Molecular free energy profiles from force spectroscopy experiments by inversion of observed committors ♀⊘

Special Collection: JCP Emerging Investigators Special Collection

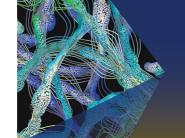
Roberto Covino [0]; Michael T. Woodside; Gerhard Hummer [0]; Attila Szabo; Pilar Cossio [0]



J. Chem. Phys. 151, 154115 (2019) https://doi.org/10.1063/1.5118362







The Journal of Chemical Physics

Special Topic:

Machine Learning for Biomolecular Modeling

Guest Editors: Pratyush Tiwary, Francesca Grisoni, Pilar Cossio

Submit Today!





Molecular free energy profiles from force spectroscopy experiments by inversion of observed committors •

Cite as: J. Chem. Phys. 151, 154115 (2019); doi: 10.1063/1.5118362 Submitted: 3 July 2019 • Accepted: 29 September 2019 • Published Online: 21 October 2019









Roberto Covino, 1,a) D Michael T. Woodside, Gerhard Hummer, 1,3 D Attila Szabo, 4 and Pilar Cossio 1,5,b) D







AFFILIATIONS

- Department of Theoretical Biophysics, Max Planck Institute of Biophysics, 60438 Frankfurt am Main, Germany
- ²Department of Physics, University of Alberta, Edmonton, Alberta T6G 2E1, Canada
- Institute of Biophysics, Goethe University Frankfurt, 60438 Frankfurt am Main, Germany
- Laboratory of Chemical Physics, National Institute of Diabetes and Digestive and Kidney Diseases, National Institutes of Health, Bethesda, Maryland 20892-0520, USA

Note: This paper is part of the JCP Emerging Investigators Special Collection.

a) Electronic mail: roberto.covino@biophys.mpg.de

b) Electronic mail: pilar.cossio@biophys.mpg.de

ABSTRACT

In single-molecule force spectroscopy experiments, a biomolecule is attached to a force probe via polymer linkers and the total extension of the molecule plus apparatus is monitored as a function of time. In a typical unfolding experiment at constant force, the total extension jumps between two values that correspond to the folded and unfolded states of the molecule. For several biomolecular systems, the committor, which is the probability to fold starting from a given extension, has been used to extract the molecular activation barrier (a technique known as "committor inversion"). In this work, we study the influence of the force probe, which is much larger than the molecule being measured, on the activation barrier obtained by committor inversion. We use a two-dimensional framework in which the diffusion coefficient of the molecule and of the pulling device can differ. We systematically study the free energy profile along the total extension obtained from the committor by numerically solving the Onsager equation and using Brownian dynamics simulations. We analyze the dependence of the extracted barrier on the linker stiffness, molecular barrier height, and diffusion anisotropy and, thus, establish the range of validity of committor inversion. Along the way, we showcase the committor of 2-dimensional diffusive models and illustrate how it is affected by barrier asymmetry and diffusion anisotropy.

Published under license by AIP Publishing. https://doi.org/10.1063/1.5118362

I. INTRODUCTION

In single-molecule pulling experiments, mechanical force is used to induce conformational transitions in biomolecules. 1,2 Suppose that the molecule of interest undergoes repeated folding and unfolding transitions under constant force. The molecular extension, i.e., the end-to-end distance of the molecule, would then jump between smaller and larger values. The interpretation of the resulting time series would be simple if the molecular extension could be directly monitored experimentally. In this hypothetical case,

the folding and unfolding force-dependent transition rates of the molecule could be directly obtained by counting the number of transitions per unit time. In addition, by binning this trajectory, one could determine the probability density of the extension, the logarithm of which is the free energy profile of the molecule, a procedure known as Boltzmann inversion. Alternatively, from the trajectory, one could determine the probability that a molecule with a specific extension folds before it unfolds. This quantity describes the most probable "fate" of the system at any given point of the trajectory and is known as committor, splitting probability, or p_{fold} . Assuming that

⁵Biophysics of Tropical Diseases Max Planck Tandem Group, University of Antioquia, Medellin, Colombia

the dynamics is diffusive, the height and shape of the free energy barrier could be found by differentiation of the committor, a procedure known as committor inversion. $^{4.5}$

In reality, one cannot directly monitor the molecular extension itself because the experimental observable is actually the position of the force probe attached to the molecule by long polymer linkers. In the case of optical trapping measurements [Fig. 1(a)], the measured extension (q) is the extension of the molecule (x)plus that of the linkers attaching the molecule to mesoscopic beads trapped by laser beams. What one measures is the time dependence of the interbead distance, yielding a trajectory of the total extension of molecule and linkers [Fig. 1(b)]. The free energy profile obtained by Boltzmann inversion of the observed trajectory of the total extension is a convolution of the molecule and linker profiles. If the properties of the linkers are known, one can obtain the free energy profile of the molecule by deconvolution.^{6,7} This methodology requires large amounts of data and works best with low molecular barriers, and it can thus be challenging to use in practice.

As a viable alternative, the group of one of us investigated the free energy profile obtained by committor inversion of the measured trajectory.⁵ If the dynamics of the total extension could be described as diffusion of the free energy profile obtained by Boltzmann inversion, then both the committor and Boltzmann inversion would give the same result. Consequently, one would still have to use deconvolution to obtain the molecular profile. However, unless the response of the apparatus is much faster than that of the molecule, the dynamics of the total extension cannot necessarily be described as a one-dimensional diffusive process.⁸ Committor inversion may, therefore, lead to a different free energy profile than Boltzmann inversion. Experimentally, committor inversion has been applied to DNA hairpin folding, successfully recovering the free energy profile obtained from deconvolution of the Boltzmann inverted one.⁵ It has

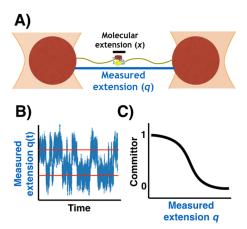


FIG. 1. Schematic of a single-molecule pulling experiment. (a) Example of the experimental setup in force spectroscopy using optical tweezers. A small biomolecule is attached via polymer linkers to two beads trapped by laser beams. The molecular extension x is hidden within the observed extension q. (b) Trajectory of the measured extension q(t) as a function of time. (c) The committor estimated from the observed trajectory of the measured extension q(t) in the interval between the red lines.

also been used to extract free energy barriers encountered when bacteriorhodopsin is pulled out of a membrane. This procedure has the potential to become widely used as a viable alternative to deconvolution of the profile obtained by Boltzmann inversion. However, the range of validity of committor inversion in light of the limitations imposed by probe/linker attachments to the molecule has not been investigated.

Committor inversion yields the exact molecular free energy profile in the limit of very stiff polymer linkers (i.e., when the linker force constant is much larger than those of the molecular extrema). However, for such linkers, the free energy profile obtained from Boltzmann inversion is already quite close to the molecular one. Moreover, in this limit, the measured transition or hopping rates become proportional to the diffusion coefficient not of the molecule but of the probe (e.g., mesoscopic beads) attached to the molecule. Onsequently, here, we shall primarily consider soft linkers for which the measured rates are meaningful. As first pointed out by Thirumalai and co-workers, free energy profiles are most easily found using stiff linkers, but reliable estimates of the hopping rates can only be made by using flexible handles.

We will investigate whether transition path theory can aid in the reconstruction of molecular free energy profiles from the information encoded in the committor estimated from observed trajectories. This will be done in the framework of a simple model where the molecular and total extensions diffuse anisotropically on a twodimensional free energy surface. We previously used such surfaces to determine the influence of the mesoscopic pulling device on the observed rates and transition paths. 12,13 Here, we obtain the committor both by analyzing Brownian dynamics trajectories of the total extension—as in experiments—and by numerically solving the Onsager equation.³ Additionally, we derive and validate analytic expressions for the committor obtained in the high-barrier limit. We then investigate how the extracted barriers depend on the stiffness of the linker, the shape of the molecular free energy profile, and the diffusion anisotropy. We find that although in some realistic cases this procedure yields useful estimates of the heights of molecular barriers, it is challenging to establish its validity in many other cases of practical interest.

II. THEORY

Let x be the molecular (hidden) extension and q be the total (observable) extension [Fig. 1(a)]. Let a constant force be exerted on the system so that the resulting free energy surface has the form

$$G(q,x) = G_o(x) + \frac{\kappa_l}{2}(x-q)^2. \tag{1}$$

Here, the first term on the rhs is the molecular free energy in the presence of force and the second describes the coupling due to a harmonic linker with spring constant κ_l . For the sake of simplicity, we will assume the constant force to be subsumed in $G_o(x)$, which is symmetric about its maximum at x=0 and has two minima, corresponding to metastable states, at $x_1=-x_0$ and $x_2=x_0$ (Fig. 2). It is straightforward to generalize the results presented below to an asymmetric $G_o(x)$ and to anharmonic (e.g., wormlike chain) linkers, albeit at the expense of complicating the analytical expressions.

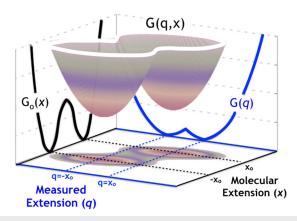


FIG. 2. Two-dimensional potential surface G(q, x). We assume that $G_o(x)$ (black solid line) is symmetric about its maximum at x = 0 and has minima at $-x_0$ and x_0 . The potential of mean force along q, G(q), is shown as the blue solid line. The extensions are shifted by constants so that the barrier occurs at zero.

We assume that the dynamics on the surface in Eq. (1) is diffusive with position-independent diffusion coefficients D_x and D_q along the x and q coordinates, respectively. The value of D_q is essentially determined by the Stokes-Einstein diffusion coefficient of the beads in a laser tweezer experiment and for large beads may thus be slower than D_x . By simulating Brownian dynamics, one can obtain long trajectories describing the evolution of the system on the twodimensional surface in Eq. (1). The system will spend most of the time in one of the two metastable states, rarely but rapidly jumping from one to the other. We can now mimic the typical situation of force spectroscopy experiments and assume that only the component q(t) of the simulated trajectory is observable [see Fig. 1(b) for an example trajectory]. From such trajectories, one can then calculate the "observed" committor $\phi(q)$ [Fig. 1(c)], defined as the probability of reaching the folded minimum before the unfolded minimum, starting from a given value of q.

Alternatively, one can obtain the exact $\phi(q)$ as a conditional equilibrium average of the two-dimensional committor, $\phi(q, x)$, which can be accurately obtained by solving the two-dimensional Onsager equation³ on a grid with the appropriate boundary conditions (see Sec. III). Specifically, the observed committor $\phi(q)$ is given by

$$\phi(q) = \frac{\int_{-\infty}^{\infty} dx \, \phi(q, x) e^{-\beta G(q, x)}}{\int_{-\infty}^{\infty} dx \, e^{-\beta G(q, x)}},\tag{2}$$

where $\beta = 1/k_BT$, k_B is Boltzmann's constant, and T is the absolute temperature. The denominator in Eq. (2) is the exponential of the negative of the free energy profile G(q) along q, given within a

$$e^{-\beta G(q)} = \int_{-\infty}^{\infty} dx \, e^{-\beta G(q,x)} = \int_{-\infty}^{\infty} dx \, e^{-\beta (G_0(x) + \kappa_l(q-x)^2/2)}.$$
 (3)

G(q) can be obtained from the observed trajectory by Boltzmann inversion. If the linker spring-constant is known, then $G_o(x)$ can, in principle, be obtained from G(q) by deconvolution, which amounts to a numerically challenging inverse Weierstrass transform.

For a one-dimensional diffusive process on $G_0(x)$ with position-independent diffusion coefficient, the committor $\phi_o(x)$ is given by

$$\phi_o(x) = \frac{\int_x^{x_o} dy \, e^{\beta G_o(y)}}{\int_{-x_o}^{x_o} dy \, e^{\beta G_o(y)}}.$$
 (4)

Thus, by differentiating both sides with respect to x, one can obtain the following inversion formula:

$$\beta[G_o(x) - G_o(x_0)] = \ln\left[\frac{\phi'_o(x)}{\phi'_o(x_0)}\right],\tag{5}$$

which expresses the molecular free energy profile in the interval $-x_0 \le x \le x_0$ in terms of the derivatives of the committor, denoted as primes. Note that $\phi_o(-x_0) = 1$ and $\phi_o(x_0) = 0$ by definition.

For multidimensional diffusive dynamics, there is no analytic relation between the free energy surface and the committor analogous to Eq. (5). Nevertheless, one can formally use this relation to define a new free energy profile $G_{CI}(q)$ (CI = committor inversion) using the committor $\phi(q)$ obtained from the experimental trajectory in the interval $-x_o \le q \le x_o$,

$$\beta[G_{CI}(q) - G_{CI}(q_0)] = \ln\left[\frac{\phi'(q)}{\phi'(q_0)}\right].$$
 (6)

Since the dynamics along q cannot be, in general, described by onedimensional diffusion, 8,12 then in general, $^4G_{CI}(q) \neq G(q)$. In other words, the committor-inverted and Boltzmann-inverted profiles are not necessarily the same. It has been conjectured that, in fact, $G_{CI}(q)$ is very similar to the hidden molecular free energy profile G_0 in the barrier region. This assumption does not have any obvious theoretical justification, and in the following, we will systematically explore its validity.

We shall now derive approximate analytic expressions for $\phi(q)$ and $G_{CI}(q)$ when the molecular free energy is symmetric and has a high barrier. We begin with the calculation of G(q). When the barrier of $G_o(x)$ is high, the major contribution to the integral in Eq. (3) comes when x is near to the minima of $G_o(x)$, which are located at $x = \pm x_0$. Thus, one can approximate the integral from $-\infty$ to ∞ as a sum of two integrals, one around $x_1 = -x_0$ and the other around $x_2 = x_0$. Then, we expand G(q, x) in Eq. (2) around x_i (i = 1, 2) to the second order as $G_i(q, x) \approx G(q, x_i) + (x - x_i)G'(q, x_i)$ + $(x - x_i)^2 G''(q, x_i)/2$, where the primes denote derivatives with respect to x. By extending the range of integration in both integrals to $(-\infty, \infty)$ and evaluating the resulting Gaussian integrals, we find (to within a constant)

$$e^{-\beta G(q)} = e^{-\beta \kappa (q + x_o)^2/2} + e^{-\beta \kappa (q - x_o)^2/2},$$
 (7)

where $1/\kappa = 1/\kappa_l + 1/G''_o(x_o)$ and $G''_o(x_o) = G''_o(-x_o) > 0$. If we choose the constant in the definition of G(q) so that $G(q = \pm x_0) = 0$,

$$\beta G(q) = \ln \left[\frac{\cosh(\beta \kappa x_o^2)}{\cosh(\beta \kappa x_o q)} \right] + \frac{\beta}{2} \kappa (q^2 - x_o^2). \tag{8}$$

Let us now evaluate the integral in Eq. (2) that determines $\phi(q)$ in an analogous way. We break the integral into two parts, one around

 $-x_0$ and the other around x_0 , and expand G(q, x) about these points to the second order as before. We then approximate the committor around $-x_0$ as $\phi(q, x) = 1$ and set $\phi(q, x) = 0$ in the integral around x_0 . Evaluating the resulting Gaussian integrals, we find

$$\phi(q) = \frac{1}{1 + e^{2\beta\kappa x_o q}} \tag{9}$$

for $-x_0 \le q \le x_0$. This approximate expression is valid for high molecular barriers. In this regime, $\phi(q)$ does not depend on the diffusion anisotropy and, more importantly, it has no direct dependence on the molecular barrier height or shape (although indirect effects from correlations between the well curvature and barrier height may

Using Eqs. (9) and (6) and requiring that $G_{CI}(q = \pm x_o) = 0$, we find that

$$\beta G_{\text{CI}}(q) = 2 \ln \left[\frac{\cosh(\beta \kappa x_o^2)}{\cosh(\beta \kappa x_o q)} \right], \tag{10}$$

which using Eq. (8) for G(q) can be rewritten as

$$G_{\rm CI}(q) = 2G(q) + \kappa (x_o^2 - q^2).$$
 (11)

These approximate expressions are valid for sufficiently large molecular barriers. In this regime, $G_{CI}(q)$ contains no explicit information about the molecular barrier height and shape, as determined by $G_0(x)$.

III. METHODS

A. Free energy surfaces

We model force spectroscopy experiments at constant force as a diffusive process on the two-dimensional free energy surface G(q, q)x), given by Eq. (1) [Fig. 2], where q and x are the total and molecular extension, respectively, $G_o(x)$ is the molecular free energy in the presence of force, and κ_l is the linker stiffness. We used two analytic forms of the molecular free energy. A symmetric potential is given by a bistable matched-harmonic with $G_o(x) = \Delta G_o^{\ddagger} f(x/x^{\ddagger})$, where

$$f(x) = \begin{cases} -2x^2 & 0 \le |x| \le 1/2\\ 2(|x|-1)^2 - 1 & 1/2 < |x| \end{cases},$$
 (12)

and ΔG_0^{\ddagger} and x^{\ddagger} are the activation barrier and the distance to the transition state, respectively, in the presence of force. An asymmetric potential is given by the negative logarithm of a linear combination of two Gaussian distributions,

$$\beta G_o^{\text{asym}}(x) = -\ln\left(\frac{w}{\sqrt{2\pi s_1^2}}e^{-(x+x_0)^2/2s_1^2} + \frac{1-w}{\sqrt{2\pi s_2^2}}e^{-(x-x_0)^2/2s_2^2}\right), (13)$$

where s_1 , s_2 , and $\pm x_0$ are the Gaussian widths and centers, respectively. In particular, we considered a potential displaying a small barrier by using $s_1 = 0.15$, $s_2 = 1$, and w = 0.4 and a potential displaying a larger barrier by using $s_1 = 0.2$, $s_2 = 0.6$, and w = 0.5. In both cases, the minima are located at $\pm x_0$, with $x_0 = 1.5$.

B. Two-dimensional committor using the Onsager equation

The Onsager equation³ for a n-dimensional diffusive process z

$$\nabla \cdot \mathbf{D}(\mathbf{z}) \exp[-\beta G(\mathbf{z})] \nabla \phi(\mathbf{z}) = 0, \tag{14}$$

where D(z) is a position-dependent diffusion tensor. If we assume a two-dimensional diffusion on the free energy surface G(q,x), with a diagonal and position-independent diffusion tensor, then the Onsager equation becomes

$$-D_x \partial_x \beta G(q, x) \partial_x \phi(q, x) + D_x \partial_x^2 \phi(q, x)$$
$$-D_q \partial_q \beta G(q, x) \partial_q \phi(q, x) + D_q \partial_q^2 \phi(q, x) = 0.$$
(15)

After discretizing this equation, an iterative relaxation method can provide an accurate numerical solution $\phi(q, x)$. We thus consider a mesh on the plane (q, x) such that both continuous variables take N+ 1 and M + 1 discrete values, respectively, $q_i \equiv i\Delta q$ and $x_j \equiv j\Delta x$, with $\Delta q = (q_{\text{max}} - q_{\text{min}})/N$ and $\Delta x = (x_{\text{max}} - x_{\text{min}})/M$. We evaluate the committor on the mesh, $\phi_{ij} \equiv \phi(q_i, x_j)$, by solving the (central) finite difference version of Eq. (15),

$$\left(\frac{2D_{q}}{\Delta q^{2}} + \frac{2D_{x}}{\Delta x^{2}}\right)\phi_{ij} = D_{q}\frac{\phi_{i+1,j} + \phi_{i-1,j}}{\Delta q^{2}} + D_{x}\frac{\phi_{i,j+1} + \phi_{i,j-1}}{\Delta x^{2}} - D_{q}\partial_{q}\beta G_{ij}\frac{\phi_{i+1,j} - \phi_{i-1,j}}{2\Delta q} - D_{x}\partial_{x}\beta G_{ij}\frac{\phi_{i,j+1} - \phi_{i,j-1}}{2\Delta x}, \tag{16}$$

where $\partial_q \beta G_{ij}$ and $\partial_x \beta G_{ij}$ are the gradients of the potential evaluated on the mesh. We set boundary conditions $\phi_{ij} = 1$ for all points $(q < -x_0, x < -x_0)$ and $\phi_{ij} = 0$ for all points $(q > x_0, x > x_0)$. This definition of the boundaries assumes that an experienced practitioner will be able to separate true transitions from mere recrossing events by looking at the entire trajectory. Additionally, we set reflective boundary conditions on all remaining points on the border of the mesh, i.e., $\phi_{i,j} = \phi_{i+1,j}$ for i = 0 or i = N - 1 and $\phi_{i,j} = \phi_{i,j+1}$ for j = 0 or j = M - 1. We then solve Eq. (16) iteratively by initially setting all $\phi_{i,j}$ on the rhs of the equation inside the boundaries equal to 1/2. Figure 3(a) shows an example of the numerical solution $\phi(q, x)$ of Eq. (16).

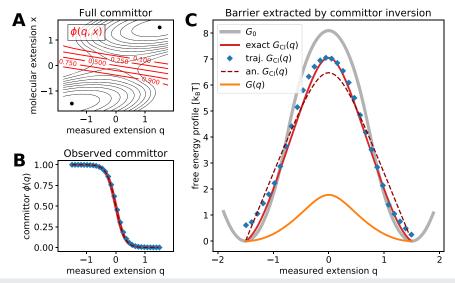
C. Brownian dynamics simulations

We generated trajectories along q and x using

$$q(t + \Delta t) = -\beta \partial_q G(q, x) D_q \Delta t + (2D_q \Delta t)^{1/2} R_q(t),$$

$$x(t + \Delta t) = -\beta \partial_x G(q, x) D_x \Delta t + (2D_x \Delta t)^{1/2} R_x(t),$$
(17)

where $R_q(t)$, $R_x(t)$ are independent Gaussian random numbers with zero mean and unit variance, respectively, and Δt is the time step. The diffusion coefficient D_x of the molecule is kept constant, and that of the apparatus D_q is varied such that D_q/D_x ranges from 10 to 10^{-2} . We chose the time step such that $D_x \Delta t = 5 \times 10^{-4}$. Figure 1(b) shows an example of the measured extension as a function of time.



D. Estimating the committor from diffusive trajectories

To calculate the committor directly from a trajectory, we followed the procedure described by Chodera and Pande.⁴ For a trajectory of duration τ , the committor is estimated by

$$\phi_{\text{traj}}(q) = \frac{\int_0^{\tau} dt \, \delta(q - q(t)) c(t)}{\int_0^{\tau} dt \, \delta(q - q(t))}, \tag{18}$$

where the hitting function c(t) keeps track of whether q(t) hits the folded state before the unfolded one immediately following time t and assumes a value of unity if so, and zero otherwise. This implies that c(t) uses q only and does not make use of any indirect information about the hidden variable x. In practice, we discretized the trajectory q(t) in space and time and considered the resulting discrete chain j(k), where $k=0,\ldots,N$ is the time index and $j=0,\ldots,M$ labels the bins along the extension q. The discretized committor estimated from the trajectory is, therefore, given by

$$\phi_{\text{traj}}(i) = \frac{\sum_{k=0}^{N} \delta_{ij(k)} c(k)}{\sum_{k=0}^{N} \delta_{ij(k)}},$$
(19)

where δ_{ij} is the Kronecker delta. Thus, for each bin i (along q), $\phi_{\rm traj}(i)$ is the ratio between the population committed to the folded state and the total population. In order to calculate $G_{\rm CI}(q)$ with Eq. (6), we discretized the observed trajectory in 30 bins between $q=-x_0$ and $q=x_0$, numerically evaluated the gradient, and smoothened it with a Savitzky–Golay filter.

E. Code

We generated, analyzed, and visualized data with custom code based on Numpy, ¹⁴ Scipy, ¹⁵ Ipython, ¹⁶ Numba, ¹⁷ and Matplotlib. ¹⁸

IV. RESULTS AND DISCUSSION

We first verified that the observed committor $\phi(q)$ is an equilibrium conditional average of the full two-dimensional committor $\phi(q,x)$. In order to model a typical force spectroscopy experiment, we performed Brownian dynamics simulations on the two-dimensional potential G(x, q) with $D_q/D_x = 1$ [see Eq. (2)]. For $G_0(x)$, we used the matched-harmonic potential of Eq. (12) and parameters similar to those experimentally obtained for the 20TS06/T4 DNA hairpin.¹⁹ We estimated the observed committor by using Eq. (19) from the Brownian dynamics trajectories, which contained 54 transitions between the minima $q = -x_0$ and $q = x_0$. Following a completely independent route, we calculated $\phi(q, x)$ by numerically solving the Onsager equation [Eqs. (15) and (16), Fig. 3(a)] and obtained $\phi(q)$ as the conditional average in Eq. (2). Figure 3(b) shows these two independent ways to estimate $\phi(q)$ and compares them to the analytic prediction from Eq. (9) (dashed line). We find that the results from the Brownian dynamics simulations, accurate numerical calculations, and the analytic prediction are in excellent agreement.

We used Eq. (6) to invert the mean committor $\phi(q)$ and extracted a free energy profile $G_{\text{CI}}(q)$, both from the accurate numerical solution and the one estimated from simulated trajectories.

Figure 3(c) reports the results for the 20TS06/T4 DNA hairpin parameters and shows a good agreement between the two independent ways for extracting $G_{CI}(q)$. We compared these results to the potential obtained from Boltzmann inversion G(q) without deconvolution (orange solid line). As reported in Ref. 5, the Boltzmann profile has a much lower barrier and the two profiles differ significantly. This difference indicates that the dynamics along the observed extension q cannot be described as one-dimensional diffusion on G(q).^{4,8} Interestingly, $G_{CI}(q)$ is very similar to the hidden molecular profile G_0 (gray solid line) and the values of their barriers differ only by approximately 10%, consistent with the results found in Ref. 5. However, Fig. 3(c) also shows that $G_{CI}(q)$ is in good agreement with the analytic approximation from Eq. (11) (red dashed line). As shown in Eq. (11), the analytic expression does not explicitly depend on $G_o(x)$ but only on G(q) and the stiffnesses of the molecule and linker. This raises the possibility that the agreement is fortuitous, which motivated us to further assess the validity of the committor inversion to extract the hidden molecular profile for a large number of scenarios.

We investigated how well the free energy profile obtained by inversion of the observed committor, $G_{\rm CI}(q)$, reproduces the hidden molecular profile, G_o , for a number of cases of practical interest. Having shown that the observed committor is accurately reproduced by numerical solutions of Onsager's equation, we used the latter to systematically investigate the influence of parameters of our model. In Fig. 4, we report the dependence of the exact $G_{\rm CI}(q)$ on the linker stiffness (solid lines) and on the height of the hidden molecular barrier. The results show that the accuracy of predicting G_0 depends on all the examined parameters. For instance, using the linker stiffness $\kappa_l = 1.5 \ k_{\rm B} T/[q]^2$, where [x] = [q] indicates the units of extension, guarantees acceptable results for $\Delta G_0^{\dagger} = 3 \ k_{\rm B} T$ but works rather poorly for larger barriers, as can be seen for $\Delta G_0^{\dagger} = 8.1 \ k_{\rm B} T$. For $\Delta G_0^{\dagger} = 16 \ k_{\rm B} T$, only a very stiff linker gives an acceptable reconstruction.

We tested the validity of the analytic approximation [Eq. (11)] (dashed lines in Fig. 4). We found that Eq. (11) reproduces accurately the exact solutions for sufficiently large molecular barriers (≥ 5 $k_{\rm B}T$) and for soft linkers. The lowest barrier considered, 3 $k_{\rm B}T$, is outside the range where the analytic approximation can be applied with confidence. Since the analytic formula does not contain explicit information about the hidden molecular profile (only information about the potential wells), whenever this approximation accurately reproduces $G_{\rm CI}(q)$ the reconstruction of G_0 by committor inversion is likely to be invalid. Notably, in these cases, the barrier height obtained by committor inversion is systematically lower than the molecular barrier.

We then investigated the cases in which the analytic approximation does not correctly reproduce the free energy profile obtained by the exact numerical solution of the committor inversion, even when the barrier seems sufficiently high. This can be seen, for instance, in Fig. 4 for $\Delta G_o^{\dagger} = 8.1 \, k_B T$ and $\kappa_l = 5 \, k_B T / [q]^2$. We compared the exact $\phi(q)$ to the analytic prediction from Eq. (9) (Fig. 5). These two quantities are in striking agreement over most of the reaction coordinate range and deviate only close to the minima by exponentially small amounts, which causes the slopes, $\phi'(q)$, of the two curves to be different [Fig. 5(a), inset]. These differences in $\phi'(q)$ are amplified by the logarithm in Eq. (6), causing large errors in

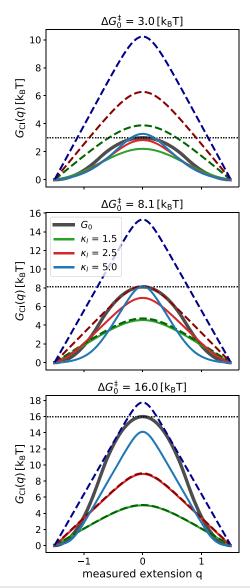


FIG. 4. Free energy barrier extracted by committor inversion, $G_{\text{Cl}}(q)$, using the observed committor (solid lines) and the analytic approximation of Eq. (11) (dashed lines and darker shade of color). We varied the height of the hidden molecular barrier ($\Delta G_o^{\dagger}=3$, 8.1, and 16 $k_{\text{B}}T$, from top to bottom) and the linker stiffness ($\kappa_{\text{I}}=1.5, 2.5, \text{ and } 5~k_{\text{B}}T/[q]^2$, green, red, and blue lines, respectively). We used the matched-harmonic free energy function from Eq. (12) with $\Delta x^{\ddagger}=1.5~[x]$. For reference, each panel shows the respective hidden molecular profile G_o (gray solid line).

the inverted free energy barrier $G_{CI}(q)$ at the well bottom [Fig. 5(b)] that lead to systematic underestimation of the barrier height. In fact, Eq. (9) accurately reproduces the exact $G_{CI}(q)$ around the top of the barrier but poorly describes how $G_{CI}(q)$ approaches its minima. Figure 5 indicates that the mean committor close to the stable states encodes crucial information about the height of the extracted barrier and must be estimated with very high precision. This requirement

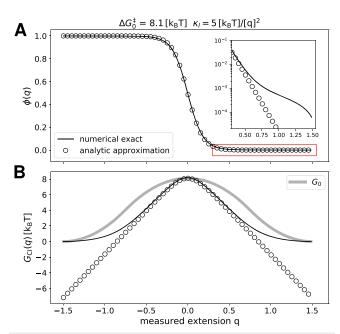


FIG. 5. (a) Observed committor $\phi(q)$ and (b) barrier $G_{\mathrm{CI}}(q)$ obtained by committor inversion for the exact numerical solution (solid line) and the analytic approximation of Eq. (11) (empty circles—to highlight agreement with exact solution). We used parameters $\Delta G_o^{\dagger}=8.1\,k_{\mathrm{B}}T$ and $\kappa_{\mathrm{I}}=5\,k_{\mathrm{B}}T/[q]^2$. The inset is the zoomed-in view of $\phi(q)$ in the range highlighted by the red square box. For reference, G_o is shown in Fig. 5(b) as a gray solid line. All curves in Fig. 5(b) are aligned on the barrier top. A seemingly insignificant difference in the committors shown in Fig. 5(a) leads to an overestimation of the barrier height by a factor of 2.

poses a major challenge for practical attempts to reconstruct barriers by committor inversion.

We also investigated how well committor inversion allows one to estimate the shape of the hidden molecular barrier for asymmetric molecular energy profiles. In Fig. 6(a), we show the twodimensional free energy surface G(q, x) and two-dimensional committor for small (left) and large (right) asymmetric barriers. On both free energy surfaces, the well to the left of the barrier is narrower and deeper than that to the right of the barrier. The asymmetry of the barrier is reflected in the full two-dimensional committor. In fact, the committor isoline of 0.5 is not located at the barriertop but displaced toward the shallower state, whereas points on the top of the barrier are actually highly committed. $\phi(q)$ obtained by solving Onsager's equation and the profiles extracted by committor inversion are shown in Figs. 6(b) and 6(c), respectively, for different linker stiffnesses. The comparison to the hidden molecular free energy (gray line) shows that the asymmetry of the molecular free energy can be captured only qualitatively under the conditions used here. Indeed, the accuracy in determining both the location of the barrier top and the relative stability of the two states depends on the stiffness of the linker and improves with increasingly stiffer linkers. For low linker stiffness, the barrier from committor inversion is systematically lower and closer to q = 0 than in G_o^{asym} .

Finally, we studied the effects of diffusion anisotropy on $\phi(q,x)$, $\phi(q)$, and $G_{\text{CI}}(q)$ by numerically obtaining the solutions of

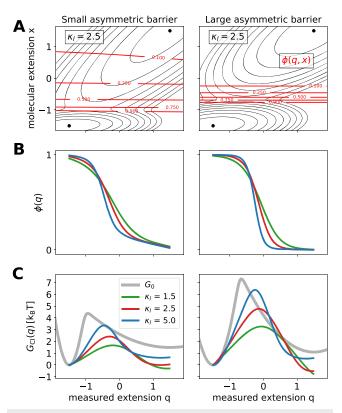


FIG. 6. Asymmetric molecular barriers and free energy profiles extracted by committor inversion. (a) Full two-dimensional committor $\phi(q,x)$ for the free energy surfaces G(q,x) with an asymmetric molecular free energy $G_o^{\rm asym}$ [Eq. (13)], $\kappa_l = 2.5 \ k_B T / [q]^2$, and $D_q / D_x = 1$. Isolines of the free energy surfaces are shown as black solid lines separated by 1 $k_B T$. Isolines of the committor are shown as resolid lines. We considered free energy surfaces with a small and a large barrier, left and right, respectively. (b) The observed committor $\phi(q)$ is obtained by numerically solving Onsager's equation. $\phi(q)$ is shown for different linker stiffnesses (1.5, 2.5, and 5.0 $k_B T / [q]^2$). (c) Free energy profiles $G_C(q)$ extracted by committor inversion. The respective hidden molecular profiles $G_o^{\rm asym}$ are shown as gray solid lines.

Eq. (15) as a function of the ratio D_q/D_x over four orders of magnitude using different molecular barrier heights ΔG_0^{\ddagger} (Fig. 7). For small molecular barriers, reducing D_q (corresponding to slower diffusion of the force probe attached to the molecule) induces a "rotation" of the full committor $\phi(q, x)$ around the barrier [Fig. 7(a)]. For $D_q/D_x = 10$, the isolines of the committor are almost parallel to the q-axis, indicating transitions that are dominated by the dynamics along x [Fig. 7(a), orange lines]. As D_q decreases, the isolines rotate, until they are perpendicular to the *q*-axis for very small D_q , indicating that transitions are dominated by the much slower dynamics along q (blue lines). This phenomenon is most clearly observed for $\Delta G_0^{\ddagger} = 3 k_B T$. For $\Delta G_0^{\ddagger} = 8.1 k_B T$, larger values of diffusion anisotropy are required to induce rotations of the isolines of the full committor, which are mostly suppressed on the barrier. For the largest barrier, $\Delta G_0^{\ddagger} = 16 k_B T$, diffusion anisotropy has no sizable effect on the committor, which is completely determined by the free energy surface.

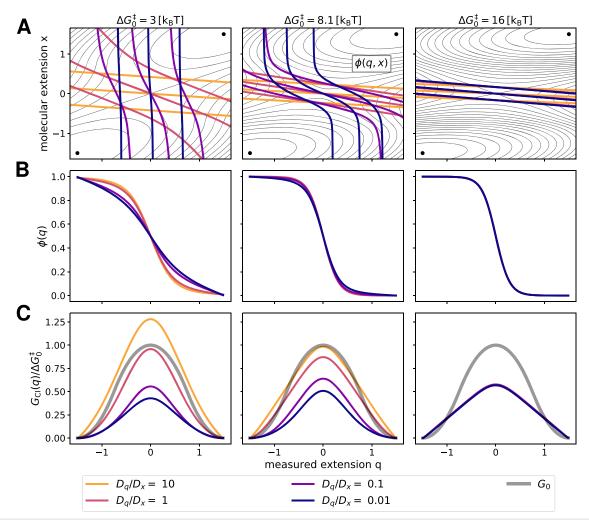


FIG. 7. Effect of diffusion anisotropy on committors and free energy barriers extracted by committor inversion. (a) Free energy surfaces and corresponding full committor $\phi(q,x)$ calculated for different diffusion anisotropies D_q/D_x and increasing values of the molecular barrier height ($\Delta G_o^{\ddagger} = 3$, 8.1, and 16 k_BT , from left to right) with $\kappa_l = 2.6$ $k_BT/[q]^2$ in each case. Isolines of the free energy surfaces are shown as black solid lines separated by 1 k_BT . Isolines of the committor corresponding to 0.2, 0.5, and 0.8 (from top-right to bottom-left) are shown as colored solid lines. (b) Corresponding observed committor $\phi(q)$ as a function of diffusion anisotropy D_q/D_x . (c) Barrier $G_{\text{Cl}}(q)$ obtained by inversion of the observed committors shown in (b). The hidden molecular barrier G_0 is reported as a gray solid line. In each panel of (c), free energies are measured in units of the corresponding value of ΔG_0^{\ddagger} . Color code for diffusion anisotropy D_q/D_x : 10 orange, 1 red, 0.1 purple, and 0.01 blue. In the rightmost panel, the curves for $G_{\text{Cl}}(q)$ are superimposed.

Consequently, diffusion anisotropy has a significant impact on $\phi(q)$ and $G_{\rm CI}(q)$ for low and medium barrier heights, but no effect for very large ones. For such high barriers, the 2-dimensional committor function depends only weakly on diffusion anisotropy because crossing the barrier effectively commits trajectories. For decreasing values of D_q/D_x , the barrier reconstructed by committor inversion tends to increasingly underestimate the hidden molecular barrier ΔG_0^{\ddagger} . We note, however, that the relevant parameters are constrained by the molecular system and the experimental apparatus. This represents a further challenge for practical applications of the committor inversion method to experiments, since the probe diffusion may well be much slower than the molecular diffusion (depending

on the molecule being studied and the design of the probe). As shown already in Fig. 5, small variations in the observed committor arising from diffusion anisotropy can have dramatic effects on the barrier height estimated by inversion [Figs. 7(b) and 7(c), middle panel].

V. CONCLUDING REMARKS

We have assessed the validity of committor inversion to extract molecular free energy profiles from single-molecule force spectroscopy experiments. Within the framework of a two-dimensional model for the coupled dynamics of the molecular and measured extensions, we obtained approximate analytic expressions for the measured committor and the extracted free energy profile from its inversion. We compared these analytic results with those obtained from Brownian dynamics simulations and accurate numerical solutions of the Onsager equation for various linker stiffness values and molecular barrier heights. We found that for isotropic diffusion, the committor inversion gives reasonable results for small and mediumhigh barriers and that the accuracy depends on the stiffness of the linker. When the apparatus diffuses much more slowly than the molecule, or when the barrier is high, the reconstruction is far less accurate. We have also shown that due to the logarithms in the inversion formula, even exponentially small inaccuracies in the observed committor lead to large errors in the barrier height of the reconstructed molecular free energy profile. This may represent a serious challenge for practical applications of the committor inversion approach. Although in some situations molecular free energy profiles estimated by committor inversion from single-molecule experiments can be informative, systematically ascertaining their validity is challenging and they should, in general, be regarded with caution.

ACKNOWLEDGMENTS

R.C., G.H., and P.C. acknowledge the support of the Max Planck Society. P.C. was also supported by Colciencias, University of Antioquia, and Ruta N, Colombia. A.S. was supported by the Intramural Research Program of the National Institute of Diabetes and Digestive and Kidney Diseases of the National Institutes of Health. M.T.W. acknowledges support from the John Simon Guggenheim Foundation. The authors acknowledge Dr. John Chodera for useful discussion.

REFERENCES

- ¹ W. J. Greenleaf, M. T. Woodside, and S. M. Block, Annu. Rev. Biophys. Biomol. Struct. **36**, 171 (2007).
- ²K. C. Neuman and A. Nagy, Nat. Methods 5, 491 (2008).
- ³L. Onsager, Phys. Rev. **54**, 554 (1938).
- ⁴J. D. Chodera and V. S. Pande, Phys. Rev. Lett. 107, 098102 (2011).
- ⁵A. P. Manuel, J. Lambert, and M. T. Woodside, Proc. Natl. Acad. Sci. U. S. A. 112, 7183 (2015).
- ⁶M. T. Woodside, P. C. Anthony, W. M. Behnke-Parks, K. Larizadeh, D. Herschlag, and S. M. Block, Science **314**, 1001 (2006).
- ⁷M. Hinczewski, Y. von Hansen, and R. R. Netz, Proc. Natl. Acad. Sci. U. S. A. 107, 21493 (2010).
- ⁸G. Hummer and A. Szabo, Proc. Natl. Acad. Sci. U. S. A. 107, 21441 (2010).
- ⁹ H. Yu, M. G. W. Siewny, D. T. Edwards, A. W. Sanders, and T. T. Perkins, Science 355, 945 (2017).
- ¹⁰D. E. Makarov, J. Chem. Phys. **141**, 241103 (2014).
- ¹¹C. Hyeon, G. Morrison, and D. Thirumalai, Proc. Natl. Acad. Sci. U. S. A. 105, 9604 (2008).
- ¹²P. Cossio, G. Hummer, and A. Szabo, Proc. Natl. Acad. Sci. U. S. A. **112**, 14248 (2015).
- ¹³P. Cossio, G. Hummer, and A. Szabo, J. Chem. Phys. **148**, 123309 (2018).
- ¹⁴T. E. Oliphant, *Guide to NumPy*, 2nd ed. (Create Space Independent Publishing Platform, USA, 2015), ISBN: 151730007X; 9781517300074.
- ¹⁵E. Jones, T. Oliphant, P. Peterson *et al.*, SciPy: Open source scientific tools for Python, URL: http://www.scipy.org/.
- ¹⁶F. Perez and B. E. Granger, Comput. Sci. Eng. **9**, 21 (2007).
- ¹⁷S. K. Lam, A. Pitrou, and S. Seibert, in *Proceedings of Second Workshop on the LLVM Compiler Infrastructure HPC-LLVM'15* (ACM Press, New York, New York, USA, 2015), pp. 1–6, ISBN: 9781450340052.
- ¹⁸J. D. Hunter, Comput. Sci. Eng. **9**, 90 (2007).
- ¹⁹ K. Neupane and M. T. Woodside, Biophys. J. **111**, 283 (2016).