

Calculation of methane production from volumetric measurements

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1 BMP-methods

File version 1.2. This file is from the GitHub repository BMP-methods. For more information, visit BMP-methods at <https://github.com/sashahafner/BMP-methods>.

2 Description

This document describes calculations for volumetric measurement of biogas. As with manometric methods, two methods are commonly used and both are described here: one based on normalized CH₄ concentrations (method 1) and one that explicitly includes estimation of CH₄ in the bottle headspace (method 2). Expected results from the two methods are identical; differences are due only to error in measurement of biogas composition or headspace volume. Both methods are available through the `cumBg()` function in the biogas package [1] and through the web application OBA (<https://biotransformers.shinyapps.io/oba1/>) and can be easily added to, e.g., a spreadsheet template.

3 Standardization of measured gas volume

Both methods use the same approach for standardization of gas volume. Dry biogas volume in a bottle's headspace before and after venting is calculated by correcting for water vapor, temperature, and pressure. First the measured gas volume (e.g., in a syringe or hanging water column) is converted to dry conditions at standard pressure:

$$V_{dry} = V_{headspace}(P_{meas} - P_{H_2O})/101.325 \text{ kPa} \quad (1)$$

where P_{meas} is the measured headspace pressure and P_{H_2O} the water vapor partial pressure (both in kPa). Eq. (1) is an expression of Boyle's law. The

value of P_{H_2O} is assumed to be the saturation vapor pressure, and can be calculated using, e.g., the Magnus-form equation given below (Eq. 21 in [2]):

$$P_{H_2O} = 0.61094e^{(17.625T/(243.04+T))} \quad (2)$$

where T is temperature in °C. Volume is then further standardized to 273.15 K by application of Charles's law:

$$V_{std} = V_{dry}273.15\text{ K}/T_{meas} \quad (3)$$

where V_{std} is the standardized volume of gas within a bottle's headspace at the time of pressure measurement. Interval biogas production $V_{biogas,i}$ is taken as this standardized volume v_{std} . Cumulative production is taken as the cumulative sum of interval values.

4 Calculation of CH₄ production

4.1 Method 1

In the first method, biogas is assumed to consist of only CH₄ and CO₂ at the time of production (i.e., as produced by the microbial community) and CH₄ production is calculated from vented (removed) biogas only. This method is described in [3]. Coupled with the assumption that all gas production is biogas, this provides the simplest approach for calculating CH₄ production.

First, concentrations of CH₄ and CO₂ are adjusted so they sum to 1.0:

$$x_{CH_4,n} = x_{CH_4}/(x_{CH_4} + x_{CO_2}) \quad (4)$$

where x_{CH_4} and x_{CO_2} are the measured CH₄ and CO₂ concentrations as volume (mole) fraction (possibly including a correction for water vapor—this has no effect here) and $x_{CH_4,n}$ is the normalized CH₄ volume fraction.

Methane production in an interval i is then calculated as

$$V_{CH_4,i} = x_{CH_4,n}V_{biogas,i} \quad (5)$$

Cumulative production is taken as the cumulative sum of interval values.

4.2 Method 2

Method 2 relies on fewer assumptions, but requires the true concentration of CH₄ (volume fraction) of CH₄ within the bottle headspace, with correction only for water vapor. Here, CH₄ production in an interval has two components: a vented part that is naturally interval, and a residual headspace part, that is naturally cumulative:

$$V_{CH_4,i} = V_{CH_4,v,i} + (V_{CH_4,HSR,i} - V_{CH_4,HSR,i-1}) \quad (6)$$

where the subscript v indicates vented volume and HSR = residual headspace volume (post-venting).

Vented CH_4 is calculated from:

$$V_{CH_4,v,i} = x_{CH_4,n,i} V_{biogas,i} \quad (7)$$

Headspace CH_4 is calculated from:

$$V_{CH_4,HSR,i} = x_{CH_4,n,i} V_{post,i} \quad (8)$$

where V_{post} is the post-venting standardized volume of gas in the bottle headspace.

Cumulative production is taken as the cumulative sum of interval values.

5 Example Calculations

In the following example, CH_4 production is calculated from a single interval measurement made on a single bottle in a BMP trial. Calculations are made using both volumetric method 1 and 2.

For both methods standardized gas volume is calculated from Eq. (3) by correcting for water vapor, temperature, and pressure. Measured biogas volume ($V_{biogas,i}$) was 618 mL at a temperature (T_{meas}) of 20°C and atmospheric pressure (P_{meas}) of 101.325 kPa. First water vapor pressure is calculated at the measured headspace temperature using Eq (2).

$$P_{H_2O} = 0.61094 \cdot e^{\frac{17.625 \cdot 20^\circ\text{C}}{243.04 + 20^\circ\text{C}}} = 2.333 \text{ kPa}$$

Secondly, the headspace volume is converted to dry conditions at standard pressure using Eq. (1).

$$V_{dry} = \frac{618 \text{ mL} \cdot (101.325 \text{ kPa} - 2.333 \text{ kPa})}{101.325 \text{ kPa}} = 592.158 \text{ mL}$$

Then, volume is further standardized following Eq. (3).

$$V_{std} = \frac{592.158 \text{ mL} \cdot 273.15 \text{ K}}{293.15 \text{ K}} = 551.758 \text{ mL}$$

$V_{biogas,i}$ is taken as this V_{std} . Cumulative production is taken as the cumulative sum of interval values.

5.1 Method 1

The mole fraction of CH_4 (x_{CH_4} , dimensionless) normalized for CH_4 and CO_2 can be calculated according Eq. (4), but was given as 0.627. Then, following Eq. (5), CH_4 production in the interval is calculated from interval biogas production.

$$V_{CH_4,i} = 0.672 \cdot 551.758 \text{ mL} = 345.952 \text{ mL}$$

5.2 Method 2

Post-venting dry and standardized volume for current interval (i) and the previous interval ($i-1$) is required for method 2 and can be calculated following Eq. (1) and (3), respectively. Post headspace pressure (P_{post}) was assumed to be constant at 1.01 bar throughout the BMP trial.

Assuming constant biogas composition, CH_4 production in the interval can be calculated following Eq. (6). Under the assumptions of constant post headspace pressure and gas composition, Eq. (6) reduces to Eq. (7). Meaning that CH_4 production in the interval is equal to vented CH_4 volume and hence, method 2 equals method 1:

$$V_{CH_4,i} = V_{CH_4,v,i} = 345.952 \text{ mL}$$

References

- [1] Hafner, S.D., Koch, K., Carrere, H., Astals, S., Weinrich, S., Rennuit, C. 2018 Software for biogas research: Tools for measurement and prediction of methane production. *SoftwareX* 7: 205-210
- [2] Alduchov, O.A., Eskridge, R.E. 1996 Improved Magnus form approximation of saturation vapor pressure. *Journal of Applied Meteorology* 35: 601-609
- [3] Richards, B.K., Cummings, R.J., White, T.E., Jewell, W.J. 1991 Methods for kinetic-analysis of methane fermentation in high solids biomass digesters. *Biomass and Bioenergy* 1: 65-73