PART 5 Biochemical Reaction Systems

This part is highly similar to Part I. Though the kinetic is a little bit different, it is still homogeneous. All you need is to pay attention to something unique in this part, like washout.

Chap 27 Enzyme Fermetation

M-M Kinetics

Enzyme Fermentation $A \xrightarrow{E(\text{Enzyme})} R \qquad \boxed{-r_A = r_R = k \frac{C_{E0}C_A}{C_M + C_A}} \leftarrow \text{M-M Kinetic}$

k: Rate constant $C_{\text{\tiny EO}}$: Michaelis constant $C_{\text{\tiny EO}}$: total enzyme

Analysis: ① When $C_A \gg C_M$ $-r_A = r_R \approx k \frac{C_{E0}C_A}{C_A} = kC_{E0}$ zero-order

② When $C_{\rm A} \ll C_{\rm M}$ $-r_{\rm A} = r_{\rm R} \approx k \frac{C_{\rm E0}}{C_{\rm M}} C_{\rm A}$ first-order

Mechanism: $A + E = \frac{1}{2} \times X = R + E$... and ... $C_{E0} = C_E + C_X$ intermediate to reactant

	Batch or Plug Flow Fermentor	Mixed Flow Fermentor
Integrated Form	$C_{\rm M} \ln \frac{C_{\rm A0}}{C_{\rm A}} + (C_{\rm A0} - C_{\rm A}) = k_3 C_{\rm E0} t$	$k_3 C_{\text{E0}} \tau = \frac{(C_{\text{A0}} - C_{\text{A}})(C_{\text{M}} + C_{\text{A}})}{C_{\text{A}}}$
Linear Form	$\frac{C_{A0} - C_{A}}{\ln \frac{C_{A0}}{C_{A}}} = -C_{M} + k_{3}C_{E0} \frac{t}{\ln \frac{C_{A0}}{C_{A}}}$	$C_{A} = -C_{M} + \frac{k_{3}C_{E0}C_{A}\tau}{C_{A0} - C_{A}}$

二 Inhibition

Competitive Inhibition	Noncompetitive Inhibition
A B B	Note: when B is on the surface it will not let A enter or leave
$A + E \stackrel{1}{\rightleftharpoons} X \stackrel{3}{\rightleftharpoons} R + E \qquad B + E \stackrel{4}{\rightleftharpoons} Y$	$A + E \underset{2}{\overset{1}{\rightleftharpoons}} X \xrightarrow{3} R + E \qquad B + E \underset{5}{\overset{4}{\rightleftharpoons}} Y \qquad B + X \underset{7}{\overset{6}{\rightleftharpoons}} Z$
$r_{\rm R} = \frac{k_{\rm 3}C_{\rm E0}C_{\rm A}}{C_{\rm M}(1 + NC_{\rm B0}) + C_{\rm A}}$	$r_{\rm R} = \frac{\frac{k_{\rm 3}}{1 + LC_{\rm B0}} C_{\rm E0} C_{\rm A}}{C_{\rm M} (\frac{1 + NC_{\rm B0}}{1 + LC_{\rm B0}}) + C_{\rm A}}$
$C_{\rm M} = \frac{k_2 + k_3}{k_1}$	$N = \frac{k_4}{k_5} \qquad L = \frac{k_6}{k_7}$
Competitive $C_{BO} = 0 \text{Low} \text{High } C_{BO}$ Only this line from Fig. 6 $C_{A} C_{M} C_{M} C_{M}(1 + NC_{BO})$ $C_{M} k_{3} C_{M}(1 + NC_{BO})$	Noncompetitive $C_{BO} = 0 \qquad \text{Low} \qquad \text{High } C_{BO}$ Only this line from Fig. 6 $Slope = \frac{k_3}{1 + LC_{BO}}$ $C_{EO}C_A\tau$ $C_{AO} - C_A$ $C_{CO}C_A\tau$

Chap 28 Microbial Fermentation

Microbial Fermentation

$$A \xrightarrow{C} C + R$$

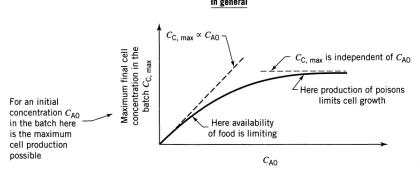
A: food C: Microbes R: Waste material (can be product)

When cell grow in a friendly constant environment with food C_A :

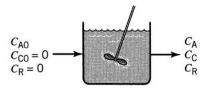
Monod Equation:

$$r_{\rm C} = \frac{kC_{\rm A}C_{\rm C}}{C_{\rm A} + C_{\rm M}}$$

In batch the cells will start dying after a exponential growth for lack of food or too much poisonous waste.



Thus we usually use MFR to operate at optimum.



☐ Fractional Yield

Relationship

$$\begin{array}{c}
R/A = R/C \cdot C/A \\
R/A = 1/C/A
\end{array}$$

$$\begin{array}{c}
r_C = (-r_A) C/A \\
r_R = (-r_A) R/A \\
r_R = (r_C) R/C
\end{array}$$

Suppose all φ values remain constant at all composition :

$$\begin{split} & C_{\rm C} - C_{\rm C0} = \textcircled{C/A} \, \left(C_{\rm A0} - C_{\rm A} \right) \quad ... \text{or} ... \quad C_{\rm C} = C_{\rm C0} + \textcircled{C/A} \, \left(C_{\rm A0} - C_{\rm A} \right) \\ & C_{\rm R} - C_{\rm R0} = \textcircled{R/A} \, \left(C_{\rm A0} - C_{\rm A} \right) \quad ... \text{or} ... \quad C_{\rm R} = C_{\rm R0} + \textcircled{R/A} \, \left(C_{\rm A0} - C_{\rm A} \right) \\ & C_{\rm R} - C_{\rm R0} = \textcircled{R/C} \, \left(C_{\rm C} - C_{\rm C0} \right) \quad ... \text{or} ... \quad C_{\rm R} = C_{\rm R0} + \textcircled{R/C} \, \left(C_{\rm C} - C_{\rm C0} \right) \end{split}$$

Chap 29 Substrate-Limiting Microbial Fermentation

Kinetics and Max Rate

There is no slowing of rate so the kinetics would be monod equation

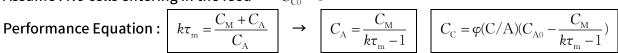
$$r_{\rm C} = \varphi({\rm C/A})(-r_{\rm A}) = \frac{kC_{\rm A}C_{\rm C}}{C_{\rm A} + C_{\rm M}}$$

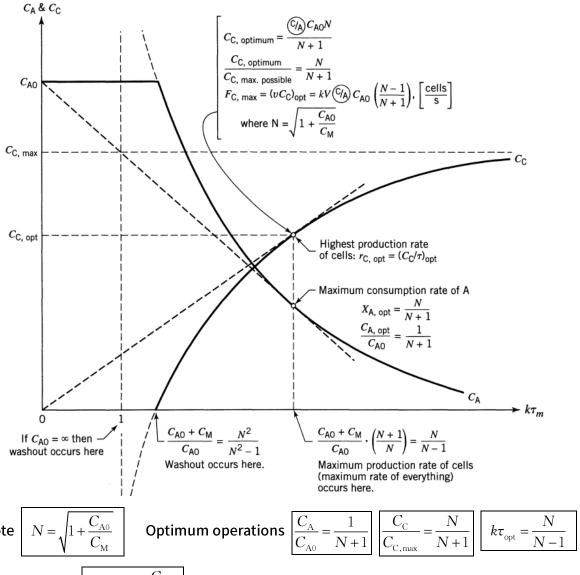
$$C_{\rm C} - C_{\rm C0} = \varphi({\rm C/A})(C_{\rm A0} - C_{\rm A})$$

When $C_A = \sqrt{C_M^2 + C_M(C_{A0} + \varphi(A/C)C_{C0})} - C_M$, r_C reaches to its maximum

Mixed Flow Fermentor **

Assume : No cells entering in the feed $\rightarrow C_{\text{CO}} = 0$





Note

There must be $k\tau_{\rm m} > 1 + \frac{C_{\rm M}}{C_{\rm AO}}$ otherwise washout would occur

↑ the microbes in fermentor start to decrease

Chap 30 Product-Limiting Microbial Fermentation

Kinetics

In many cases, microbes' waste products like alcohol inhibit their growth So the monod equation need to be extended:

$$r_{\rm C} = \varphi({\rm C/R})r_{\rm R} = k(1 - \frac{C_{\rm R}}{C_{\rm R}^*})^n \frac{C_{\rm A}C_{\rm C}}{C_{\rm A} + C_{\rm M}}$$

 $C_{\rm R}^*$: When $C_{\rm R}$ reaches it, reactions stop

We focus on special case of sufficient food $C_{\rm A}\gg C_{\rm M}$, $C_{\rm R0}=C_{\rm C0}=0$ and n=1

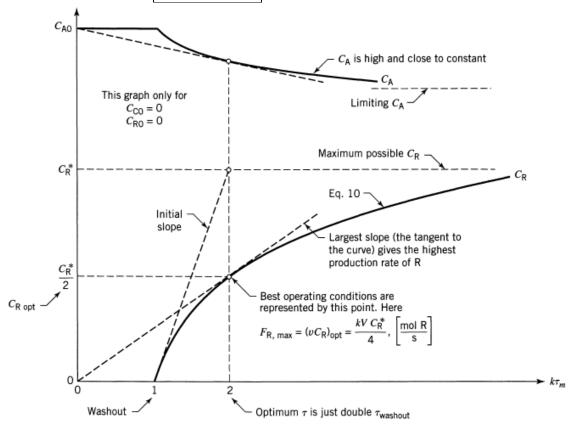
$$r_{\rm C} = \varphi({\rm C/R})r_{\rm R} = k(1-\frac{C_{\rm R}}{C_{\rm R}^*})C_{\rm C} \qquad r_{\rm R} = k(1-\frac{C_{\rm R}}{C_{\rm R}^*})C_{\rm R}$$
 So $r_{\rm R}$ reach maximum when
$$\boxed{C_{\rm R,maxrate} = \frac{1}{2}C_{\rm R}^*}$$

$$C_{\rm R, maxrate} = \frac{1}{2} C_{\rm R}^*$$

MFR For n = 1 %

Performance Equation: $k\tau_{\rm m} = \frac{C_{\rm R}^*}{C_{\rm p}^* - C_{\rm p}}$

$$k\tau_{\rm m} = \frac{C_{\rm R}^*}{C_{\rm R}^* - C_{\rm R}}$$



- Washout occurs when $k\tau_{\rm m} = 1$
- · Maximum production rate obtained at $k\tau_{\rm m}=2$ when $C_{\rm R}=C_{\rm R}^*/2$
- Maximum production $F_{
 m R,max} = \varphi({
 m R/C})F_{
 m C,max} = kVC_{
 m R}^* \ / \ 4$