \rho g h_1$
 - $F_d = a \rho g (h - h_2)$
 - Three wooden blocks of different masses and sizes float in a container of water, as shown in the figure. Each of the masses has a weight on top. Which of the

- following correctly ranks the buoyancy force on the wooden blocks?
- (A) $A > B = C$
(B) $A = B > C$
(C) $B > A = C$
(D) $B > A > C$
8. Two blocks of the same dimensions are floating in a container of water, as shown in the figure. Which of the following is a correct statement about the two blocks?
- (A) The net force on both blocks is the same.
(B) The buoyancy force exerted on both blocks is the same.
(C) The density of both blocks is the same.
(D) The pressure exerted on the bottom of each block is the same.
9. The figure shows four cubes of the same volume at rest in a container of water. Cube C is partially submerged. Cubes A, B, and D are fully submerged, with B resting on the bottom of the container. Which of the following correctly ranks the densities (ρ) of the cubes? Assume the water to be incompressible.
- (A) $\rho_C > \rho_D > \rho_A > \rho_B$
(B) $\rho_B > \rho_A > \rho_D > \rho_C$
(C) $\rho_B > \rho_A = \rho_D > \rho_C$
(D) $\rho_B > \rho_A = \rho_D = \rho_C$
10. A beaker of water sits on a balance. A metal block with a mass of 70 g is held suspended in the water by a spring scale in position 1, as shown in the figure. In this position, the reading on the balance is 1260 g, and the spring scale reads 120 g. When the block is lifted from the water to position 2, what are the readings on the balance and spring scale?
11. Blood cells pass through an artery that has a buildup of plaque along both walls, as shown in the figure. Which of the following correctly describes the behavior of the blood cells as they move from the right side of the figure through the area of plaque? Assume the blood cells can change volume.
- (A) The blood cells increase in speed and expand in volume.
(B) The blood cells increase in speed and decrease in volume.
(C) The blood cells decrease in speed and expand in volume.
(D) The blood cells decrease in speed and decrease in volume.
12. Firefighters use a hose with a 2 cm exit nozzle connected to a hydrant with an 8 cm diameter opening to attack a fire on the second floor of a building 6 m above the hydrant, as shown in the figure. What pressure must be supplied at the hydrant to produce an exit velocity of 15 m/s? (Assume the density of water is 1000 kg/m^3 , and the exit pressure is $1 \times 10^5 \text{ Pa}$.)
- (A) $1.7 \times 10^5 \text{ Pa}$
(B) $2.0 \times 10^5 \text{ Pa}$
(C) $2.6 \times 10^5 \text{ Pa}$
(D) $3.2 \times 10^5 \text{ Pa}$
13. A 1-cm-diameter pipe leads to a showerhead with twenty 1-mm diameter exit holes. The velocity of the water in the pipe is v . What is the velocity of the water exiting the holes?
- (A) $0.05v$
(B) $0.5v$
(C) $5v$
(D) $100v$
14. Air is made up primarily of nitrogen and oxygen. In an enclosed room with a constant temperature, which of the following statements is correct concerning the nitrogen and oxygen gases?
- (A) The nitrogen gas molecules have a higher average kinetic energy than the oxygen gas molecules.
(B) The nitrogen gas molecules have the same average kinetic energy as the oxygen gas molecules.
(C) The nitrogen gas molecules have a lower average kinetic energy than the oxygen gas molecules.
(D) More information is necessary to compare the average kinetic energies of the two gases.
15. Air is made up primarily of nitrogen and oxygen. In an enclosed room with a constant temperature, which of the following statements is correct concerning the nitrogen and oxygen gases?
- (A) The nitrogen gas molecules have a higher velocity than the oxygen gas molecules.

- (B) The nitrogen gas molecules have the same velocity as the oxygen gas molecules.
- (C) The nitrogen gas molecules have a lower velocity than the oxygen gas molecules.
- (D) It is impossible to compare the velocity of the two gases without knowing the temperature of the air and the percentage of nitrogen and oxygen in the room.
16. In an experiment, a gas is confined in a cylinder with a movable piston. Force is applied to the piston to increase the pressure and change the volume of the gas. Each time the gas is compressed, it is allowed to return to a room temperature of 20°C . The data gathered from the experiment is shown in the table. What should be plotted on the vertical and horizontal axes so the slope of the graph can be used to determine the number of moles of gas in the cylinder?
- (A) P and V_2
- (B) P and V
- (C) P and $(V)^{1/2}$
- (D) P and $1/V$
17. In an experiment, a sealed container with a volume of 100 ml is filled with hydrogen gas. The container is heated to a variety of temperatures, and the pressure is measured. The data from the experiment is plotted in the figure. Which of the following methods can be used to determine additional information regarding the gas? Select two answers.
- (A) The slope can be used to calculate the number of atoms in the gas.
- (B) The area under the graph can be used to calculate the work done by the gas.
- (C) The vertical axis can be used to calculate the force the gas exerts on the container.
- (D) The x-intercept can be used to estimate the value of absolute zero.
18. Two identical rooms are connected by an open door. The temperature in one room is greater than the temperature in the other. Which room contains the most gas molecules?
- (A) The warmer room.
- (B) The colder room.
- (C) The number of gas molecules will be the same in both rooms.
- (D) It is impossible to determine without more information.
19. On a hiking trip in the mountains, where the air temperature is cool and has a lower concentration of oxygen, you seal an empty water bottle. You return to your home near sea level where the air temperature is warm and has a higher concentration of oxygen. You notice that the sealed bottle appears partially crushed. Which of the following would contribute to the decrease in volume of the bottle?
- (A) The change in temperature
- (B) The change in atmospheric pressure
- (C) The change in oxygen concentration
- (D) The change in temperature, pressure, and oxygen concentration
20. The figure shows the pressure and volume of a gas at four different states. Which of the following correctly ranks the temperature of the gas at the different states?
- (A) $T_A > T_B > T_C > T_D$
- (B) $T_B = T_C > T_A = T_D$
- (C) $T_C > T_B = T_D > T_A$
- (D) $T_D > T_C > T_B > T_A$
21. Which of the following is correct concerning the two processes shown in the figure?
- (A) $\Delta U_1 = \Delta U_2$ and $W_1 = W_2$
- (B) $\Delta U_1 = \Delta U_2$ and $W_1 > W_2$
- (C) $\Delta U_1 > \Delta U_2$ and $W_1 = W_2$
- (D) $\Delta U_1 > \Delta U_2$ and $W_1 \geq W_2$
22. The figure shows four samples of gas being taken through four different processes. Process 1 is adiabatic. In which process is heat being transferred to the gas sample from the environment?
- (A) 1
- (B) 2
- (C) 3
- (D) 4
23. Two sealed cylinders holding different gases are placed one on top of the other so heat can flow between them. Cylinder A is filled with hydrogen. Cylinder B is filled with helium moving with an average speed that is half that of the hydrogen atoms. Helium atoms have four times the mass of hydrogen

atoms. Which of the following best describes the transfer of heat between the two containers by conduction?

- (A) Net heat flows from cylinder A to cylinder B, because heat flows from higher kinetic energy atoms to lower kinetic energy atoms.
- (B) Net heat flows from cylinder B to cylinder A, because heat flows from higher kinetic energy atoms to lower kinetic energy atoms.
- (C) There is no net heat transfer between the two cylinders, because both gases have the same average atomic kinetic energy.
- (D) There is no net heat transfer between the two cylinders, because heat conduction requires the movement of atoms between the cylinder, and the cylinders are sealed.

Questions 24 and 25

A gas beginning at point O on the graph can be taken along four paths to different ending conditions.

24. Which of the following are the same for processes 2 and 3? Select two answers.
- (A) Q
 - (B) ΔT
 - (C) ΔU
 - (D) W
25. Along which of the paths is the most thermal energy removed from the gas?
- (A) 1
 - (B) 2

- (C) 3
- (D) 4

26. The graph shows the distribution of speeds for one mole of hydrogen at temperature T , pressure P , and volume V . How would the graph change if the sample was changed from one mole hydrogen to one mole of argon at the same temperature, pressure, and volume?
- (A) The peak will shift to the left
 - (B) The peak will shift upward and to the left
 - (C) The peak will shift to the right
 - (D) The peak will shift downward and to the right
27. The graph shows the pressure and volume of a gas being taken from state #1 to state #2. Which of the following correctly indicates the sign of the work done by the gas, and the change in temperature of the gas?

Questions 115 and 116

A resistor of resistance (R) is sealed in a closed container with n moles of gas inside. A battery of emf (\mathcal{E}) is connected to the resistor

1. Which of the following graphs shows the correct relationship between the gas atoms average velocity (v_{avg}) and electrical energy (E) supplied to the resistor?
2. Which of the following graphs shows the correct relationship between the gas atoms average velocity (v_{avg}) and electrical energy (E) supplied to the resistor?

AP[®] Physics 2: Fluid Mechanics & Thermodynamics
Student Answer Sheet for Multiple-Choice Section

No.	Answer
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PHYSICS 2: Fluid Mechanics & Thermodynamics**SECTION II****4 Questions**

Directions: Answer all questions. The suggested time is about 10 minutes for answering each of the questions. The parts within a question may not have equal weight. All final numerical answers should include appropriate units. Credit depends on the quality of your solutions and explanations, so you should show your work. Credit also depends on demonstrating that you know which physical principles would be appropriate to apply in a particular situation. Therefore, you should clearly indicate which part of a question your work is for.