## **Opening of Terminal Epoxides:**

• Nucleophilic opening of terminal epoxides is often highly regioselective

Behrens, C. H.; Sharpless, K. B.; Aldrichimica Acta, 1983, 16, 67-80.

## Payne Rearrangement

HO CH<sub>3</sub> 
$$H_3$$
C CH<sub>3</sub>  $H_2$ O  $H_3$ C  $H_3$ C

• Steric factors permitting, equilibrium generally favors the more substituted epoxide.

Payne, G. B. J. Org. Chem. 1962, 27, 3819-3822

## Payne Rearrangement-Opening Sequence:

• β-Hydroxy sulfides are readily converted into terminal epoxides.

Behrens, C. H.; Sharpless, K. B.; Aldrichimica Acta, 1983, 16, 67-80.

## 2,3-Epoxy alcohols:

• Ti(O-i-Pr)<sub>4</sub> can catalyze the addition of nucleophiles to C3 of 2,3-epoxy alcohols:

	Ti(O <i>i-</i> Pr) <sub>4</sub>		
Nucleophile	(equiv)	C3 : C2	yield
Et <sub>2</sub> NH	0	3.7 : 1	4
Et <sub>2</sub> NH	1.5	20 : 1	90
<i>i</i> -PrOH	0	-	0
<i>i</i> -PrOH	1.5	100 : 1	88
(allyl) <sub>2</sub> NH	1.5	100 : 1	96
allyl alcohol	1.5	100 : 1	90
NH <sub>4</sub> OBz	1.5	100 : 1	74
NH <sub>4</sub> OAc	1.5	65 : 1	73
KCN	1.7	2.4 : 1	76
	Et <sub>2</sub> NH Et <sub>2</sub> NH <i>i</i> -PrOH <i>i</i> -PrOH (allyl) <sub>2</sub> NH allyl alcohol NH <sub>4</sub> OBz NH <sub>4</sub> OAc	Nucleophile         (equiv) $Et_2NH$ 0 $Et_2NH$ 1.5 $i$ -PrOH         0 $i$ -PrOH         1.5           (allyl) $_2$ NH         1.5           allyl alcohol         1.5           NH $_4$ OBz         1.5           NH $_4$ OAc         1.5	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

Caron, M.; Sharpless, K. B. J. Org. Chem. 1985, 50, 1557-1560.

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• Regioselectivity of uncatalyzed nucleophilic opening of 2,3-epoxy alcohols varies with the substrate and reaction conditions.

substrate	nucleophile	regioselectivity C3 : C2	combined yield (%)
H <sub>3</sub> C, CH <sub>3</sub>	NaN <sub>3</sub>	1 : 10	90
H <sub>3</sub> C, CH <sub>3</sub>	NaSPh H	1:>10	76
	H NaN <sub>3</sub>	1.4 : 1	71
	H NaSPh	1 : 1.4	72

Behrens, C. H.; Sharpless, K. B.; J. Org. Chem. 1985, 50, 5696-5704.

• Phenyl substitution at C3 of 2,3-epoxy alcohols can lead to high C3-regioselectivity.

Ph 
$$\stackrel{\bullet}{\longrightarrow}$$
 OH  $\stackrel{\bullet}{\longrightarrow}$  Nu  $\stackrel{\bullet}{\longrightarrow}$  OH  $\stackrel{\bullet}{\longrightarrow}$  OH  $\stackrel{\bullet}{\longrightarrow}$  OH  $\stackrel{\bullet}{\longrightarrow}$  Ph  $\stackrel{\bullet}{\longrightarrow}$  OH  $\stackrel{\bullet}{\longrightarrow}$  OH

Hanson, R. M. Chem. Rev. 1991, 91, 437-575, and references therein.

 $\bullet \ \text{C2 reduction of 2,3-epoxy alcohols using Red-Al is highly selective when C4 is oxygenated. } \\$ 

Red-AI =  $[(CH_3OCH_2CH_2O)_2AIH_2]^-Na^+$ 

epoxy alcohol	C2 reduction C2 : C3	C3 reduction yield (%)
n-C <sub>6</sub> H <sub>13</sub>	1:1	94
BnOOOOOO	5 : 1	89
BnO	40 : 1	98
H <sub>3</sub> C OH	>100 : 1	78
CH <sub>3</sub> OBn OH	100 : 1	95

Ma, P.; Martin, V. S.; Masamune, S.; Sharpless, K. B.; Viti, S. M. *J. Org. Chem.* **1982**, *47*, 1378–1380.

Finan, J.; Kishi Y. Tetrahedron Lett. 1982, 23, 2719-2722.

• 1,3-Bis-epoxides:

Ma, P.; Martin, V. S.; Masamune, S.; Sharpless, K. B.; Viti, S. M. *J. Org. Chem.* **1982**, *47*, 1378–1380.

Allylic epoxides:

Nicolaou, K. C.; Uenishi, J. J. Chem. Soc., Chem. Commun. 1982, 1292–1293.

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• The regioselectivity of epoxide opening can vary with the organometallic reagent.

BnO OH 
$$\frac{1. \text{"M(CH}_3)_n\text{"}}{2. \text{NaIO}_4 \text{THF:H}_2\text{O}}$$
 BnO CH<sub>3</sub>  $+$  BnO OH  $\frac{\text{CH}_3}{\text{OH}}$   $+$  BnO OH  $\frac{\text{CH}_$ 

Johnson, M. R.; Nakata, T.; Kishi, Y. *Tetrahedron Lett.* **1979**, 4343–4346. Roush, W. R.; Adam, M. H.; Peseckis, S. M. *Tetrahedron Lett.* **1983**, 1377–1380.

- AE of allyl alcohol followed by in situ derivatization affords versatile chiral building blocks, such as glycidol tosylate (now commercially available).
- Reactions of glycidol tosylate:

Klunder, J. M.; Onami, T.; Sharpless, K. B. *J. Org. Chem.* **1989**, *54*, 1295–1304. Hanson, R. M. *Chem. Rev.* **1991**, *91*, 437–475.

• Internal nucleophiles may be used to open 2,3-epoxy alcohols:

Corey, E. J.; Hopkins, P. B.; Munroe, J. E.; Marfat, A.; Hashimoto, S.-I. *J. Am. Chem. Soc.* **1980**, *102*, 7986–7987

Minami, N.; Ko, S. S.; Kishi, Y. J. Am. Chem. Soc. 1982, 104, 1109-1111.

Trost, B. M.; Sudhakar, A. R. J. Am. Chem. Soc. 1987, 109, 3792-3794.

HO 
$$\stackrel{\text{(H}_2\text{CO)}_n}{\text{Cs}_2\text{CO}_3}$$
 OH  $\stackrel{\text{OH}}{\text{CH}_3\text{CN}, 23 °C}$  Ph

McCombie, S. W.; Metz, W. A. Tetrahedron Lett. 1987, 28, 383-386.

Myers, A. G.; Widdowson, K. L. Tetrahedron Lett. 1988, 29, 6389-6392.

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