Derivative Crusher Algorithm
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Idea: A partial derivative with thermodynamically significant variables

$$\left(\frac{\partial x}{\partial y}\right)_z \text{ where } \{x, y, z\} \text{ can be any of } \{S, T, P, V, E, F, G, H\}$$

can be expressed in terms of in terms of first and second derivatives of fundamental relations. This re-arranging of expressions is very useful in thermodynamics and statistical mechanics, as it allows us to find simple relations between physical quantities in a thermodynamical system. The Derivative Crusher Algorithm is an algorithm that "crushes" a thermodynamically significant partial derivative into useful and easily measureable quantities. Developed and taught by Professor Ratindranath Akhoury.

1   Gibbs Free Energy

Variable	Easy to keep fixed?	Easy to measure change?
S	**Yes**, Adiabatic processes	**No**, change in S depends on change in Q, which is hard to measure
T	**Yes**, with a heat bath or feedback system	**Yes**, by using a thermometer
V	**No**, not easily, thermal expansion of the container is a byproduct of the container being thermally conductive, the volume changes with heat	**Yes**
P	**Yes**, keeping equilibrium with atmosphere	**Yes**, barometer

Table 1: T, P are the easiest quantities to measure and control, making $G(T, P)$ the easiest fundamental relationship to work with.

2   Operations

The Derivative Crusher Algorithm is made of three operations and six relations. Looking at the derivatives of $G(T, P) = E - TS - PV$ more closely, certain constants can be defined:

$$\left(\frac{\partial G}{\partial T}\right)_P = -S \qquad\qquad \left(\frac{\partial^2 G}{\partial T^2}\right)_P = -\left(\frac{\partial S}{\partial T}\right)_P \equiv \frac{C_p}{T} \tag{C1}$$

$$\left(\frac{\partial G}{\partial P}\right)_T = V \qquad\qquad \left(\frac{\partial^2 G}{\partial P^2}\right)_T = \left(\frac{\partial V}{\partial P}\right)_T \equiv \kappa V \tag{C2}$$

$$\left[\frac{\partial}{\partial T}\left(\frac{\partial G}{\partial P}\right)_T\right]_P = \left[\frac{\partial}{\partial P}\left(\frac{\partial G}{\partial T}\right)_P\right]_T = \left(\frac{\partial V}{\partial T}\right)_P \equiv \alpha V \tag{C3}$$

C_p is specific heat at constant pressure.
κ is isothermal compressibility.
α is volume coefficient of expansion.
The derivative crusher algorithm is made of three partial derivative operations:

1. Bringing y to the numerator:

$$\left(\frac{\partial x}{\partial y}\right)_z = \frac{1}{\left(\frac{\partial y}{\partial x}\right)_z} \tag{D1}$$

2. Bringing z to the numerator:

$$\left(\frac{\partial x}{\partial y}\right)_z = -\frac{\left(\frac{\partial z}{\partial y}\right)_x}{\left(\frac{\partial z}{\partial x}\right)_y} \tag{D2}$$

3. Introduce new variable w:

$$\left(\frac{\partial x}{\partial y}\right)_z = \frac{\left(\frac{\partial x}{\partial w}\right)_z}{\left(\frac{\partial y}{\partial w}\right)_z} \tag{D3}$$

The last crucial component are the Maxwell Relations between S, V, T, P:

$$\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial P}{\partial T}\right)_V \tag{M1}$$

$$\left(\frac{\partial V}{\partial S}\right)_P = \left(\frac{\partial T}{\partial P}\right)_S \tag{M2}$$

$$\left(\frac{\partial V}{\partial T}\right)_P = -\left(\frac{\partial S}{\partial P}\right)_T \tag{M3}$$

3   Steps

1. Bring potentials E, F, G, H to the numerator and eliminate them using:

$$dE = TdS - PdV \qquad\qquad\qquad dF = -SdT - PdV$$

$$dG = -SdT + VdP \qquad\qquad\qquad dH = TdS + VdP$$

2. Bring S to numerator using Maxwell Relations depending on what z is in $\left(\dfrac{\partial x}{\partial y}\right)_z$

$$z = T \qquad \boxed{\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial P}{\partial T}\right)_V \quad , \quad \left(\frac{\partial S}{\partial P}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_P}$$

$$z = P, V \qquad \boxed{\left(\frac{\partial S}{\partial T}\right)_P = \frac{C_P}{T} \quad , \quad \left(\frac{\partial S}{\partial T}\right)_V = \frac{C_v}{T}}$$

3. Bring V to numerator in order to get α or κ.

4. Eliminate C_v in favor of C_p, α, κ