

Assignment X

1. $4H \rightarrow {}^4He + \text{energy (in positrons, neutrinos and gamma rays)}$

$$E = (\Delta m)c^2 = (m_{4H} - m_{{}^4He})c^2 = \text{energy in positrons, etc.}$$

$$= (4 \cdot 1.6737 \cdot 10^{-27} \text{ kg} - 6.6447 \cdot 10^{-27} \text{ kg}) \cdot (3 \cdot 10^8 \text{ m/s})^2$$

$$= 4.51 \cdot 10^{-12} \text{ J (28 MeV)}$$

The sun is $\frac{3}{4}$ H by mass, so it has $0.75 M_\odot / m_H$

$$= 0.75 \cdot \frac{1.9891 \cdot 10^{30} \text{ kg}}{1.6737 \cdot 10^{-27} \text{ kg}} = 8.91 \cdot 10^{56} \text{ H atoms}$$

It takes 4H nuclei for one reaction, so $2.22 \cdot 10^{56}$ rxn.

$$\text{so } 2.22 \cdot 10^{56} \text{ rxn} \cdot 4.51 \cdot 10^{-12} \text{ J/rxn}$$

$$= 1.00 \cdot 10^{45} \text{ J}$$

This is the total energy emitted by the sun during its life.

$$\frac{1.00 \cdot 10^{45} \text{ J}}{3.84 \cdot 10^{26} \text{ W}} = 2.62 \cdot 10^{18} \text{ s} = \underline{\underline{82.9 \text{ Gyr}}}$$

The true lifetime of the sun is ~ 10 Gyr, because it doesn't burn all its hydrogen.

2. $T \propto \frac{1}{a} \propto 1+z$ (since $1+z(t) = \frac{a_0}{a(t)}$, so $1+z(t) \propto \frac{1}{a(t)}$)

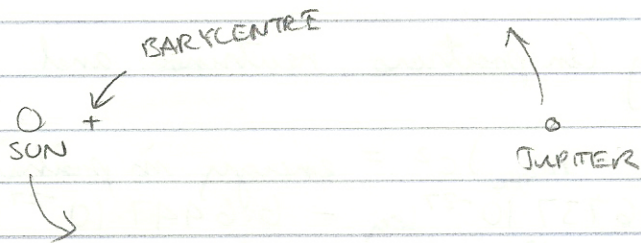
$$\frac{T_{SF}}{T_{CMB}} = \frac{1+z_{SF}}{1+z_{CMB}}$$

$$z_{SF} = \left(\frac{T_{SF}}{T_{CMB}} \right) (1+z_{CMB}) - 1$$

$$= \frac{100\text{K}}{3000\text{K}} (1+1100) - 1 = \underline{\underline{35.7}}$$

This is higher than the age redshift of HD1523-0901

3.



If the CM is fixed
in space, then
 $V_{CM} = 0$

$$V_{CM} = \frac{M_{\odot} V_{\odot} + M_J V_J}{M_{total}} = 0$$

$$M_{\odot} V_{\odot} = -M_J V_J$$

$$V_{\odot} = -\frac{M_J}{M_{\odot}} V_J \rightarrow |V_{\odot}| = \text{magnitude of vel.} = \frac{M_J}{M_{\odot}} V_J$$

$$|V_{\odot}| \approx \frac{1}{1000} \cdot 13 \text{ km/s} = 0.013 \text{ km/s} = 13 \text{ m/s}$$

$$\frac{\Delta \lambda}{\lambda} = \frac{v}{c} = \frac{13 \text{ m/s}}{3 \cdot 10^8 \text{ m/s}} = \underline{4.33 \cdot 10^{-8}}$$

We can actually do better than this with today's telescopes.