Solid-State Battery

Definition

A **solid-state battery** is a rechargeable battery that uses a solid electrolyte instead of the liquid or gel-based electrolytes found in conventional batteries. This design improves safety, energy density, and performance, making it a key technology for next-generation energy storage.

Examples

- 1. Thin-Film Solid-State Batteries: Used in wearables and medical devices.
- 2. **Bulk Solid-State Batteries**: Designed for electric vehicles (EVs) and grid storage.



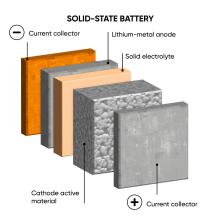
Advantages

- Safety: Non-flammable solid electrolytes reduce the risk of fire.
- **Higher Energy Density**: Stores more energy, enabling longer use in devices or driving range in EVs.
- **Durability**: Can withstand more charge-discharge cycles.
- Wide Temperature Range: Operates in extreme conditions (-50°C to 125°C).

Disadvantages

- Cost: High manufacturing and material costs.
- Scalability Issues: Challenges in producing large-scale batteries consistently.
- Interfacial Resistance: Difficulty ensuring efficient contact between components.

Cell Reactions



1. At the Anode (Oxidation):

$$Li \rightarrow Li++e-$$

Lithium releases electrons and ions.

2. At the Cathode (Reduction):

$$Li++e-+$$
 Cathode Material \rightarrow Li(Cathode

Material)

Lithium ions combine with electrons and the cathode material to store energy.

Performance Statistics

• Energy Density:

o Thin-film: 300–900 Wh/kg

o Bulk: 250–500 Wh/kg

• Cycle Life: 10,000–100,000 cycles.

• Operating Voltage: 2.5 V to 4.6 V.

• **Temperature Range**: -50°C to 125°C.

Applications



Electric Vehicles (EVs):

• Increases driving range and enables faster charging (~15 minutes to 80%).



Consumer Electronics:

• Powers compact devices like smartphones and laptops.

Medical Devices:

• Reliable and safe for pacemakers and wearable health monitors.

Aerospace and Industrial Equipment:

• Functions under extreme conditions for satellites and drones.



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Submitted By

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