Corrosion humide des métaux

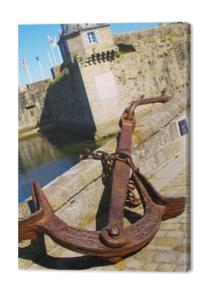
Agrégation 2020

La corrosion dans notre quotidien

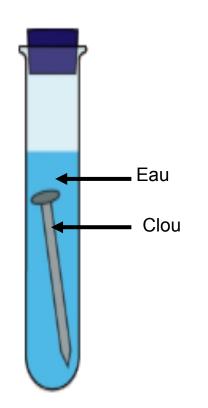


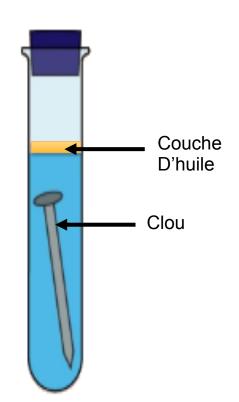


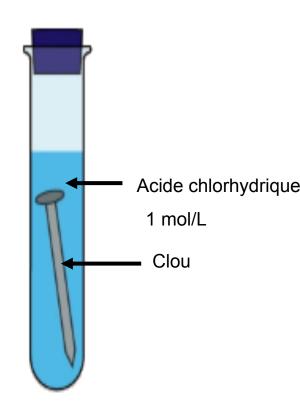




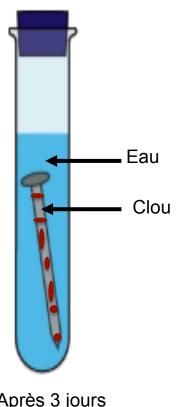
Corrosion du fer

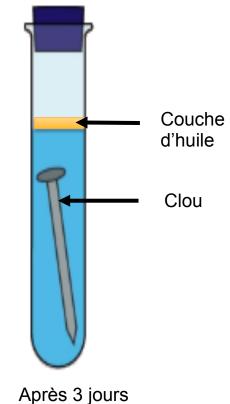


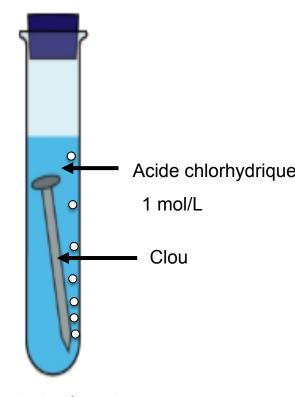




Corrosion du fer







Après 3 jours Après 3 jour

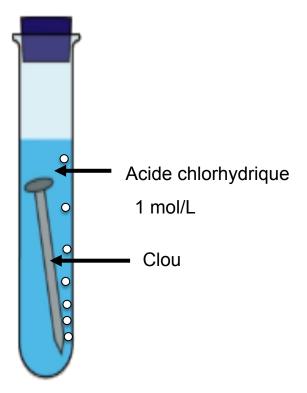
Instantanément

Corrosion uniforme d'un clou en fer



Photo d'un clou rouillé

Caractérisation des produits formés



- Combustion du gaz formé
- Ajout de quelques gouttes d'une solution d'hexacyanatoferrate (III) de potassium.

Diagramme E-pH du fer superposé au diagramme de l'eau

Convention de tracé:

c=1.10⁻⁶ mol.L⁻¹

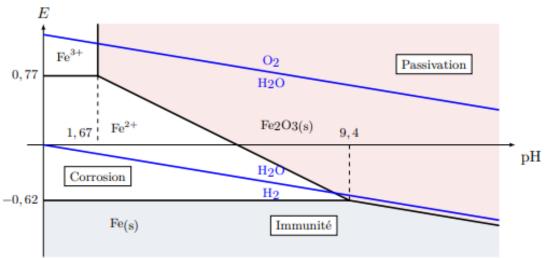
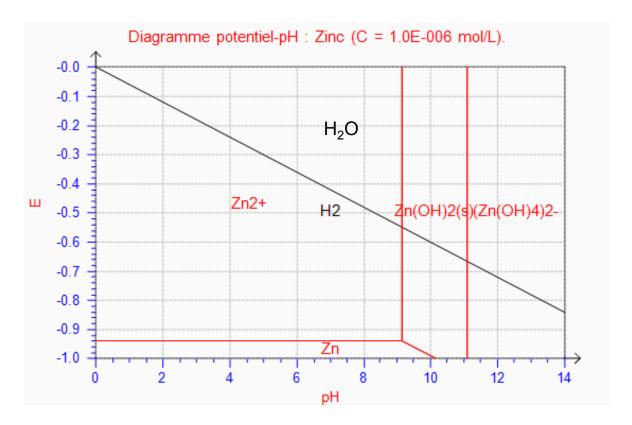
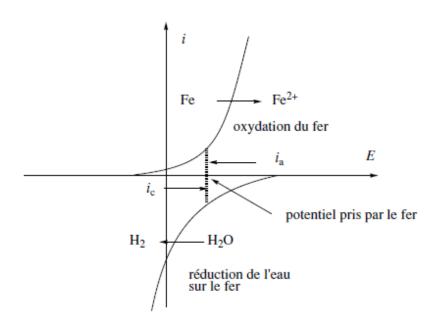


Diagramme E-pH du zinc superposé à celui de l'eau

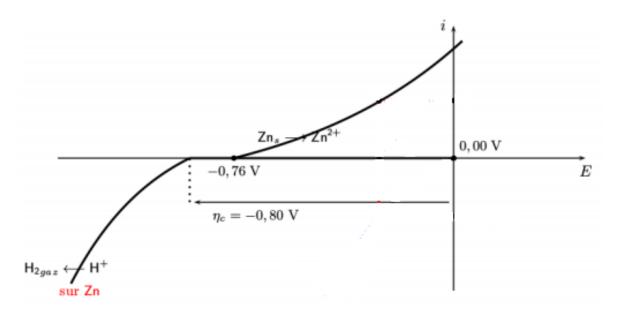
Convention de tracé : c=1.10⁻⁶ mol.L⁻¹



Courbe courant-potentiel



Courbe courant-potentiel : blocage cinétique

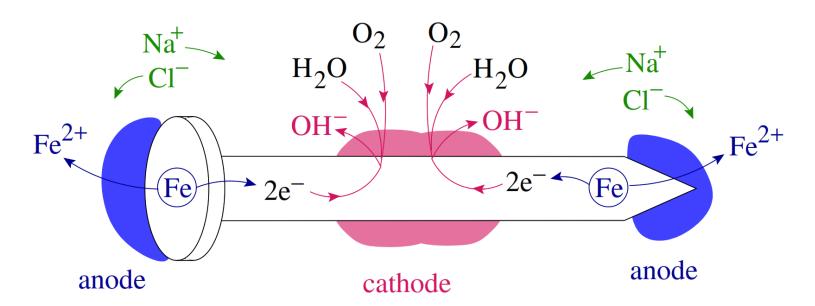


Corrosion différentielle



<u>Clou placé dans un gel agar-agar auquel est ajouté de la phénophtaléine et l'hexacyanatoferrate (III) de potassium.</u>

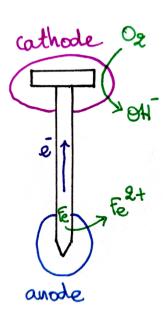
Corrosion différentielle : structure du clou



Aération différentielle

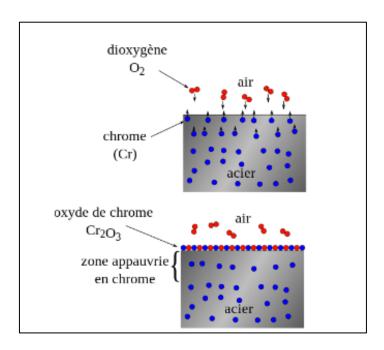


Photo de l'expérience

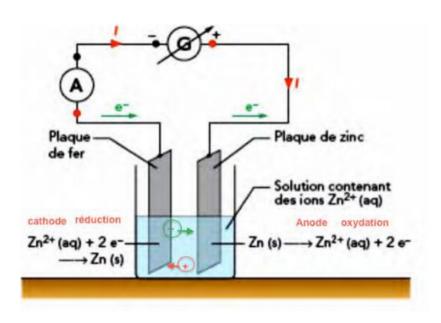


<u>Schéma</u>

Passivation d'un acier inoxydable

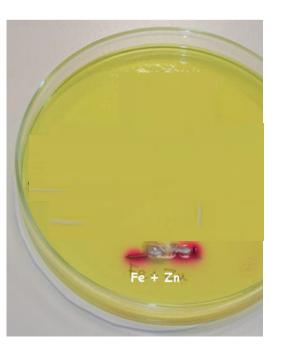


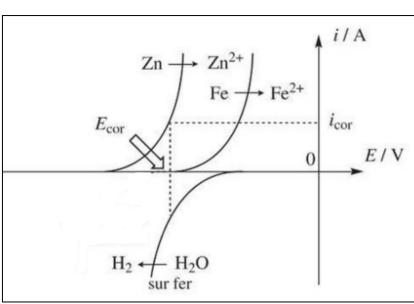
Protection du fer : par électro zingage



Protection par anode sacrificielle

Chimie Tout-en-un MP. Fosset et al. Dunod. 2004





Expérience au laboratoire

Anodes sacrificielles sur un bateau

Courbes i-E