# **Humidity at a Glance**

## Most Relevant Equations with Sample Code

This summary provides on overview on the most-used humidity-related formulas. The sample code is optimized for microprocessors (e.g. the common logarithm "log10" is used rather than the natural logarithm "ln"). For an in-depth study of the equations please refer to our complimentary paper "Introduction to Humidity" available on

## 1 Relative humidity

Condition: Constant absolute humidity (e.g. closed systems).

$$RH_2 = RH_1 \exp \left[ m \cdot T_n \frac{t_1 - t_2}{(T_n + t_1)(T_n + t_2)} \right]$$

```
RH₁
        relative humidity at position 1
        relative humidity at position 2
        temperature in °C at position 1
        temperature in °C at position 2
        17.62
        243.12 °C
```

Sample code: RH2 = RH1\*exp(4283.78\*(t1-t2)/(243.12+t1)/(243.12+t2));

## 2 Dew point

Definition: The dew point is the temperature to which a given parcel of air must be cooled, at constant barometric pressure, for water vapor to condense into water.

$$t_{d}(t,RH) = T_{n} \cdot \frac{\ln\left(\frac{RH}{100\%}\right) + \frac{m \cdot t}{T_{n} + t}}{m - \left[\ln\left(\frac{RH}{100\%}\right) + \frac{m \cdot t}{T_{n} + t}\right]}$$

$$t_{d} \text{ dew point temperature in °C}$$

$$RH \text{ actual relative humidity in %}$$

$$m \quad 17.62$$

$$T_{n} \quad 243.12 °C$$

```
dew point temperature in °C
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Sample code:  $H = (\log 10(RH) - 2.0) / 0.4343 + (17.62*t) / (243.12+t);$ td = 243.12\*H/(17.62-H);

## 3 Absolute humidity

Definition: The absolute humidity is the mass of water vapor in a particular volume of dry air. The unit is g/m<sup>3</sup>.

$$d_{v}(t,RH) = 216.7 \cdot \begin{bmatrix} \frac{RH}{100\%} \cdot A \cdot \exp\left(\frac{m \cdot t}{T_{n} + t}\right) \\ 273.15 + t \end{bmatrix}$$

$$d_{v} \text{ absolute his actual tem} \\ RH \text{ actual relation} \\ m \text{ 17.62} \\ T_{n} \text{ 243.12 °C} \\ A \text{ 6.112 hPa}$$

```
absolute humidity in g/m3
actual temperature in °C
actual relative humidity in %
```

Sample code: dv = 216.7\*(RH/100.0\*6.112\*exp(17.62\*t/(243.12+t))/(273.15+t));

## 4 Mixing ratio

Definition: The mixing ratio is the mass of water vapor in a particular mass of dry air. The unit is g/kg.

$$r(t,RH) = \frac{622 \cdot \frac{RH}{100\%} \cdot A \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right)}{p - \frac{RH}{100\%} \cdot A \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right)}$$

r mixing ratio in g/kg
 t actual temperature in °C
 RH actual relative humidity in %
 p barometric air pressure in hPa
 m 17.62
 T<sub>0</sub> 243.12 °C

6.112 hPa

Sample code: e = RH/100.0\*6.112\*exp(17.62\*t/(243.12+t));r = 622.0\*e/(p-e);

#### 5 Heat index

Definition: The heat index is determined according to the National Weather Service and Weather Forecast Office of the National Oceanic and Atmospheric Administration (NOAA).

$$HI_{Celsius}(t,RH) = t + \frac{5}{9} \cdot \left[ \frac{RH}{100\%} \cdot \exp\left(\frac{m \cdot t}{T_n + t}\right) - 10 \right]$$

$$HI_{Fahrenheit}(t,RH) = \frac{9}{5} \cdot HI_{Celsius} + 32$$

 $\begin{array}{lll} \textit{HI}_{\textit{Celsius}} & \textit{Heat index in °C} \\ \textit{HI}_{\textit{Fahrenhei}\,t} & \textit{Heat index in °F} \\ \textit{t} & \textit{actual temperature in °C} \\ \textit{RH} & \textit{actual relative humidity in \%} \\ \textit{m} & \textit{17.62} \\ \textit{T}_{\textit{n}} & \textit{243.12 °C} \end{array}$ 

Sample code:

p = RH/100.0\*exp(17.62\*t/(243.12+t));HIC = t+5.0/9.0\*(p-10.0); // this is the heat index in Celsius HIF = 9.0/5.0\*HIC+32.0; // this is the heat index in Fahrenheit

#### **Revision History**

Date	Revision	Changes
Aug. 20, 2008	1.0	Initial version