

## Experiment 1

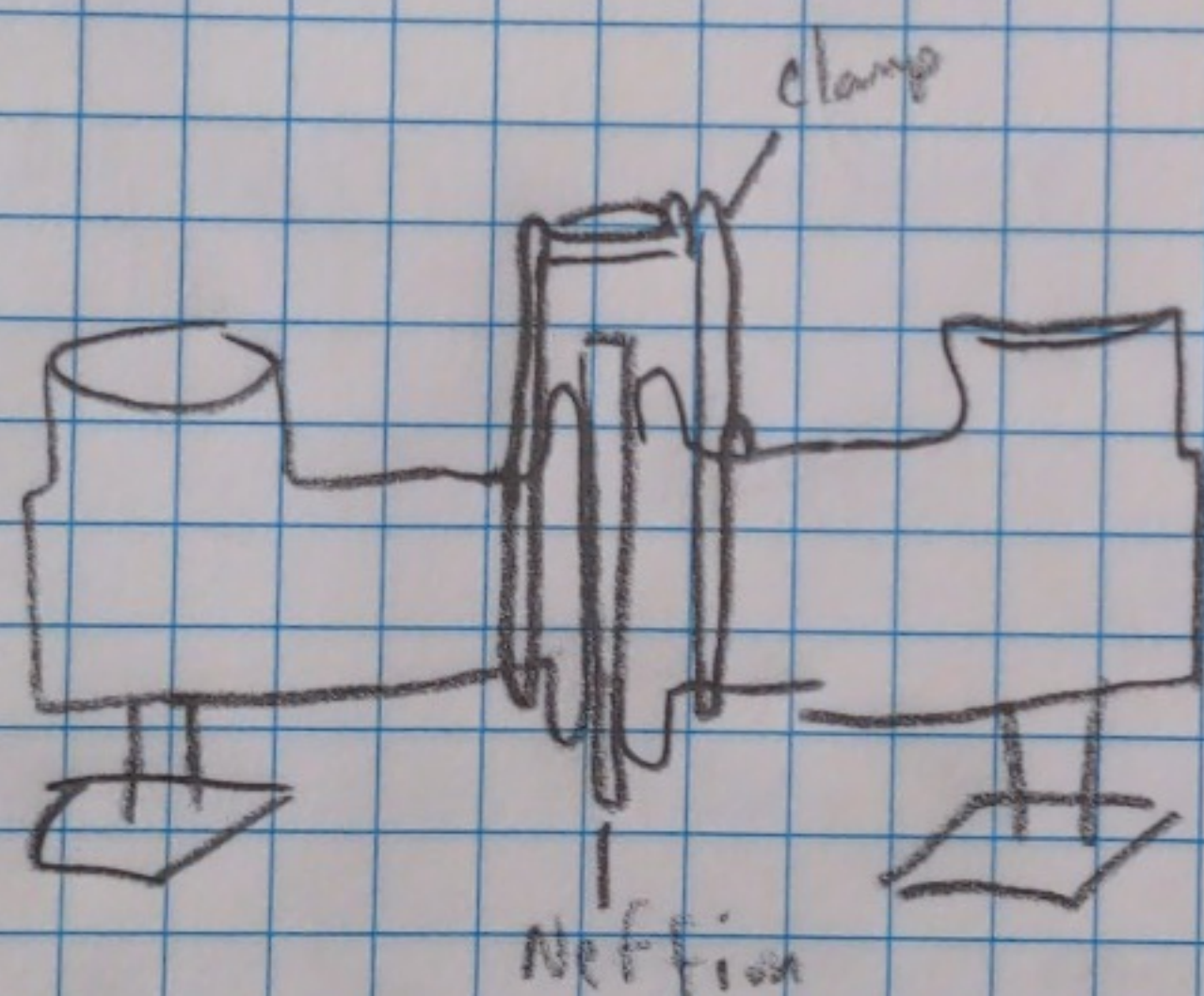
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1. Clean all cell components using high  $\Sigma$  di  $H_2O$

- Figure out how to clean glassware
  - soak in 2M KOH, then HCl to remove base
  - then  $H_2O$  to remove HCl, then heat w/  $H_2O$  to remove all Cl
- Ensure all glassware is completely dry

Naffion 117 - perfluorinated membrane  
supports  $H^+$  diffusion

- Place membrane between glass cells with washer. Clamp shut



- Fill with 25 mL acetate buffer

- This is the electrolyte

The amount of charge is a conversion is related to the amount/concentration of electrolyte

- Place WE in electrolyte - Knock  $O_2$  off WE by tapping in electrolyte
- Cover with cap.

- Fill counter cell with 25 mL acetate buffer
  - make sure level is equal to ensure same pressure

- Spurge WE cell with  $N_2$  → 40 mL/min Air  
Flow Rate



## Experiment 1

- While sparging, turn on stir
- Set up potentiostat electrodes
  - ensure ease of access
  - any disturbance of electrodes might change IR drop in solution.
- Prepare RE, clean with ultrapure DF, dry sufficiently
  - Fill with acetate buffer
  - tap out air bubbles
  - Insert RE into glassware
- Place CE in CE cell
- Tap cell to hood floor to prevent vibration effects
- Connect RE and CE to potentiostat

## Note Files

date - catalyst - electrolyte - sparge

Perform CV - Background CV (look up)

Sweep Rate 50mV/s

$E_{1,2} = .022 - 1.2$  vs RHE