

Table 2. Considerations for the FFZ. Fluxes that should be considered for each spatial regime in the FFZ and recommended direction.

Spatial Regime	Fluxes to consider in a comprehensive accounting	Recommended near-term path forward (minimum viable product)
Lower vadose zone and groundwater flow paths	 All the fluxes in the NFZ also apply in this region. It will be important to understand and identify permanent alkalinity sinks and changes to counterfactual alkalinity generation. Exchange with the atmosphere is limited or nonexistent. 	Largely a priority area for research and development: Not currently feasible to require monitoring or modeling of processes in the deep vadose zone and along groundwater flow paths in commercial ERW deployments. Undertaking deep vadose zone monitoring or installing groundwater monitoring wells in a select subset of commercial deployments, where applicable and feasible, is recommended.
Surface water systems	 Outgassing caused by DIC system equilibration. Carbonate mineral burial. Changes to organic matter respiration and metabolic activity in stream/river systems, including the influence on aquatic vegetation that directly takes up bicarbonate. Authigenic clay formation and reverse weathering. Bedrock interaction with solution and changes to counterfactual alkalinity generation. Changes to nutrient export and nitrogen cycling. 	Focus primarily on: Outgassing caused by DIC system equilibration. Carbonate mineral precipitation (assume it doesn't re-dissolve) It is recommended that practitioners assess the potential for net CO ₂ loss due to these processes at a minimum in two spatial regimes: in the immediate discharge zone (i.e., the surface water system the weathering flux from a deployment is draining into), and in the major river system of the deployment catchment (i.e., the highest order segment in the deployment catchment).
Ocean	 CO₂ evasion due to <u>carbonic acid</u> <u>system equilibration</u> in the coastal ocean. Carbon and alkalinity feedbacks due to changes to surface ocean chemistry. Carbon and alkalinity feedbacks due to changes to diagenetic reactions in shallow marine sediments. Long timescale marine carbonate burial. 	Focus primarily on: CO ₂ evasion due to <u>carbonic acid</u> <u>system equilibration</u> in the coastal ocean. At a minimum, practitioners should derive a conservative assumption of evasion from carbonic acid system equilibration by considering the thermodynamic storage efficiency as a worst-case scenario, assuming complete equilibration with atmospheric CO ₂ at representative temperature, salinity, and current atmospheric pCO ₂ .