Physics 406 Homework

$\mathbb{E}$VAN $\mathbb{C}$ARPENTER

Winter 2022

1 Homework 1

Problem 1.

problem 1

Problem 2.

problem 2

Problem 3.

problem 3

Problem 4.

Suppose that a particle moving in one dimension is confined to $x > 0$, and it's energy is $E = \frac{p^2}{2m} + mgx$ Make a sketch to indicate what region of classical phase space is accessible to this particle if its energy lies between E_0 and $E_0 + \delta E_0$.

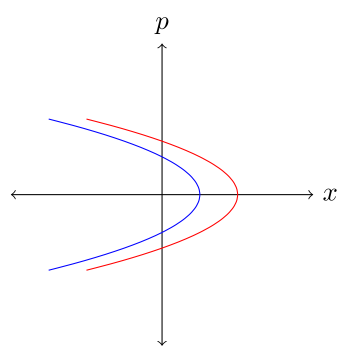

Figure 1: Particle constrained

2 Homework 2

Problem 1.

(a) Show that the number of states $\phi(E)$ with energy less than E, for a particle of mass m in a cubical box of side L is:

$$\phi(E) = \frac{\pi}{6}\left(\frac{L}{\pi\hbar}\right)^3 (2mE)^{3/2}$$

Hint: Use the energy levels 2.1.3 in Reif and treat the n as continuous variables.

(b) Calculate $\Omega(E)$

(c) A nitrogen molecule at room temperature has a typical energy of 6×10^{-14}ergs. Calculate $\phi(E)$ for a particle in a box of side length 10cm. Also calculate $\Omega(E)$ assuming $\delta E = 10^{-24}$ergs

Problem 2. *Reif 2.4*

Consider an isolated system consisting of a large number N of weakly interacting localized particles of spin $\frac{1}{2}$. Each particle has a magnetic moment μ which can point either parallel or antiparallel to an applied field H. The energy od the system is then $E = -(n_1 - n_2)\mu H$, where n_1 is the number of spins aligned parallel to H and n_2 is the number of spins aligned antiparallel to H.

(a) Consider the energy range between $E + \delta E$ where δE is much smaller than E, but E is still microscopically large, so $\mu H \ll \delta E \ll E$. What is $\Omega(E)$ (the total number of states in the energy range)?

(b) Write down an expression for $\ln(\Omega(E))$ as a function of E. Simplify this expression by using Stirling's formula in it's simplest form:
$$\ln(n!) \approx n \ln(n) - n$$

(c) Assume that the energy E is in a region where $\Omega(E)$ is appreciable $\rightarrow$ that it is not close to the extreme possible values $\pm N\mu H$ which it can assume. In this case apply a Gaussian approximation to part (a) to obtain a simple expression for $\Omega(E)$ as a function of E.

Problem 3. *Reif 2.5*

Consider the infinitesimal quantity

$$A(x, y)dx + B(x, y)dy \equiv đF$$

(a) Suppose $đF$ is an exact differential so that $F = F(x, y)$. Show that A, B must satisfy the condition:

$$\frac{\partial A}{\partial y} = \frac{\partial B}{\partial x}$$

(b) If $đF$ is an exact differential, show that the integral $\int đF$ evaluated along any clsoed path on the xy plane must vanish.

Problem 4. *Reif 2.7*

(a) Consider a particle confined to a cubical box. The possible energy levels are given by

$$E = \frac{(\hbar\pi)^2}{2m}\left[\left(\frac{n_x}{L_x}\right)^2 + \left(\frac{n_y}{L_y}\right)^2 + \left(\frac{n_z}{L_z}\right)^2\right]$$

Show that the force exerted by the particle in this state on a wall perpendicular to the x axis is given by

$$F_x = -\frac{\partial E}{\partial L_x}$$

while the length L_x is changed quasi-statically by an amount dL_x.

(b) Calculate explicitly the pressure on this wall. By averaging over all posible states, find an expression for the mean pressure on this wall (Hint: Exploit the property that $\overline{n_x^2} = \overline{n_y^2} = \overline{n_z^2}$ must be true by symmetry.) Show that the mean pressure can be simply expressed in terms of mean energy $\overline{E}$ of the particle and the volume $V = L_x L_y L_z$ of the box.

3 Homework 3

Problem 5. *Reif 2.11*

In a quasi-static process $A \to B$ (Add diagram) in which no heat is exchanged with the environment, the mean pressure $\overline{p}$ of a certain amount of gas is found to change with its volume V according to the relation:

$$\overline{p} = \alpha V^{-5/3}$$

where α is a constant. Find the quasi-static work done and the net heat absorbed by the system in each of the following three processes, all of which take the system from macrostate A to macrostate B.

(a) The system is expanded from its original to its final volume, heat being added to maintain the pressure constant. The volume is then kept constant, and heat is extracted to reduce the pressure to 10^{-6} dynes cm^{-2}.

(b) The volume is increased and heat is supploed to cause the pressure to decrease linearly with the volume.

(c) The two steps in process (a) are performed in the oppisite order.

Problem 6. *Reif 3.2*

Consider a system of N localized weakly-interacting particles, each of spin 1/2 and magnetic moment μ located in an external magnetic field H. [1]

(a) Using the expression for $\ln(\Omega(E))$ calculated in Reif 2.4b and the definition $\beta = \frac{\partial \ln \Omega}{\partial E}$ find the relation between the absolute temperature T and the total energy E of this system.

(b) Under what circumstances is T negative?

(c) The total magnetic moment M of this system is related to its energy E. Use the result of part (a) to find M as a function of H and the absolute temperature T.

[1] This system was already discussed in Reif 2.4

Problem 7. *Reif 3.4*

Suppose a system A is places into thermal contact with a heat reservoir A' which is at an absolute temperature T' and that A absorbs an amount of heat Q in this process. Show that the entropy increase ΔS of A in this process satisfies the inequality

$$\Delta S \geq \frac{Q}{T'}$$

Where the = case is only valid if the initial temperature of A differs infinitesimally from T'.

Problem 8. *Reif 3.5*

A system consists of N_1 molecules of type 1, and N_2 molecules of type 2 confined within a box of volume V. The molecules are supposed to interact very weakly so that they constitute an ideal gas mixture.

(a) How does $\Omega(E)$ (the total number of states between $E, E+\delta E$) depend on V in this system? You may treat the problem clasically.

(b) Use this result to find the equation of the state of this system$\rightarrow$ find the mean pressure $\overline{p}$ as a function of V, T.