### Reviews:

Roy, M.-N.; Lindsay, V. N. G.; Charette, A. B. *Stereoselective Synthesis: Reactions of Carbon–Carbon Double Bonds (Science of Synthesis)*; de Vries, J. G., Ed.; Thieme: Stuttgart, **2011**, *Vol 1.*; 731–817.

Lebel, H.; Marcoux, J.-F.; Molinaro, C.; Charette, A. B. *Chem. Rev.* **2003**, *103*, 977–1050. Davies, H. M. L.; Beckwith, R. E. J. *Chem. Rev.* **2003**, *103*, 2861–2903. Li, A-H.; Dai, L. X.; Aggarwal, V. K. *Chem. Rev.* **1997**, 97, 2341–2372.

· Applications of Cyclopropanes in Synthesis

Carson, C. A.; Kerr, M. A. *Chem. Soc. Rev.* **2009**, *38*, 3051–3060. Reissig, H.-U.; Zimmer, R. *Chem. Rev.* **2003**, *103*, 1151–1196. Gnad, F.; Reiser, O. *Chem. Rev.* **2003**, *103*, 1603–1624.

Cyclopropane Biosynthesis

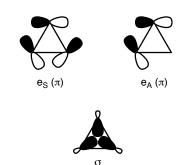
Thibodeaux, C. J.; Chang, W.-c.; Liu, H.-w. Chem. Rev. 2012, 112, 1681-1709.

#### Introduction

$$= \prod_{H} \prod_$$

- Cyclopropanes are stable but highly strained compounds (ring strain ~29 kcal/mol).
- C–C bond angles =  $60^{\circ}$  (vs  $109.5^{\circ}$  for normal  $C_{sp3}$ – $C_{sp3}$  bonds).
- Substituents on cyclopropanes are eclipsed. H–C–H angle is ~120°. As a result, the C–H bonds have higher s character compared to normal sp³ bonds.
- Because of their inherent strain, the reactivity of cyclopropanes is more closely analogous to that of alkenes than that of alkanes.

· Bonding Orbitals in Cyclopropane (Walsh Model):



de Meijere, A. Angew. Chem. Int. Ed. 1979, 18, 809-886.

## **General Strategies for Cyclopropanation:**

· via carbenoids

via carbenes generated by decomposition of diazo compounds

· via Michael addition and ring closure

$$\overline{RCH_2}$$
  $\overline{RCH_2}$   $\overline{RCH_2}$ 

Lebel, H.; Marcoux, J.-F.; Molinaro, C.; Charette, A. B. Chem. Rev. 2003, 103, 977-1050.

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## Simmons-Smith Reaction — Zinc Reagents in Cyclopropanation

· Original Report:

Simmons, H. E.; Smith, R. D. J. Am. Chem. Soc. 1958, 80, 5323-5324.

· Reaction Overview:

$$\begin{array}{c} R_1 \\ R_2 \end{array} \xrightarrow{"ZnCH_2|"} \begin{array}{c} Z_{1} \\ \vdots \\ R_{1} \\ \vdots \\ R_{2} \end{array} \xrightarrow{R_1} \begin{array}{c} R_1 \\ \vdots \\ R_2 \end{array} \end{array}$$

$$\begin{array}{c} R_1 \\ \vdots \\ R_2 \end{array}$$

• The reaction is proposed to proceed through a "butterfly" transition state.

Simmons, H. E. Org. React. 1973, 20, 1-133.

- Zinc cyclopropanating reagents can be generated in various ways. Both Zn metal and ZnEt<sub>2</sub> can be used.
- Many zinc reagents for cyclopropanation have been developed:

Diastereoselective cyclopropanation is possible in the presence of directing groups:

Dauben, W. G.; Berezin, G. H. J. Am. Chem. Soc. 1963, 85, 468-472.

• Interestingly, excess carbenoid can reverse the directing effect of alcohols.

Charette, A. B.; Marcoux, J. F. Synlett, 1995, 1197-1207.

Directed cyclopropanation is also possible in acyclic systems:

Charette, A. B.; Lebel, H. J. Org. Chem. 1995, 60, 2966-1967.

Diastereoselectrive cyclopropanation has been used in tandem asymmetric organozinc additions:

Ph 1. HBEt<sub>2</sub>, PhCH<sub>3</sub>, 23 °C 
2. Et<sub>2</sub>Zn, *i*-PrCHO 
Ph CH<sub>3</sub> 
CH<sub>3</sub>

$$CH_3$$
 $CH_3$ 
 $CH_3$ 

Kim, H. Y.; Lurain, A. E.; Garcia-Garcia, P.; Carroll, P. J.; Walsh, P. J. *J. Am. Chem. Soc.* **2005**, 127, 13138–13139. James Mousseau, Fan Liu

- · Asymmetric Simmons-Smith Reaction Using Chiral Auxiliaries
- · Allylic alcohols:

$$\begin{array}{c} \text{OBn} \\ \text{BnO} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{OBn} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{Ph} \\ \\ \text{Ioluene, } -35 \rightarrow 0 \text{ °C} \\ \\ \text{98\%, } dr > 50:1 \\ \end{array} \begin{array}{c} \text{BnO} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{OBn} \\ \text{Ph} \\ \text{2. DMF, } H_2\text{O, } C_5\text{H}_5\text{N} \\ \text{2. DMF, } H_2\text{O, } C_5\text{H}_5\text{N} \\ \text{160 °C, } 90\% \\ \end{array} \\ \begin{array}{c} \text{OBn} \\ \text{HO} \\ \text{Ph} \\ \end{array} \begin{array}{c} \text{OBn} \\ \text{BnO} \\ \text{OC} \\ \end{array} \begin{array}{c} \text{OBn} \\ \text{OBn} \\ \text{OBn} \\ \end{array} \\ \begin{array}{c} \text{OBn} \\ \text{OBn} \\ \text{OBn} \\ \end{array} \\ \begin{array}{c} \text{OBn} \\ \text{OBn} \\ \text{OBn} \\ \end{array} \end{array}$$

Charette, A. B.; Côté, B.; Marcoux, J.-F. J. Am. Chem. Soc. 1991, 113, 8166-8167.

· Allylic amines:

Ph OH 
$$H_3C$$
 Ph  $Et_2Zn, CH_2I_2$   $Ph$  OH  $H_3C$   $Ph$  OH  $H_3C$ 

Aggarwal, V. K.; Fang, G. Y.; Meek, G. Org. Lett. 2003, 5, 4417-4420.

α,β-Unsaturated carbonyls:

$$\begin{array}{c} \text{H}_{3}\text{C} \\ \text{H}_{3}\text{C} \\ \text{O} \\ \text{H} \\ \text{O} \\ \text{O} \\ \text{EtOH, 23 °C} \\ \\ \text{2.} \\ \text{O} \\ \text$$

Mash, E. A.; Nelson, K. A. *J. Am. Chem. Soc.* **1985**, *107*, 8256–8258. Mori, A.; Arai, I.; Yamamoto, H. *Tetrahedron* **1986**, *42*, 6447–6458.

## • Stoichiometric Promoter for Asymmetric Simmons-Smith Cyclopropanation

- A chiral dioxaborolane auxiliary prepared from tetramethyltartramide and butylboronic acid has been shown to be effective in the asymmetric cyclopropanation of allylic alcohols.
- A hydrogen peroxide work-up is employed to remove boron side-products.

A (1.1 equiv)
$$\frac{Zn(CH_{2}I)_{2}, DME, CH_{2}CI_{2}}{0 \rightarrow 23 \, ^{\circ}C}$$
then 30% aq  $H_{2}O_{2}$ 

$$Via H_{3}C$$

Charette, A. B.; Juteau, H.; Lebel, H.; Molinaro, C. J. Am. Chem. Soc. 1998, 120, 11943-11952.

Allylic alcohols are cyclopropanated selectively:

Nicolaou, K. C.; Sasmal, P. K.; Rassais, G.; Reddy, M. V.; Altmann, K. H.; Wartmann, M.; O'Brate, A.; Giannakakou, P. *Angew. Chem. Int. Ed.* **2003**, *42*, 3515–3520.

The cyclopropanation of allenic alcohols affords spiropentane derivatives:

Charette, A. B.; Jolicoeur, E.; Bydlinkski, G. A. S. Org. Lett. 2001, 3, 3293–3295

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- The dioxaborolane promoter can be used to prepare 1,2,3-trisubstituted cyclopropanes.
- In the example shown, the intermediate borinate was used directly for Suzuki coupling:

Ph OH 
$$\frac{1. \operatorname{Et_2Zn}}{\operatorname{CH_2Cl_2}, 0 \, {}^{\circ}\operatorname{C}}$$
 Ph OZnEt  $\frac{2. \, \mathbf{A} \, (1.2 \, \operatorname{equiv})}{\operatorname{Ph}}$  Ph  $\frac{1. \, \operatorname{Et_2Zn}}{\operatorname{Ch_2Cl_2}, 0 \, {}^{\circ}\operatorname{C}}$  Ph OZnEt  $\frac{2. \, \mathbf{A} \, (1.2 \, \operatorname{equiv})}{\operatorname{Ch_2Cl_2}, 0 \, {}^{\circ}\operatorname{C}}$ 

B-Zn Exchange 
$$(H_3C)_2N$$
  $N(CH_3)_2$   $N($ 

Zimmer, L. E.; Charette, A. B. J. Am. Chem. Soc. 2009, 131, 15624-15626.

• Homoallylic ethers can be cyclopropanated using a zinc phosphate, prepared *in situ* from a chiral phosphoric acid:

 Unfunctionalized alkenes can undergo asymmetric Simmons–Smith cyclopropanation in presence of a valine/proline dipeptide.

Selectivity is higher for trisubstituted alkenes than for disubstituted alkenes.

Long, J.; Yuan, Y.; Shi, Y. J. Am. Chem. Soc. 2003, 125, 13632-13633.

# · Catalytic Enantioselective Simmons-Smith Cyclopropanation Reactions

Chiral bis-sulfonamides have been used to direct asymmetric Simmons

—Smith reactions of allylic alcohols:

Denmark, S. E.; O'Connor, S. P. J. Org. Chem. 1997, 62, 584-594.

"Taddolates" can also be used:

Charette, A. B.; Molinaro, C.; Brochu, C. J. Am. Chem. Soc. 2001, 123, 12168-12175.

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Lacasse, M.-C.; Poulard, C. Charette, A. B. J. Am. Chem. Soc. 2005, 127, 12440-12441.

• A bifunctional Al-complex is an effective cyclopropanation catalyst and is believed to bind both the Zn and the allylic alcohol:

Shitama, H.; Katsuki, T. Angew. Chem. Int. Ed. 2008, 47, 2450-2453.

# Cyclopropanation via Michael Addition and Ring Closure — Asymmetric Cyclopropanation Through Chiral Ylides

 In an early report, optically enriched oxosulfonium F was prepared in 3 steps, which stereoselectively cyclopropanates Michael-acceptors:

Johnson, C. R.; Schroeck, C. W. J. Am. Chem. Soc. 1968, 90, 6852-6854.

· Telluronium ylides can also be used:

Liao, W.-W.; Li, K. Tang, Y. J. Am. Chem. Soc. 2003, 125, 13030-13031.

Camphor-derived sulfur ylides can be employed for stereoselective cyclopropanations of Michael-acceptor olefins:

$$H_3C$$
 $CH_3$ 
 $CO_2CH_3$ 
 $CO_2Me$ 
 $CO_2Me$ 
 $CO_2CH_3$ 
 $CO_2C$ 

Deng, X.-M.; Cai, P.; Ye, S.; Sun, X.-L.; Liao, W.-W.; Li, K.; Tang, Y.; Wu, Y.-D.; Dai, L.-X. *J. Am. Chem. Soc.* **2006**, *128*, 9730–9740.

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# · Catalytic Enantioselective Ylide Cyclopropanations

 Cinchona alkaloids have been employed to generate chiral ammonium ylides, which stereoselectively cyclopropanates Michael-acceptors:

Papageorgiou, C. D.; Cubillo de Dios, M. A.; Ley, S. V.; Gaunt, M. J. Angew. Chem. Int. Ed. 2004, 43. 4641–4644.

Johansson, C. C. C.; Bremeyer, N.; Ley, S. V.; Owen, D. R.; Smith, S. C.; Gaunt, M. J. *Angew. Chem. Int. Ed.* **2006**, *45*, 6024–6028.

· Lanthanum complexes were also found to be effective:

OH OH OH 
$$\frac{1. \text{La}(\text{O-}i\text{-Pr})_3}{\text{THF, 0 °C} \rightarrow 23 °C}$$
  $\frac{\text{Li}}{\text{O}}$   $\frac{\text{Li}}{\text{O}}$   $\frac{\text{C}}{\text{C}}$   $\frac{\text{Li}}{\text{O}}$   $\frac{\text{C}}{\text{C}}$   $\frac{\text{C}}{\text{C}}$   $\frac{\text{Li}}{\text{C}}$   $\frac{\text{C}}{\text{C}}$   $\frac{\text{C}}{\text{C$ 

Kakei, H.; Sone, T.; Sohtome, Y.; Matsunaga, S.; Shibasaki, M. *J. Am. Chem. Soc.* **2007**, *129*, 13410–13411.

· Chiral sulfoxonium intermediates can be generated in situ by trapping of a rhodium carbenoid:

Aggarwal, V. K.; Alonso, E.; Fang, G.; Ferrara, M.; Hynd, G.; Porcelloni, M. *Angew. Chem. Int. Ed.* **2001**, *40*, 1433–1436.

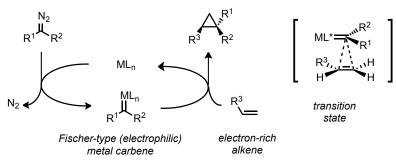
· Asymmetric cyclopropanation through a chiral iminium intermediate:

Kunz, R. K.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 3240-3241.

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### **Asymmetric Cyclopropanation using Metal Carbenes**

- Transition metals catalyze the cyclopropanation of electron-rich olefins via carbenoids formed from electron-deficient diazo compounds.
- The catalytic cycle proceeds via a Fischer-type (electrophilic) metal carbene formed from diazo precursors:



- This methodology is most effective for three classes of diazo substrates:
- · Diazo substrates with one electron-withdrawing group
- · Diazo substrates with two electron-withdrawing groups
- Diazo substrates with one electron-withdrawing group and one electron-donating group

### · Diazo Substrates with One Electron-Withdrawing Group

 The use of C<sub>2</sub>-symmetric Cu(I) oxazoline complexes for cyclopropanation was first reported by Evans:

Evans, D. A.; Woerpel, K. A.; Hinman, M. M.; Faul, M. M. J. Am. Chem. Soc. 1991, 113, 726-728.

 Since the initial report, extensive research has been done to develop other C<sub>2</sub>-symmetric Cu(I) and Ru(II) oxazoline complexes.

- Many alkenes can be used, with styrenes and enol ethers being the most common.
- · The reaction proceeds with retention of the olefin geometry:

$$R = \text{aryl, alkyl, OR}$$

- · Up to three stereocenters can be formed.
- Cu(I) and Ru(II) oxazoline complexes typically give trans-1,2-cyclopropanes selectively:

$$F = \frac{1}{N_2} + \frac{CuOTf \cdot I}{Ot-Bu} = \frac{CuOTf \cdot I}{S6\%}$$

$$\frac{CuOTf \cdot I}{S6\%}$$

$$\frac{F}{I} + \frac{Cu}{N} = \frac{CuOTf \cdot I}{S6\%}$$

$$\frac{F}{I} + \frac{F}{I} + \frac{F}{$$

Haufe, G.; Rosen, T. C.; Meyer, O. G. J.; Fröhlich, R.; Rissanen, K. *J. Fluorine Chem.* **2002**, *114*, 189–198.

• Examples of Cu(I)- and Ru(II)-catalyzed enantioselective cyclopropanations:

CO<sub>2</sub>CH<sub>3</sub>

$$\begin{array}{c}
O \\
N_2
\end{array}$$
OEt
$$\begin{array}{c}
I (2.5 \text{ mol\%}) \\
CuOTf (2 \text{ mol\%}) \\
PhNHNH2 (2 \text{ mol\%}) \\
\hline
CH2Cl2, 0 \rightarrow 20 °C \\
91\% \text{ ee (crude)}
\end{array}$$
after recrystallization:
$$53\%, >99\% \text{ ee}$$

Böhm, C.; Reiser, O. Org. Lett. 2001, 3, 1315-1318.

· Intramolecular reactions can also proceed with high enantioselectivities:

Doyle, M. P.; Peterson, C. S.; Parker Jr., D. L. Angew. Chem. Int. Ed. 1996, 35, 1334-1336...

• Chiral Ir(III)-salen complexes afford cis-1,2-cyclopropanes with high enantioselectivities:

$$p\text{-CF}_3\text{C}_6\text{H}_4$$

| III (1 mol%)
| THF, -78 °C | 73%, 97% ee | dr = 97:3

| Ot-Bu | THF, -78 °C | 87%, 94% ee | dr = 96:4

| Ot-Bu | THF, -78 °C | 87%, 94% ee | dr = 96:4

Suematsu, H.; Kanchiku, S.; Uchida, T.; Katsuki, T. J. Am. Chem. Soc. 2008, 130, 10327-10337.

· Co(II)-porphyrin complexes can cyclopropanate electron-deficient alkenes enantioselectively:

Chen, Y.; Ruppel, J. V.; Zhang, X. P. J. Am. Chem. Soc. 2007, 129, 12074–12075.

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• At low temperatures, rhodium(III) catalysts are compatible with higher  $\alpha$ -alkyl- $\alpha$ -diazoesters, which otherwise often undergo undesired  $\beta$ -hydride elimination upon metal carbene formation:

$$\begin{array}{c} \text{Ph} & \text{EtO} & \begin{array}{c} \text{Rh}_2[(S)\text{-PTTL}]_4 \\ \text{(0.5 mol\%)} \\ \text{hexane, } -78 \, ^{\circ}\text{C} \\ \text{93\%, } 96\% \text{ ee} \\ dr = 99:1 \; (trans:cis) \end{array} \\ \end{array} \quad \begin{array}{c} \text{EtO}_2\text{C} \\ \text{Ph} \\ \text{Ph} \\ \end{array} \quad \begin{array}{c} \text{O-Rh} \\ \text{O-Rh} \\ \text{O-Rh} \\ \text{O-RTL}]_3\text{TPA} \end{array}$$

Boruta, D. T.; Dmitrenko, O.; Yap, G. P. A.; Fox, J. M. Chem. Sci. 2012, 3, 1589-1593.

- Diazo Substrates with Two Electron-Withdrawing Groups
- While symmetrical diazomalonates give poor selectivities, unsymmetrical diazomalonates are excellent substrates:

Marcoux, D.; Charette, A. B. Angew. Chem. Int. Ed. 2008, 47, 10155-10158.

 For α-nitro-α-diazo carbonyls, the diastereoselectivity is sensitive to the nature of the carbonyl substituent:

Charette, A. B.; Wurz, R. P.; Ollevier, T. Helv. Chim. Acta 2002, 85, 4468-4484.

α-nitro-α-diazo aryl ketones give cis cyclopropanes selectively:

$$O_{2}N + O_{1}O_{2}N + O_{2}N + O_{3}O_{2}N + O_{4}O_{2}N + O_{5}O_{2}N + O_{5}O_{1}O_{2}N + O_{5}O_{2}N + O_{5}$$

Lindsay, V. N. G.; Lin, W.; Charette, A. B. J. Am. Chem. Soc. 2009, 131, 16383-16385.

•  $\alpha$ -nitro- $\alpha$ -diazo esters give *trans* cyclopropanes selectively. In the example below, the oxidant iodosobenzene can be used to form the carbene precursor from  $\alpha$ -nitro esters *in situ*:

Moreau, B.; Charette, A. B. J. Am. Chem. Soc. 2005, 127, 18014-18015.

Alternatively, α-nitro-α-diazo acetates give cis cyclopropanes with cobalt(II)-porphyrin catalysts:

Zhu, S.; Perman, J. A.; Zhang, X. P. Angew. Chem. Int. Ed. 2008, 47, 8460-8463.

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- · Diazo Substrates with One Electron-Withdrawing Group and One Electron-Donating Group
- Metal carbenes with adjacent electron-donating and electron-withdrawing groups ("push-pull" systems) are relatively stable and reactive.
- Rhodium(II) complexes using chiral ligands derived from proline have often been employed:

Doyle, M. P.; Zhou, Q.-L.; Charnsangavej, C.; Longoria, M. A.; McKervey, M. A.; Garcla, C. F. *Tetrahedron Lett.* **1996**, *37*, 4129–4132.

Davies, H. M. L.; Bruzinski, P. R.; Fall, M. J. Tetrahedron Lett. 1996, 37, 4133-4136.

• Styrenes can also be considered as electron-donating groups on the diazo substrate:

Davies, H. M. L.; Bruzinski, P. R.; Lake, D. H.; Kong, N.; Fall, M. J. *J. Am. Chem. Soc.* **1996**, *118*, 6897–6907.

· A wide range of electron-withdrawing groups within the diazo substrate are tolerated:

Rh-O N-O Rh-O Rh<sub>2</sub>[(S)-PTAD]<sub>4</sub>

Reddy, R. P.; Lee, G. H.; Davies, H. M. L. Org. Lett. 2006, 8, 3437-3440.

Denton, J. R.; Sukumaran, D.; Davies, H. M. L. Org. Lett. 2007, 9, 2625-2628.

NC Ph  

$$N_2$$
 Ph  
 $N_2$  Ph

Denton, J. R.; Cheng, K.; Davies, H. M. L. Chem. Commun. 2008, 1238-1240.

• N-sulfonyl-1,2,3-triazoles serve as alternatives to diazo compounds as carbene precursors:

Chuprakov, S.; Kwok, S. W.; Zhang, L.; Lercher, L.; Fokin, V. V. J. Am. Chem. Soc. 2009, 131, 18034.

Grimster, N.; Zhang, L.; Fokin, V. V. J. Am. Chem. Soc. **2010**, 132, 2510.

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### **Selected Transformations of Cyclopropanes**

- Cyclopropanes can undergo direct C-H functionalization enantioselectively by Pd(II) complexes with chiral amino acid ligands.
- A special directing group is required on the cyclopropane substrate to coordinate the Pd(II) complex and direct insertion into the adjacent cis C-H bond.
- The directing group is readily available and can be hydrolyzed to give the corresponding acid:

Wasa, Y.; Engle, K. M.; Lin, D. W.; Yoo, E.-J.; Yu, J.-Q. J. Am. Chem. Soc. 2011, 133, 19598–19601.

• Cyclopropanes can undergo vinylcyclopropane-cyclopentene rearrangements:

TBSO 
$$H_3$$
C  $O$   $N_2$   $t$ -Bu  $t$ -Bu

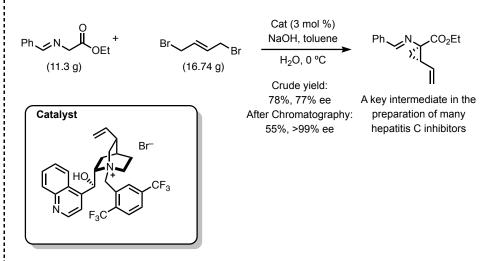
Corey, E. J.; Myers, A. G. *J. Am. Chem. Soc.* **1985**, *107*, 5574–5576. Corey, E. J.; Myers, A. G. *Tetrahedron Lett.* **1984**, *25*, 3559–3562.

### Selected Industrial Examples

 A copper-catalyzed diazo decomposition led to the asymmetric cyclopropanation of 2,5-dimethyl-2,4-hexadiene in good yield and high enantioselectivities:

Itagaki, M.; Masumoto, K.; Suenobu, K.; Yamamoto, Y. Org. Proc. Res. Dev. 2006, 10, 245-250.

• Cinchona alkaloids were applied to the synthesis of (1*R*,2*S*)-1-amino-2-vinylcyclopropanecarboxylic acid ethyl esters in good yield and modest ee's.



Belyk, K. M.; Xiang, B.; Bulger, P. G.; Leonard, W. R.; Balsells, J.; Yin, J.; chen, C.-y. *Org. Proc. Res. Dev.* **2010**, *14*, 692–700.

James Mousseau, David W. Lin, Fan Liu