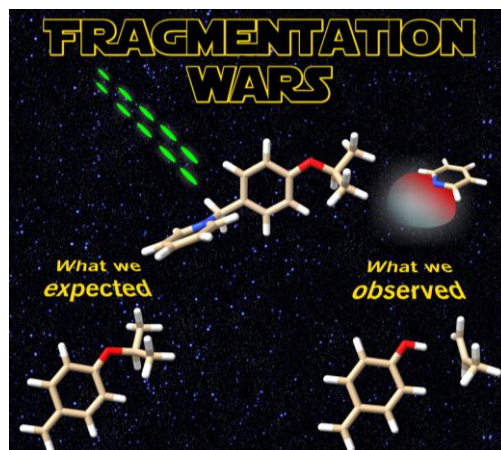


# Benzylpyridinium 'Thermometer' Ions can Fragment Through an Unexpected Intramolecular Elimination: These are Not the Fragments You are Looking For

Christian Ieritano, W. Scott Hopkins

Hopkins Group

37<sup>th</sup> Annual Trent Conference for Mass Spectrometry



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2021, 12, 5994-5999

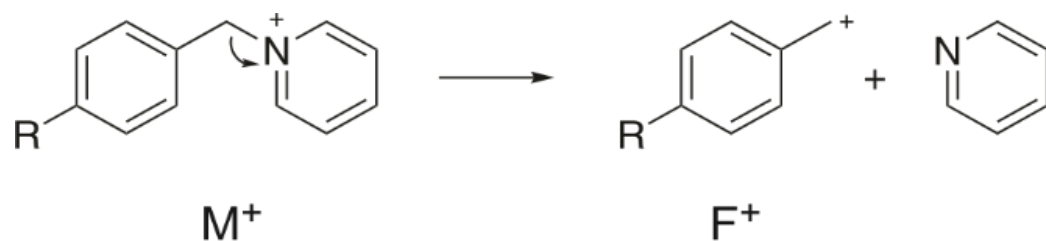


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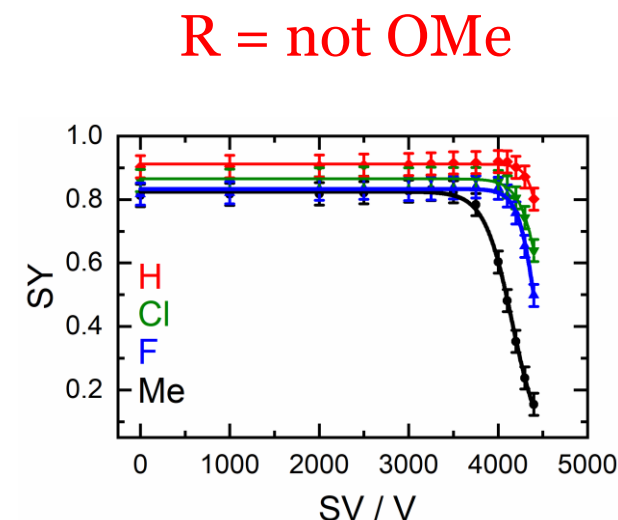
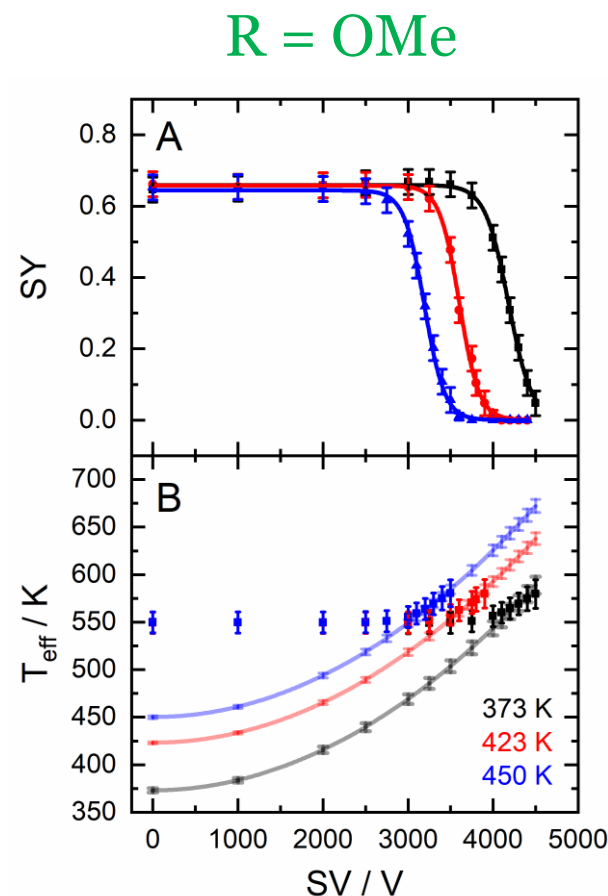
FACULTY  
OF SCIENCE

# A long time ago, in a galaxy far away when COVID wasn't a thing...

We were able to characterize the effective temperature induced by the separation field in differential mobility spectrometry (DMS) using benzyropyridinium thermometer ions



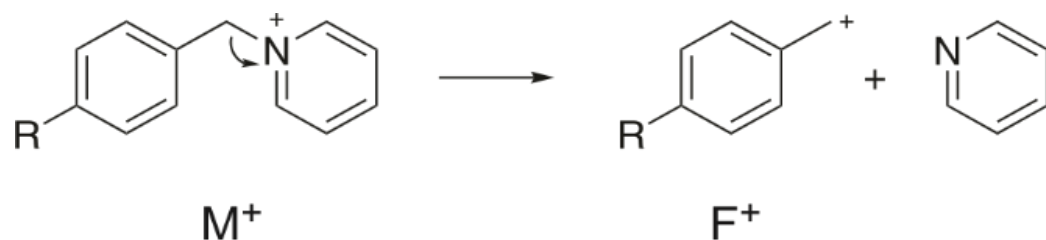
R	$E_o / \text{kJ mol}^{-1}$
<b>OMe</b>	<b>178.5</b>
Me	219.0
F	230.6
Cl	228.7
H	241.2
CN	264.4



Requires high bath gas temp AND high SV

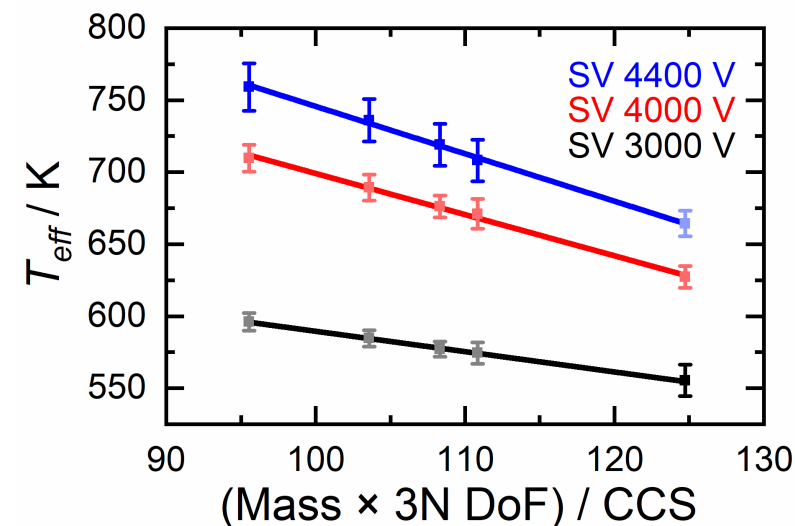
# A long time ago, in a galaxy far away when COVID wasn't a thing...

We were able to characterize the effective temperature induced by the separation field using benzyropyridinium thermometer ions



R	$E_o / \text{kJ mol}^{-1}$
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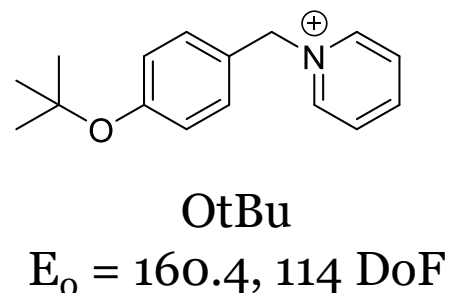
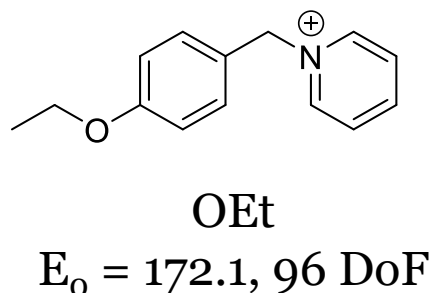
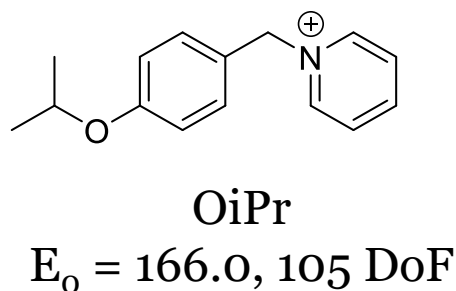
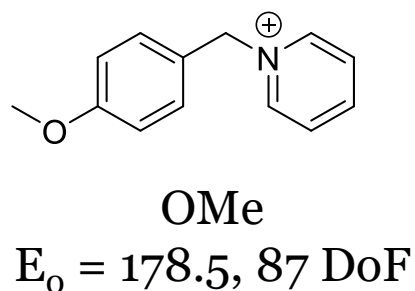
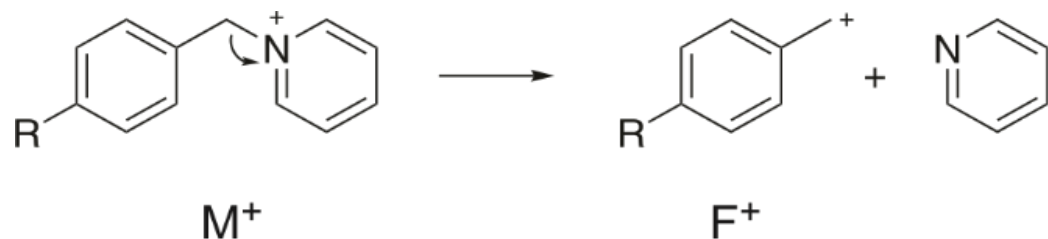
Linear relationship between ion  $T_{\text{eff}}$  and its density of states / collision probability



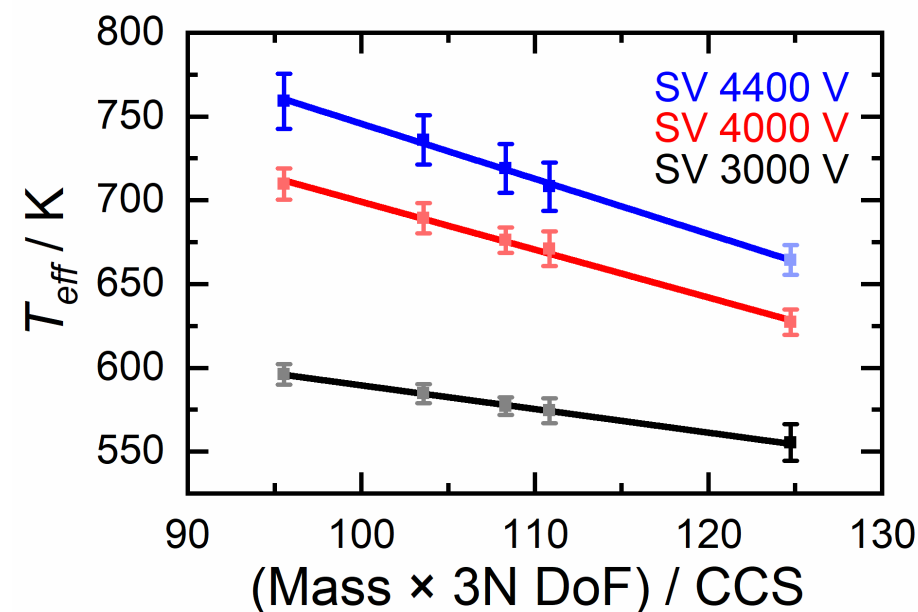
How can we extend this to lower bath gas temperatures?

# A long time ago, in a galaxy far away when COVID wasn't a thing...

Give the methoxy BP ions more rovibrational states to funnel energy into

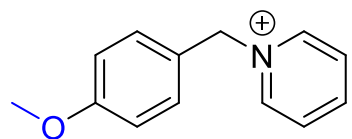
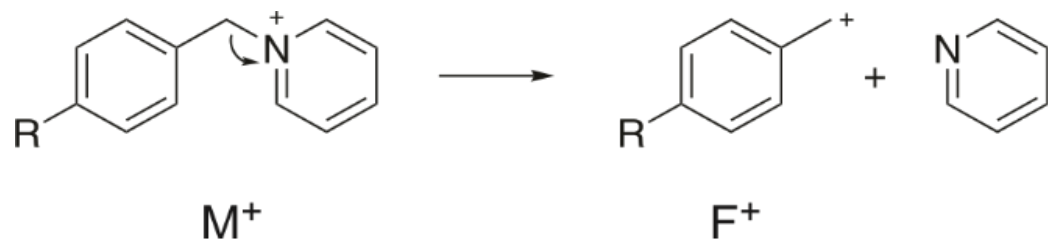


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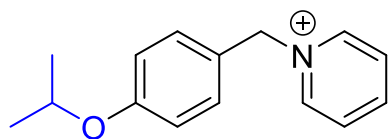
# And that's when things go crazy

Loss of pyridine **is not the dominant fragmentation pathway**



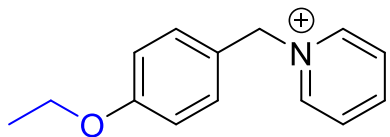
OMe

$E_0 = 178.5$ , **87 DoF**



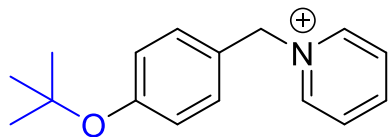
OiPr

$E_0 = 166.0$ , **105 DoF**



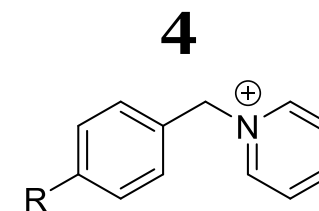
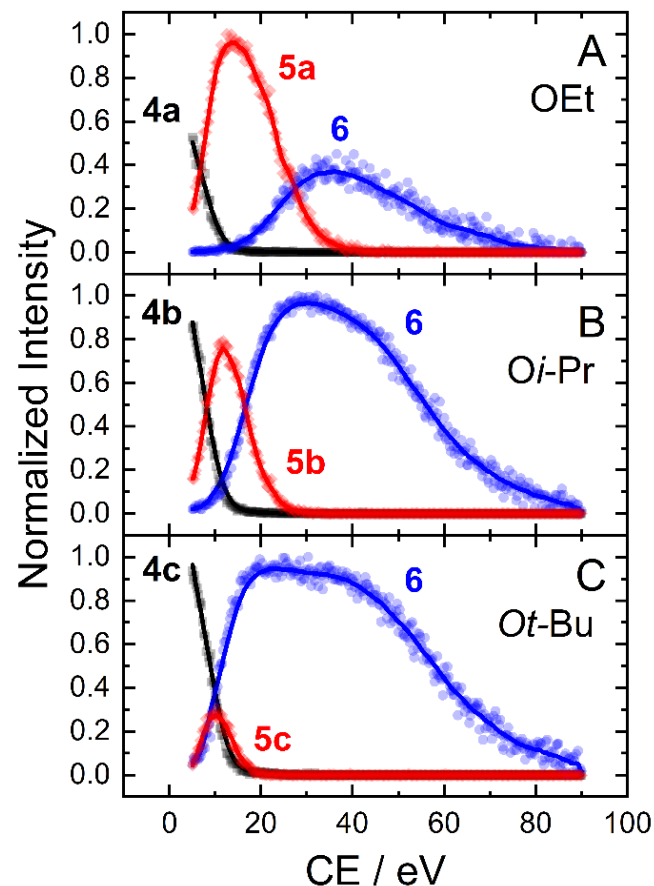
OEt

$E_0 = 172.1$ , **96 DoF**

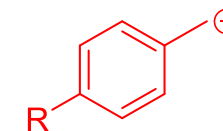


OtBu

$E_0 = 160.4$ , **114 DoF**



**5**

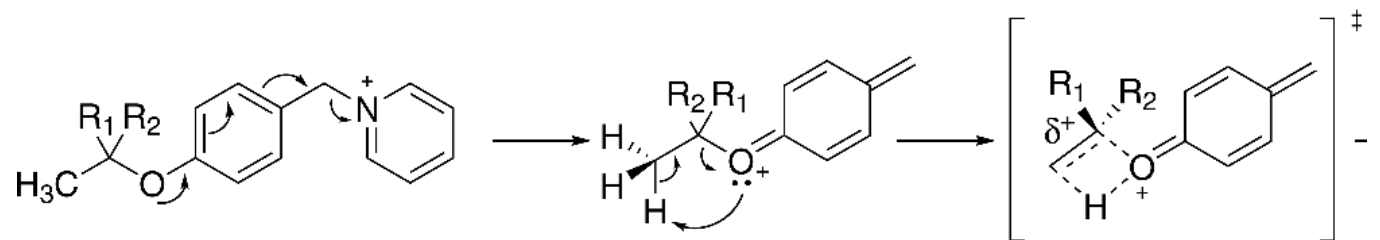


**6**

**$m/z$  107 ???**

# Breakdown behaviour gives insight into what's happening

Formation of 6 is **greater** for BP ions with bulkier side chains



**4a** ( $R_1 = R_2 = \text{H}$ )

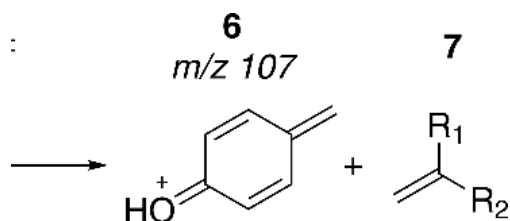
**4b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

**4c** ( $R_1 = R_2 = \text{Me}$ )

**5a** ( $R_1 = R_2 = \text{H}$ )

**5b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

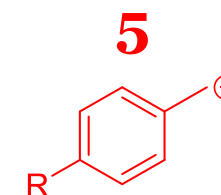
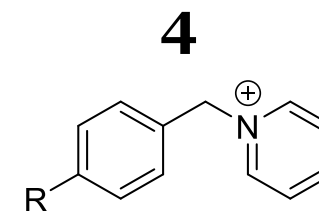
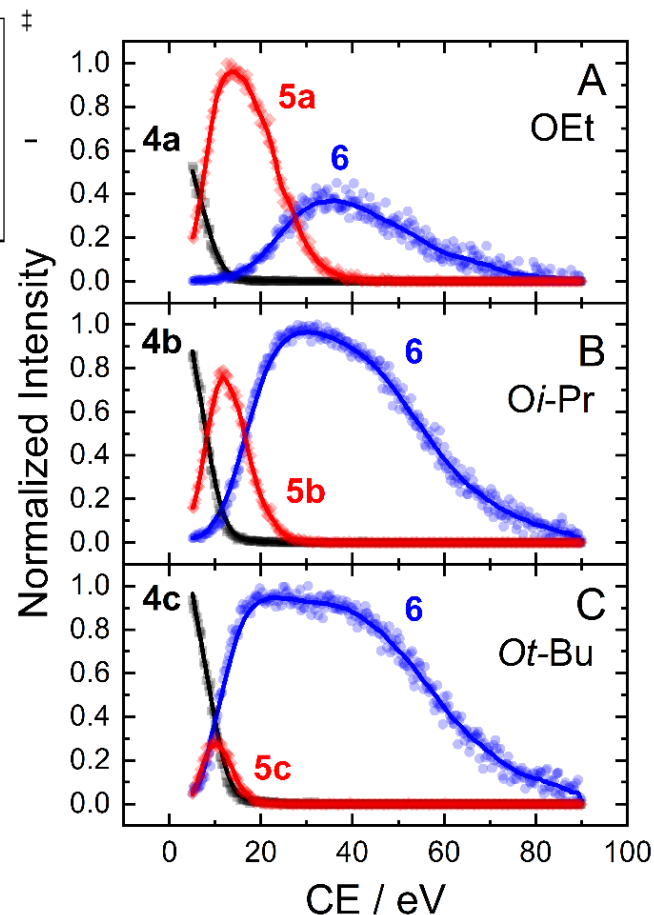
**5c** ( $R_1 = R_2 = \text{Me}$ )



**7a** ( $R_1 = R_2 = \text{H}$ )

**7b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

**7c** ( $R_1 = R_2 = \text{Me}$ )



**6**  
*m/z* 107 ???

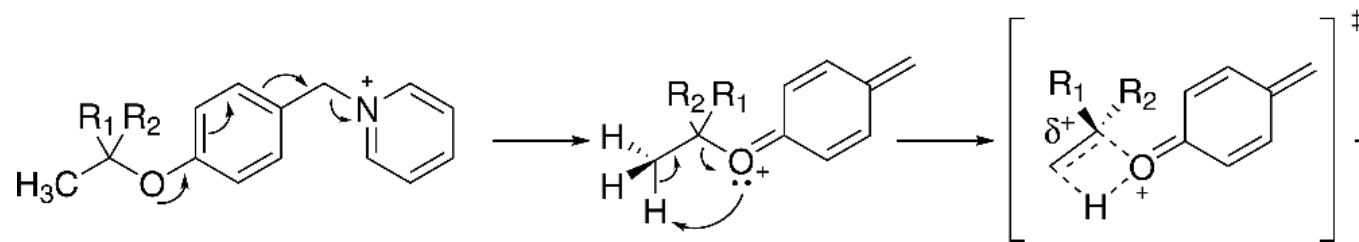
# Figuring out wtf is going on

The only way to form 107 is via an intramolecular elimination that passes through a 4-membered ring TS.

2 major implications:

$$1. \quad \Delta G^\ddagger \text{ OEt} > \text{OiPr} > \text{OtBu}$$

Hyperconjugation effect ( $E_i$  like)



**4a** ( $R_1 = R_2 = \text{H}$ )

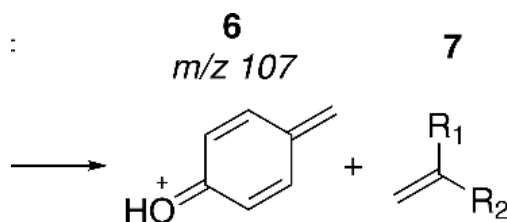
**4b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

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**5a** ( $R_1 = R_2 = \text{H}$ )

**5b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

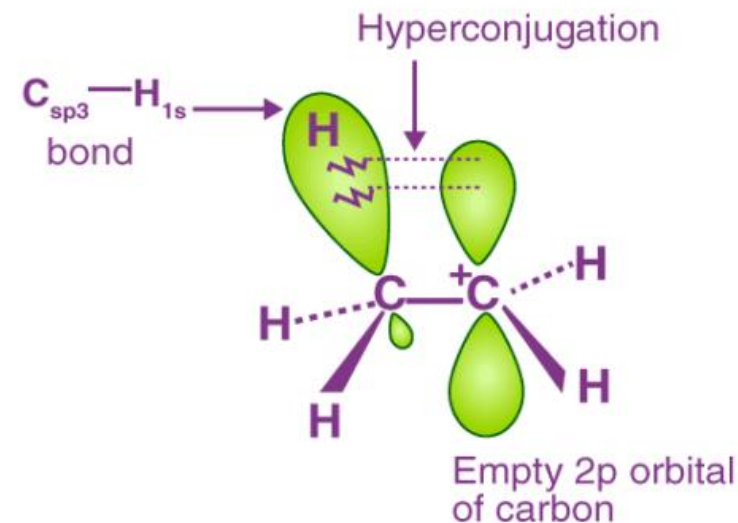
**5c** ( $R_1 = R_2 = \text{Me}$ )



**7a** ( $R_1 = R_2 = \text{H}$ )

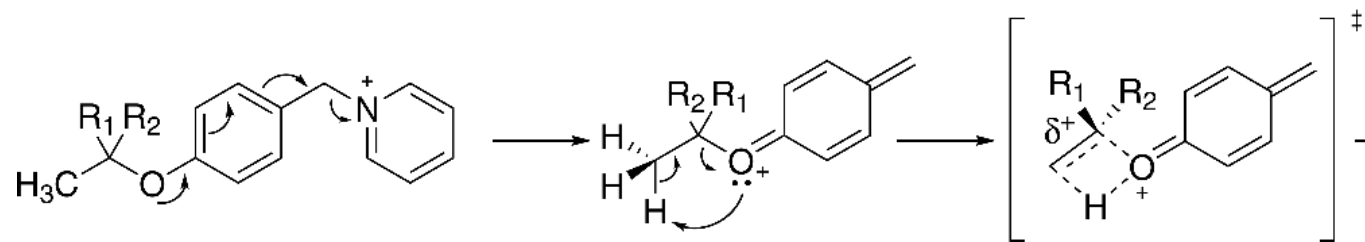
**7b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

**7c** ( $R_1 = R_2 = \text{Me}$ )



# Figuring out wtf is going on

The only way to form 107 is via an intramolecular elimination that passes through a 4-membered ring TS.



**4a** ( $R_1 = R_2 = \text{H}$ )

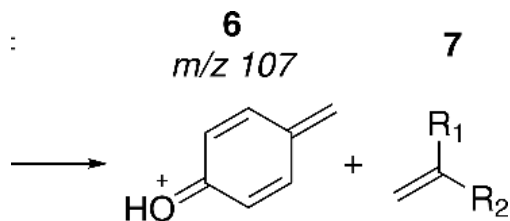
**4b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

**4c** ( $R_1 = R_2 = \text{Me}$ )

**5a** ( $R_1 = R_2 = \text{H}$ )

**5b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

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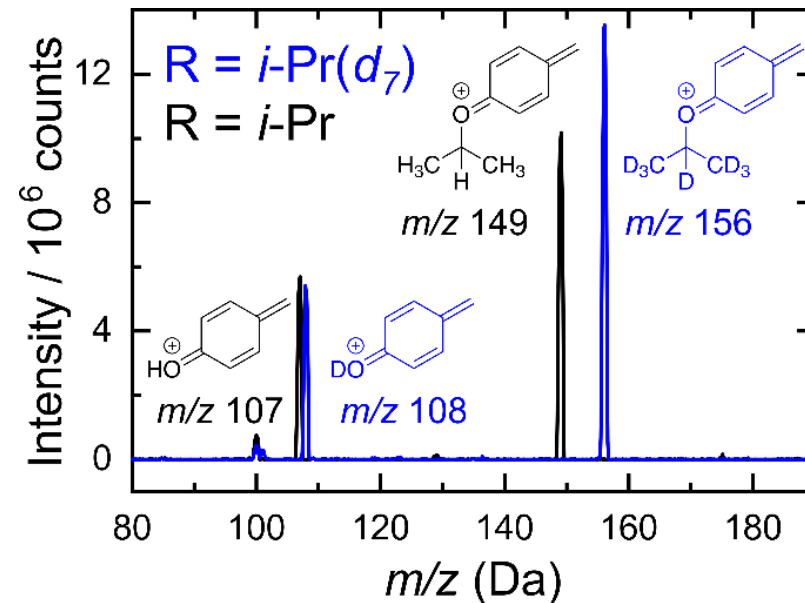
**7a** ( $R_1 = R_2 = \text{H}$ )

**7b** ( $R_1 = \text{H}; R_2 = \text{Me}$ )

**7c** ( $R_1 = R_2 = \text{Me}$ )

2 major implications:

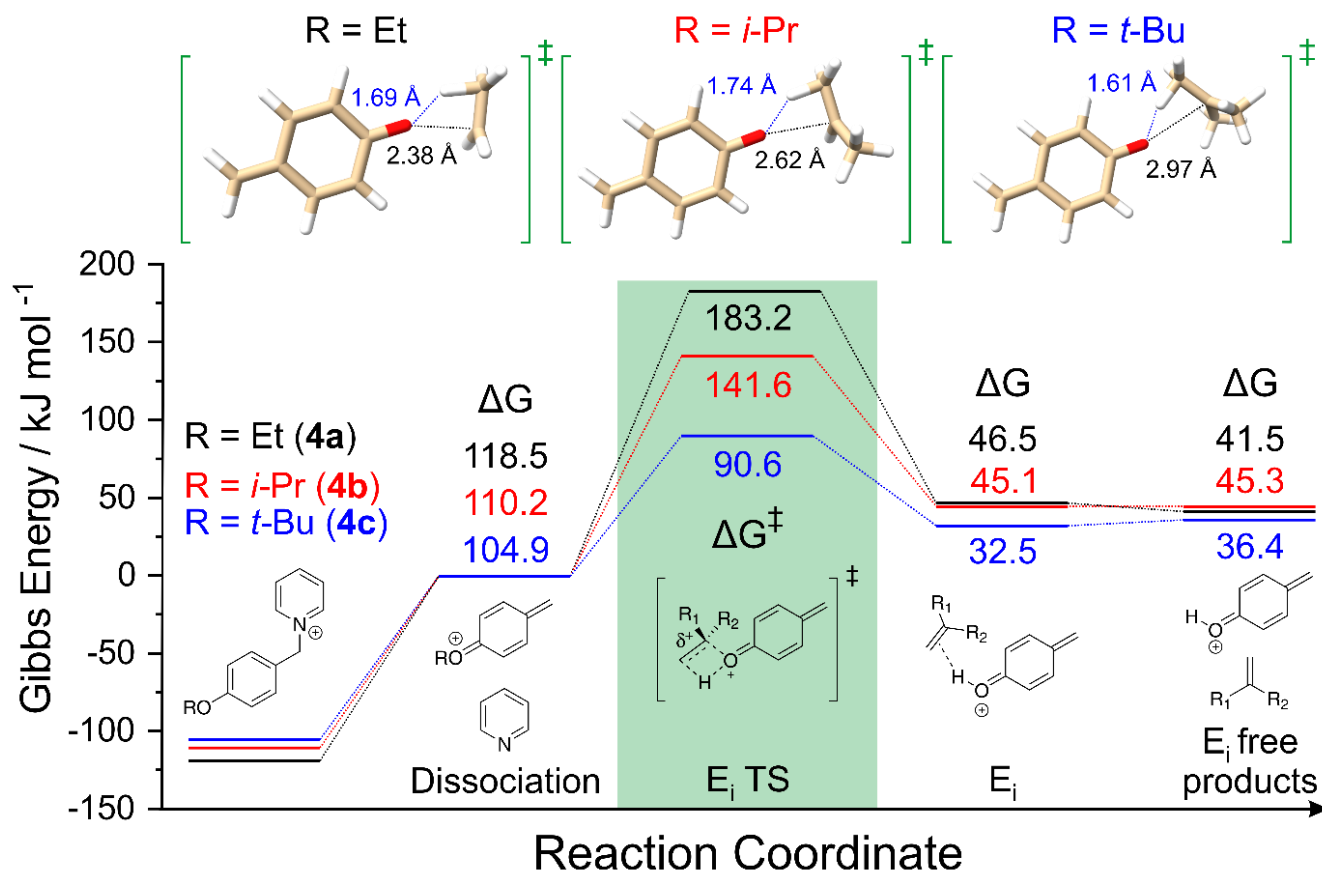
- Intramolecular elimination requires proton incorporation into the product



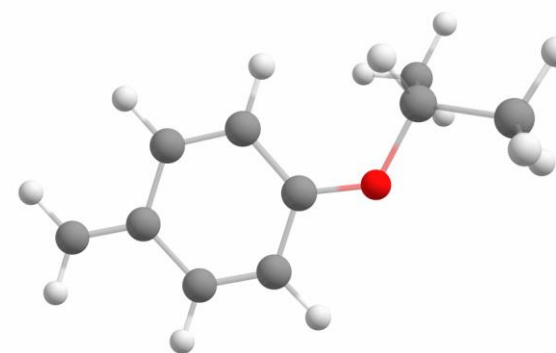


# Finally understanding what was going on

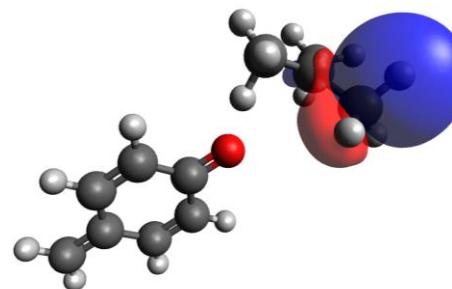
Mapping the reaction coordinate using the nudged elastic band confirms our proposed fragmentation mechanism



Reaction pathway



Hyperconjugation in the TS



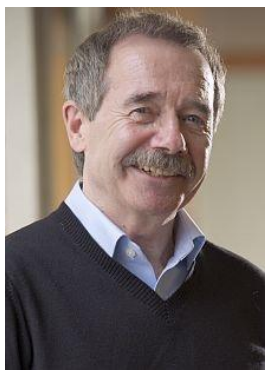
OEt: 24.0  $\text{kJ mol}^{-1}$   
*Oi*Pr: 206.0  $\text{kJ mol}^{-1}$   
*Ot*Bu: 759.8  $\text{kJ mol}^{-1}$

# Acknowledgements

## PhD Committee



Prof. W. Scott  
Hopkins



Prof. Terry B.  
McMahon



Adj. Prof. J.  
Larry Campbell

## Hopkins Lab

Dan Rickert  
Dr. Joshua Featherstone

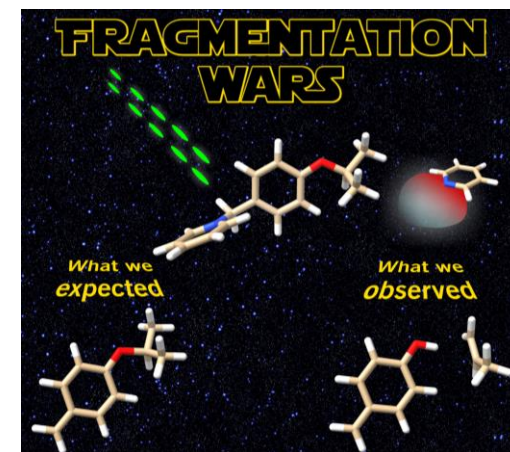
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Justine Bissonnette  
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Dr. Mircea Guna

For more details, see  
our publication:

*J. Phys. Chem. Lett.*,  
2021, 12, 5994-5999



## Collaborators, Resources, and Funding



compute CANADA

