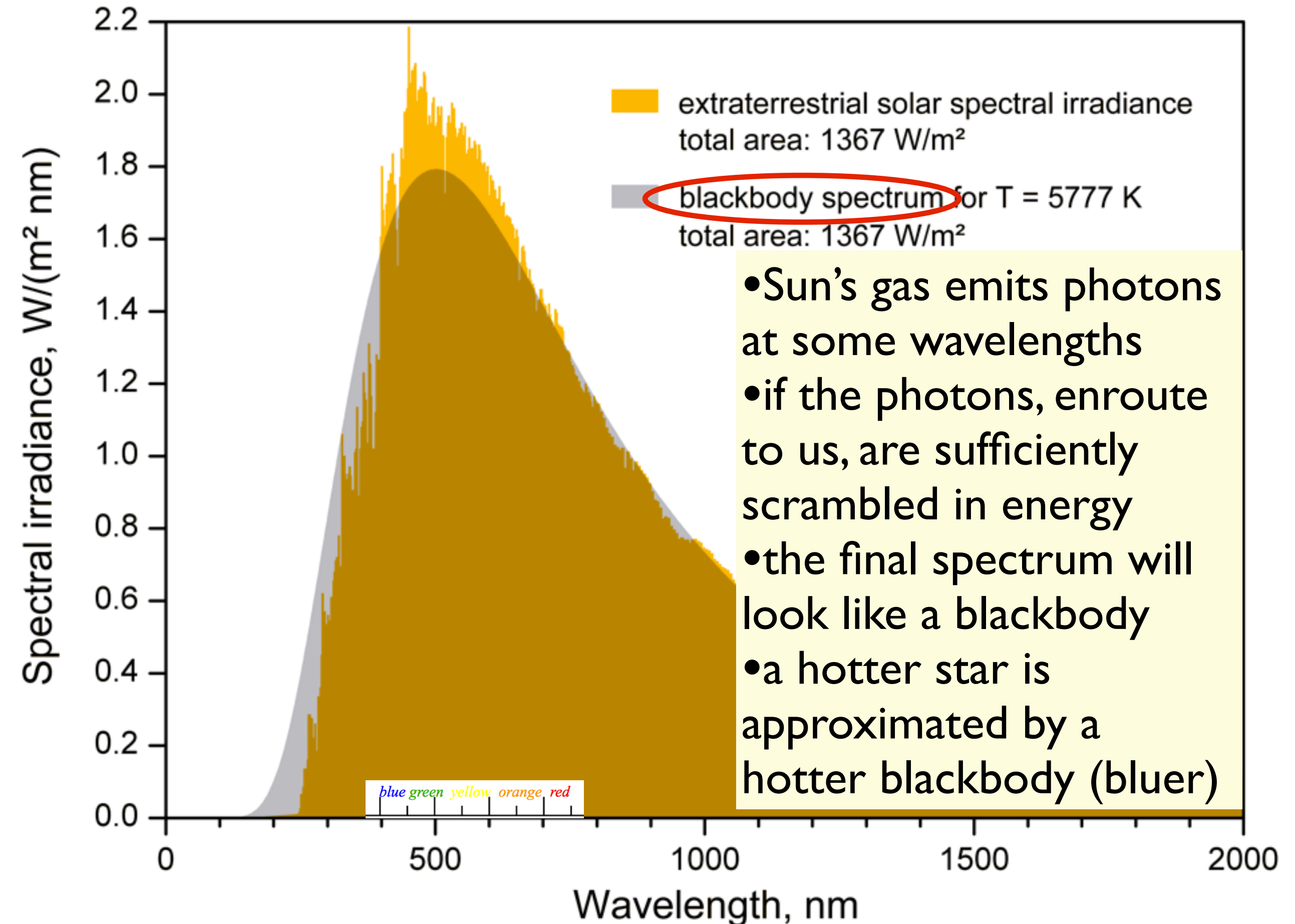
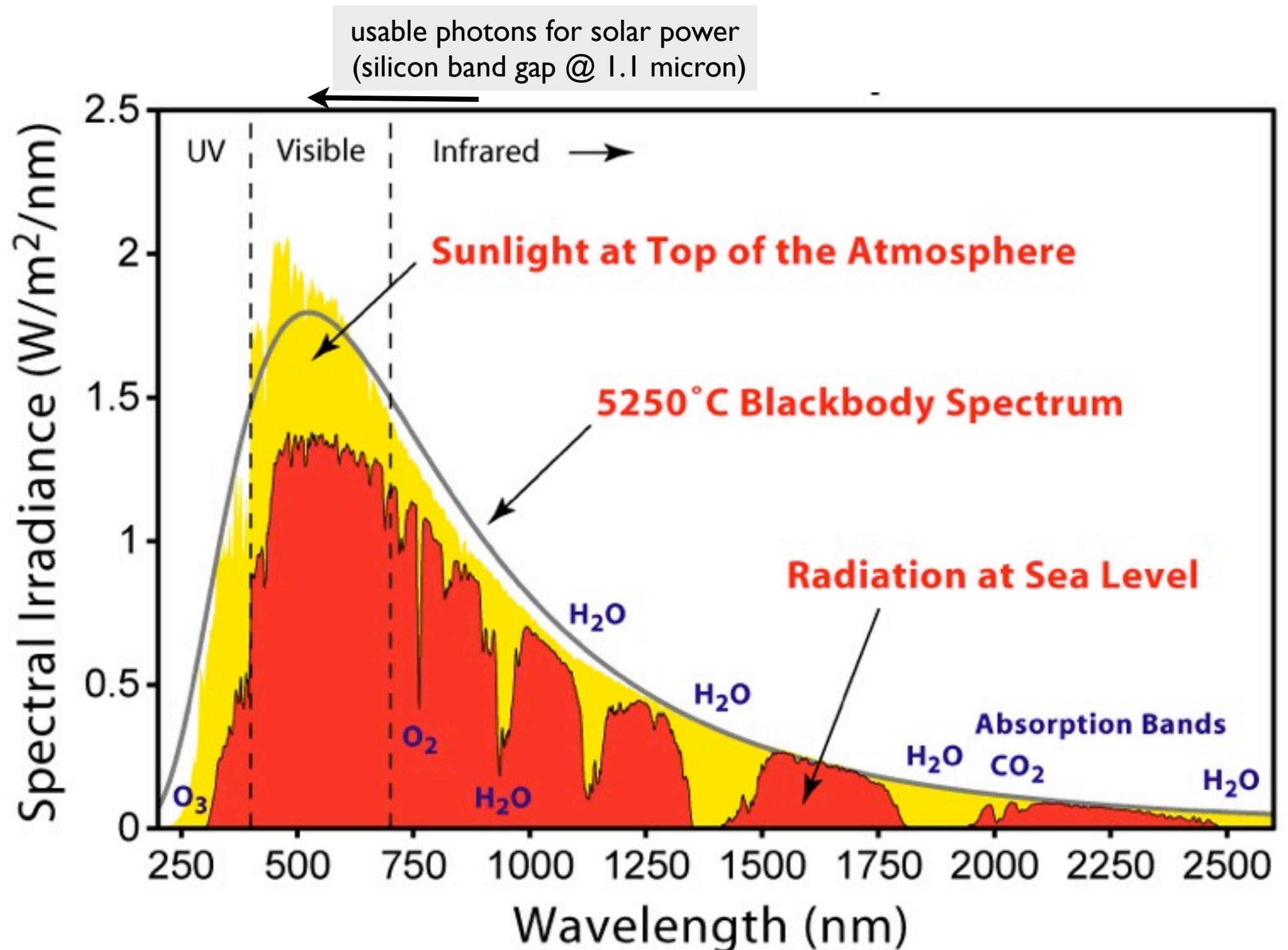


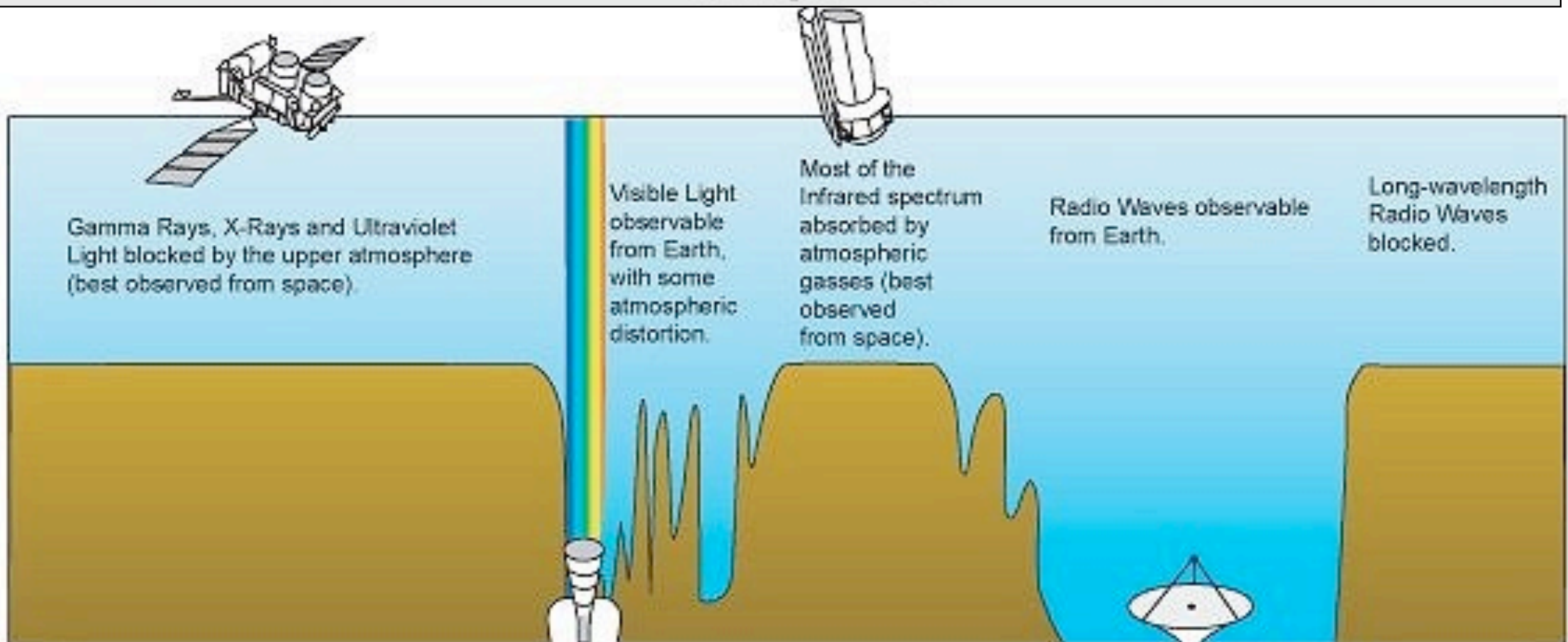
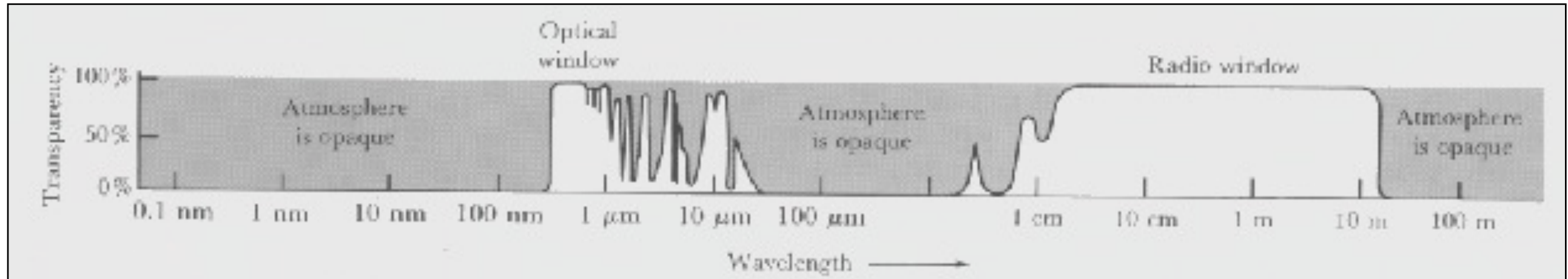
# the “blackbody” spectrum is an approximation



- Molecules in the Earth atmosphere absorb in certain wavelengths (filtering)
- Ozone ( $O_3$ ) absorbs in UV
- $H_2O$ ,  $CO_2$  absorbs in infrared, “the Greenhouse effect”



Due to these absorptions, our atmosphere is transparent only in optical & radio.



Atoms/molecules absorb or emit at specific wavelengths.

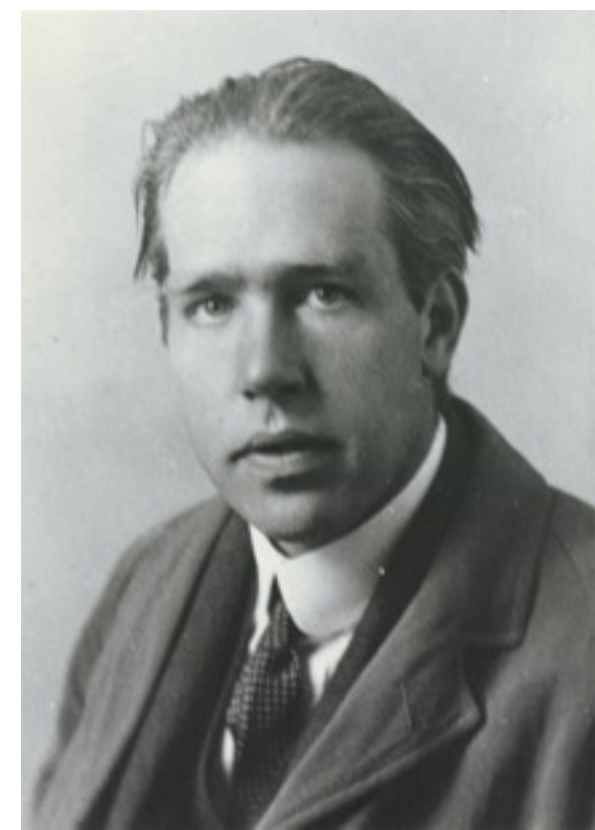
This allows us to study the make-up of celestial bodies as if they are on our table-top.

The mechanism behind this was first revealed by Niels Bohr in the 1920s.

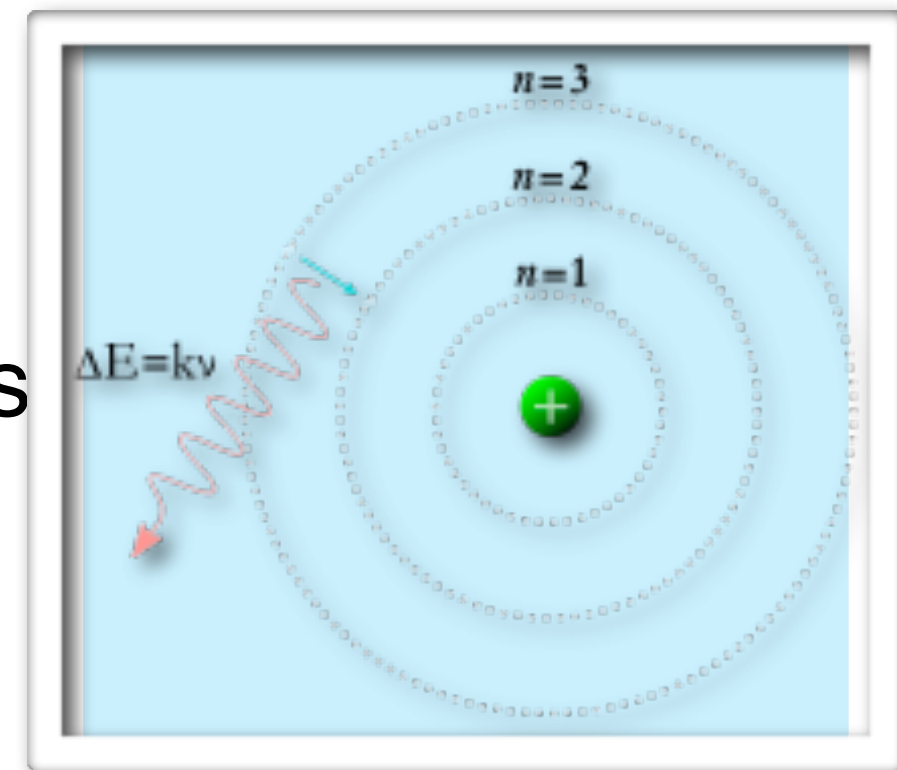


# Quantum mechanical behaviour of matter

## -- Niels Bohr's hydrogen atom (1913)



- classical orbits, e.g., planetary orbits around the Sun, can be of any shape/size.
- electron is both a wave and a particle. inside an atom, its wave-like nature determines that it can only have some special orbits.
- Electrons can only move between these allowed orbits, by interacting with photons. Otherwise they remain stable.



# What are these special orbits?

- electrons are particles as well as waves, with wavelength (the de Broglie wavelength)

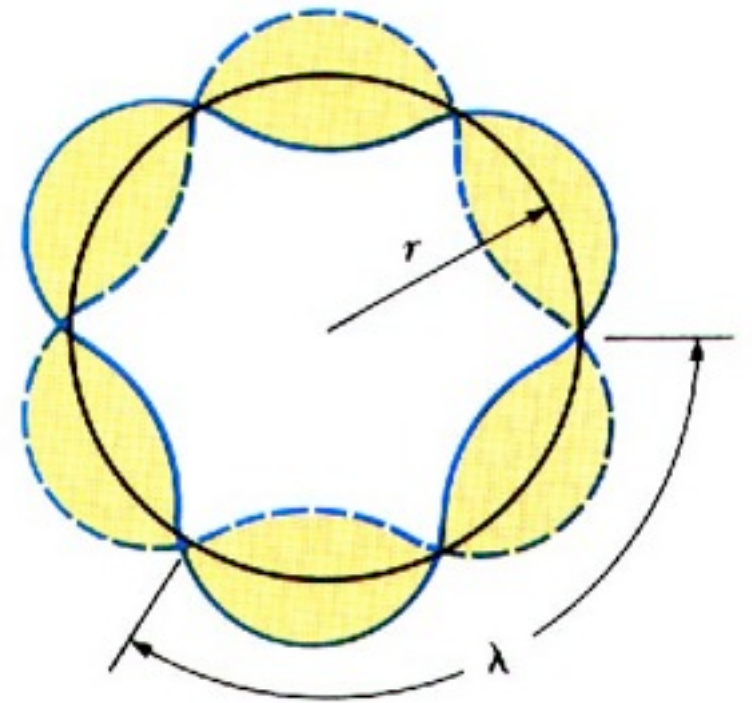
$$\lambda = h/p = h/(m v)$$

- **assume**: the circumference of the orbit has to be an integer number times the wavelength (standing wave)

$$2 \pi r = n \lambda = n h/(m v)$$

- $r$  and  $v$  are related: electrostatic attraction between the nucleus and the electron has to be balanced by the centrifugal force

$$\frac{k_e e^2}{r^2} = \frac{m_e v^2}{r}$$



# Energy of an orbit with quantum number n

- size of the orbit is quantized as

$$r = n^2 \left( \frac{h}{2\pi} \right)^2 \frac{1}{m_e k_e e^2} = n^2 \frac{\hbar^2}{m_e k_e e^2} = n^2 \times (0.5 \text{\AA})$$

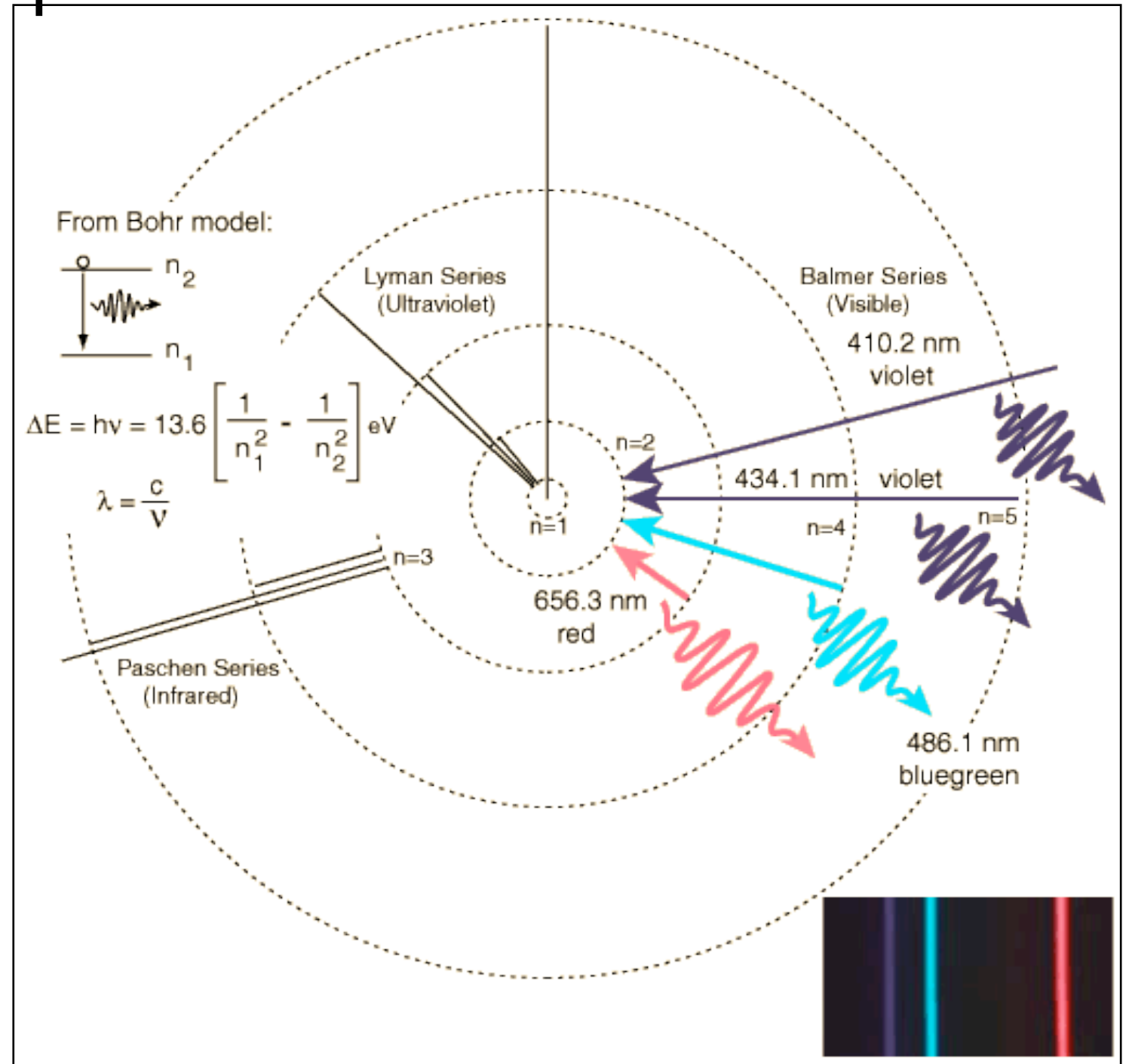
- and the energy is quantized as

$$E = \frac{1}{2} m_e v^2 + \left( -\frac{k_e e^2}{r} \right) = -\frac{1}{n^2} \left( \frac{k_e^2 e^4 m_e}{2\hbar^2} \right) \approx -\frac{1}{n^2} \times (13.6 \text{ eV})$$

- this is a semi-classical theory that gives the right result. Modern quantum mechanics posits that electrons are spherical clouds of probabilities.

- . electron orbits have discrete energies
- . an electron at the  $n=2$  orbit, decays to the  $n=1$  orbit: emits a photon
- . to go backward: absorbs a photon

.....



Lyman, Balmer, Paschen...series



“Of all objects, the [stars and] planets are those which appear to us under the least varied aspect. We see how we may determine their forms, their distances, their bulk, and their motions. Be we can never know anything of their chemical or mineralogical structure.

– August Comte (1842)  
Famous French Philosopher

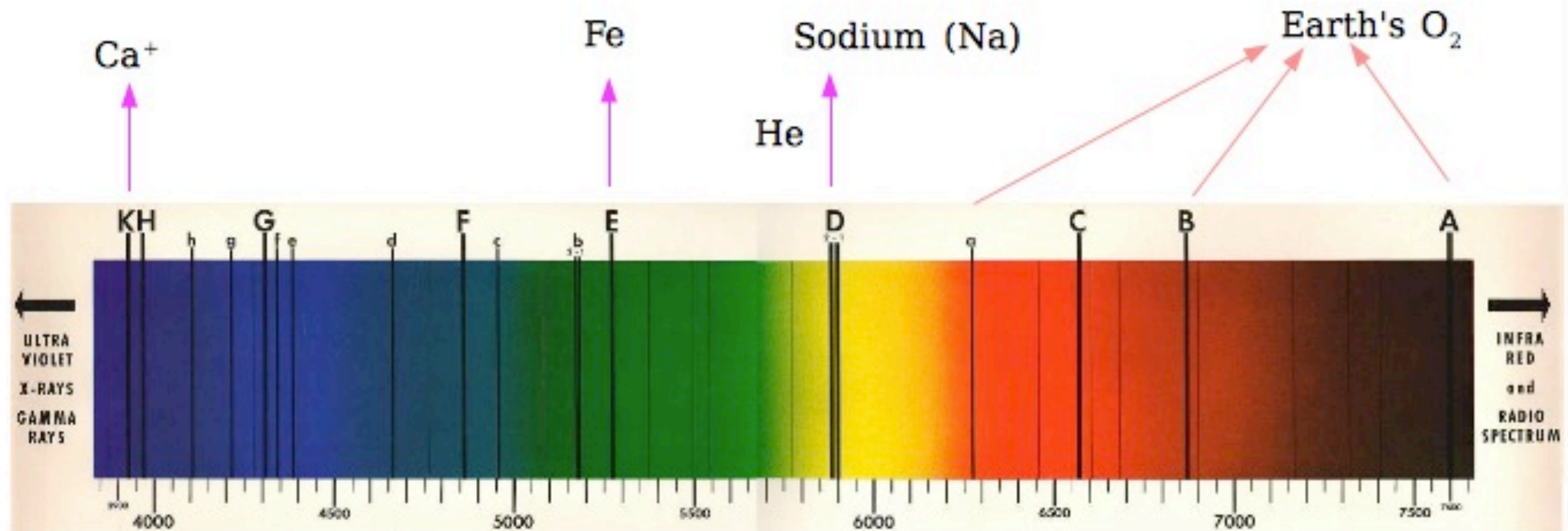


Eight years later (1850), Fraunhofer found this when looking at the Sun through a prism (don't do this at home):

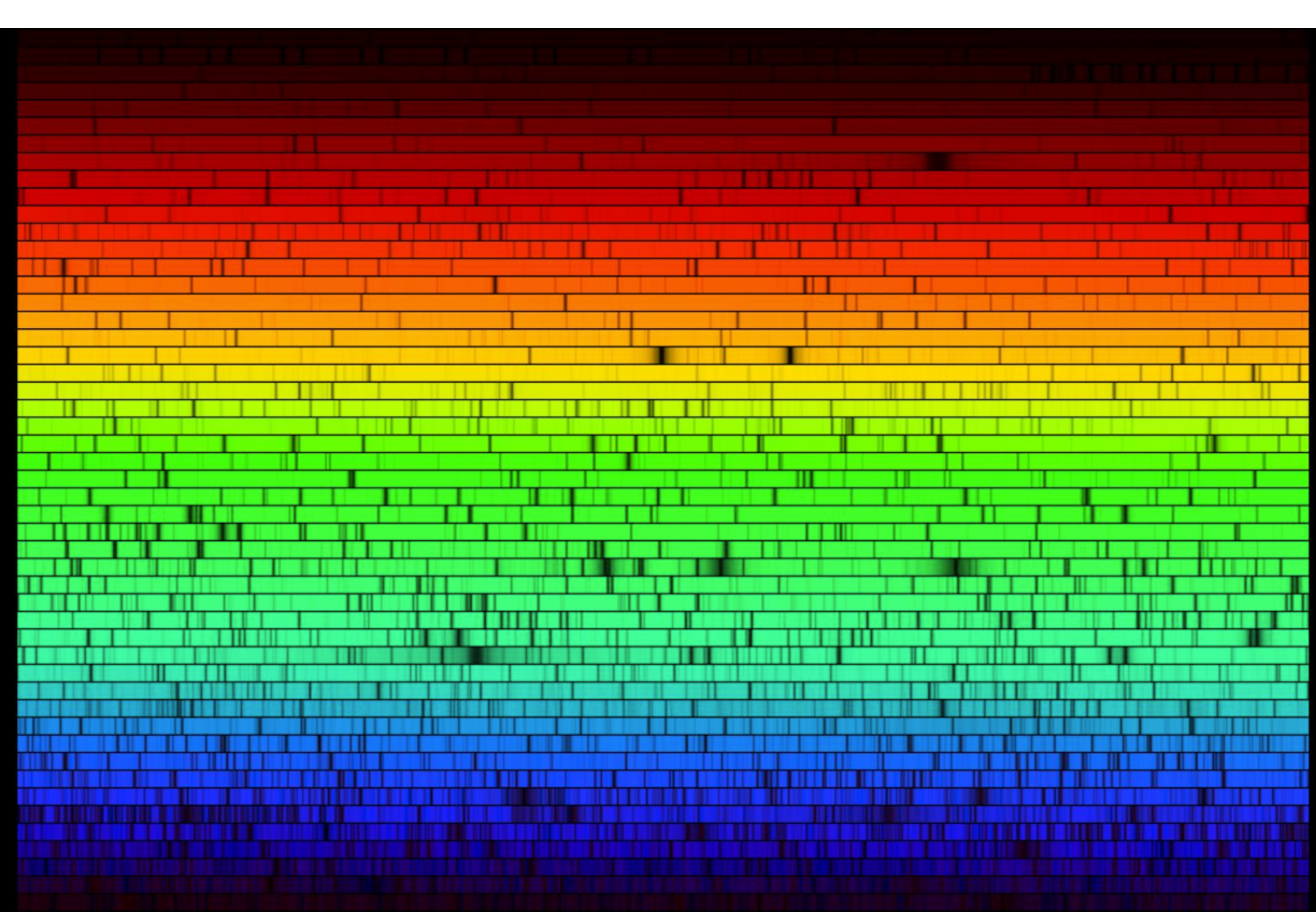
## Optical Spectrum of the Solar lights ---- Fraunhofer lines

*from [www.harmsy.freeuk.com/fraunhofer.html](http://www.harmsy.freeuk.com/fraunhofer.html)*

absorption features: flux deficit in the radiation spectrum





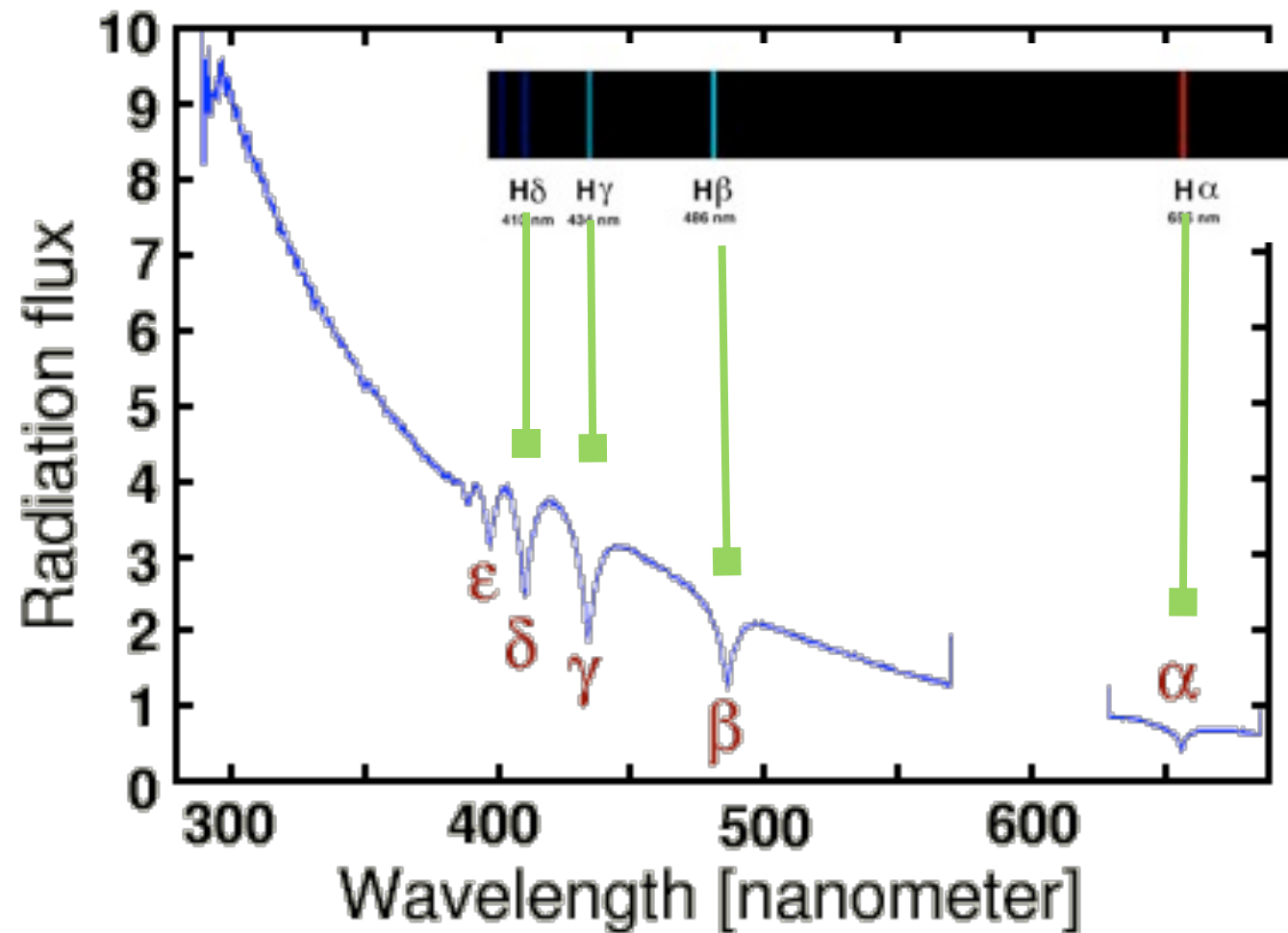


a very high resolution spectrum of the Sun

Friday, January 25, 2013



# Spectrum of Sirius B (a white dwarf)

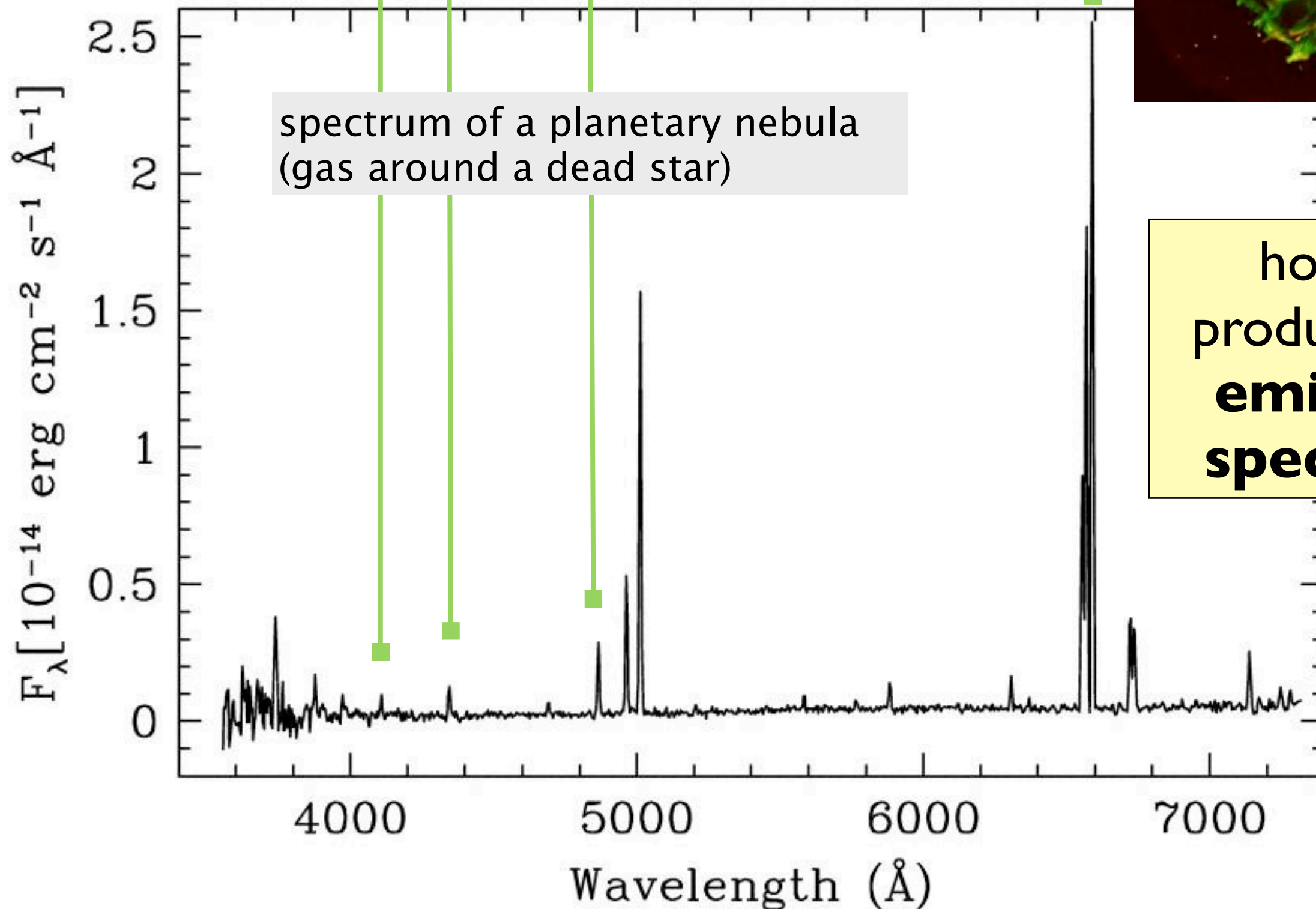
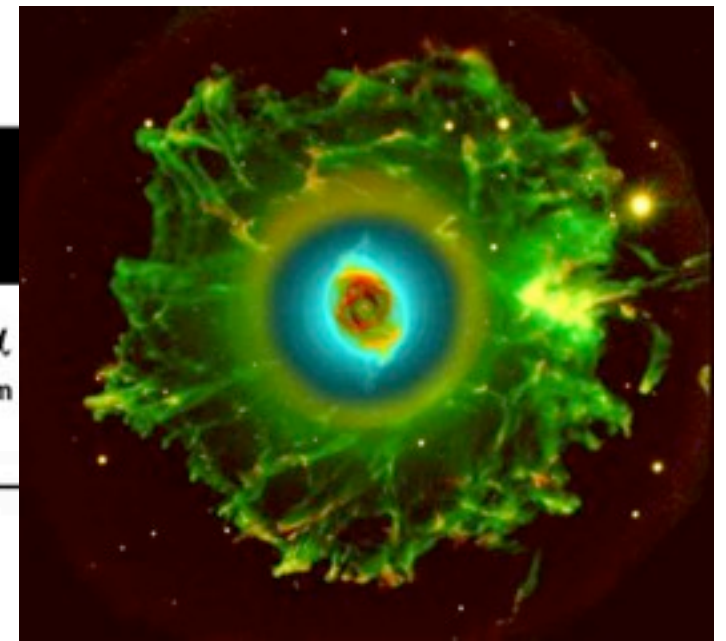


Hydrogen gas is in absorption

blackbody radiation from deeper, hotter layers of the star is absorbed by the cooler outer layer of gas at specific wavelengths, producing an **absorption spectrum**.

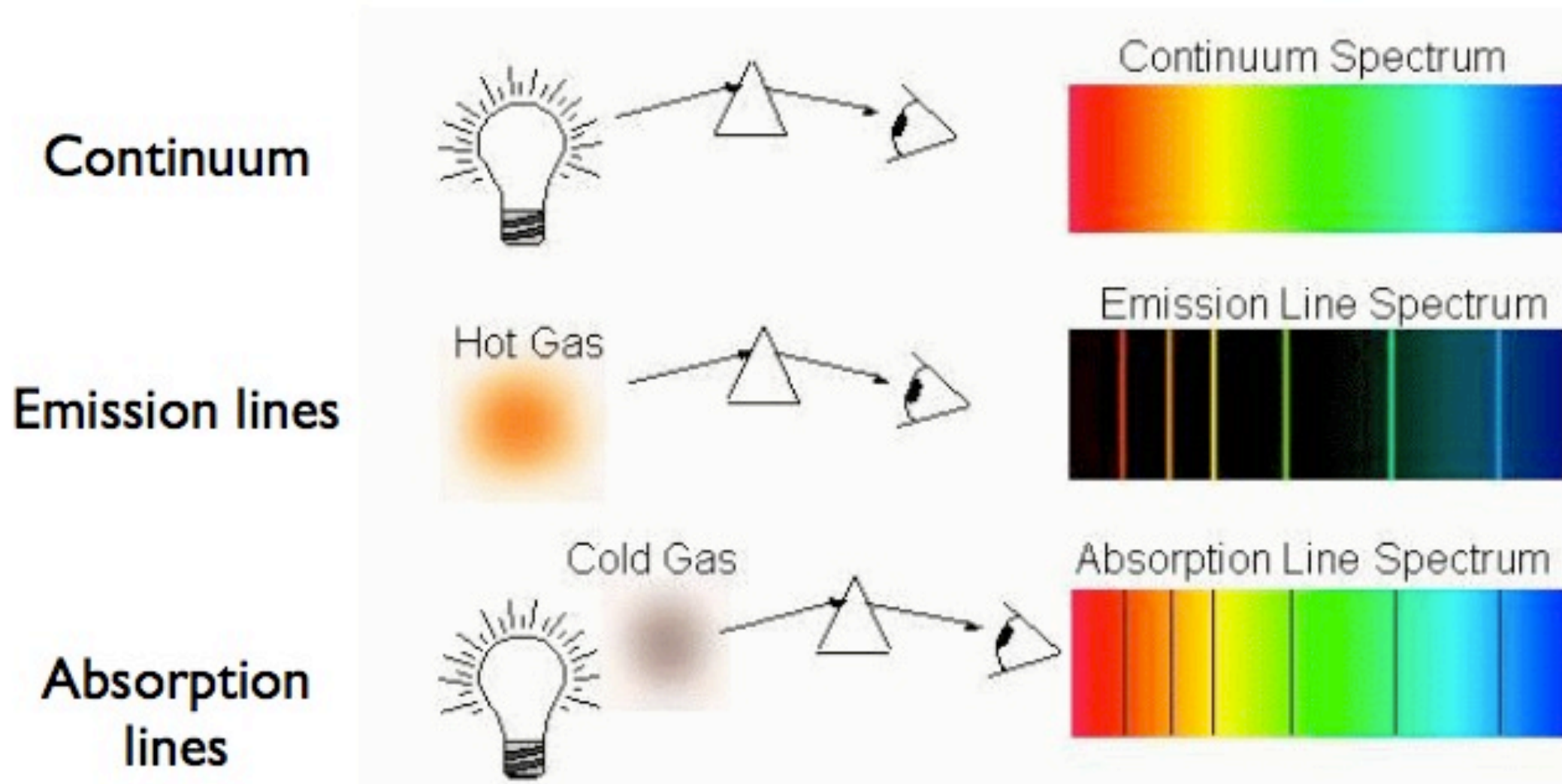


hot hydrogen  
gas in lab



hot gas  
producing an  
**emission  
spectrum**

# 3 Types of astronomical spectra



continuum spectrum: e.g., blackbody

emission spectrum: a hot gas that radiates only in spectral lines

absorption spectrum: continuum radiation being partially absorbed by intervening colder gas

# emission spectrum

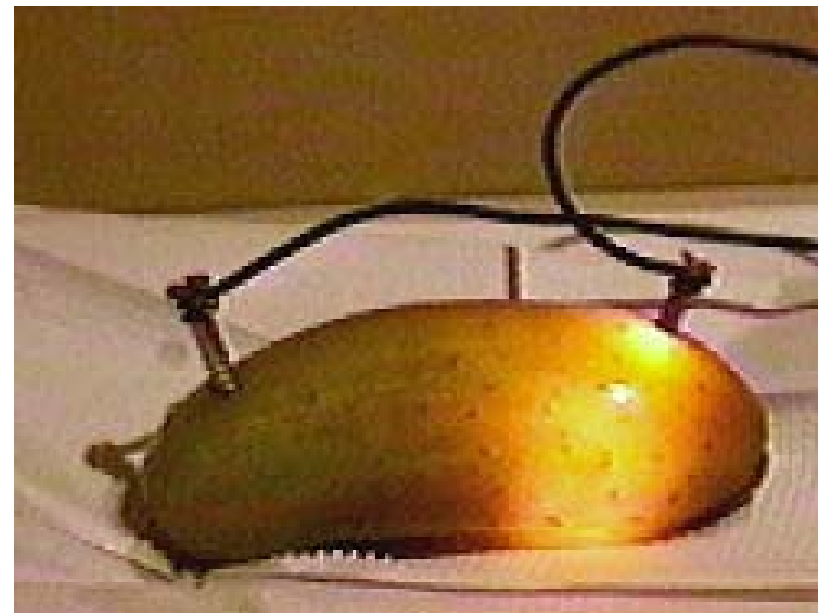
helium



sodium

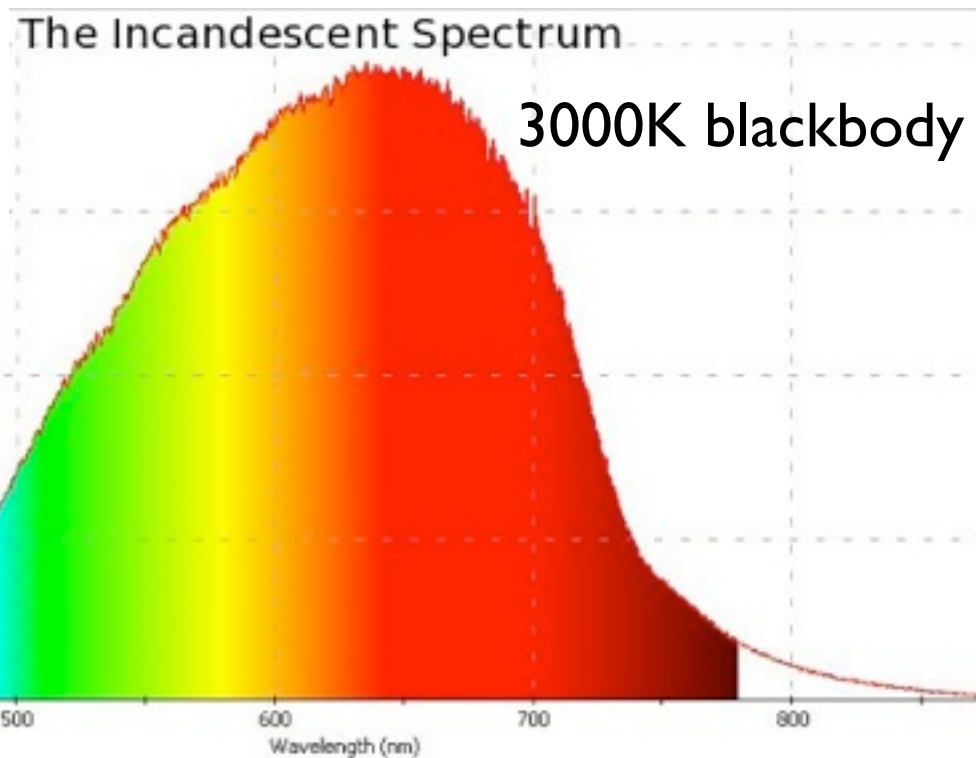


neon



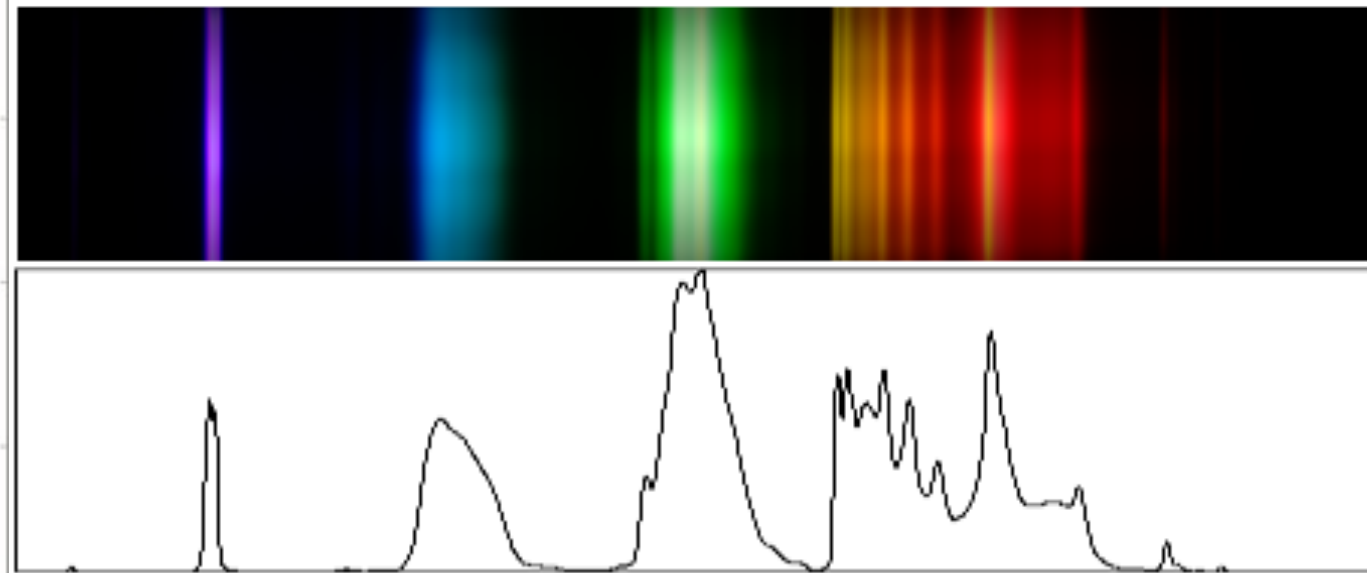
# Real World Applications:

why do compact fluorescent bulbs cut your monthly energy bills?



Spectrum of Compact Fluorescent Light  
GREENLITE 18W/ELS-M 2700K FCC ID: N6AFJEE0404

photo: 20D  
J. Beale 9/2007



Energy efficient lights work by emitting in only a narrow range of wavelength (emission lines), and fool our brains into thinking the light is a broad spectrum.

They are more efficient because the bulbs don't have to be heated up to high temperature, unlike conventional bulbs (and wasting energy in infrared lights)

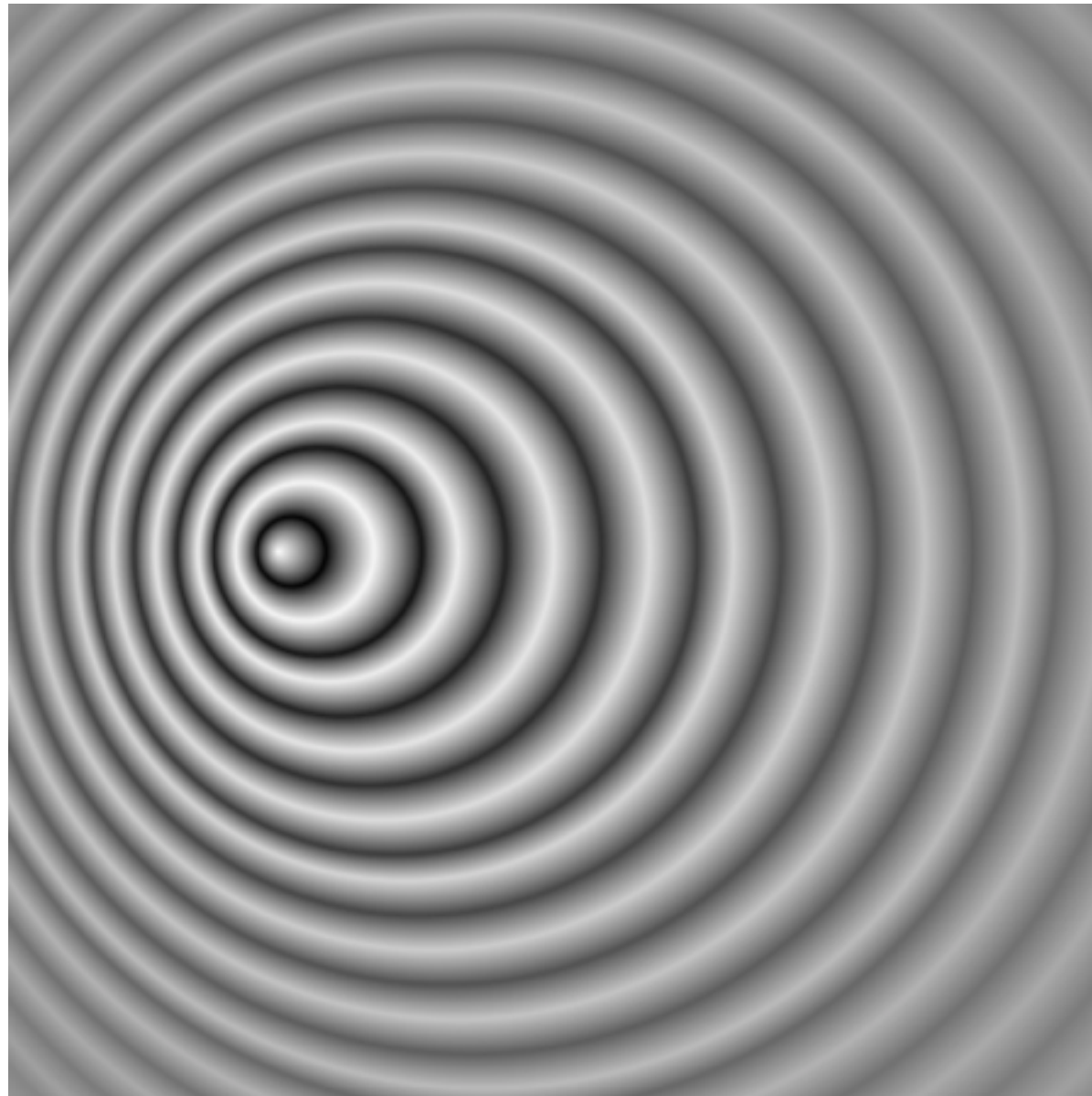


# The (Classical) Doppler effect

Light is a wave, just like sound wave, or water wave.

light source  
moving toward  
you

$\lambda = \lambda_0 * (1 - v/c)$   
light blue-shifted

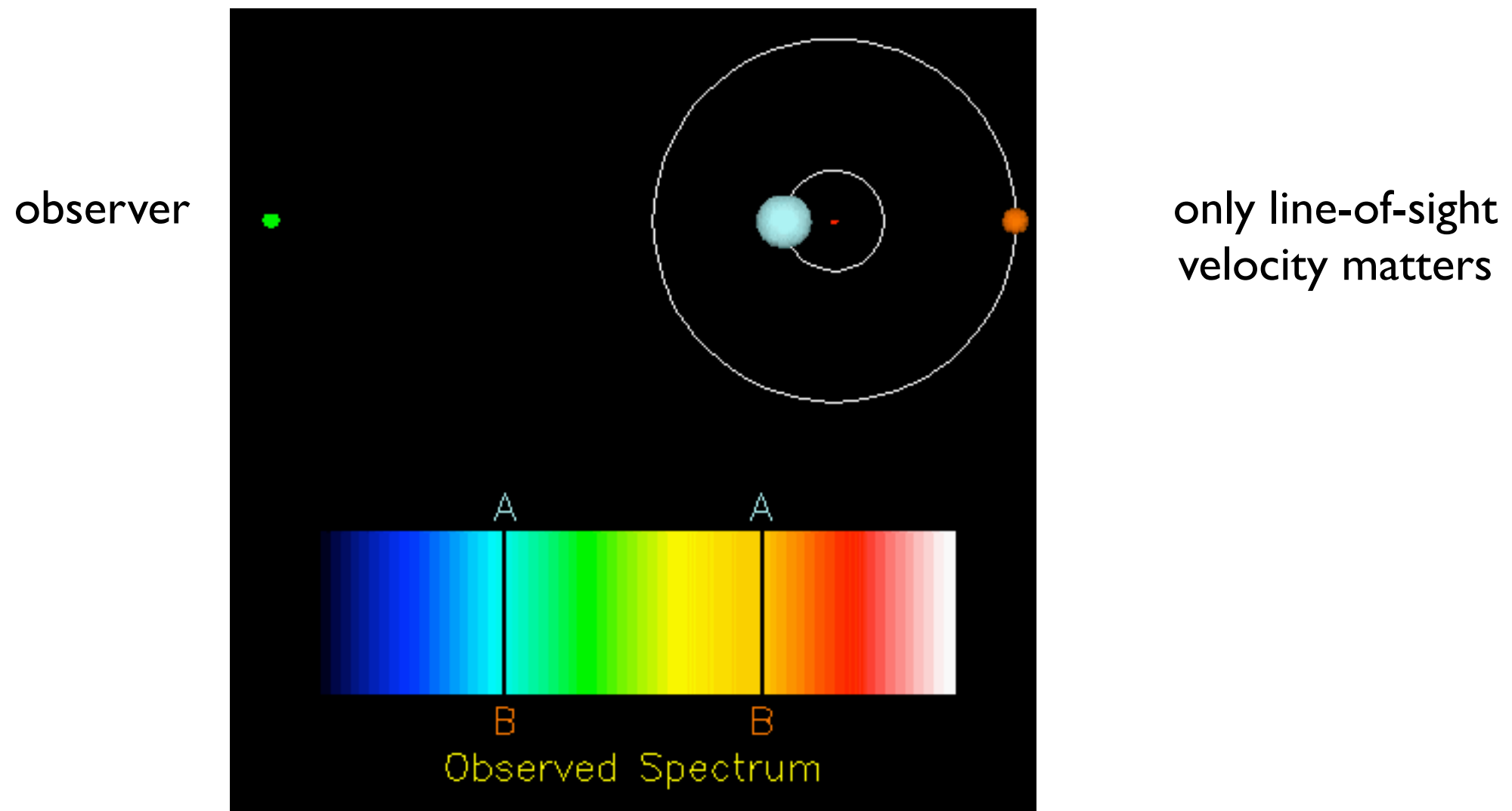


moving away from  
you

$\lambda = \lambda_0 * (1 + v/c)$   
light red-shifted

perpendicular direction  
 $\lambda = \lambda_0$  (rest wavelength)  
light unchanged

# Doppler effect revealing motion in a binary star system

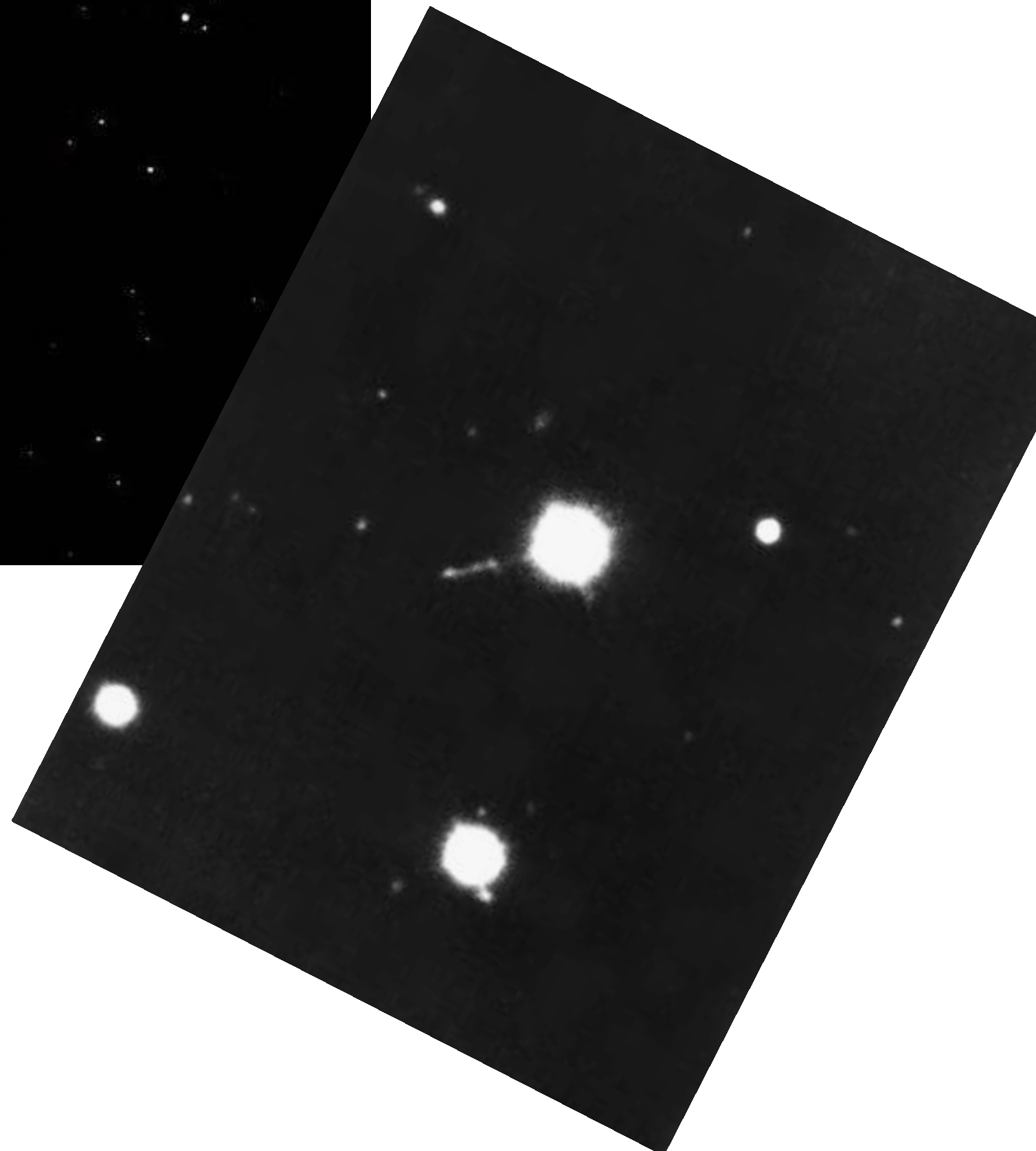
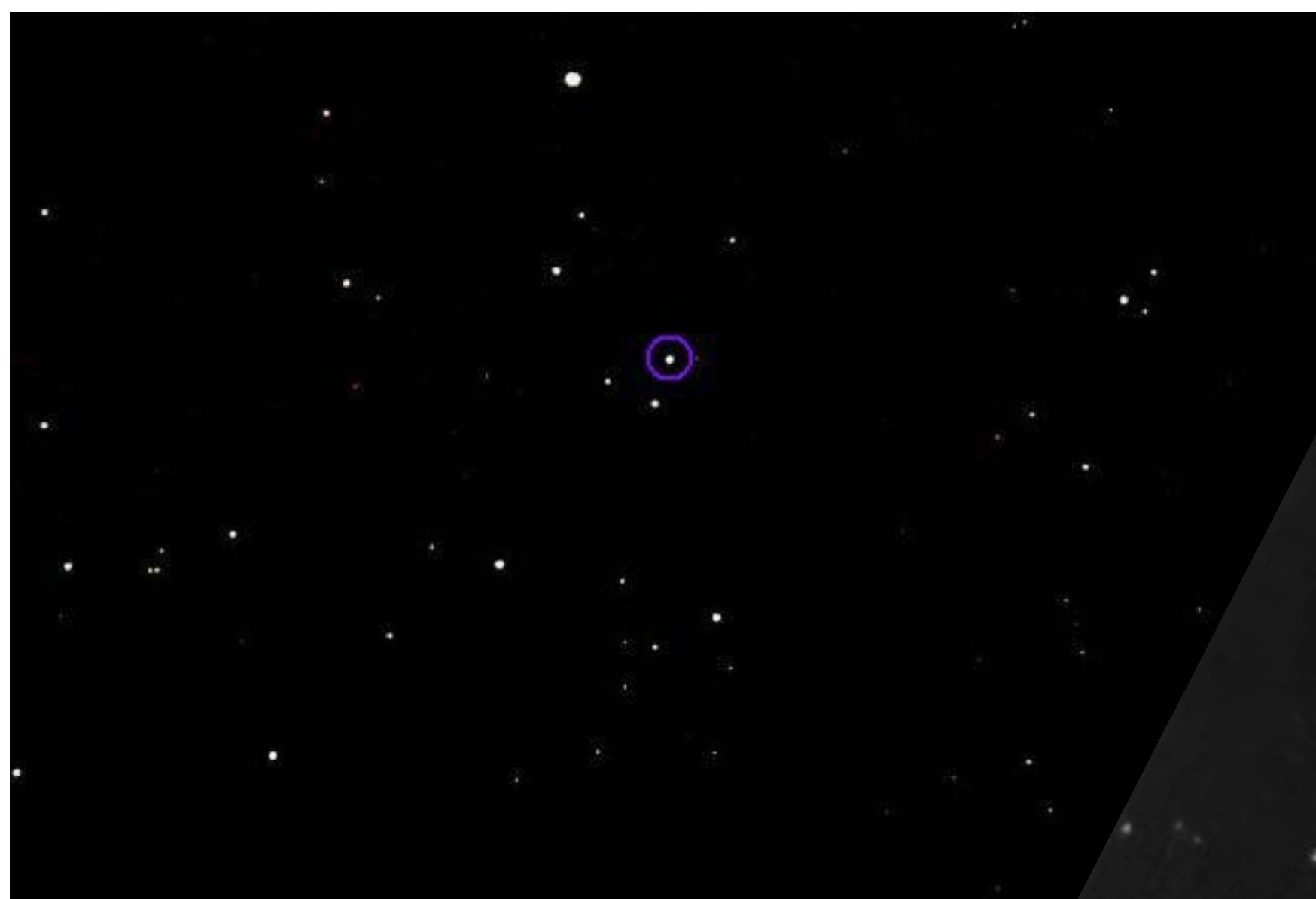


moving away from  
you

$$\lambda = \lambda_0 * (1 + v/c)$$

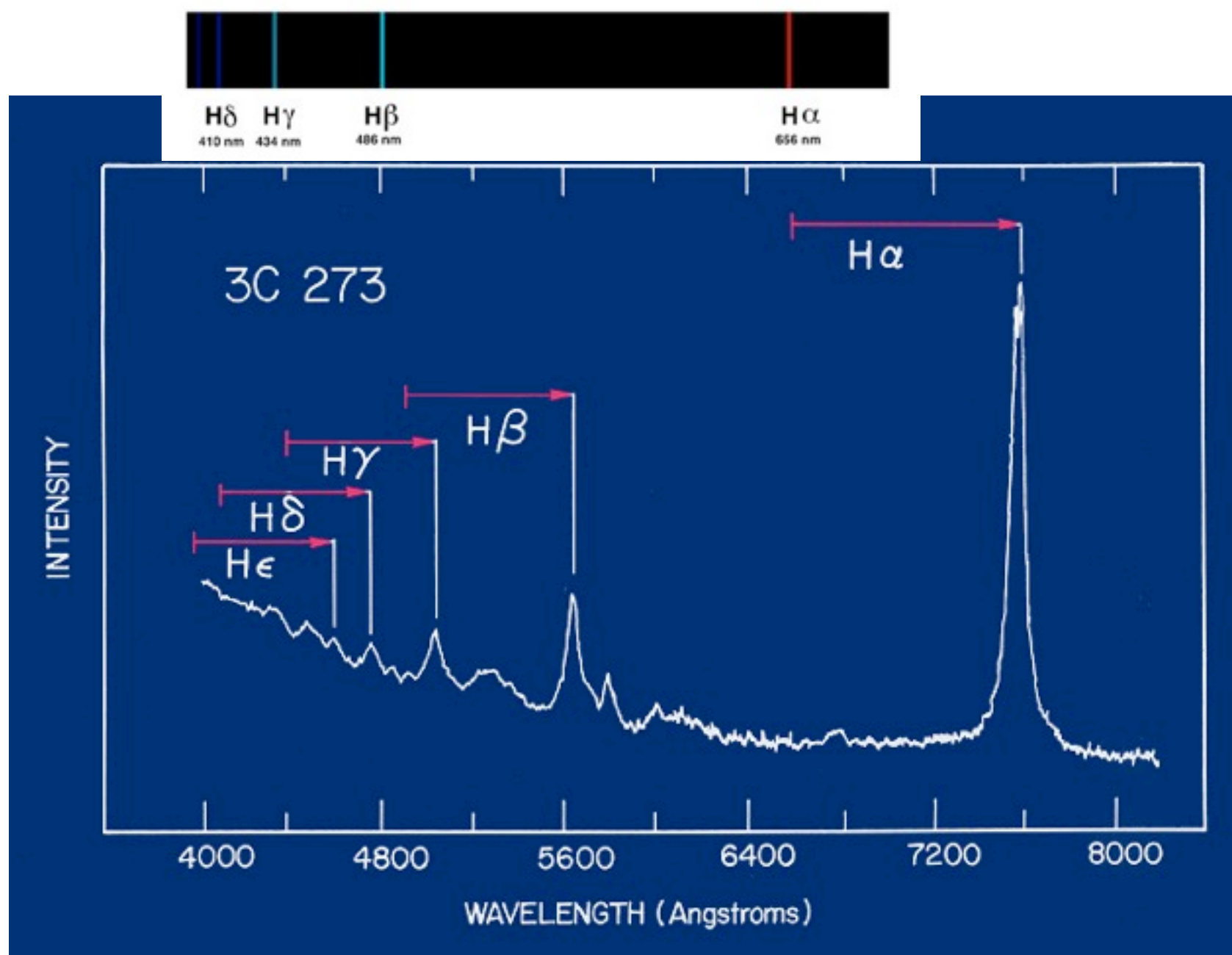
light red-shifted

# Quasar 3C 273



quasar (quasi-stellar object):  
star-like point sources but at  
cosmological distances. Their  
extremely high intrinsic  
luminosities are powered by  
massive blackholes.

# Spectrum of Quasar 3C 273



quasar discovered by Maarten Schmidt (1963)  
in 3C 273: all hydrogen lines red-shifted by 15%:

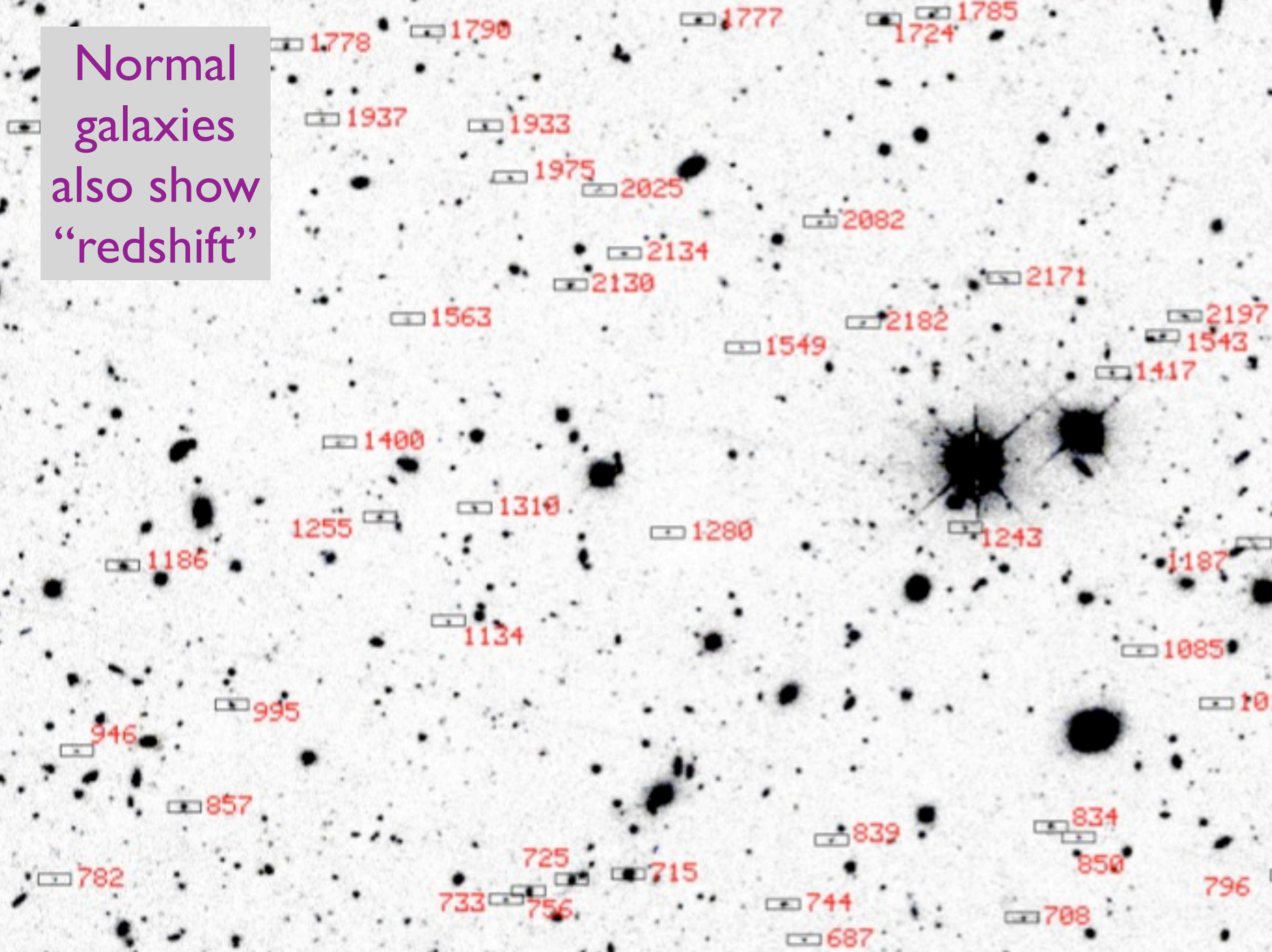
$$z = 0.15$$

define redshift:

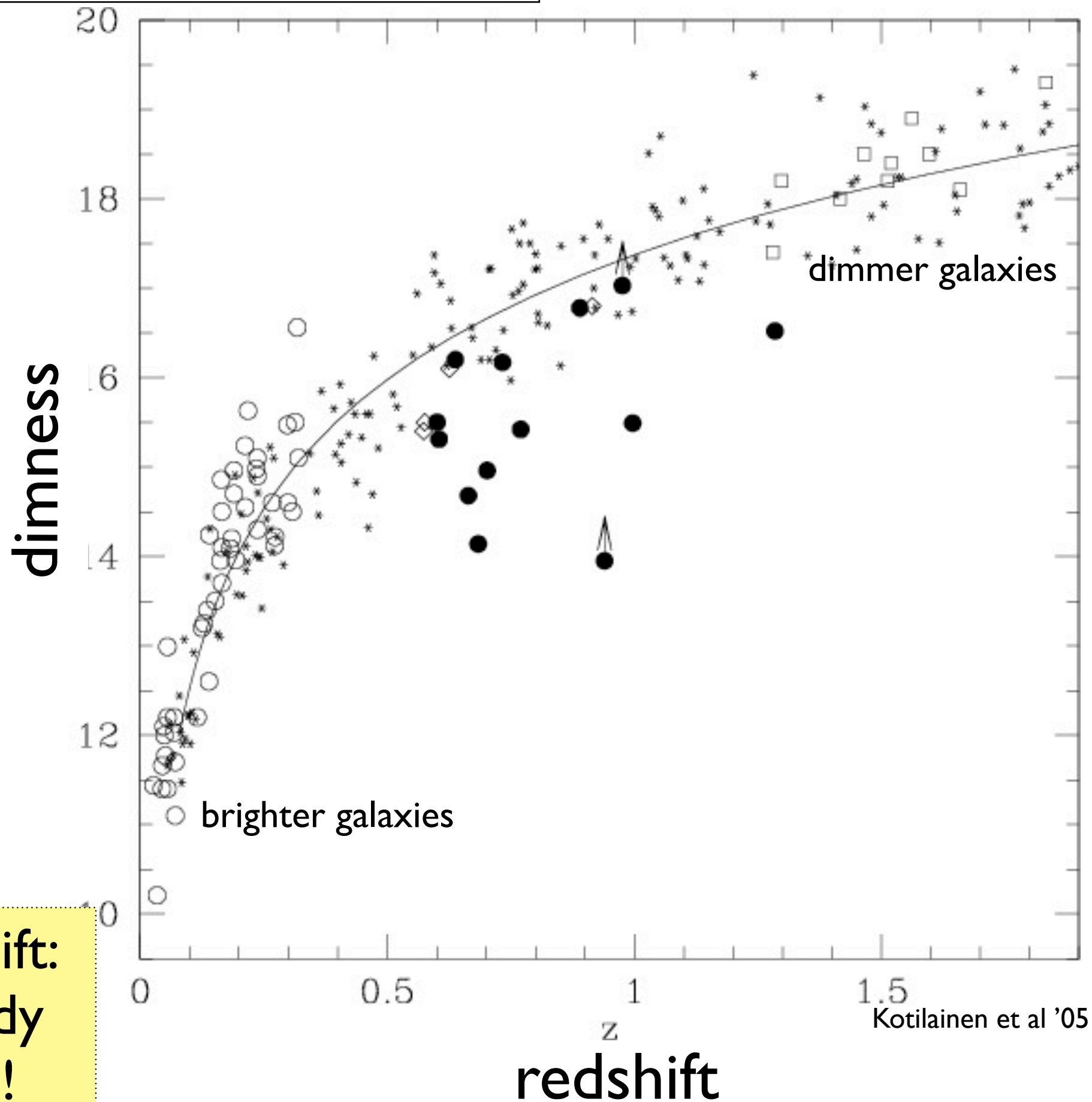
$$z = \lambda/\lambda_0 - 1 = v/c$$



Normal galaxies also show “redshift”



define redshift:  $z = \lambda/\lambda_0 - 1$



blueshift:  
nobody  
here!



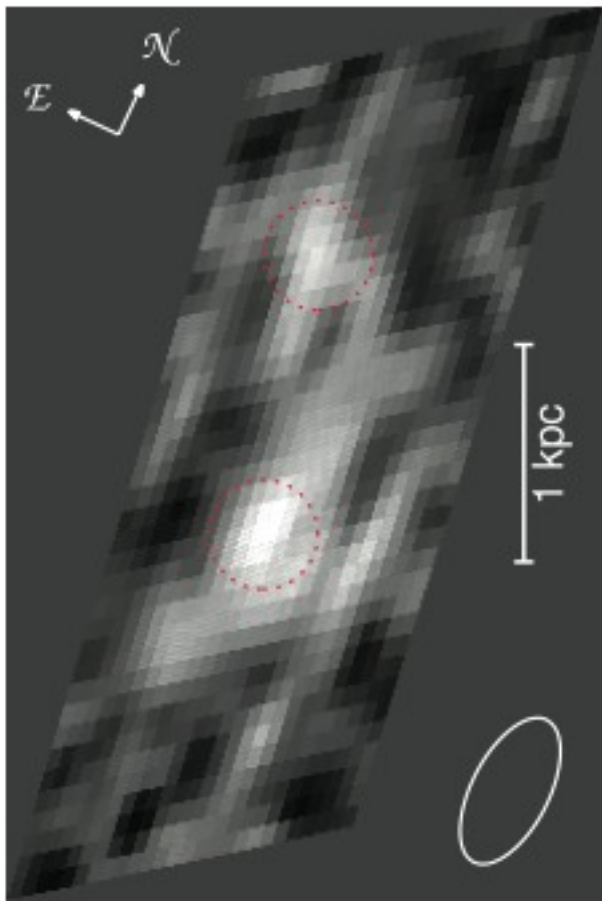
# galaxy holding the redshift record:

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## DISCOVERY OF A VERY BRIGHT STRONGLY LENSED GALAXY CANDIDATE AT $z \approx 7.6$ <sup>1</sup>

L. D. BRADLEY,<sup>2</sup> R. J. BOUWENS,<sup>3</sup> H. C. FORD,<sup>2</sup> G. D. ILLINGWORTH,<sup>3</sup> M. J. JEE,<sup>4</sup> N. BENÍTEZ,<sup>5</sup> T. J. BROADHURST,<sup>6</sup>  
M. FRANX,<sup>7</sup> B. L. FRYE,<sup>8</sup> L. INFANTE,<sup>9</sup> V. MOTTA,<sup>10</sup> P. ROSATI,<sup>11</sup> R. L. WHITE,<sup>12</sup> AND W. ZHENG<sup>2</sup>

*Received 2007 September 21; accepted 2008 January 8*



the Cosmic  
Microwave  
Background is  
at  $z \sim 1100$

# cosmological redshift

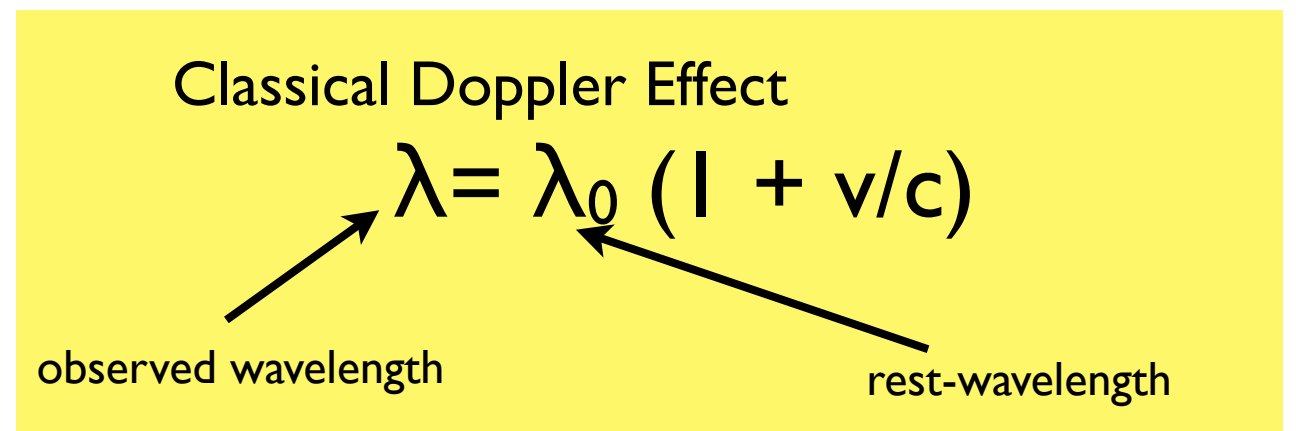
$$z = \lambda/\lambda_0 - 1$$

. how fast are these objects receding from us? can it be faster than speed of light?

Classical Doppler Effect

$$\lambda = \lambda_0 (1 + v/c)$$

observed wavelength      rest-wavelength

A yellow rectangular box containing the text 'Classical Doppler Effect' at the top. Below it is the equation  $\lambda = \lambda_0 (1 + v/c)$ . An arrow points from the text 'observed wavelength' to the variable  $\lambda$ . Another arrow points from the text 'rest-wavelength' to the variable  $\lambda_0$ .

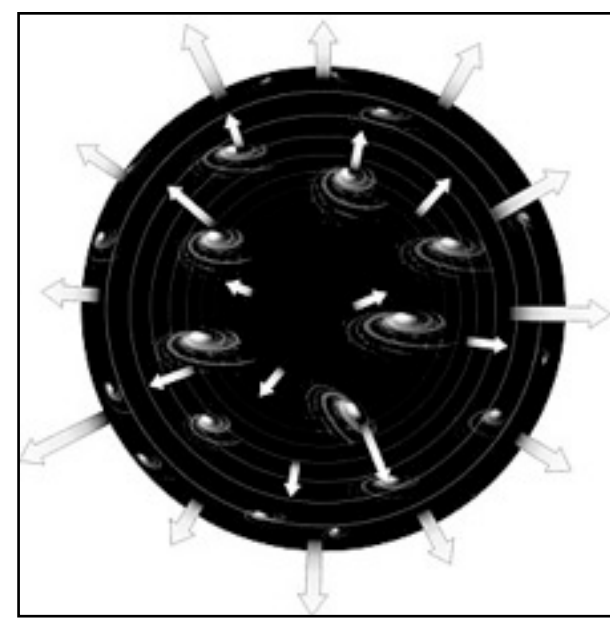
. why are we left in the middle?

we will discuss these in later lectures



All are **red**shifted.

The universe is expanding!



this was first pointed out by Edwin Hubble in the 1920s. It is one of the greatest scientific discoveries of the 20th century.

