

**CHEM 3PC3****Quiz #4**

October 7, 2025

Name: \_\_\_\_\_

Student Number: \_\_\_\_\_

**SUBMISSION INSTRUCTIONS — READ CAREFULLY**

To receive full credit, you must follow these steps:

1. Answer all the questions.
2. Write your solutions on new, separate pages. Do not write your solutions in the margins of this paper.
3. At the top of each solution page, clearly write the corresponding question number (e.g., “**Question 5**”). If you use more than one page for a question, write the question number on each page (e.g., “**Question 5 (Page 1 of 2)**”).
4. **If you cannot solve a question, still attach a page with the question number and write “Blank” or “No Answer” to indicate you attempted it.**
5. **If you are unsure of a complete answer, still attempt the question:** attach a page with the question number and write down any relevant thoughts, formulas, or initial steps. Partial credit may be awarded for demonstrated effort and correct reasoning, whereas a blank answer will receive no credit.
6. Show all your work clearly and legibly. Unorganized or illegible work may not receive credit.

**1 Problems**

1. Do the following integrals:

a.

$$\int_0^{\pi} \sin^3(x) dx$$

b.

$$\int_0^{\pi} x \sin^3(x) dx$$

c.

$$\int_0^1 \ln(x) dx$$

d.

$$\int_5^{10} \frac{1}{V^{1.5}} dV$$

2. The Redlich–Kwong equation of state is given by:

$$P_{RK}(V, T) = \frac{RT}{V - b} - \frac{a}{\sqrt{T} V(V + b)},$$

where  $R$ ,  $a$  and  $b$  are constants.

Derive the analytical expression of the integral,

$$\int V \left( \frac{\partial P_{RK}(V, T)}{\partial V} \right)_T dV.$$