

H1a: MD-simulation of aluminium properties

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Introduction

A physicist studying matter interactions on a macroscopic scale will often encounter problems that either can't be solved analytically, or where it is not practical to do so. In such cases the physicist must resort to numerical methods. One such method, molecular dynamics simulation, or MD-simulations for short is a technique where interacting particle trajectories are generated by numerically integrating Newton's equation of motion [1]. This paper examines static properties of aluminium undergoing phase changes, by using MD-simulation together with the velocity-verlet-algorithm for integrating particle trajectories. Sections () to () outlines different problems and simulations, together with their theoretical background and analysis of the simulation results. The last section, Section , contains a discussion of the respective results, and the advantages and disadvantages of the MD-simulation technique. The purpose of this paper is to theoretically outline, and apply a basic MD-simulation technique which can later be generalized to study more complex molecular and lattice behaviours.

Problem 1

Solid aluminium, Al, organizes as a face-centered-cubic-crystal, or fcc-crystal. The length between two opposing sides of the fcc-crystal is called the lattice-parameter. The total binding energy of the fcc-crystal depends on the aluminium particles inter-atomic potentials, which in turn is a function of the aforementioned lattice parameter. One can vary the lattice parameter and examine how this changes the total binding energy. The value of the lattice parameter that minimizes the total binding energy is then the lattice parameter that is found in naturally occurring aluminium. The first problem in this paper is to identify this equilibrium lattice parameter numerically.

The first step is to initialize an FCC-structure. This was done using a function included in the course material, which can be found in appendix (C.2). The created FCC-structure is a supercell of four unit cells in each direction. The number of atoms for one unit cell is one eighth per corner, and one half per face, for a total of four atoms. Hence the supercell contains $4 \times 4^3 = 256$ atoms. The reason for using a supercell instead of just examining one unit cell, is to capture effects from particle interactions over distances longer than the lattice parameter. But these are usually weak over distances longer than a few unit cells and thus interactions between particles longer is approximately zero. Assuming that a particle in a homogenous material has a spherical range of interactions, the particles on the sides of a unit cell should interact with particles in neighbouring cells. These effects can be taken into account by using the supercell and applying periodic boundary conditions. Figuratively, the supercell can be seen as periodically repeating "images" of the unit cell, see figure (1), and implemented using the minimum image convention [2]. Results can then be calculated from considering a particles interaction with the nearest images of surrounding particles, which is then averaged and reported in units of per unit-cell.

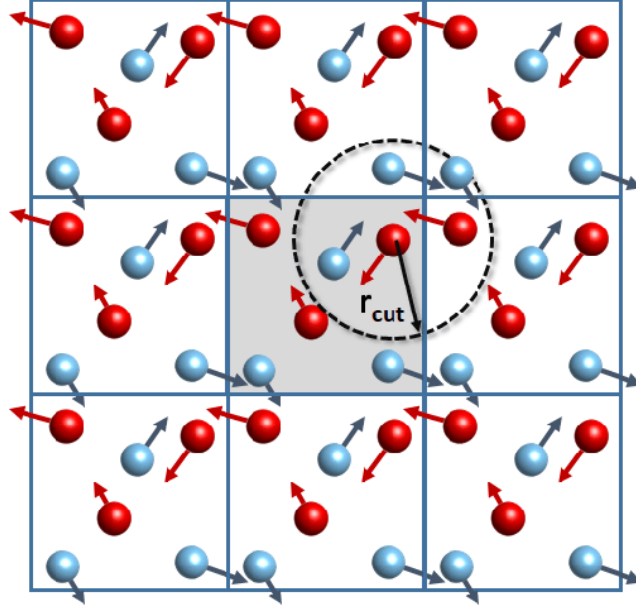


Figure 1: Periodic boundary conditions used to model interactions between particles within each others interaction range but placed in different unit cells. In this problem, a supercell of $4 \times 4 \times 4$ was used. Picture by Bo Yang [3]

The FCC-cell was initialized with 20 lattice parameters uniformly spaced out between 3.98 and 4.075 and their respective potential energies were calculated. The equilibrium lattice parameter was determined to be $(4.030 \pm 0.005) \text{ \AA}$ and the resulting plot is seen in Figure (2). A slight jump in energy for the measured datapoints can be seen around value of the lattice par for the local minimum. One possible cause could be the use of the minimum image convention, but it is uncertain.

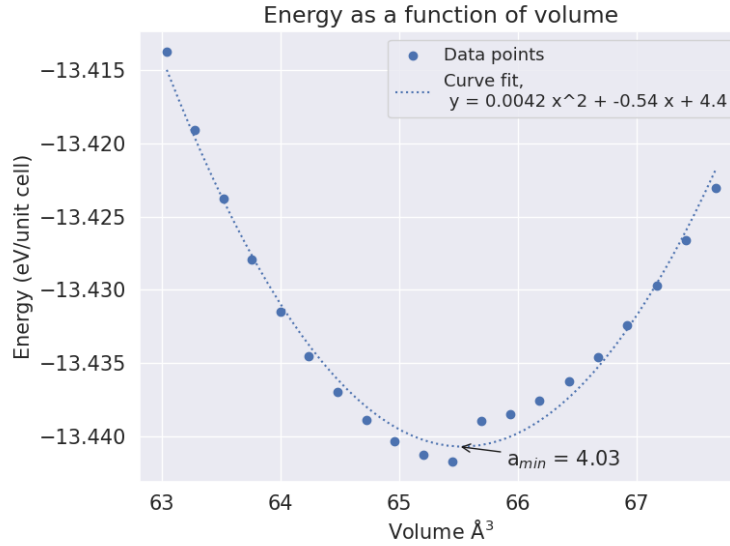


Figure 2: Potential energy of an aluminium fcc as a function of the lattice parameter. The local minimum is the equilibrium lattice parameter, here identified as $(4.030 \pm 0.005) \text{ \AA}$. A slight jump in measured energy of measured datapoints can be seen around this value of the lattice parameter.

Problem 2

A common algorithm to perform the numerical integration of particle trajectories is the Velocity-Verlet-algorithm. The algorithm is derived from Taylor expanding the position coordinate forward and backwards in time, and is summarized in the equation relating the velocity in a following timestep to the acceleration as

$$\mathbf{v}(t + dt) = \mathbf{v}(t) + \frac{1}{2}[\mathbf{a}(t) + \mathbf{a}(t + dt)]dt. \quad (1)$$

[2]. A clever implementation of the algorithm, made to reduce the error, divides this timestep into two parts. In the first half the contribution of the first acceleration term, $\mathbf{a}(t)$, is used to update the velocity at $\mathbf{v}(t + dt/2)$. This velocity is then used to update the particle position, \mathbf{r} , after which a new acceleration, $\mathbf{a}(t + dt)$, is retrieved, and used to update the velocity at $\mathbf{v}(t + dt)$. This can be written in pseudocode as

$$\begin{aligned} \mathbf{v}(t + dt/2) &= \mathbf{v}(t) + \frac{1}{2}\mathbf{a}(t)dt \\ \mathbf{r}(t + dt) &= \mathbf{r}(t) + \mathbf{v}(t + dt/2)dt \\ \mathbf{a}(t) &\rightarrow \mathbf{a}(t + dt) \\ \mathbf{v}(t + dt) &= \mathbf{v}(t + dt/2) + \frac{1}{2}\mathbf{a}(t + dt)dt \end{aligned} \quad (2)$$

The benefits of the Velocity-Verlet-algorithm is that it is time-reversible, has a small error-term, $O((\Delta)^3)$, and is self-starting, meaning that the state in the next time-step only depends on the state in the previous [4]. But it also assumes that the forces acting on the particles only depend on their position and not their velocity. This means that the algorithm has to be adapted to calculate trajectories for charged particles in an electromagnetic field for example.

In this problem the particles of the equilibrated aluminium FCC-cell is displaced and their trajectories, numerically integrated with the Velocity-Verlet-algorithm, is examined to calculate the temperature and pressure per unit-cell. The particles is first displaced with a factor of $\pm 6.5\%$ of the lattice-parameter. The forces on each particle was retrieved with a function provided in the course material, see appendix (C.1), after which followed a sequence of timesteps. One timestep consisted of first calculating the velocities half a timestep forward, then the positions after which forces were updated and used to update the velocities another half timestep forward, see equation (2). The velocities after the timestep was used to calculate the particles kinetic energy as

$$E_{\text{kin}} = \left\langle \sum_{i=1}^N \frac{p_i^2(t)}{2m_i} \right\rangle_t. \quad (3)$$

And from the kinetic energy the temperature, T , is calculated using the equipartition theorem,

$$T = \frac{2}{3Nk} E_{\text{kin}}, \quad (4)$$

where N is the total number of particles and k is Boltzmanns constant. The pressure, P , is in turn given by

$$PV = NkT + W, \quad (5)$$

where W is called the virial, which was retrieved with a provided function and accounts for some intra-molecular forces, see appendix (C.1) [5]. The virial is given by

$$W = \langle \mathcal{W} \rangle_{NVT} = \left\langle \frac{1}{3} \sum_{i=1}^N \mathbf{r}_i \cdot \mathbf{F}_i \right\rangle_{NVT}. \quad (6)$$

Lastly the potential energy was retrieved with a provided function, from which the total energy is simply given by the sum of the kinetic and potential energy. The full source code of the implementation can be seen in Appendix (A.3).

One of the parameters for the algorithm is the size of the timestep, dt . An optimal timestep will be small enough that energy will be conserved during simulation, but also big enough to not unnecessarily prolong simulation time. Figures (3) to (3) shows a timeseries over measured energy for different value of dt . The rightmost graph showing total energy can be seen to show a relative conservation for dt equals to $1e-2$, complete divergence for $2e-2$ and a slight drift for $1.5e-2$ which could indicate an imminent divergence. Hence the value of $1e-2$ was chosen as an upper limit for dt .

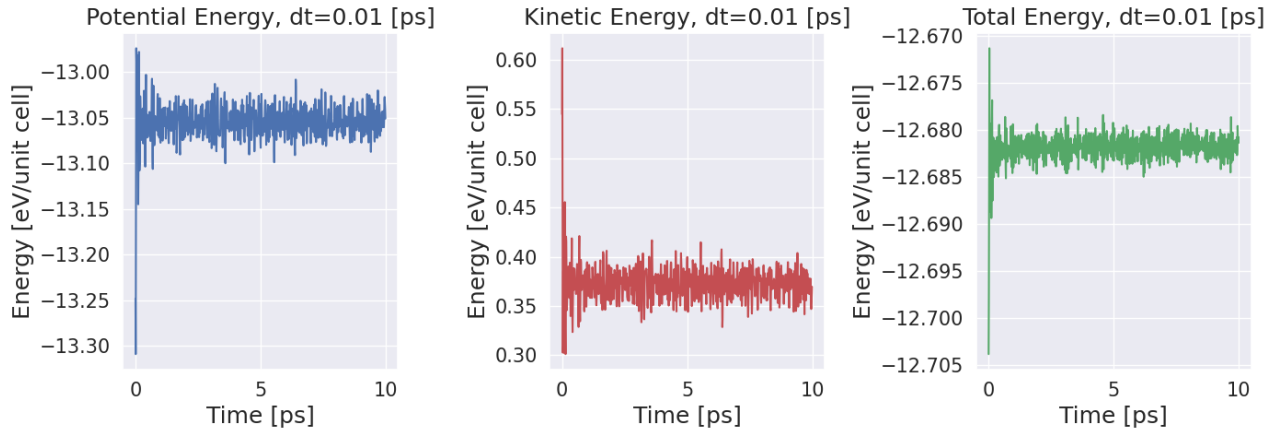


Figure 3: Energy over time for an aluminium fcc lattice. In the graph to the right, the total energy can be seen to be relatively conserved.

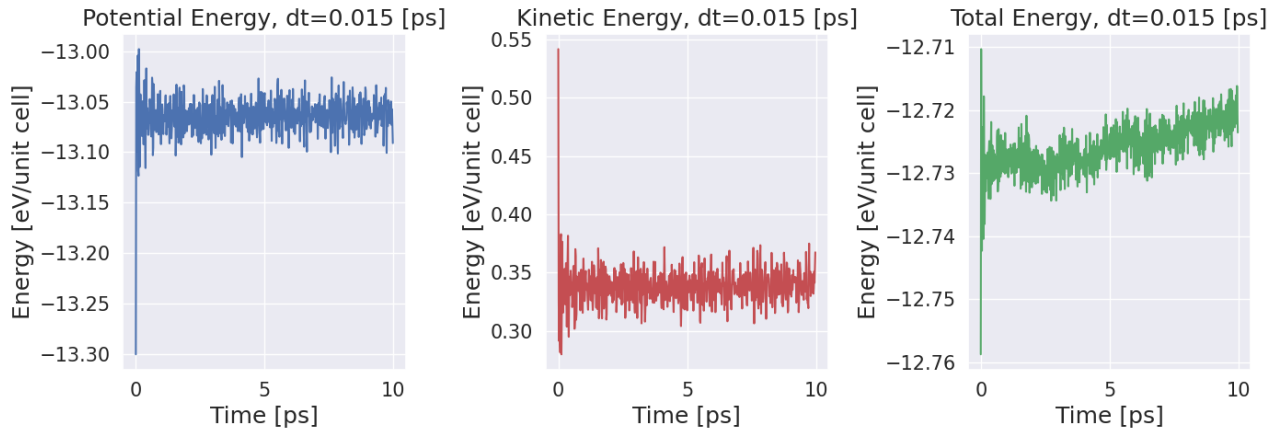


Figure 4: Energy over time for an aluminium FCC lattice. In the graph to the right, the total energy can be seen to be drifting, which could indicate an imminent divergence.

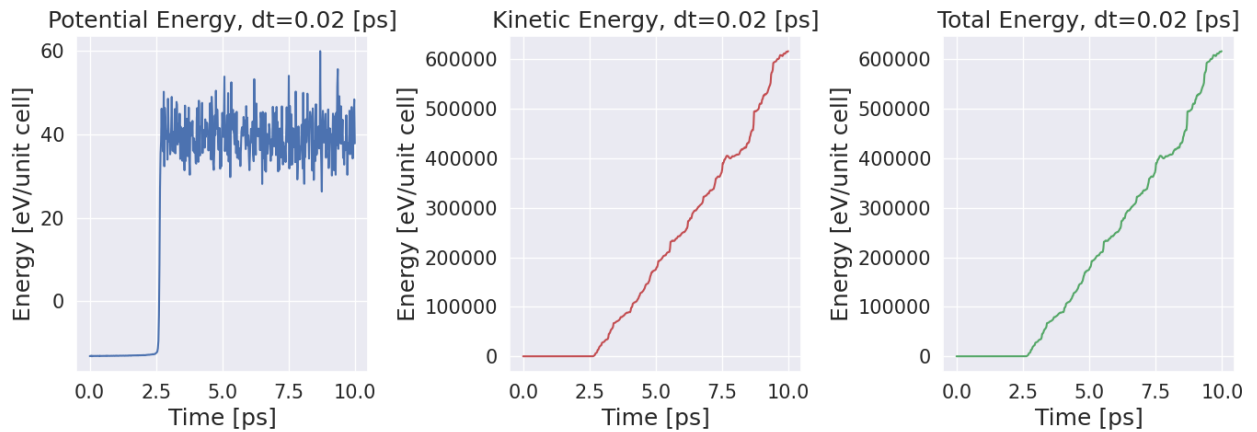


Figure 5: Energy over time for an aluminium FCC lattice. In the graph to the right, conservation of the total energy is violated.

The temperature per unit cell of the simulation with $dt = 1e-2$ can be seen in figure (6), where it shows fluctuating behaviour around 720.93 K. It can be noted that the fluctuation in the temperature and energy graphs is natural as an initial displacement in a frictionless environment should lead to the particle-oscillations around their equilibrium-position. The magnitude of the average temperature is affected by the size of the random displacements, here $\pm 6.5\%$ of the lattice parameter, with a larger displacement leading to a higher average temperature as the oscillations would be larger. The higher temperatures in the first few timesteps is natural as the particles because all particles start away from their equilibrium and need a few timesteps of interaction before they occupy a more stable state. Note also that here quantities are measured at the end of the velocity verlet timestep, and therefore kinetic energy and temperature doesn't start from zero.

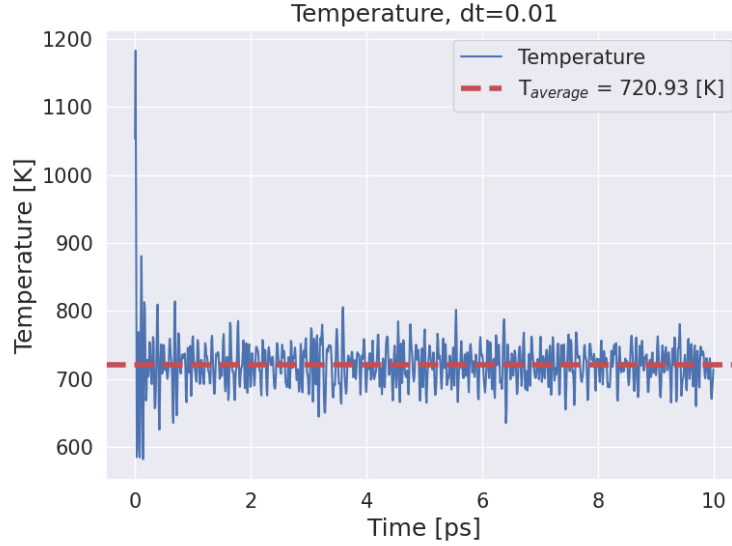


Figure 6: Temperature over time for an aluminium FCC lattice. The temperature can be seen to fluctuate around 720.93 K.

Problem 3

When performing a molecular-dynamics simulation we often are interested in studying a system with certain macroscopic properties such as temperature or pressure. When specifying initial conditions for the particles being simulated and equilibrating the system it will tend to a certain temperature and pressure, but we are not able to obtain a specific macroscopic state by choosing certain initial positions and velocities for the particles. For this problem we were tasked with simulating aluminium in a solid phase at the temperature 500°C (773.15 K) and pressure 1 Bar. A common technique to circumvent this problem is to scale coordinates for position and velocities during the equilibration after which a production run is performed from which average quantities can be calculated. For this project the following algorithm was used in combination with velocity verlet for scaling the velocity coordinates

$$\mathbf{v}_i^{\text{new}} = \sqrt{\alpha_T} \mathbf{v}_i^{\text{old}} \quad (7)$$

with

$$\alpha_T = 1 + \frac{2dt}{\tau_T} \frac{T_{\text{eq}} - \mathcal{T}(t)}{\mathcal{T}(t)} \quad (8)$$

where $\mathcal{T}(t)$ is the instantaneous temperature, τ_T is a time constant regulating the speed with which the temperature is re-scaled, dt is the size of the time step and T_{eq} is the equilibrium temperature. The position coordinates and the lattice parameter a_0 were re-scaled by

$$\mathbf{r}_i^{\text{new}} = \alpha_P^{1/3} \mathbf{r}_i^{\text{old}}, \quad a_0^{\text{new}} = \alpha_P^{1/3} a_0^{\text{old}} \quad (9)$$

with

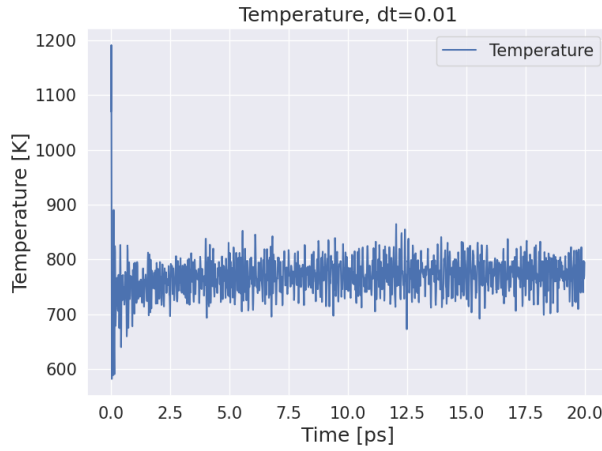
$$\alpha_P = 1 - \kappa_T \frac{dt}{\tau_P} \frac{P_{eq} - \mathcal{P}(t)}{\mathcal{P}(t)} \quad (10)$$

where $\mathcal{P}(t)$ is the instantaneous pressure, P_{eq} is the equilibrium pressure, τ_P is a time constant regulating the speed of the pressure scaling. The κ_T is the isothermal compressibility, being 0.01385 Bar^{-1} for aluminium [6]. The size of parameters τ_T and τ_P were chosen to be $100 \cdot dt$ and $300 \cdot dt$ by trial and error according to what yielded good results in terms of liquidization and energy conservation of the system. A summary of the parameters used for simulation of solid aluminium can be seen in table (1).

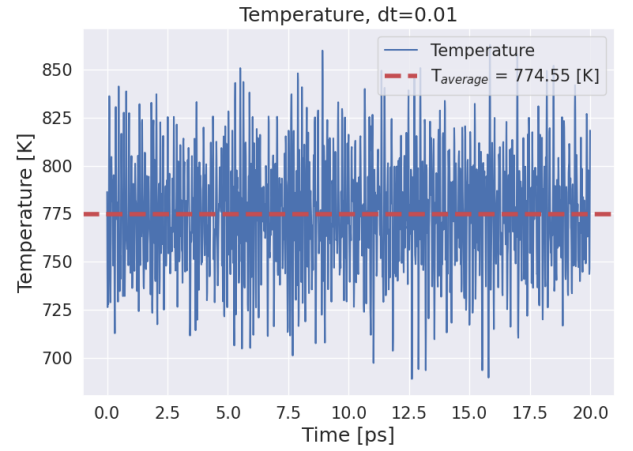
The resulting graphs of temperature and pressure during equilibration and production can be seen in figures (11) and (12) respectively. The time average of the temperature for the production run was calculated to be $\langle T \rangle = 774.55 \text{ K}$, with fluctuations of up to 100 K.

dt [ps]	κ_T [1/Bar]	T_{eq} [K]	P_{eq} [Bar]	τ_T [ps]	τ_P [ps]
0.01	$0.01385 \cdot 10^4$	773.15	1	$100 \cdot dt$	$300 \cdot dt$

Table 1: Table presenting parameter values used for simulation of solid aluminium. Value for κ_T was retrieved from [6] and then converted to inverse bar.



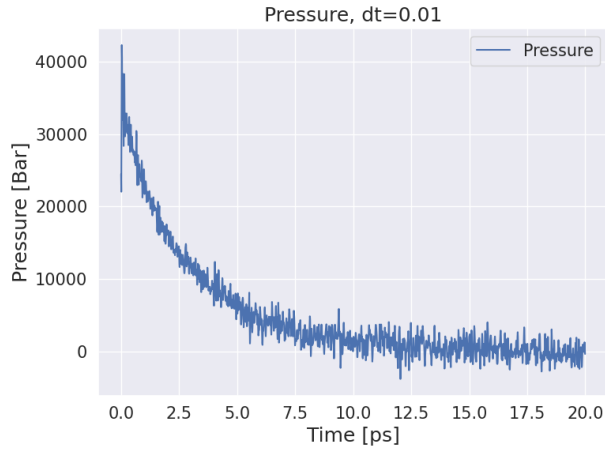
(a) Temperature during equilibration.



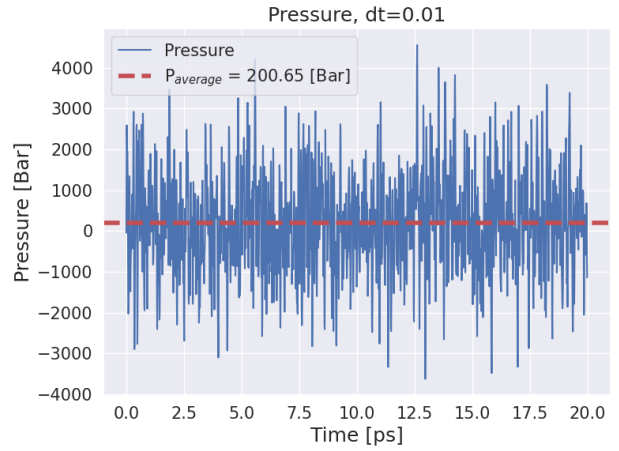
(b) Temperature during following production run.

Figure 7: Equilibration and production temperature during simulation of solid aluminium.

Pressure as a function of time can be seen in figure (8a) and (8b). The time average of the pressure for the production run was calculated to be $\langle P \rangle = 200.65 \text{ Bar}$, a bit larger than the desired pressure of 1 Bar. One reason could be that the temperature is a macroscopic quantity, that is not very well defined for smaller systems, such as this one with only 256 particles. One should also keep in mind that this computer simulated system doesn't take all possible particle interactions into account, and won't give an entirely accurate representation of a real physical system.



(a) Pressure during equilibration.



(b) Pressure during production run.

Figure 8: Equilibration and production pressure during simulation of solid aluminium. Pressure during production can be seen fluctuating heavily.

One consequence of the large fluctuations of pressure is that it is difficult to see whether the system is fully equilibrated. Therefore one can look at the lattice parameter that is scaled with α_P , see equation 9, and define an adequate pressure equilibration as one in which the lattice parameter has stabilized. The measurement of the lattice parameter for the solid aluminium simulation is shown in figure (9). There the lattice parameter can be seen stabilize and become approximately constant around 17.5 ps, which can be compared to pressure during equilibration also stabilizing, see figure (8a). The final equilibrated lattice parameter used for the production run was 4.092 Å.

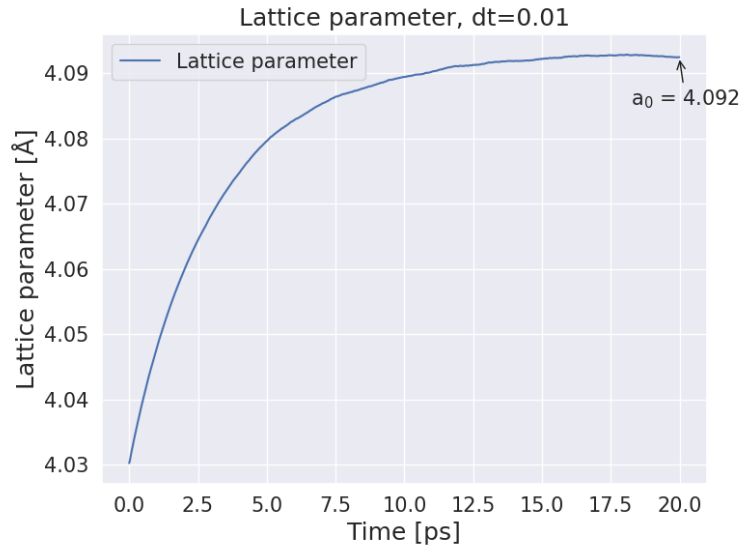


Figure 9: Lattice parameter as a function of time during equilibration. The last measured value of 4.092 Å was used for the production run.

Lastly, to show that the system was in a solid state the trajectories of three random particles were plotted in figure (10). These plots clearly shows the particles remaining close to their initial positions.

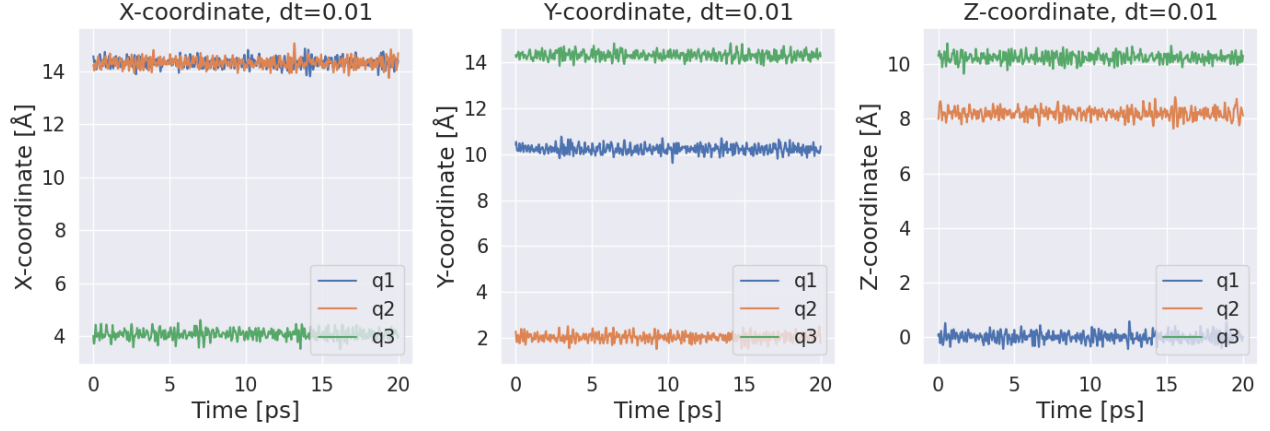


Figure 10: Positions of three random particles during production run of solid aluminium.

Problem 4

Simulation of aluminium in a liquid phase, at a temperature of 973.15 K and pressure of 1 Bar, was implemented using a similar equilibration procedure to the one discussed in the previous section, but with an extra equilibration step. The first equilibration step was performed at a higher temperature of 5000 K for 20 ps after which an equilibration run was performed at 973.15 K for another 20 ps. The reason for heating up the system to a temperature above the desired, in contrast to how a real-world procedure would be conducted, is that in the real world materials often need a nucleation point to start melting. And this is not present in the idealised version realized in a simulation. Also, the timestep was changed from 10^{-2} ps to 10^{-3} ps to ensure that the system was well behaved as the increased dynamics of the system needed the smaller timestep for energy to be conserved during the production run. After these initial steps of heating up and cooling down the system, a production run was performed for 20 ps where time averages of the pressure and temperature could be calculated. The time average of the temperature was calculated to $\langle T \rangle = 966.17$ K and the time dependence of the temperature is displayed in figures (11a) and (11b) for the equilibration and production steps respectively.

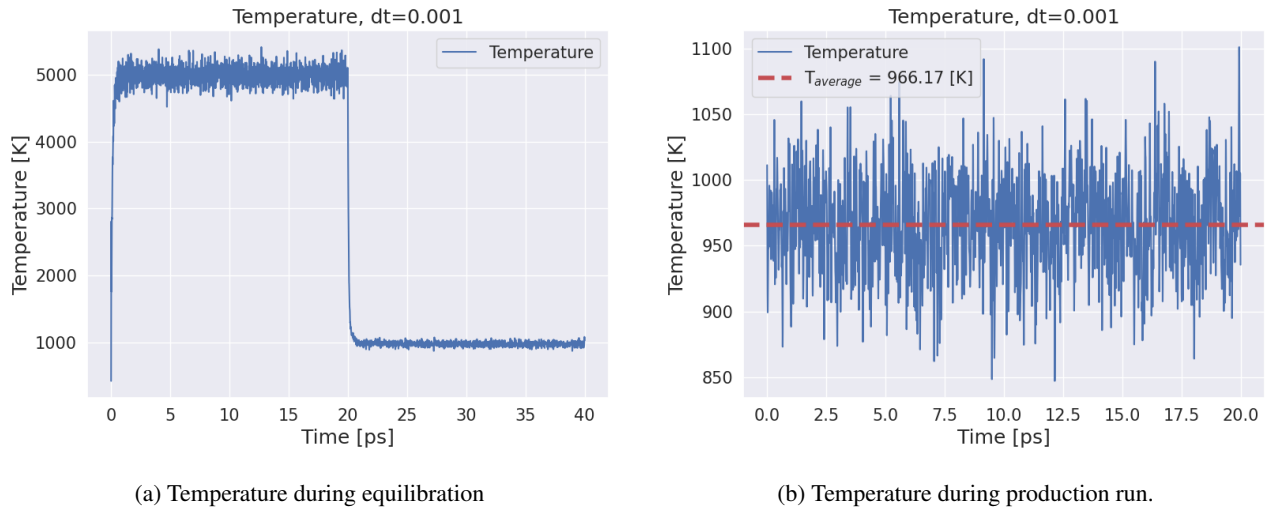
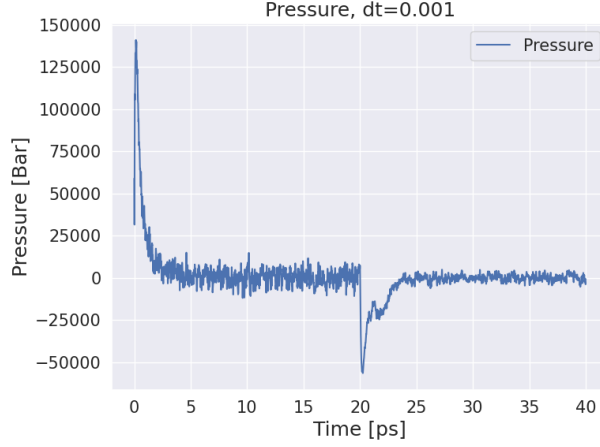


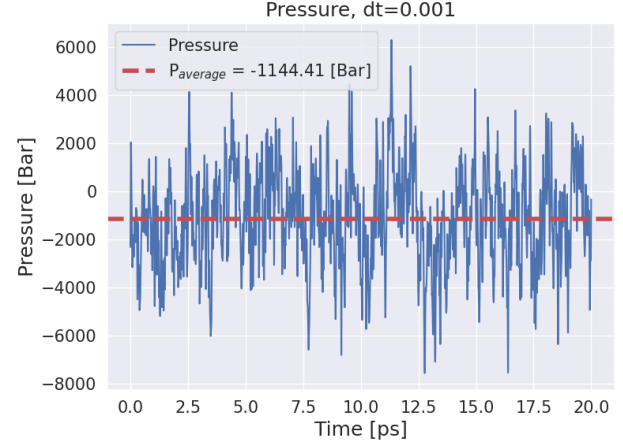
Figure 11: Equilibration and production temperature during simulation of liquid aluminium. During equilibration temperature is first increased heavily, before a cool-down period is initiated, where the system is equilibrated against the desired temperature of 973.15 K. In the following production run, the average temperature was measured to be 966.17 K.

The time average of the pressure for the liquid aluminium simulation was calculated during the production run to be

$\langle P \rangle = -1144.41$ Bar and the pressure as a function of time can be seen for the equilibration and the production run in figures (12a) and (12b). The corresponding measurement for the lattice parameter can be seen in figure 13, where the lattice parameter can be seen to be stabilized at the end the heating step at 20 ps, and again at the end of the cooling step at 40 ps. The final lattice parameter used for the production run was 4.254 \AA , slightly bigger than the lattice parameter of solid aluminium simulation of 4.092 \AA . One should note that a liquid doesn't have a crystal lattice structure, and thus the concept of a lattice parameter is not well defined. But it could be argued that the lattice parameter in this simulated system indicates that the atoms in the equilibrated state are spaced further apart, than in the initial state. Which is true for most newtonian solids and fluids when temperature is increased.



(a) Pressure during equilibration



(b) Pressure during production run.

Figure 12: Equilibration and production pressure during simulation of liquid aluminium. During equilibration temperature is first increased heavily, before a cool-down period is initiated, where the system is equilibrated against the desired pressure of 1 Bar. In the following production run, the average pressure was measured to be -1144 Bar and can be seen to be heavily fluctuating.

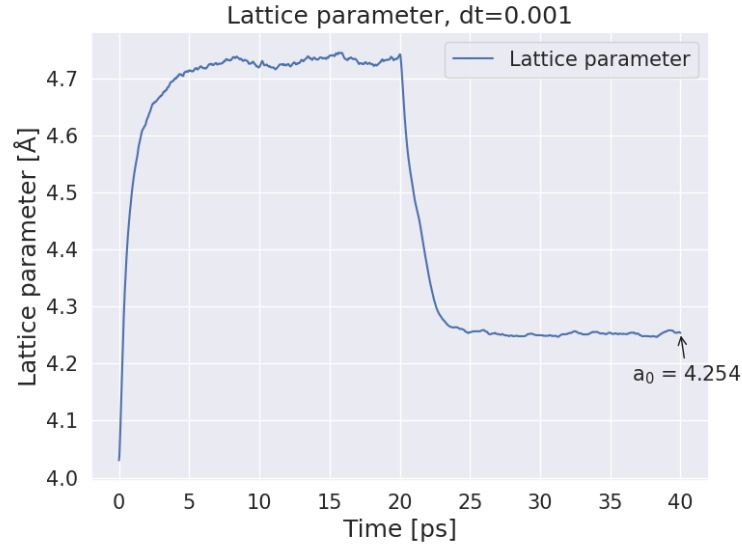


Figure 13: Lattice parameter as a function of time during equilibration of a liquid aluminium simulation. The last measured value of 4.254 \AA was used for production run.

To determine that the system had indeed entered a liquid phase the trajectories of a few random particles during the

production run were plotted in figure 14. There it can be seen that the particles travel a distance of a few to a few dozen Ångström from their initial positions in a few picoseconds. In contrast to the more constant positions seen in the solid aluminium simulation in figure 10, the particles could be said to be in a liquid state.

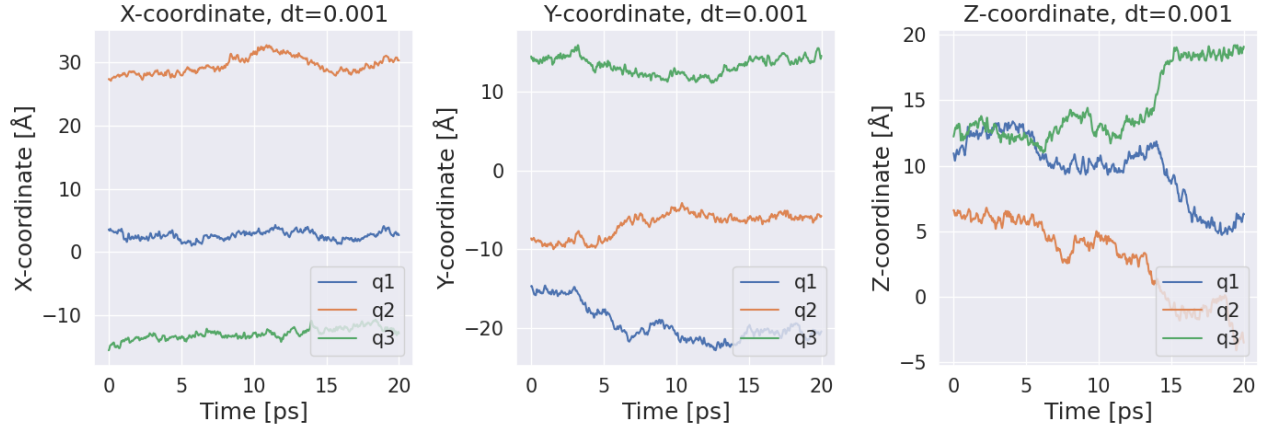


Figure 14: Positions of three random particles during production run of solid aluminium.

Problem 5

For a system with constant number of particles N , constant volume V and constant energy E the heat capacity at constant volume C_V is related to the fluctuations of the instantaneous values of the potential and kinetic values. The heat capacity can be calculated as

$$C_{V,pot} = \frac{3Nk_B}{2} \left[1 - \frac{2}{3Nk_B^2 T^2} \langle (\delta \mathcal{E}_{pot})^2 \rangle_{NVE} \right]^{-1}$$

$$C_{V,kin} = \frac{3Nk_B}{2} \left[1 - \frac{2}{3Nk_B^2 T^2} \langle (\delta \mathcal{E}_{kin})^2 \rangle_{NVE} \right]^{-1} \quad (11)$$

where $\langle (\delta \mathcal{E}_{pot})^2 \rangle_{NVE}$ and $\langle (\delta \mathcal{E}_{kin})^2 \rangle_{NVE}$ are the variance of the instantaneous values of the potential and kinetic energy along the phase-space trajectory of the simulations. Values for the heat capacity were calculated for solid aluminium at $T = 500^\circ \text{C}$ at $P = 1, \text{bar}$ and liquid aluminium at $T = 700^\circ \text{K}$ and $P = 1 \text{ bar}$. The values calculated from the simulations are presented in table (2). The values calculated using fluctuations in kinetic and potential energy are close which would be expected for a system with conserved total energy as a fluctuation in kinetic energy would imply a fluctuation in potential energy and vice versa. One can also notice that the values for the heat capacity for the solid and melted system also are very close, which is not expected as the experimental value of the specific heat capacity for solid aluminium is 921.096 J/(kg K) [7] and 1.18 J/(kg K) [8]. The simulated values are a lot closer to the value the experimental value of aluminum, this implies that there are some problems simulating the liquid state of aluminum in a way that is true to the real physical system.

	$C_{V,pot} [\text{eV/K}]$	$C_{V,kin} [\text{eV/K}]$
$T = 500^\circ \text{C}, P = \text{bar}$	0.0661900	0.0661893
$T = 700^\circ \text{C}, P = 1 \text{ bar}$	0.0661895	0.0661895

Table 2: Value for the heat capacity calculated from fluctuations in the potential and kinetic energies during the simulation for a solid and liquid state for aluminium.

Problem 6

For this problem we were tasked with determining the radial distribution function $g(r)$ for the aluminium in a liquid state with the temperature $T = 700^\circ \text{C}$ and pressure $P = 1 \text{ bar}$. The pair distribution function for a spatially homogeneous system

is given by

$$g(\mathbf{r}', \mathbf{r}'') = \left\langle \sum_{i=1}^N \sum_{j \neq i}^N \delta(\mathbf{r}' - \mathbf{r}_i) \delta(\mathbf{r}'' - \mathbf{r}_j) \right\rangle \quad (12)$$

and for a isotropic system we can average over angles and not lose information and thus calculate the radial distribution function for the system. By integrating the radial distribution function from zero to its first minimum r_m we obtain the coordination number

$$I(r_m) = n \int_0^{r_m} g(r) 4\pi r^2 dr, \quad (13)$$

where n is the density of particles. The calculation of $g(r)$ was implemented in the Velocity-Verlet where at each time step the distance between every pair of particles were calculated. To handle the periodic boundary conditions the minimum image convention was used and the distances were sorted into a histogram and then by averaging over all particles and each time step a value for $g(r)$. The histogram was constructed by creating bins according to

$$r_k = (k - 0.5)\Delta r, \quad k = 1, 2, 3, \dots 300. \quad (14)$$

What bin to count a recorded distance r_{ij} between two particles was determined by rounding down the expression $r_{ij}/(\Delta r) + 0.5$ to the closest integer and adding a count to the an array at the corresponding index. The resulting value for the radial distribution function can be seen in figure (15). The coordination number was calculated to $I(r_m) = 12.521$ using `scipy.integrate.trapezoid()` in python from the measured minimum of 4.22 Å, see figure (15). The resulting value is relatively close to the value of solid aluminiums coordination number 12 and according to [9] a value of 11.5 was obtained for liquid aluminum so this result isn't unreasonable.

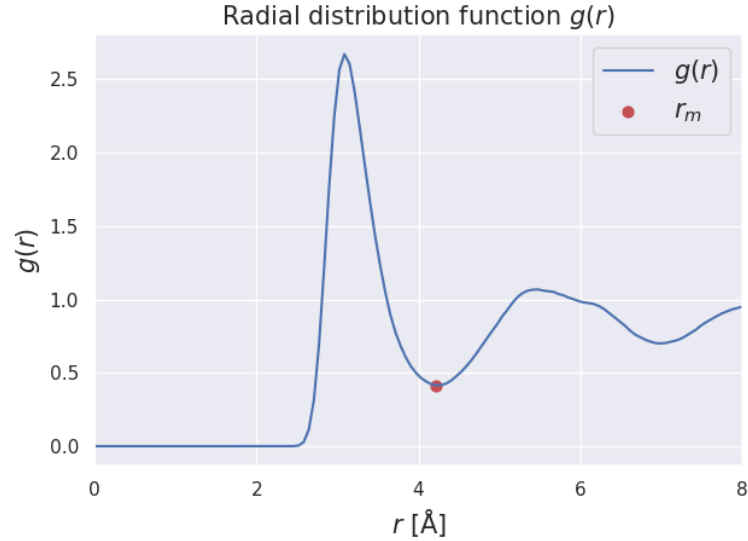


Figure 15: Figure displaying the radial distribution function obtained from simulating aluminum at $T = 700^\circ\text{C}$ and $P = 1$ bar. First local minium used to calculate coordination number was 4.22 Å.

Concluding Discussion

In general the results seen indicates the utility of computer simulations using the velocity verlet algorithm. Using the produced and provided code, quantities such as the lattice parameter, temperature, pressure, heat capacity and coordination number has been examined for solid and liquid aluminium respectively. The simulated values of the lattice parameter for solid aluminium at 773.15 K and liquid aluminium at 973.15 K has been measured to 4.092 and 4.254 Å respectively. These can be compared to experimental values of 4.0478 Å [10], which is relatively close. However, the lattice parameter for a

liquid is not well defined, and it is thus interesting that such a reasonable result was obtained. One reason could be the short timescale used in the simulation, and so the lattice parameter for the liquid aluminium simulation could be predicted to diverge during a longer simulation. While measured temperatures and pressures were consequences of equilibration, it was interesting to note the heavy fluctuations of pressure, compared to the relatively more well behaved fluctuations of temperature. This was discussed to partly be a consequence of the small system of just 256 particles, but this was not examined further and would be an interesting continuation of the experiment. The measured heat capacity for the solid and liquid aluminium simulations were measured to be very similar, which was not expected as the experimental values for these differ significantly. Lastly the coordination number was measured to be 12.521 which was relatively close to experimental values [9].

These results indicate the advantages of molecular dynamics simulations using the velocity verlet algorithm, as it allows the comparison between theoretical results that otherwise couldn't be obtained with great accuracy, to experimental results. One could also use results obtained to identify which parameters are more likely to impact the outcomes and use that to design experiments. For example one could use the scale of interactions to guide the selection of accuracy of measurement devices. In conclusion one could say that the results shown in this paper show the utility of molecular dynamics simulations with the velocity verlet algorithm.

References

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A Source Code

A.1 Running program and setting parameters for MD-simulations: run.c

```
1  #include <stdio.h>
2  #include <math.h>
3  #include <stdlib.h>
4  #include <stdbool.h>
5
6  #include "lattice.h"
7  #include "potential.h"
8  #include "tools.h"
9  #include "try_lattice_constants.h"
10 #include "velocity_verlet.h"
11
12 #include <gsl/gsl_rng.h>
13 #include <gsl/gsl_randist.h>
14 #include <time.h>
15
16 void H1_task1(), H1_task2(), H1_task3(), H1_task4(), H1_task6();
17
18 int
19 run(
20     int argc,
21     char *argv[]
22 )
23 {
24     H1_task1();
25     H1_task2();
26     H1_task3();
27     H1_task4();
28     H1_task6();
29
30     return 0;
31 }
32
33 void
34 H1_task1()
35 {
36     int nbr_atoms = 256; int n_rows = nbr_atoms; int n_cols = 3; int n_unitcells = 4;
37
38     // Testing lattice parametersdhwudh
39     bool get_small_lattice_param = true;
40     if(get_small_lattice_param == true){
41         double smallest_lattice_param = 0;
42         smallest_lattice_param = try_lattice_constants((int) nbr_atoms, (int) n_rows, (int) n_cols);
43         printf("%f\n", smallest_lattice_param);
44     }
45 }
46
47 void
48 H1_task2()
49 {
50     int nbr_atoms = 256; int n_rows = nbr_atoms; int n_cols = 3; int n_unitcells = 4;
51
52     // Initialising position and velocity arrays
53     double position[nbr_atoms][n_cols];
54     double velocity[nbr_atoms][n_cols];
55     for(int ix = 0; ix < nbr_atoms; ix++){
56         for(int jx = 0; jx < n_cols; jx++){
57             velocity[ix][jx] = 0;
58         }
59     }
60
61     // Choosing lattice param
62     double lattice_param = 4.03; // True is around 4.0478 (Masahiko Morinaga, https://bit.ly/3ERRFt3)
63     double cell_length = 4 * lattice_param;
64
65     // Initialice and displace fcc
66     init_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param); // 4 unit cells in each ↔
67     // direction
68
69     displace_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param);
70
71     // Declaring parameters for velocity verlet.
72     // If temp/press_scaling = false scaling is turned off and scaling factors remains = 1
73     int end_time; double dt;
74     bool temp_scaling, press_scaling, write_not_append;
75     double temp_eq, press_eq, tau_T, tau_P;
```

```

76 // Production run
77 end_time = 10; dt = 2e-2;
78 temp_scaling = false; press_scaling = false;
79 temp_eq = 773.15; press_eq = 1; //773.15 K och 1 Bar
80 tau_T = 1 * dt; tau_P = 1*dt;
81 write_not_append = true;
82
83 cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
84 (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
write_not_append, (double) tau_P, (double) tau_T);
85 }
86
87 void
88 H1_task3()
89 {
90     int nbr_atoms = 256; int n_rows = nbr_atoms; int n_cols = 3; int n_unitcells = 4;
91
92     // Initialising position and velocity arrays
93     double position[nbr_atoms][n_cols];
94     double velocity[nbr_atoms][n_cols];
95     for(int ix = 0; ix < nbr_atoms; ix++){
96         for(int jx = 0; jx < n_cols; jx++){
97             velocity[ix][jx] = 0;
98         }
99     }
100
101     // Choosing lattice param
102     double lattice_param = 4.03; // True is around 4.0478 (Masahiko Morinaga, https://bit.ly/3ERRFt3)
103     double cell_length = 4 * lattice_param;
104
105     // Initialice and displace fcc
106     init_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param); // 4 unit cells in each ←
direction
107
108     displace_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param);
109
110     // Declaring parameters for velocity verlet.
111     // If temp/press_scaling = false scaling is turned off and scaling factors remains = 1
112     int end_time; double dt;
113     bool temp_scaling, press_scaling, write_not_append;
114     double temp_eq, press_eq, tau_T, tau_P;
115
116     // end_time = 25; dt = 1e-3;
117     // tau_T = 100*dt; tau_P = 50*dt;
118
119     // Equalibration run
120     end_time = 20; dt = 1e-2;
121     temp_scaling = true; press_scaling = true;
122     temp_eq = 773.15; press_eq = 1; //773.15 K och 1 Bar
123     tau_T = 100*dt; tau_P = 300*dt; //dt; 50,5
124     write_not_append = true;
125
126     cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
127 (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
write_not_append, (double) tau_P, (double) tau_T);
128
129
130     // Production run
131     end_time = 20; dt = 1e-2;
132     temp_scaling = false; press_scaling = false;
133     temp_eq = 773.15; press_eq = 1; //773.15 K och 1 Bar
134     write_not_append = true;
135     cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
136 (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
write_not_append, (double) tau_P, (double) tau_T);
137 }
138
139 void
140 H1_task4()
141 {
142     int nbr_atoms = 256; int n_rows = nbr_atoms; int n_cols = 3; int n_unitcells = 4;
143
144     // Initialising position and velocity arrays
145     double position[nbr_atoms][n_cols];
146     double velocity[nbr_atoms][n_cols];
147     for(int ix = 0; ix < nbr_atoms; ix++){
148         for(int jx = 0; jx < n_cols; jx++){
149             velocity[ix][jx] = 0;
150         }
151     }

```

```

151 }
152
153 // Choosing lattice param
154 double lattice_param = 4.03; // True is around 4.0478 (Masahiko Morinaga, https://bit.ly/3ERRFt3)
155 double cell_length = 4 * lattice_param;
156
157 // Initialice and displace fcc
158 init_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param); // 4 unit cells in each ←
159 // direction
160
161 displace_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param);
162
163 // Declaring parameters for velocity verlet.
164 // If temp/press_scaling = false scaling is turned off and scaling factors remains = 1
165 int end_time; double dt;
166 bool temp_scaling, press_scaling, write_not_append;
167 double temp_eq, press_eq, tau_T, tau_P;
168
169 // Melting run
170 end_time = 20; dt = 1e-3;
171 temp_scaling = true; press_scaling = true;
172 temp_eq = 5000; press_eq = 1; //773.15 K och 1 Bar
173 tau_T = 100*dt; tau_P = 300*dt; //50*dt;
174 write_not_append = true;
175
176 cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
177 // double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
178 // (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
179 // write_not_append, (double) tau_P, (double) tau_T);
180
181 // Cooling run
182 end_time = 20; dt = 1e-3;
183 temp_scaling = true; press_scaling = true;
184 temp_eq = 973.15; press_eq = 1; //773.15 K och 1 Bar
185 write_not_append = false;
186
187 cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
188 // double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
189 // (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
190 // write_not_append, (double) tau_P, (double) tau_T);
191
192 // Production run
193 end_time = 20; dt = 1e-3;
194 temp_scaling = false; press_scaling = false;
195 temp_eq = 973.15; press_eq = 1; //773.15 K och 1 Bar
196 write_not_append = true;
197
198 cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
199 // double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
200 // (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
201 // write_not_append, (double) tau_P, (double) tau_T);
202
203 }
204
205 void
206 H1_task6()
207 {
208     int nbr_atoms = 256; int n_rows = nbr_atoms; int n_cols = 3; int n_unitcells = 4;
209
210     // Initialising position and velocity arrays
211     double position[nbr_atoms][n_cols];
212     double velocity[nbr_atoms][n_cols];
213     for(int ix = 0; ix < nbr_atoms; ix++){
214         for(int jx = 0; jx < n_cols; jx++){
215             velocity[ix][jx] = 0;
216         }
217     }
218
219     // Choosing lattice param
220     double lattice_param = 4.03; // True is around 4.0478 (Masahiko Morinaga, https://bit.ly/3ERRFt3)
221     double cell_length = 4 * lattice_param;
222
223     // Initialice and displace fcc
224     init_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param); // 4 unit cells in each ←
225     // direction
226
227     displace_fcc((double (*)[3]) position, (int) n_unitcells, (double) lattice_param);
228
229     // Declaring parameters for velocity verlet.

```



```

225 // If temp/press_scaling = false scaling is turned off and scaling factors remains = 1
226 int end_time; double dt;
227 bool temp_scaling, press_scaling, write_not_append;
228 double temp_eq, press_eq, tau_T, tau_P;
229
230
231 // Melting run
232 end_time = 20; dt = 1e-3;
233 temp_scaling = true; press_scaling = true;
234 temp_eq = 5000; press_eq = 1; //773.15 K och 1 Bar
235 tau_T = 100*dt; tau_P = 300*dt; //dt; 50,5
236 write_not_append = true;
237
238 cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
239 (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
write_not_append, (double) tau_P, (double) tau_T);
240
241 // Cooling run
242 end_time = 20; dt = 1e-3;
243 temp_scaling = true; press_scaling = true;
244 temp_eq = 973.15; press_eq = 1; //773.15 K och 1 Bar
245 write_not_append = false;
246
247 cell_length = velocity_verlet((double (*)[3]) position, (double (*)[3]) velocity, (double) lattice_param, (←
double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
248 (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
write_not_append, (double) tau_P, (double) tau_T);
249
250
251 // Production run
252 end_time = 20; dt = 1e-3;
253 temp_scaling = false; press_scaling = false;
254 temp_eq = 973.15; press_eq = 1; //773.15 K och 1 Bar
255 write_not_append = true;
256
257 int number_of_bins = 150;
258 double *radial_distribution_vector = calloc(sizeof(double), number_of_bins);
259 char filename_radial_dist[] = {"../csv/radial_distribution.csv"};
260 double normalisation_factor_radial_dist = (double) nbr_atoms*end_time/(dt);
261
262
263 cell_length = velocity_verlet_with_radial((double (*)[3]) position, (double (*)[3]) velocity, (double) ←
lattice_param, (double) cell_length, (int) end_time, (double) dt, (int) n_cols, (int) nbr_atoms, \
264 (bool) temp_scaling, (bool) press_scaling, (double) temp_eq, (double) press_eq, (bool) ←
write_not_append, (double) tau_P, (double) tau_T, radial_distribution_vector, ←
number_of_bins);
265
266
267 for(int bin = 0; bin<number_of_bins; ++bin)
268 {
269
270 //also need to divide by number of time steps
271 radial_distribution_vector[bin] /=normalisation_factor_radial_dist;
272 }
273
274 bool is_empty = true;
275 save_vector_to_csv(radial_distribution_vector, number_of_bins, filename_radial_dist, is_empty);
276 free(radial_distribution_vector);
277 }

```

A.2 Calculating equilibrium lattice parameter: try_lattice_constants.c

```

1 double
2 try_lattice_constants(int N_atoms, int n_rows, int n_cols){
3 // Initializing values needed for task
4 double E_pot = 0; int n_lattice_params = 8;
5 char filename_result[] = {"try_lattice_constants.csv"};
6 double lattice_params[n_lattice_params]; double lattice_param_init = 4.05; // 4.0478; // Lattice_param, ←
denoted a0 in document. Should be 4.0478 (Masahiko Morinaga, https://bit.ly/3ERRft3)
7
8 // Set print_q=true if we want to print the lattice_parameters code is looping through.
9 // Lattice params is an array around lattice_param_init with lattice_param_spacing
10 bool print_q = false; double lattice_param_spacing = 0.05;
11 if(print_q = true){
12 printf("n/2-ix, lattice_param \n");
13 for(int ix = 0; ix < n_lattice_params; ix++){
14 printf("%i, ", n_lattice_params/2-ix);

```

```

15     lattice_params[ix] = lattice_param_init - (n_lattice_params/2-ix)*lattice_param_spacing;
16     printf("%f\n", lattice_params[ix]);
17 }
18 printf("\n");
19 } else{
20     for(int ix = 0; ix < n_lattice_params; ix++){
21         lattice_params[ix] = lattice_param_init - (n_lattice_params/2-ix)*lattice_param_spacing;
22     }
23     printf("\n");
24 }
25
26 double smallest_E_potperunitcell = 0; double smallest_lattice_param = 0;
27 // For loop to run over lattice parameters
28 for(int ix = 0; ix < n_lattice_params; ix++){
29     // Initialize matrix and update values
30     double pos_matrix[256][3]; //double pos_matrix[4*256][3];
31     double lattice_param = lattice_params[ix];
32     double lattice_volume = pow(lattice_param, 3);
33
34     // Retrieve fcc with already made function
35     init_fcc((double (*)[3]) pos_matrix, (int) 4, (double) lattice_param); // 4 unit cells in each direction
36
37     // Retrieve potential energy with ready made function
38     double cell_length = 4 * lattice_param; // 4-unit cells
39     E_pot = get_energy_AL((double (*)[3]) pos_matrix, (double) cell_length, (int) N_atoms);
40
41     // Scaling result with number of unit cells and initializing result vector
42     double E_pot_per_unitcell = E_pot/pow(4,3);
43     double result_vec[] = {ix, lattice_param, lattice_volume, E_pot, E_pot_per_unitcell};
44
45     //Saving smallest lattice param if it results in lowest potential energy per unit cell
46     if(E_pot_per_unitcell < smallest_E_potperunitcell){
47         smallest_E_potperunitcell = E_pot_per_unitcell;
48         smallest_lattice_param = lattice_param;
49     }
50
51     // Saving results to csv file "try_lattice_constants.csv"
52     if(ix==0){
53         save_vector_to_csv(result_vec, 5, filename_result, true); // true -> fopen with "w"
54     } else {
55         save_vector_to_csv(result_vec, 5, filename_result, false); // false -> fopen with "a"
56     }
57
58     //Printing results in terminal if print_book is set to true
59     bool print_bool = true;
60     if(print_bool == true) {
61         printf("Lattice volume: %f\n", lattice_volume);
62         printf("E_pot: %f\n", E_pot);
63         printf("E_pot_perunitcell: %f\n", E_pot_per_unitcell);
64         printf("\n");
65     }
66 }
67
68 return smallest_lattice_param;
69 }

```

A.3 Velocity-Verlet-algorithm, with get_radial_distribution: *velocity_{erlet}.c*

```

1
2 #include <stdio.h>
3 #include <math.h>
4 #include <stdlib.h>
5 #include <stdbool.h>
6
7 #include "lattice.h"
8 #include "potential.h"
9 #include "tools.h"
10 #include "try_lattice_constants.h"
11
12 #include <gsl/gsl_rng.h>
13 #include <gsl/gsl_randist.h>
14 #include <time.h>
15
16 double
17 velocity_verlet(double positions[][3], double v[][3], double lattice_param, double cell_length, int end_time, \
18 double dt, int n_cols, int nbr_atoms, bool temp_scaling, bool press_scaling, double temp_eq, double press_eq, \
19 bool write_not_append, \
20 double tau_P, double tau_T)

```

```

20 {
21     // Initialize variables
22     double cell_volume = pow(cell_length, 3.);
23     double lattice_volume = pow(lattice_param, 3.);
24     double aluminium_amu = 26.98; double m_asu = 9649;
25     double aluminium_asu = aluminium_amu/m_asu; //Mass in atomic simulation units
26
27     double E_kinetic = 0; double E_kinetic_per_unitcell = 0;
28     double E_potential = 0; double E_potential_per_unitcell = 0;
29     double E_total_per_unitcell = 0;
30
31     double virial = 0; double virial_per_unitcell = 0;
32     double temp_inst = 0; double temp_inst_per_unitcell = 0;
33     double press_inst = 0; double press_inst_per_unitcell = 0;
34     double kB = 8.61733 * 1e-5; // Boltzmann constant in eV/K
35
36     //char *filename_result, *filename_pos, *filename_param;
37     char filename_result[] = {"../csv/vel_verlet_eq.csv"};
38     char filename_pos[] = {"../csv/position_track_eq.csv"};
39     char filename_param[] = {"../csv/parameters_eq.csv"};
40
41     char filename_result_prod[] = {"../csv/vel_verlet_prod.csv"};
42     char filename_pos_prod[] = {"../csv/position_track_prod.csv"};
43     char filename_param_prod[] = {"../csv/parameters_prod.csv"};
44     char filename_radial_distribution[] = {"../csv/radial_distribution.csv"};
45     bool is_write;
46
47     //Creating empty arrays
48     //double v[nbr_atoms][n_cols];
49     double f[nbr_atoms][n_cols];
50
51     for(int ix = 0; ix < nbr_atoms; ix++){
52         for(int jx = 0; jx < n_cols; jx++){
53             //v[ix][jx] = 0;
54             f[ix][jx] = 0;
55         }
56     }
57
58     // Variables for duration of measurement
59     int n_timesteps = end_time / dt;
60
61     // Declaring scaling variables, if temp/press_scaling = false scaling is turned off and alpha_T/P just ←
62     // remains 1
63     double alpha_T = 1; double alpha_P = 1;
64
65     //Velocity verlet algorithm as in E1
66     get_forces_AL((double (*)[3]) f, (double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
67
68     for(int tx = 1; tx < n_timesteps + 1; tx++){
69         {
70             /* v(t+dt/2) */
71             for(int ix = 0; ix < nbr_atoms; ix++){
72                 {
73                     for(int jx = 0; jx < n_cols; jx++){
74                         {
75                             v[ix][jx] += 0.5 * dt * f[ix][jx] / aluminium_asu;
76                             //printf("v[%i][%i] %f\n", ix, jx, v[ix][jx]);
77                         }
78                     }
79                 }
80             }
81             /* q(t+dt) */
82             for(int ix = 0; ix < nbr_atoms; ix++){
83                 {
84                     for(int jx = 0; jx < n_cols; jx++){
85                         {
86                             // alpha_P for scaling. If press_scaling==false this is 1.
87                             positions[ix][jx] += dt * v[ix][jx];
88                             positions[ix][jx] *= cbrt(alpha_P);
89                             //positions[ix][jx] *= pow(alpha_P, (double) 1./3.);
90                         }
91                     }
92                 }
93             }
94             // After scaling positions lattice_parameter is scaled to change pressure
95             cell_length *= cbrt(alpha_P); cell_volume = cell_length * cell_length * cell_length;
96
97             /* a(t+dt) */
98             get_forces_AL((double (*)[3]) f, (double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
99
100             // Resetting kinetic energy for every timestep to ensure summing over particles and not over timesteps
101             E_kinetic = 0;

```

```

101
102 /* v(t+dt) */
103 for(int ix = 0; ix < nbr_atoms; ix++)
104 {
105     for(int jx = 0; jx < n_cols; jx++)
106     {
107         // alpha_T for scaling. If temp_scaling==false this is 1.
108         v[ix][jx] += 0.5 * dt * f[ix][jx] / aluminium_asu;
109         v[ix][jx] *= sqrt(alpha_T);
110         E_kinetic += 0.5 * aluminium_asu * v[ix][jx] * v[ix][jx];
111     }
112 }
113
114 // Calculate potential, kinetic and total energy per unitcell
115 E_potential = get_energy_AL((double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
116 E_potential_per_unitcell = E_potential / pow(4.,3.);
117 E_kinetic_per_unitcell = E_kinetic / pow(4.,3.);
118 E_total_per_unitcell = E_potential_per_unitcell + E_kinetic_per_unitcell;
119
120 // Calculate virial to use for pressure calculation
121 virial = get_virial_AL((double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
122 virial_per_unitcell = virial / pow(4.,3.);
123
124 double N = 256;
125 double N1 = 4;
126 temp_inst_per_unitcell = 2 / (3 * N * kB) * E_kinetic;
127 //temp_inst_per_unitcell = 2 / (3 * N1 * kB) * E_kinetic_per_unitcell;
128
129 // Change scaling parameter for temperature
130 if(temp_scaling == true){
131
132     alpha_T = 1 + 2 * dt / tau_T * (temp_eq - temp_inst_per_unitcell)/temp_inst_per_unitcell;
133
134
135     // Calculate pressure in eV/ ^3
136     press_inst_per_unitcell = ((N * kB * temp_inst_per_unitcell) + virial) / cell_volume; //lattice_volume;
137
138     //press_inst_per_unitcell = ((N * kB * temp_inst_per_unitcell) + virial_per_unitcell) / cell_volume; //←
139     lattice_volume;
140
141     press_inst_per_unitcell *= 1.60219*1e2*1e4; // 1eV/ ^3 = 160.2 Gpa, 1 GPa = 10^4 Bar
142
143     // Changing scaling parameter for pressure
144     if(press_scaling == true){
145
146         double kappa_T = 0.01385*1e-4; // Neg eller pos?
147
148         // Unsure if plus or minus. Scaling with "correct" sign seems to change pressure in wrong direction
149         alpha_P = 1 - kappa_T * dt / tau_P * (press_eq - press_inst_per_unitcell);
150     }
151
152     // Creating vectors so to save results in csv files. Can be plotted with python files plot_energy.py and ←
153     plot_position_track.py
154     double parameter_vec[] = {end_time, dt, lattice_param, temp_scaling, press_scaling, temp_eq, press_eq, ←
155     tau_T, tau_P};
156     double result_vec[] = {tx*dt, cell_length, E_potential_per_unitcell, E_kinetic_per_unitcell, ←
157     E_total_per_unitcell, temp_inst_per_unitcell, press_inst_per_unitcell, alpha_T, alpha_P};
158     double position_track_vec[] = {tx*dt, positions[23][0], positions[23][1], positions[23][2], ←
159     positions[135][0], positions[135][1], positions[135][2], ←
160     positions[189][0], positions[189][1], positions[189][2], ←
161     temp_inst_per_unitcell};
162
163     // Saving results to csv files
164     if(tx == 1){is_write = write_not_append;} else {is_write = false;};
165     if(temp_scaling == true || press_scaling == true)
166     {
167         save_vector_to_csv(result_vec, 9, filename_result, is_write); // true -> fopen with "w"
168         save_vector_to_csv(position_track_vec, 10, filename_pos, is_write); // false -> fopen with "a"
169         if(tx == n_timesteps){
170             save_vector_to_csv(parameter_vec, 9, filename_param, is_write);
171         }
172         printf("Calibration: Inst. Temp, Press at t = [%i]: %f, %f\n", tx, temp_inst_per_unitcell, ←
173         press_inst_per_unitcell);
174     } else {
175         save_vector_to_csv(result_vec, 9, filename_result_prod, is_write); // true -> fopen with "w"
176         save_vector_to_csv(position_track_vec, 10, filename_pos_prod, is_write); // false -> fopen with "a"
177
178         if(tx == n_timesteps){
179             save_vector_to_csv(parameter_vec, 9, filename_param_prod, is_write);
180         }
181         printf("Production: Inst. Temp, Press at t = [%i]: %f, %f\n", tx, temp_inst_per_unitcell, ←
182         press_inst_per_unitcell);
183     }
184 }

```

```

177 // Printing temperature for each timestep to keep track during longer measurements
178 //printf("Inst. Temp, Press at t = [%i]: %f, %f\n", tx, temp_inst_per_unitcell, press_inst_per_unitcell↵
179 }
180 return cell_length;
181 }
182
183
184 //
185
186 double
187 velocity_verlet_with_radial(double positions[][3], double v[][3], double lattice_param, double cell_length, int ↵
188 end_time, \
189 double dt, int n_cols, int nbr_atoms, bool temp_scaling, bool press_scaling, double temp_eq, double press_eq, ↵
190 bool write_not_append, \
191 double tau_P, double tau_T, double *radial_histogram_vector, int number_of_bins)
192 {
193 // Initialize variables
194 double cell_volume = pow(cell_length, 3);
195 double lattice_volume = pow(lattice_param, 3);
196 double aluminium_amu = 26.98; double m_asu = 9649;
197 double aluminium_asu = aluminium_amu/m_asu; //Mass in atomic simulation units
198
199 double E_kinetic = 0; double E_kinetic_per_unitcell = 0;
200 double E_potential = 0; double E_potential_per_unitcell = 0;
201 double E_total_per_unitcell = 0;
202
203 double virial = 0; double virial_per_unitcell = 0;
204 double temp_inst = 0; double temp_inst_per_unitcell = 0;
205 double press_inst = 0; double press_inst_per_unitcell = 0;
206 double kB = 8.61733 * 1e-5; // Boltzmann constant in eV/K
207
208 //char *filename_result, *filename_pos, *filename_param;
209 char filename_result[] = {"../csv/vel_verlet_eq.csv"};
210 char filename_pos[] = {"../csv/position_track_eq.csv"};
211 char filename_param[] = {"../csv/parameters_eq.csv"};
212
213 char filename_result_prod[] = {"../csv/vel_verlet_prod.csv"};
214 char filename_pos_prod[] = {"../csv/position_track_prod.csv"};
215 char filename_param_prod[] = {"../csv/parameters_prod.csv"};
216 char filename_radial_distribution[] = {"../csv/radial_distribution.csv"};
217 bool is_write;
218
219 //Creating empty arrays
220 //double v[nbr_atoms][n_cols];
221 double f[nbr_atoms][n_cols];
222
223 for(int ix = 0; ix < nbr_atoms; ix++){
224     for(int jx = 0; jx < n_cols; jx++){
225         //v[ix][jx] = 0;
226         f[ix][jx] = 0;
227     }
228 }
229
230 // Variables for duration of measurement
231 int n_timesteps = end_time / dt;
232
233 // Declaring scaling variables, if temp/press_scaling = false scaling is turned off and alpha_T/P just ↵
234 remains 1
235 double alpha_T = 1; double alpha_P = 1;
236
237 //Velocity verlet algorithm as in E1
238 get_forces_AL((double (*)[3]) f, (double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
239
240 get_radial_dist_AL(number_of_bins, radial_histogram_vector, (double (*)[3])positions, cell_length, nbr_atoms)↵
241 ;
242
243 for(int tx = 1; tx < n_timesteps + 1; tx++)
244 {
245     /* v(t+dt/2) */
246     for(int ix = 0; ix < nbr_atoms; ix++)
247     {
248         for(int jx = 0; jx < n_cols; jx++)
249         {
250             v[ix][jx] += 0.5 * dt * f[ix][jx] / aluminium_asu;
251             //printf("v[%i][%i] %f\n", ix, jx, v[ix][jx]);
252         }
253     }
254
255     /* q(t+dt) */
256     for(int ix = 0; ix < nbr_atoms; ix++)

```

```

254 {
255     for(int jx = 0; jx < n_cols; jx++)
256     {
257         // alpha_P for scaling. If press_scaling==false this is 1.
258         positions[ix][jx] += dt * v[ix][jx];
259         positions[ix][jx] *= cbrt(alpha_P);
260         //positions[ix][jx] *= pow(alpha_P, (double) 1/3);
261     }
262 }
263
264 // After scaling positions lattice_parameter is scaled to change pressure
265 cell_length *= cbrt(alpha_P); cell_volume = cell_length*cell_length*cell_length;
266
267 /* a(t+dt) */
268 get_forces_AL((double (*)[3]) f, (double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
269 get_radial_dist_AL(number_of_bins, radial_histogram_vector, (double (*)[3])positions, cell_length, ←
270     nbr_atoms);
271
272 // Resetting kinetic energy for every timestep to ensure summing over particles and not over timesteps
273 E_kinetic = 0;
274
275 /* v(t+dt) */
276 for(int ix = 0; ix < nbr_atoms; ix++)
277 {
278     for(int jx = 0; jx < n_cols; jx++)
279     {
280         // alpha_T for scaling. If temp_scaling==false this is 1.
281         v[ix][jx] += 0.5 * dt * f[ix][jx] / aluminium_asu;
282         v[ix][jx] *= sqrt(alpha_T);
283         E_kinetic += 0.5 * aluminium_asu * v[ix][jx] * v[ix][jx];
284     }
285 }
286
287 // Calculate potential, kinetic and total energy per unitcell
288 E_potential = get_energy_AL((double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
289 E_potential_per_unitcell = E_potential / 64; // pow(4.,3.);
290 E_kinetic_per_unitcell = E_kinetic / 64; //pow(4.,3.);
291 E_total_per_unitcell = E_potential_per_unitcell + E_kinetic_per_unitcell;
292
293 // Calculate virial to use for pressure calculation
294 virial = get_virial_AL((double (*)[3]) positions, (double) cell_length, (int) nbr_atoms);
295 virial_per_unitcell = virial / 64; //pow(4.,3.);
296
297 double N = 256;
298
299 // Calculate temperature (4 atoms in unit cell, kB in eV/K)
300 temp_inst_per_unitcell = 2 / (3 * N * kB) * E_kinetic;
301 //temp_inst_per_unitcell = 2 / (3 * N1 * kB) * E_kinetic_per_unitcell;
302
303 // Change scaling parameter for temperature
304 if(temp_scaling == true){
305     alpha_T = 1 + 2 * dt / tau_T * (temp_eq - temp_inst_per_unitcell)/temp_inst_per_unitcell;
306 }
307 press_inst_per_unitcell = ((N * kB * temp_inst_per_unitcell) + virial) / cell_volume; //lattice_volume;
308 press_inst_per_unitcell *= 1.60219*1e2*1e4; // 1eV/ Å³ = 160.2 gPa
309
310 // Changing scaling parameter for pressure
311 if(press_scaling == true){
312     // Isothermal compressability for aluminium in Gpa^-1
313     double kappa_T = 0.01385*1e-4;
314
315     alpha_P = 1 - kappa_T * dt / tau_P * (press_eq - press_inst_per_unitcell);
316 }
317
318 // Creating vectors so to save results in csv files. Can be plotted with python files plot_energy.py and ←
319 // plot_position_track.py
320 double parameter_vec[] = {end_time, dt, lattice_param, temp_scaling, press_scaling, temp_eq, press_eq, ←
321     tau_T, tau_P};
322 double result_vec[] = {tx*dt, cell_length, E_potential_per_unitcell, E_kinetic_per_unitcell, ←
323     E_total_per_unitcell, temp_inst_per_unitcell, press_inst_per_unitcell, alpha_T, alpha_P};
324 double position_track_vec[] = {tx*dt, positions[23][0], positions[23][1], positions[23][2], \
325     positions[135][0], positions[135][1], positions[135][2], \
326     positions[189][0], positions[189][1], positions[189][2], \
327     temp_inst_per_unitcell};
328
329 // Saving results to csv files
330 if(tx == 1){is_write = write_not_append;} else {is_write = false;};
331 if(temp_scaling == true || press_scaling == true)
332 { save_vector_to_csv(result_vec, 9, filename_result, is_write); // true -> fopen with "w"

```

```

332     save_vector_to_csv(position_track_vec, 10, filename_pos, is_write); // false -> fopen with "a"
333     if(tx == n_timesteps){
334         save_vector_to_csv(parameter_vec, 9, filename_param, is_write);
335     }
336     printf("Calibration: Inst. Temp, Press at t = [%i]: %f,    %f\n", tx, temp_inst_per_unitcell, ←
        press_inst_per_unitcell);
337 } else {
338     save_vector_to_csv(result_vec, 9, filename_result_prod, is_write); // true -> fopen with "w"
339     save_vector_to_csv(position_track_vec, 10, filename_pos_prod, is_write); // false -> fopen with "a"
340
341     if(tx == n_timesteps){
342         save_vector_to_csv(parameter_vec, 9, filename_param_prod, is_write);
343     }
344     printf("Production: Inst. Temp, Press at t = [%i]: %f,    %f\n", tx, temp_inst_per_unitcell, ←
        press_inst_per_unitcell);
345 }
346 }
347 return cell_length;
348 }
349
350 /* Returns the forces */
351 void get_radial_dist_AL(int number_of_bins, double *radial_histogram_vector, double positions[][3], double ←
    cell_length, int nbr_atoms)
352 {
353     int i, j;
354     double cell_length_inv, cell_length_sq, bin_length;
355     double rcut, rcut_sq;
356     //double densityi, dens, drho_dr, force;
357     //double dUpair_dr;
358     double sxi, syi, szi, sxij, syij, szij, rij, rij_sq;
359
360     double *sx = malloc(nbr_atoms * sizeof (double));
361     double *sy = malloc(nbr_atoms * sizeof (double));
362     double *sz = malloc(nbr_atoms * sizeof (double));
363
364     rcut = 6.06; // Embedded atom method potential.
365     rcut_sq = rcut * rcut;
366
367
368     cell_length_inv = 1 / cell_length;
369     cell_length_sq = cell_length * cell_length;
370     bin_length= cell_length/(double)number_of_bins;
371
372     for (i = 0; i < nbr_atoms; i++){
373         sx[i] = positions[i][0] * cell_length_inv;
374         sy[i] = positions[i][1] * cell_length_inv;
375         sz[i] = positions[i][2] * cell_length_inv;
376     }
377
378     /* Compute radial distribution on atoms. */
379     /* Loop over atoms again :-(. */
380
381     for (i = 0; i < nbr_atoms; i++) {
382         /* Periodically translate coords of current particle to positive quadrants */
383         sxi = sx[i] - floor(sx[i]);
384         syi = sy[i] - floor(sy[i]);
385         szi = sz[i] - floor(sz[i]);
386
387         /* Loop over other atoms. */
388         for (j = i + 1; j < nbr_atoms; j++)
389         {
390             /* Periodically translate atom j to positive quadrants and calculate distance to it. */
391             sxij = sxi - (sx[j] - floor(sx[j]));
392             syij = syi - (sy[j] - floor(sy[j]));
393             szij = szi - (sz[j] - floor(sz[j]));
394
395             /* Periodic boundary conditions. */
396             sxij = sxij - (int)floor(sxij + 0.5);
397             syij = syij - (int)floor(syij + 0.5);
398             szij = szij - (int)floor(szij + 0.5);
399
400             /* squared distance between atom i and j */
401             rij_sq = cell_length_sq * (sxij*sxij + syij*syij + szij*szij);
402
403             /*Add position into bin depending on radial size*/
404             rij = sqrt( rij_sq );
405
406             int bin = (int) floor( rij/bin_length + 0.5);
407
408             radial_histogram_vector[bin] +=1;
409
410         }

```

```

411 }
412
413 free(sx); free(sy); free(sz); sx = NULL; sy = NULL; sz = NULL;
414 }

```

B Python functions for plotting

B.1 Main plotting function: plot_everything.py

```

1  import numpy as np
2  import matplotlib.pyplot as plt
3  import seaborn as sns
4  sns.set()
5
6  # set default figure size
7  plt.rcParams["figure.figsize"] = [8, 6]
8
9  SMALL_SIZE = 15
10 MEDIUM_SIZE = 18
11 BIGGER_SIZE = 18
12
13 plt.rc('font', size=SMALL_SIZE)           # controls default text sizes
14 plt.rc('axes', titlesize=MEDIUM_SIZE)     # fontsize of the axes title
15 plt.rc('axes', labelsize=MEDIUM_SIZE)     # fontsize of the x and y labels
16 plt.rc('xtick', labelsize=SMALL_SIZE)     # fontsize of the tick labels
17 plt.rc('ytick', labelsize=SMALL_SIZE)     # fontsize of the tick labels
18 plt.rc('legend', fontsize=SMALL_SIZE)     # legend fontsize
19 plt.rc('figure', titlesize=BIGGER_SIZE)
20
21 for idx, val in enumerate(["eq", "prod"]):
22     print(idx)
23     print(val)
24     str = val
25
26     # load data from file
27     array = np.genfromtxt(f'../csv/vel_verlet_{str}.csv', delimiter=',', skip_header=1)
28     parameters = np.genfromtxt(f'../csv/parameters_{str}.csv', delimiter=',')
29     pos_array = np.genfromtxt(f'../csv/position_track_{str}.csv', delimiter=',', skip_header=1)
30
31     end_time = parameters[-1,0]
32     dt = parameters[-1,1]
33     lattice_param = parameters[-1,2]
34     temp_scaling = parameters[-1,3]
35     press_scaling = parameters[-1,4]
36     temp_eq = parameters[-1,5]
37     press_eq = parameters[-1,6]
38     tau_T = parameters[-1,7]
39     tau_P = parameters[-1,8]
40
41     t = dt * np.linspace(0, len(array[:,1]), len(array[:,1]))
42     cell_length = array[:,1]
43     lattice_length = cell_length/4
44     e_pot = array[:,2]
45     e_kin = array[:,3]
46     e_tot = array[:,4]
47     temp = array[:,5]
48     press = array[:,6]
49
50     q1x = pos_array[:,1]
51     q1y = pos_array[:,2]
52     q1z = pos_array[:,3]
53     q2x = pos_array[:,4]
54     q2y = pos_array[:,5]
55     q2z = pos_array[:,6]
56     q3x = pos_array[:,7]
57     q3y = pos_array[:,8]
58     q3z = pos_array[:,9]
59
60     # create figure and axes for energy plot
61     fig, axes = plt.subplots(nrows=1, ncols=3, figsize=(14, 5))
62
63     # plot potential energy on first subplot
64     axes[0].plot(t, e_pot, label='Potential energy', color="b")
65     axes[0].set_title(f'Potential Energy, dt={dt} [ps]')
66
67     # plot kinetic energy on second subplot
68     axes[1].plot(t, e_kin, label='Kinetic energy', color="r")

```



```

69 axes[1].set_title(f'Kinetic Energy, dt={dt} [ps]')
70
71 # plot total energy on third subplot
72 axes[2].plot(t, e_tot, label='Total energy', color="g")
73 axes[2].set_title(f'Total Energy, dt={dt} [ps]')
74 axes[2].ticklabel_format(useOffset=False)
75
76 for idx in range(3):
77     axes[idx].set_xlabel('Time [ps]')
78     axes[idx].set_ylabel('Energy [eV/unit cell]')
79
80 plt.tight_layout()
81 plt.savefig(f'plots/energy_{str}.png')
82
83 figlattice , ax_lattice = plt.subplots(1,1)
84
85 ##### LATTICE #####
86 # plot energy data
87 ax_lattice.plot(t, lattice_length, label='Lattice parameter')
88
89 average_lattice = np.mean(lattice_length)
90
91 # add dashed line for the average energy
92 ax_lattice.annotate(f'a${0}$ = {lattice_length[-1]:.3f}', xy=(t[-1], lattice_length[-1]), xytext=(-50, ←
    -50),
93     textcoords='offset pixels', arrowprops=dict(arrowstyle='->', color='k'))
94
95 # set labels and title
96 ax_lattice.set_xlabel('Time [ps]')
97 ax_lattice.set_ylabel('Lattice parameter [ ]')
98 ax_lattice.set_title(f'Lattice parameter, dt={dt}')
99
100 plt.legend()
101 plt.tight_layout()
102 plt.savefig(f'plots/lattice_{str}.png')
103
104
105
106 ##### POSITIONS #####
107 fig_pos, ax_pos = plt.subplots(1, 3, figsize=(14, 5))
108
109 # Iterate over dimensions
110 for i, dim in enumerate(["X", "Y", "Z"]):
111     # Plot each dimension in a subplot
112     ax_pos[i].plot(t, eval(f"q1{dim.lower()}"), label="q1")
113     ax_pos[i].plot(t, eval(f"q2{dim.lower()}"), label="q2")
114     ax_pos[i].plot(t, eval(f"q3{dim.lower()}"), label="q3")
115
116     # Set x- and y-labels and title
117     ax_pos[i].set_xlabel("Time [ps]")
118     ax_pos[i].set_ylabel(f"{dim}-coordinate [ ]")
119     ax_pos[i].set_title(f"{dim}-coordinate, dt={dt}")
120
121     # Add legend
122     ax_pos[i].legend(loc='lower right')
123
124 # Adjust layout and save figure
125 plt.tight_layout()
126 plt.savefig(f"plots/position_track_{str}.png")
127
128
129 ##### PRESSURE #####
130 figP , axP = plt.subplots(1,1)
131 axP.plot(t, press, label='Pressure')
132
133 average_pressure = np.mean(press)
134
135 # add dashed line for the average energy
136 if(str == "prod"):
137     print("Hej")
138     axP.axhline(
139         average_pressure,
140         linestyle='--',
141         color='r',
142         linewidth=4,
143         label=f'P${average}$ = {average_pressure:.2f} [Bar]'
144     )
145
146     #axP.annotate(f'P${average}$ = {average_pressure:.2f}', xy=(t[-1], average_pressure), xytext=(-150, ←
147         150),
148     #     textcoords='offset pixels', arrowprops=dict(arrowstyle='->', color='k'))

```

```

149 axP.set_xlabel('Time [ps]')
150 axP.set_ylabel('Pressure [Bar]')
151 axP.set_title(f'Pressure, dt={dt}')
152
153 plt.legend()
154 plt.tight_layout()
155 plt.savefig(f'plots/pressure_{str}.png')
156 #plt.show()
157
158
159
160 ##### TEMPERATURE #####
161
162 average_temperature = np.mean(array[:,5])
163
164 fig2 , axT = plt.subplots(1,1)
165 axT.plot(t, temp, label='Temperature')
166
167 # add dashed line for the average energy
168 if(str == "prod"):
169     axT.axhline(
170         average_temperature,
171         linestyle='--',
172         color='r',
173         linewidth=4,
174         label=f'T$_{average}$ = {average_temperature:.2f} [K]'
175     )
176
177     #axT.annotate(f'T$_{average}$ = {average_temperature:.2f}', xy=(t[-1], average_temperature), xytext←
178     #=(-110, 200),
179     # textcoords='offset pixels', arrowprops=dict(arrowstyle='->', color='k'))
180
181 plt.legend()
182 axT.set_xlabel('Time [ps]')
183 axT.set_ylabel('Temperature [K]')
184 axT.set_title(f'Temperature, dt={dt}')
185
186 plt.legend()
187 plt.tight_layout()
188 plt.savefig(f'plots/temperature_{str}.png')

```

B.2 Plotting function for task 6 with calculation of heat capacity and coordination number: plot_task6.py

```

1 import numpy as np
2 import matplotlib.pyplot as plt
3 import seaborn as sns
4 import scipy as sp
5 sns.set()
6
7 # set default figure size
8 plt.rcParams["figure.figsize"] = [8, 6]
9
10 SMALL_SIZE = 13
11 MEDIUM_SIZE = 15
12 BIGGER_SIZE = 15
13
14 plt.rc('font', size=SMALL_SIZE) # controls default text sizes
15 plt.rc('axes', titlesize=MEDIUM_SIZE) # fontsize of the axes title
16 plt.rc('axes', labelsiz=MEDIUM_SIZE) # fontsize of the x and y labels
17 plt.rc('xtick', labelsiz=SMALL_SIZE) # fontsize of the tick labels
18 plt.rc('ytick', labelsiz=SMALL_SIZE) # fontsize of the tick labels
19 plt.rc('legend', fontsize=SMALL_SIZE) # legend fontsize
20 plt.rc('figure', titlesize=BIGGER_SIZE)
21
22
23 Cell_length = 4*4.03; number_particles = 256
24 array = np.genfromtxt(f'./csv/radial_distribution.csv', delimiter=',', skip_header=0)
25 number_of_bins = len(array)
26 def N_ideal(bin_arr, Cell_length, Number_of_particles):
27     Number_of_bins = len(bin_arr)
28
29     delta_r = Cell_length/Number_of_bins
30     V= Cell_length**3
31
32     N_ideal = (Number_of_particles-1)*4*np.pi/(3*Cell_length**3) * (np.square(bin_arr) -3*bin_arr+1)*delta_r**3
33
34     return N_ideal

```

```

35
36 bin_array = np.arange(number_of_bins)
37 ideal = N_ideal(bin_array, Cell_length, number_particles)
38 delta_r = Cell_length/number_particles
39 fig, ax_radial = plt.subplots(1,1)
40 array = (array/ideal)
41 max = np.argmax(array)
42 upper = 100
43 ax_radial.plot(bin_array*delta_r, array, label='$g(r)$')
44 ax_radial.set_xlim(0,8)
45
46 minima = np.argmin(array[max:upper])
47 max = np.argmax(array)
48
49 index_r_m = max+minima
50
51 plt.rcParams['xtick.labelsize'] = 30
52 radiusvec = bin_array*delta_r
53 ax_radial.scatter(radiusvec[index_r_m], array[index_r_m], color='r', label="$r_m$", s=40)
54 ax_radial.set_title(r"Radial distribution function $g(r)$", fontsize=15)
55 ax_radial.set_xlabel("$r$ [ ]", fontsize=15)
56 ax_radial.set_ylabel(r"$g(r)$", fontsize=15)
57 ax_radial.legend(fontsize=15)
58 plt.tight_layout()
59
60 integrand = []
61
62 for i,r in enumerate(bin_array[:index_r_m]*delta_r):
63     print(r,i, array[i])
64     integrand.append(r**2* 4*np.pi* array[i]*(256/Cell_length**3))
65
66 coordination_number = sp.integrate.trapezoid(bin_array[:index_r_m]*delta_r, integrand)
67 print(coordination_number)
68 fig.savefig(f'plots/radial.png')

```

C Provided functions

C.1 Potential, included in course material: potentials.c

```

1 #include <stdio.h>
2 #include <math.h>
3 #include <stdlib.h>
4
5 /*Parameters for the AL EAM potential */
6 #define PAIR_POTENTIAL_ROWS 18
7 const double pair_potential[90] = {2.0210, 2.2730, 2.4953, 2.7177, 2.9400, 3.1623, 3.3847, 3.6070, 3.8293, ←
    4.0517, 4.2740, 4.4963, 4.7187, 4.9410, 5.1633, 5.3857, 5.6080, 6.0630, 2.0051, 0.7093, 0.2127, 0.0202, ←
    -0.0386, -0.0492, -0.0424, -0.0367, -0.0399, -0.0574, -0.0687, -0.0624, -0.0492, -0.0311, -0.0153, -0.0024, ←
    -0.0002, 0, -7.2241, -3.3383, -1.3713, -0.4753, -0.1171, 0.0069, 0.0374, 0.0122, -0.0524, -0.0818, -0.0090, ←
    0.0499, 0.0735, 0.0788, 0.0686, 0.0339, -0.0012, 0, 9.3666, 6.0533, 2.7940, 1.2357, 0.3757, 0.1818, -0.0445, ←
    -0.0690, -0.2217, 0.0895, 0.2381, 0.0266, 0.0797, -0.0557, 0.0097, -0.1660, 0.0083, 0, -4.3827, -4.8865, ←
    -2.3363, -1.2893, -0.2907, -0.3393, -0.0367, -0.2290, 0.4667, 0.2227, -0.3170, 0.0796, -0.2031, 0.0980, ←
    -0.2634, 0.2612, -0.0102, 0};
8
9
10 #define ELECTRON_DENSITY_ROWS 15
11 const double electron_density[75] = {2.0210, 2.2730, 2.5055, 2.7380, 2.9705, 3.2030, 3.4355, 3.6680, 3.9005, ←
    4.1330, 4.3655, 4.5980, 4.8305, 5.0630, 6.0630, 0.0824, 0.0918, 0.0883, 0.0775, 0.0647, 0.0512, 0.0392, ←
    0.0291, 0.0186, 0.0082, 0.0044, 0.0034, 0.0027, 0.0025, 0.0000, 0.0707, 0.0071, -0.0344, -0.0533, -0.0578, ←
    -0.0560, -0.0465, -0.0428, -0.0486, -0.0318, -0.0069, -0.0035, -0.0016, -0.0008, 0, -0.1471, -0.1053, ←
    -0.0732, -0.0081, -0.0112, 0.0189, 0.0217, -0.0056, -0.0194, 0.0917, 0.0157, -0.0012, 0.0093, -0.0059, 0, ←
    0.0554, 0.0460, 0.0932, -0.0044, 0.0432, 0.0040, -0.0392, -0.0198, 0.1593, -0.1089, -0.0242, 0.0150, ←
    -0.0218, 0.0042, 0};
12
13 #define EMBEDDING_ENERGY_ROWS 13
14 const double embedding_energy[65] = {0, 0.1000, 0.2000, 0.3000, 0.4000, 0.5000, 0.6000, 0.7000, 0.8000, 0.9000, ←
    1.0000, 1.1000, 1.2000, 0, -1.1199, -1.4075, -1.7100, -1.9871, -2.2318, -2.4038, -2.5538, -2.6224, -2.6570, ←
    -2.6696, -2.6589, -2.6358, -18.4387, -5.3706, -2.3045, -3.1161, -2.6175, -2.0666, -1.6167, -1.1280, -0.4304, ←
    -0.2464, -0.0001, 0.1898, 0.2557, 86.5178, 44.1632, -13.5018, 5.3853, -0.3996, 5.9090, -1.4103, 6.2976, ←
    0.6785, 1.1611, 1.3022, 0.5971, 0.0612, -141.1819, -192.2166, 62.9570, -19.2831, 21.0288, -24.3978, 25.6930, ←
    -18.7304, 1.6087, 0.4704, -2.3503, -1.7862, -1.7862};
15
16
17 /* Evaluates the spline in x. */
18
19 double splineEval(double x, const double *table, int m) {
20     /* int m = mxGetM(spline), i, k;*/

```

```

21     int i, k;
22
23     /*double *table = mxGetPr(spline);*/
24     double result;
25
26     int k_lo = 0, k_hi = m;
27
28     /* Find the index by bisection. */
29     while (k_hi - k_lo > 1) {
30         k = (k_hi + k_lo) >> 1;
31         if (table[k] > x)
32             k_hi = k;
33         else
34             k_lo = k;
35     }
36
37     /* Switch to local coord. */
38     x -= table[k_lo];
39
40     /* Horner's scheme */
41     result = table[k_lo + 4*m];
42     for (i = 3; i > 0; i--) {
43         result *= x;
44         result += table[k_lo + i*m];
45     }
46
47     return result;
48 }
49
50 /* Evaluates the derivative of the spline in x. */
51
52 double splineEvalDiff(double x, const double *table, int m) {
53     /*int m = mxGetM(spline), i, k;
54     double *table = mxGetPr(spline);
55     */
56     int i, k;
57     double result;
58
59     int k_lo = 0, k_hi = m;
60
61     /* Find the index by bisection. */
62     while (k_hi - k_lo > 1) {
63         k = (k_hi + k_lo) >> 1;
64         if (table[k] > x)
65             k_hi = k;
66         else
67             k_lo = k;
68     }
69
70     /* Switch to local coord. */
71     x -= table[k_lo];
72
73     /* Horner's scheme */
74     result = 3*table[k_lo + 4*m];
75     for (i = 3; i > 1; i--) {
76         result *= x;
77         result += (i-1)*table[k_lo + i*m];
78     }
79
80     return result;
81 }
82
83 /* Returns the forces */
84 void get_forces_AL(double forces[][3], double positions[][3], double cell_length, int nbr_atoms)
85 {
86     int i, j;
87     double cell_length_inv, cell_length_sq;
88     double rcut, rcut_sq;
89     double densityi, dens, drho_dr, force;
90     double dUpair_dr;
91     double sxi, syi, szi, sxij, syij, szij, rij, rij_sq;
92
93     double *sx = malloc(nbr_atoms * sizeof (double));
94     double *sy = malloc(nbr_atoms * sizeof (double));
95     double *sz = malloc(nbr_atoms * sizeof (double));
96     double *fx = malloc(nbr_atoms * sizeof (double));
97     double *fy = malloc(nbr_atoms * sizeof (double));
98     double *fz = malloc(nbr_atoms * sizeof (double));
99
100     double *density = malloc(nbr_atoms * sizeof (double));
101     double *dUembed_drho = malloc(nbr_atoms * sizeof (double));
102

```

```

103 rcut = 6.06;
104 rcut_sq = rcut * rcut;
105
106 cell_length_inv = 1 / cell_length;
107 cell_length_sq = cell_length * cell_length;
108
109 for (i = 0; i < nbr_atoms; i++){
110     sx[i] = positions[i][0] * cell_length_inv;
111     sy[i] = positions[i][1] * cell_length_inv;
112     sz[i] = positions[i][2] * cell_length_inv;
113 }
114
115 for (i = 0; i < nbr_atoms; i++){
116     density[i] = 0;
117     fx[i] = 0;
118     fy[i] = 0;
119     fz[i] = 0;
120 }
121
122 for (i = 0; i < nbr_atoms; i++) {
123     /* Periodically translate coords of current particle to positive quadrants */
124     sxi = sx[i] - floor(sx[i]);
125     syi = sy[i] - floor(sy[i]);
126     szi = sz[i] - floor(sz[i]);
127
128     densityi = density[i];
129
130     /* Loop over other atoms. */
131     for (j = i + 1; j < nbr_atoms; j++) {
132         /* Periodically translate atom j to positive quadrants and calculate distance to it. */
133         sxij = sxi - (sx[j] - floor(sx[j]));
134         syij = syi - (sy[j] - floor(sy[j]));
135         szij = szi - (sz[j] - floor(sz[j]));
136
137         /* Periodic boundary conditions. */
138         sxij = sxij - (int)floor(sxij + 0.5);
139         syij = syij - (int)floor(syij + 0.5);
140         szij = szij - (int)floor(szij + 0.5);
141
142         /* squared distance between atom i and j */
143         rij_sq = cell_length_sq * (sxij*sxij + syij*syij + szij*szij);
144
145         /* Add force and energy contribution if distance between atoms smaller than rcut */
146         if (rij_sq < rcut_sq) {
147             rij = sqrt(rij_sq);
148             dens = splineEval(rij, electron_density, ELECTRON_DENSITY_ROWS);
149             densityi += dens;
150             density[j] += dens;
151         }
152     }
153     density[i] = densityi;
154 }
155
156 /* Loop over atoms to calculate derivative of embedding function
157 and embedding function. */
158 for (i = 0; i < nbr_atoms; i++) {
159     dUembed_drho[i] = splineEvalDiff(density[i], embedding_energy, EMBEDDING_ENERGY_ROWS);
160 }
161
162 /* Compute forces on atoms. */
163 /* Loop over atoms again :-(. */
164
165 for (i = 0; i < nbr_atoms; i++) {
166     /* Periodically translate coords of current particle to positive quadrants */
167     sxi = sx[i] - floor(sx[i]);
168     syi = sy[i] - floor(sy[i]);
169     szi = sz[i] - floor(sz[i]);
170
171     densityi = density[i];
172
173     /* Loop over other atoms. */
174     for (j = i + 1; j < nbr_atoms; j++) {
175         /* Periodically translate atom j to positive quadrants and calculate distance to it. */
176         sxij = sxi - (sx[j] - floor(sx[j]));
177         syij = syi - (sy[j] - floor(sy[j]));
178         szij = szi - (sz[j] - floor(sz[j]));
179
180         /* Periodic boundary conditions. */
181         sxij = sxij - (int)floor(sxij + 0.5);
182         syij = syij - (int)floor(syij + 0.5);
183         szij = szij - (int)floor(szij + 0.5);
184

```

```

185     /* squared distance between atom i and j */
186     rij_sq = cell_length_sq * (sxij*sxij + syij*syij + szij*szij);
187
188     /* Add force and energy contribution if distance between atoms smaller than rcut */
189     if (rij_sq < rcut_sq) {
190         rij = sqrt(rij_sq);
191         dUpair_dr = splineEvalDiff(rij, pair_potential, PAIR_POTENTIAL_ROWS);
192         drho_dr = splineEvalDiff(rij, electron_density, ELECTRON_DENSITY_ROWS);
193
194         /* Add force contribution from i-j interaction */
195         force = -(dUpair_dr + (dUembed_drho[i] + dUembed_drho[j])*drho_dr) / rij;
196         fx[i] += force * sxij * cell_length;
197         fy[i] += force * syij * cell_length;
198         fz[i] += force * szij * cell_length;
199         fx[j] -= force * sxij * cell_length;
200         fy[j] -= force * syij * cell_length;
201         fz[j] -= force * szij * cell_length;
202     }
203 }
204 }
205
206 for (i = 0; i < nbr_atoms; i++){
207     forces[i][0] = fx[i];
208     forces[i][1] = fy[i];
209     forces[i][2] = fz[i];
210 }
211
212 free(sx); free(sy); free(sz); sx = NULL; sy = NULL; sz = NULL;
213 free(fx); free(fy); free(fz); fx = NULL; fy = NULL; fz = NULL;
214 free(density); density = NULL;
215 free(dUembed_drho); dUembed_drho = NULL;
216
217 }
218
219 /* Returns the potential energy */
220 double get_energy_AL(double positions[][3], double cell_length, int nbr_atoms)
221 {
222     int i, j;
223     double cell_length_inv, cell_length_sq;
224     double rcut, rcut_sq;
225     double energy;
226     double densityi, dens;
227     double sxi, syi, szi, sxij, syij, szij, rij, rij_sq;
228
229     double *sx = malloc(nbr_atoms * sizeof (double));
230     double *sy = malloc(nbr_atoms * sizeof (double));
231     double *sz = malloc(nbr_atoms * sizeof (double));
232
233     double *density = malloc(nbr_atoms * sizeof (double));
234
235     rcut = 6.06;
236     rcut_sq = rcut * rcut;
237
238     cell_length_inv = 1 / cell_length;
239     cell_length_sq = cell_length * cell_length;
240
241     for (i = 0; i < nbr_atoms; i++){
242         sx[i] = positions[i][0] * cell_length_inv;
243         sy[i] = positions[i][1] * cell_length_inv;
244         sz[i] = positions[i][2] * cell_length_inv;
245     }
246
247     for (i = 0; i < nbr_atoms; i++){
248         density[i] = 0;
249     }
250
251     energy = 0;
252
253     for (i = 0; i < nbr_atoms; i++) {
254         /* Periodically translate coords of current particle to positive quadrants */
255         sxi = sx[i] - floor(sx[i]);
256         syi = sy[i] - floor(sy[i]);
257         szi = sz[i] - floor(sz[i]);
258
259         densityi = density[i];
260
261         /* Loop over other atoms. */
262         for (j = i + 1; j < nbr_atoms; j++) {
263             /* Periodically translate atom j to positive quadrants and calculate distance to it. */
264             sxij = sxi - (sx[j] - floor(sx[j]));
265             syij = syi - (sy[j] - floor(sy[j]));
266             szij = szi - (sz[j] - floor(sz[j]));

```

```

267
268     /* Periodic boundary conditions. */
269     sxij = sxij - (int)floor(sxij + 0.5);
270     syij = syij - (int)floor(syij + 0.5);
271     szij = szij - (int)floor(szij + 0.5);
272
273     /* squared distance between atom i and j */
274     rij_sq = cell_length_sq * (sxij*sxij + syij*syij + szij*szij);
275
276     /* Add force and energy contribution if distance between atoms smaller than rcut */
277     if (rij_sq < rcut_sq) {
278         rij = sqrt(rij_sq);
279         dens = splineEval(rij, electron_density, ELECTRON_DENSITY_ROWS);
280         densityi += dens;
281         density[j] += dens;
282
283         /* Add energy contribution from i-j interaction */
284         energy += splineEval(rij, pair_potential, PAIR_POTENTIAL_ROWS);
285     }
286 }
287
288 density[i] = densityi;
289 }
290
291 /* Loop over atoms to calculate derivative of embedding function
292 and embedding function. */
293 for (i = 0; i < nbr_atoms; i++) {
294     energy += splineEval(density[i], embedding_energy, EMBEDDING_ENERGY_ROWS);
295 }
296
297 free(sx); free(sy); free(sz); sx = NULL; sy = NULL; sz = NULL;
298 free(density); density = NULL;
299
300 return(energy);
301
302 }
303
304 /* Returns the virial */
305 double get_virial_AL(double positions[][3], double cell_length, int nbr_atoms)
306 {
307     int i, j;
308     double cell_length_inv, cell_length_sq;
309     double rcut, rcut_sq;
310     double virial;
311     double densityi, dens, drho_dr, force;
312     double dUpair_dr;
313     double sxi, syi, szi, sxij, syij, szij, rij, rij_sq;
314
315     double *sx = malloc(nbr_atoms * sizeof (double));
316     double *sy = malloc(nbr_atoms * sizeof (double));
317     double *sz = malloc(nbr_atoms * sizeof (double));
318
319     double *density = malloc(nbr_atoms * sizeof (double));
320     double *dUembed_drho = malloc(nbr_atoms * sizeof (double));
321
322     rcut = 6.06;
323     rcut_sq = rcut * rcut;
324
325     cell_length_inv = 1 / cell_length;
326     cell_length_sq = cell_length * cell_length;
327
328     for (i = 0; i < nbr_atoms; i++){
329         sx[i] = positions[i][0] * cell_length_inv;
330         sy[i] = positions[i][1] * cell_length_inv;
331         sz[i] = positions[i][2] * cell_length_inv;
332     }
333
334     for (i = 0; i < nbr_atoms; i++){
335         density[i] = 0;
336     }
337
338     for (i = 0; i < nbr_atoms; i++) {
339         /* Periodically translate coords of current particle to positive quadrants */
340         sxi = sx[i] - floor(sx[i]);
341         syi = sy[i] - floor(sy[i]);
342         szi = sz[i] - floor(sz[i]);
343
344         densityi = density[i];
345
346         /* Loop over other atoms. */
347         for (j = i + 1; j < nbr_atoms; j++) {
348             /* Periodically translate atom j to positive quadrants and calculate distance to it. */

```

```

349     sxij = sxi - (sx[j] - floor(sx[j]));
350     syij = syi - (sy[j] - floor(sy[j]));
351     szij = szi - (sz[j] - floor(sz[j]));
352
353     /* Periodic boundary conditions. */
354     sxij = sxij - (int)floor(sxij + 0.5);
355     syij = syij - (int)floor(syij + 0.5);
356     szij = szij - (int)floor(szij + 0.5);
357
358     /* squared distance between atom i and j */
359     rij_sq = cell_length_sq * (sxij*sxij + syij*syij + szij*szij);
360
361     /* Add force and energy contribution if distance between atoms smaller than rcut */
362     if (rij_sq < rcut_sq) {
363         rij = sqrt(rij_sq);
364         dens = splineEval(rij, electron_density, ELECTRON_DENSITY_ROWS);
365         densityi += dens;
366         density[j] += dens;
367     }
368 }
369 density[i] = densityi;
370 }
371
372 /* Loop over atoms to calculate derivative of embedding function
373 and embedding function. */
374 for (i = 0; i < nbr_atoms; i++) {
375     dUembed_drho[i] = splineEvalDiff(density[i], embedding_energy, EMBEDDING_ENERGY_ROWS);
376 }
377
378 /* Compute forces on atoms. */
379 /* Loop over atoms again :-(. */
380
381 virial = 0;
382
383 for (i = 0; i < nbr_atoms; i++) {
384     /* Periodically translate coords of current particle to positive quadrants */
385     sxi = sx[i] - floor(sx[i]);
386     syi = sy[i] - floor(sy[i]);
387     szi = sz[i] - floor(sz[i]);
388
389     densityi = density[i];
390
391     /* Loop over other atoms. */
392     for (j = i + 1; j < nbr_atoms; j++) {
393         /* Periodically translate atom j to positive quadrants and calculate distance to it. */
394         sxij = sxi - (sx[j] - floor(sx[j]));
395         syij = syi - (sy[j] - floor(sy[j]));
396         szij = szi - (sz[j] - floor(sz[j]));
397
398         /* Periodic boundary conditions. */
399         sxij = sxij - (int)floor(sxij + 0.5);
400         syij = syij - (int)floor(syij + 0.5);
401         szij = szij - (int)floor(szij + 0.5);
402
403         /* squared distance between atom i and j */
404         rij_sq = cell_length_sq * (sxij*sxij + syij*syij + szij*szij);
405
406         /* Add force and energy contribution if distance between atoms smaller than rcut */
407         if (rij_sq < rcut_sq) {
408             rij = sqrt(rij_sq);
409             dUpair_dr = splineEvalDiff(rij, pair_potential, PAIR_POTENTIAL_ROWS);
410             drho_dr = splineEvalDiff(rij, electron_density, ELECTRON_DENSITY_ROWS);
411
412             /* Add virial contribution from i-j interaction */
413             force = -(dUpair_dr + (dUembed_drho[i] + dUembed_drho[j])*drho_dr) / rij;
414
415             virial += force * rij_sq;
416         }
417     }
418 }
419
420 virial /= 3.0;
421
422 free(sx); free(sy); free(sz); sx = NULL; sy = NULL; sz = NULL;
423 free(density); density = NULL;
424 free(dUembed_drho); dUembed_drho = NULL;
425
426 return(virial);
427
428 }

```


C.2 Lattice, included in course material: lattice.c

```
1  /*
2  Hlattice.c
3  Program that arranges atoms on a fcc lattice.
4  Created by Anders Lindman on 2013-03-15.
5  */
6
7  #include <stdio.h>
8
9  /* Function takes a matrix of size [4*N*N*N][3] as input and stores a fcc lattice in it. N is the number of unit ↔
10     cells in each dimension and lattice_param is the lattice parameter. */
11 void init_fcc(double positions[][3], int N, double lattice_param)
12 {
13     int i, j, k;
14     int xor_value;
15
16     for (i = 0; i < 2 * N; i++){
17         for (j = 0; j < 2 * N; j++){
18             for (k = 0; k < N; k++){
19                 if (j % 2 == i % 2 ){
20                     xor_value = 0;
21                 }
22                 else {
23                     xor_value = 1;
24                 }
25                 positions[i * N * 2 * N + j * N + k][0] = lattice_param * (0.5 * xor_value + k);
26                 positions[i * N * 2 * N + j * N + k][1] = lattice_param * (j * 0.5);
27                 positions[i * N * 2 * N + j * N + k][2] = lattice_param * (i * 0.5);
28             }
29         }
30     }
```