

Chapter 1

Textbook questions

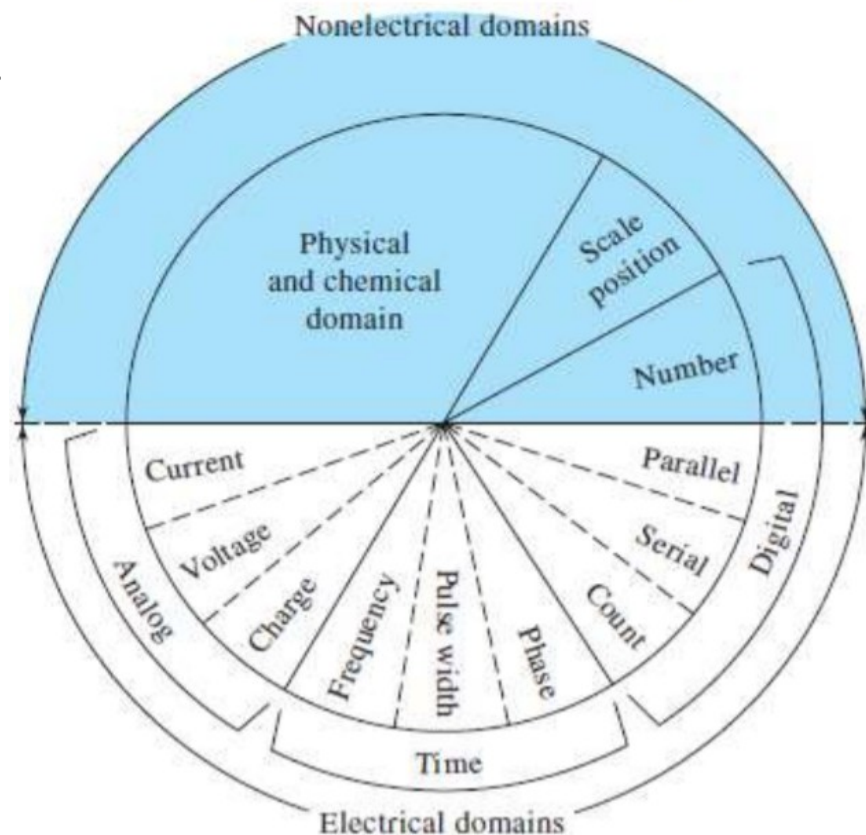
Question 1-1

What is a transducer in an analytical instrument?

換能器是一種將化學或物理信息互相轉換成電信號的設備。



Analytical data domain map

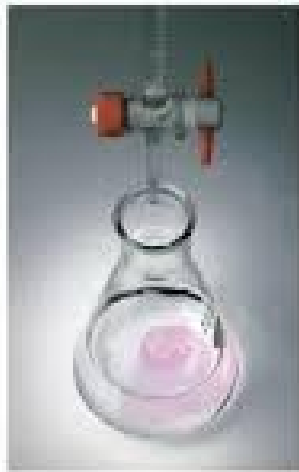


Skoog D.A., Holler F.J., Crouch S.R. 2017, Principles of Instrumental Analysis, 7th Edition. Cengage.

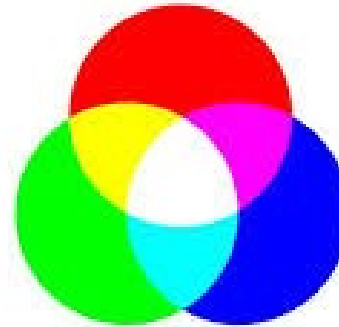
Question 1-2

What is the information processor in an instrument for measuring the colour of a solution visually?

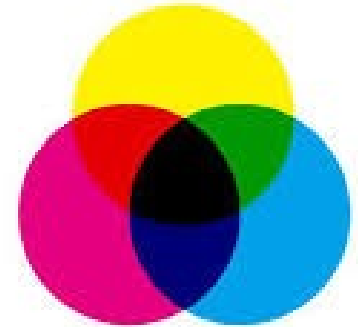
視覺色彩測量系統中的信息處理器是人的頭腦。



RGB



CMYK



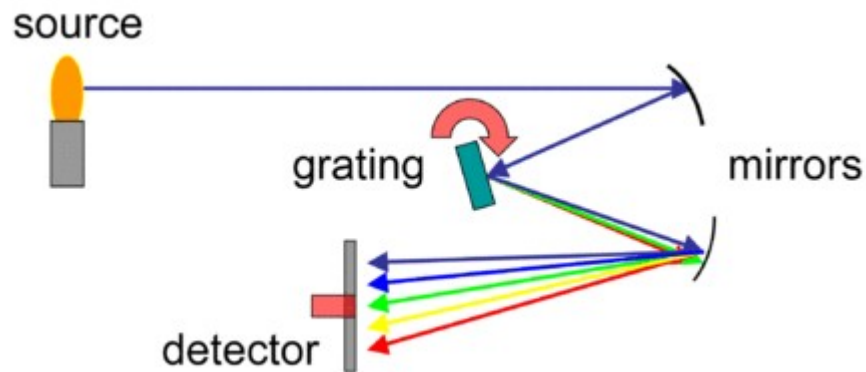
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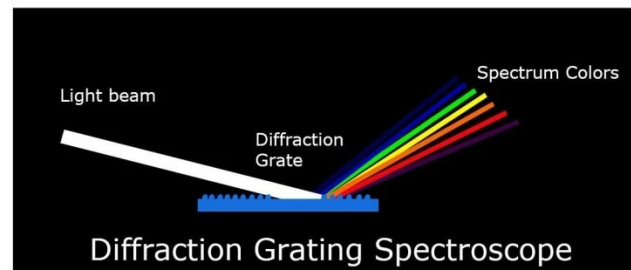
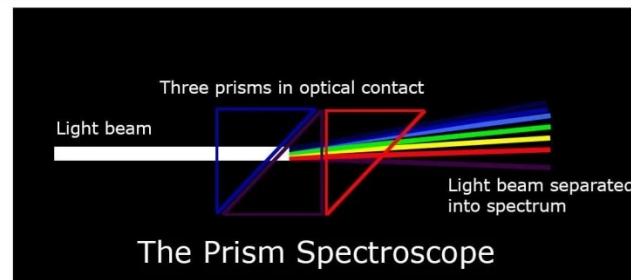
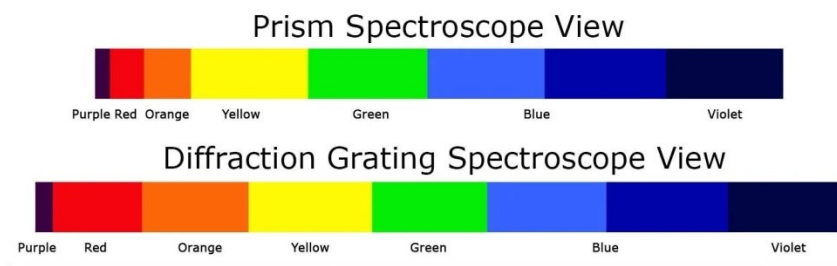
Question 1-3

What is the detector in a spectrograph in which spectral lines are recorded photographically?

光譜儀中的檢測器是攝影膠片或板。



<https://zh.wikipedia.org/wiki/%E5%85%89%E8%B0%B1%E4%BB%AA>



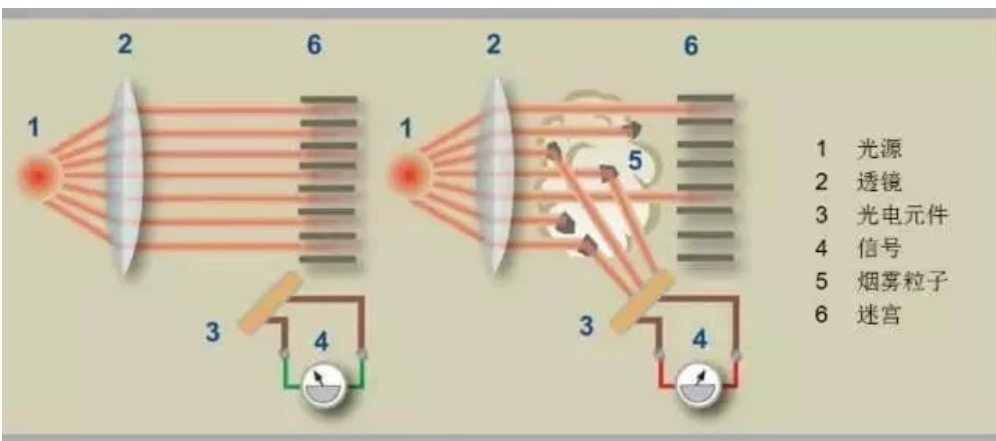
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Question 1-4

What is the transducer in a smoke detector?

煙霧探測器有兩種類型：光電探測 (photodetectors) 和電離探測器 (ionization detector)。

光電探測器



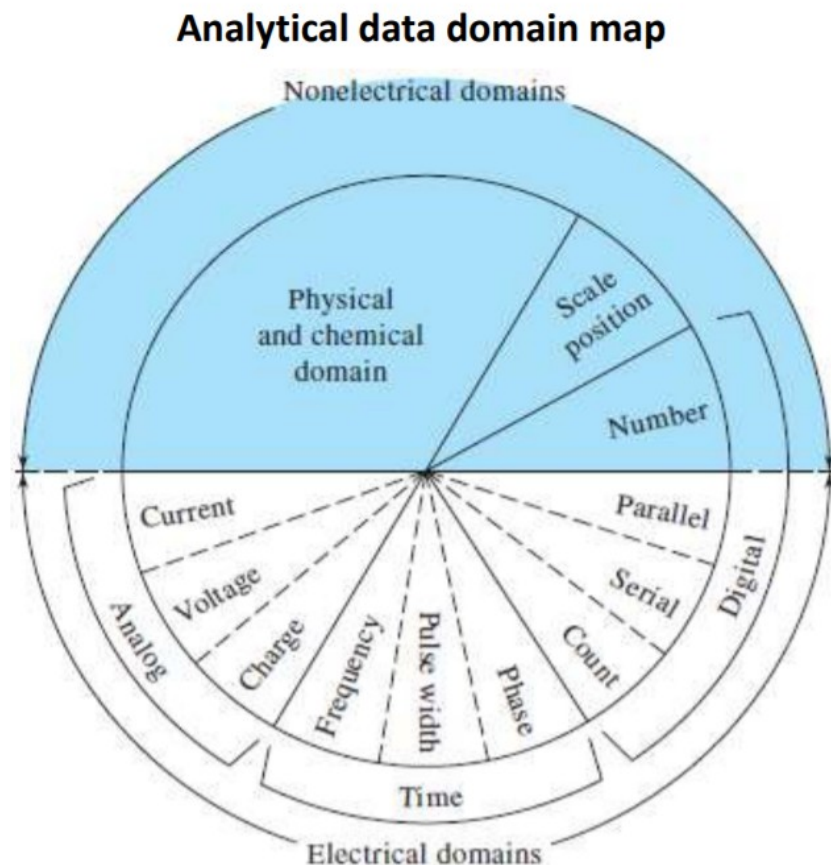
電離探測器



Question 1-5

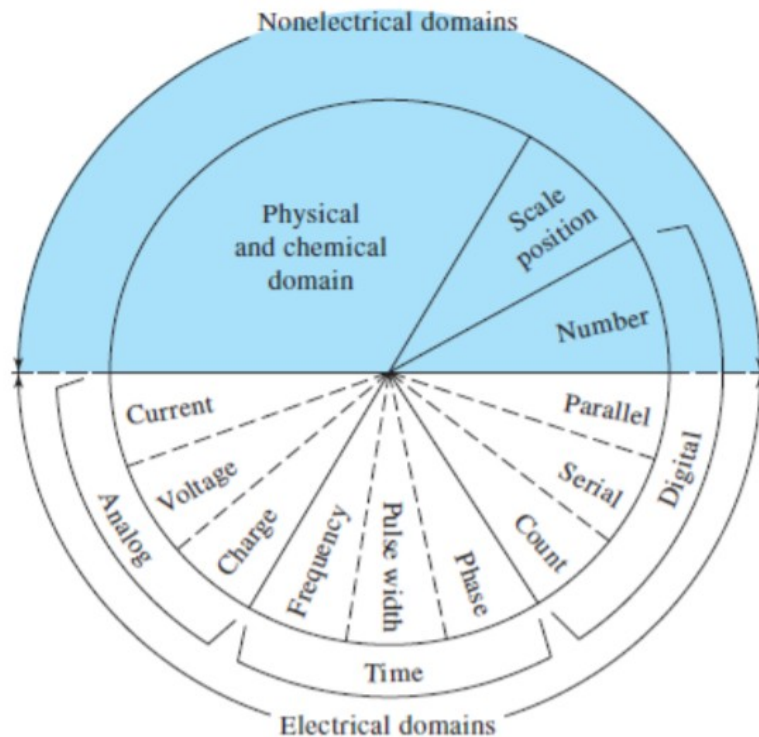
What is a data domain?

數據域，將訊息密碼化的各種方式稱為數據域，是一種包含數據的形式，可將訊息解碼為另一種訊息，其可分為電性域 (electrical domain) 和非電性域 (nonelectrical domain)，其方法有電流、電壓、數位數字、長度、密度、光強度等。



Question 1-6

Name electrical signals that are considered time-domain signals? How is the information encoded in a time-domain signal?



Time-domain signals include **period, frequency, and pulse width.**

時間域包含週期、頻率、脈衝寬。

The information is encoded in the **time relationship of signal fluctuations.**

這類訊息以訊號波動對時間來編碼。

Question 1-7

List four input transducers and describe how they are used?

Input Transducer	Use
Phototube	Convert light intensity to an analog current
Glass electrode	Convert electrode potential to a voltage related to pH
Electron multiplier	Convert ion intensity to a corresponding electric current
Thermal conductivity	Converts sample thermal conductivity to related voltage

PI

Question 1-8

What is a figure of merit?

A figure of merit(品質因數)is **a number that provides quantitative information** about some **performance criterion** for an instrument or method.

品質因數是用於表示儀器或方法的性能的數量

Example: Resolution of the image sensor in a digital camera

Question 1-9

A 25.0mL sample containing Cu^{2+} gave an instrument signal of 25.2 units (corrected for a blank). When exactly 0.500mL of 0.0275 M $\text{Cu}(\text{NO}_3)_2$ was added to the solution, the signal increased to 45.1 units. Calculate the molar concentration of Cu^{2+} assuming that the signal was directly proportional to the analyte concentration?

$$6.66 \times 10^{-4} \text{ M}$$

Let c_s = molar concentration of Cu^{2+} in standard = 0.0275 M

c_x = unknown Cu^{2+} concentration

V_s = volume of standard = 0.500 mL

V_x = volume of unknown = 25.0 mL

S_1 = signal for unknown = 25.2

S_2 = signal for unknown plus standard = 45.1

Assuming the signal is proportional to c_x and c_s , we can write

$$S_1 = Kc_x \quad \text{or} \quad K = S_1/c_x$$

After adding the standard

$$S_2 = K \left(\frac{V_x c_x + V_s c_s}{V_x + V_s} \right)$$

Substituting for K and rearranging gives,

$$c_x = \frac{S_1 V_s c_s}{S_2 (V_x + V_s) - S_1 V_x}$$

$$c_x = \frac{25.2 \times 0.500 \text{ mL} \times 0.0275 \text{ M}}{45.1(0.500 \text{ mL} + 25.0 \text{ mL}) - (25.2 \times 25.0 \text{ mL})} = 6.66 \times 10^{-4} \text{ M}$$