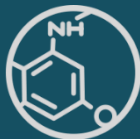




ADVANCED PROCESS  
MODELLING FORUM 2017  
London 25–26 April

## Model-based methodology for accelerated stability studies of solid formulations

Edd Close – PSE Formulated Products Senior Consultant



- Introduction
- Background
- Workflow
- Applying the framework
- Future directions

## ■ Exploratory work

- Henrique Sardinha
- make use of advanced gPROMS platform capabilities
  - parameter estimation – model calibration using data from accelerated tests
  - GSA – accounting for parameter uncertainty and variability in env. conditions
  - Optimisation

## ■ Lay the foundations for a framework for future expansions

- Development of more mechanistic (not first principles) models
  - account more explicitly for type of material
  - Implement models based on hypothesis from experts
  - use experimental data to discriminate between these models

## ■ Initially we have considered a pharmaceutical industry perspective

- We are moving forward with the understanding that the gPROMS platform capabilities and underlying framework are likely to have applications in other formulated product industries too

# Background



# Accelerated Stability Assessment Program Modelling



**Product Stability** – Product allowed storage time before any given degradation product compromises patient safety;

## Typical stability program:



- 6 months of accelerated data under ICH conditions;
- At least 2 years of long term stability testing;

## **Chemical (in)stability**

- Hydrolysis
- Photochemical reactions
- Reactions with excipients
- Oxidation

## **Physical (in)stability**

- Deliquescence;
- Dissolution;
- Polymorphic changes;
- Disintegrant expansion;

**Isoconversion** –Time required for the heterogeneous solid state reaction to achieve a specification limit or degradant concentration, regardless of the extremely complex solid state-kinetics;

## ASAP studies

-Accelerate degradation rates in order to test the stability of the formulation

## **Relative Humidity corrected Arrhenius Equation:**

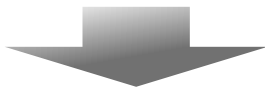
$$\ln(k) = \ln(A) - \frac{Ea}{R} \times \left(\frac{1}{T}\right) + B(RH)$$

Pseudo zero-order reactions can be considered due to the low conversion of solid dosage drugs into degradants (typically 0.5-1%)

The degradation rate (k) will be a combination of all the reactions occurring in the solid state.

## Current stability programs

- Difficulty developing mechanistic understanding of stability processes
  - Sources of chemical/ physical instability
- High pressure to succeed at first time
  - High costs of stability programs



High uncertainty

## The challenge:

- Have a better understanding of the degradation experiments
- Ensure the new drug is stable enough to start an ICH stability program
- Determine the design space and **quantify** how external factors (temperature, relative humidity) and kinetic factors ( $E_a$ ,  $\ln(A)$ ,  $B$ ) contribute to drug instability

# Workflow





# Approach to the degradation assessment

1

## Build flowsheet model of experiment



Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities.  
Click on the degradation regressor to insert the experimental conditions.

 PROMS

**Relative Humidity corrected  
Arrhenius Equation:**

$$\ln(k) = \ln(A) - \frac{Ea}{R} \times \left(\frac{1}{T}\right) + B(RH)$$





# Approach to the degradation assessment

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Relative Humidity corrected  
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$$\ln(k) = \ln(A) - \frac{Ea}{R} \times \left(\frac{1}{T}\right) + B(RH)$$

3

## Estimate parameters and analyse uncertainty

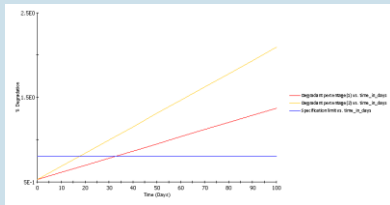


Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities. Click on the degradation regressor to insert the experimental conditions.



2

## Execute experiment & capture data



# Approach to the degradation assessment

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PROMS

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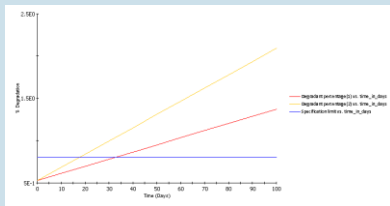


Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities. Click on the degradation regressor to insert the experimental conditions.



2

## Execute experiment & capture data



Degradation\_regressor (Degradation\_regressor)

Degradation kinetics

Experimental conditions

Initial conditions

Specification limit

Model type: Reaction Order

Model: Zero order

The rate of solid-state reaction can be described by the multiplication of the corrected Arrhenius equation and a function that represents the kinetics of the reaction.

$$dx/dt = A \cdot e^{(-Ea/RT)} \cdot f(X)$$

☒ Activation energy: 18.6578 kcal/mol

☒ Log(A): 21.9121

☒ Sensitivity to relative humidity (B): 0.00368205

OK Cancel Reset all Help

# Approach to the degradation assessment

1

Build flowsheet model of experiment



Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities. Click on the degradation regressor to insert the experimental conditions.

PROMS

Relative Humidity corrected Arrhenius Equation:

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Estimate parameters and analyse uncertainty

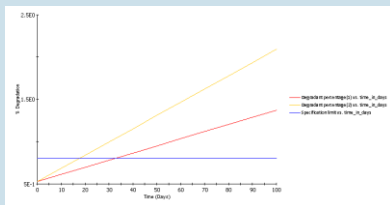


Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities. Click on the degradation regressor to insert the experimental conditions.



2

Execute experiment & capture data



Degradation\_regressor (Degradation\_regressor)

Degradation kinetics

Experimental conditions

Initial conditions

Specification limit

Model type: Diffusion

Model type: 1-D Diffusion

1-D Diffusion

2-D Diffusion

3-D Diffusion

Ginstling-Brounshtein

The rate of solid-state reaction is the multiplication of the corrected Arrhenius equation and a function that represents the kinetics of the reaction.

$$dx/dt = A \cdot e^{(-Ea/RT)} \cdot f(x)$$

☒ Activation energy: 18.6578 kcal/mol

☒ Log(A): 21.9121

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OK Cancel Reset all Help

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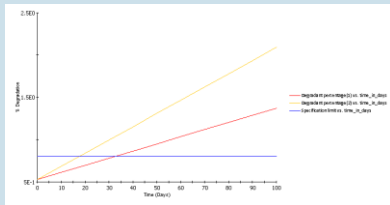


Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities. Click on the degradation regressor to insert the experimental conditions.



2

## Execute experiment & capture data



Degradation\_regressor (Degradation\_regressor)

Degradation kinetics

Experimental conditions

Initial conditions

Specification limit

☒ Temperature 298 K

☒ Relative humidity 60 %

☒ Simulation time 1500 d

OK Cancel Reset all Help

# Approach to the degradation assessment

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PROMS

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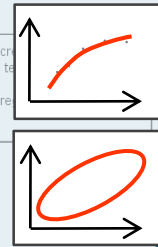
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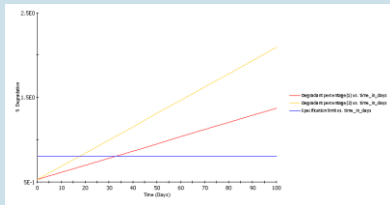


Degradant percentage increases over time when an API is exposed to high temperatures and relative humidities. Click on the degradation regressor to insert the experimental conditions.



2

## Execute experiment & capture data



Degradation\_regressor (Degradation\_regressor)

Degradation kinetics

Experimental conditions

Initial conditions

Specification limit

☒ Initial degradant percentage 0.524 %

OK Cancel Reset all Help

# Approach to the degradation assessment

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### Relative Humidity corrected Arrhenius Equation:

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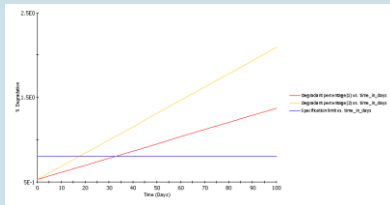


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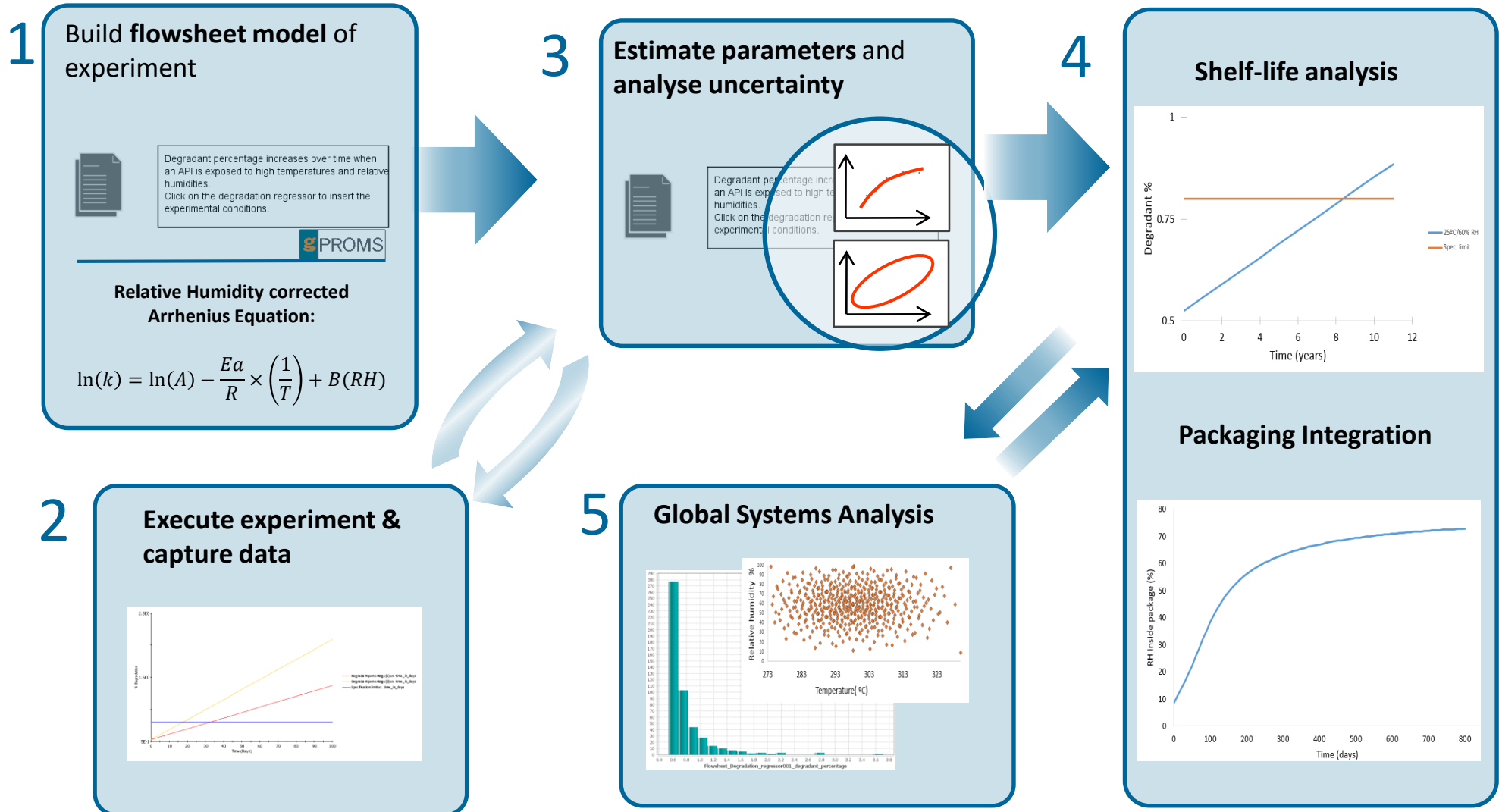
Initial conditions

Specification limit

☒ Specification limit 0.8 %

OK Cancel Reset all Help

# Approach to the degradation assessment





# Applying the framework



# Execute experiments and estimate parameters

Applying the framework



## 1) Conducted experiments

Sample degradation (%) over 30 days:

- 70°C/ 25%RH
- 70°C/ 75%RH
- 80°C/ 25%RH
- 80°C/ 75%RH
- 90°C/ 25%RH

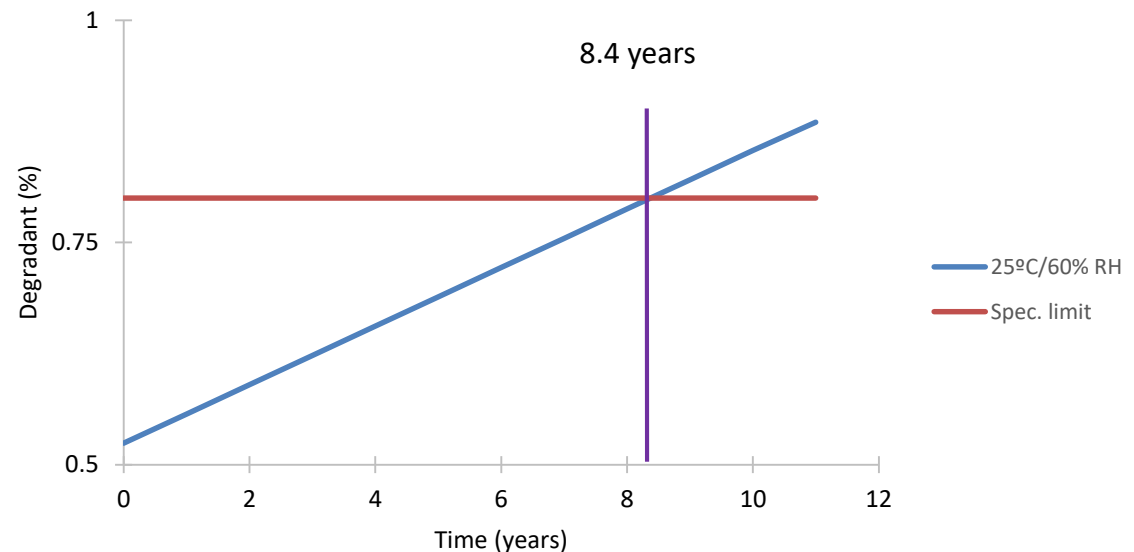
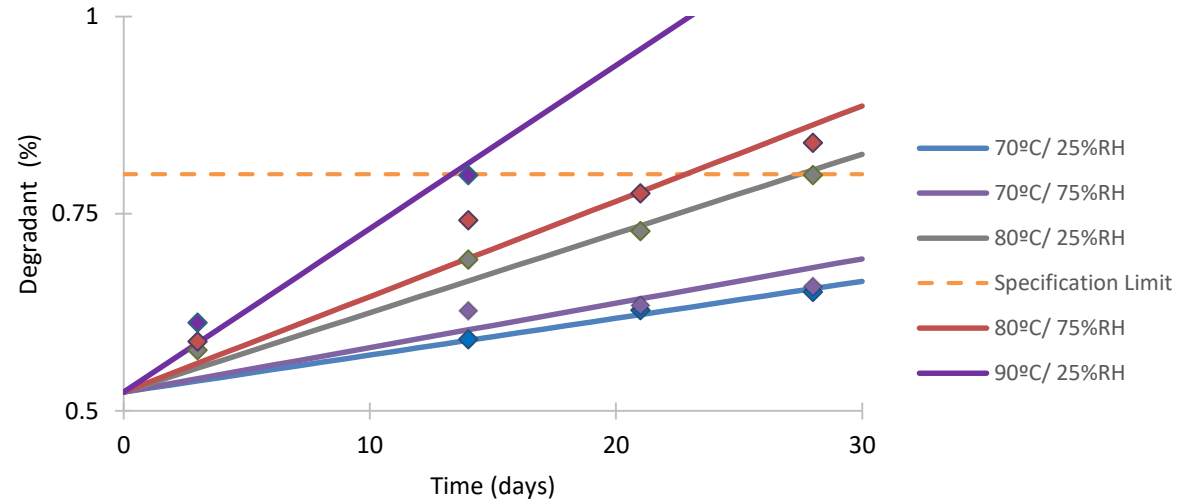
## 2) Parameter estimation

$$E_a = 18.4 \pm 1.97$$

$$\ln(A) = 21.54 \pm 2.77$$

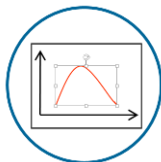
$$B = 0.0037 \pm 0.0018$$

For the estimated parameters acquired at accelerated conditions, the model predicts a **shelf-life** of **8.4 years** at **25°C/60%RH** for the tested API



# Shelf life analysis - Environmental variability

Applying the framework



## Normal Distribution

### Why?

To explore variable environmental conditions to which the sample might be exposed

### What?

Monte Carlo uncertainty analysis on **500 random samples (Sobol sampling)** of the 2 main factors:

- Temperature
- Relative humidity

### When?

**2 year shelf-life (standard targeted shelf life for pharmaceutical drugs)**

Response:

- Degradation percentage

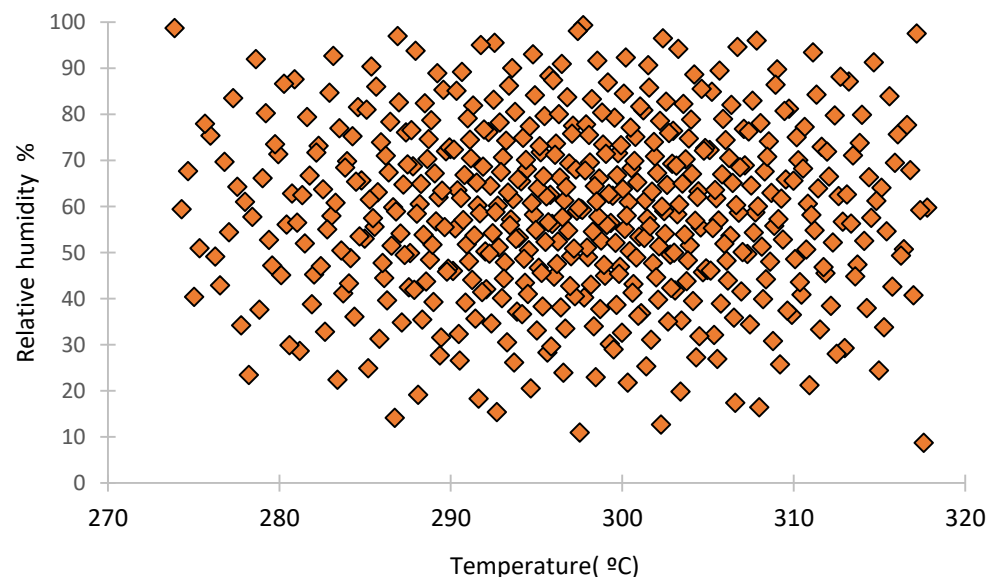
Global system analysis took 4 seconds.  
Total CPU time: 0.905s (64% system time)

### Temperature:

- Mean= 298K
- STD= 12K

### Relative humidity:

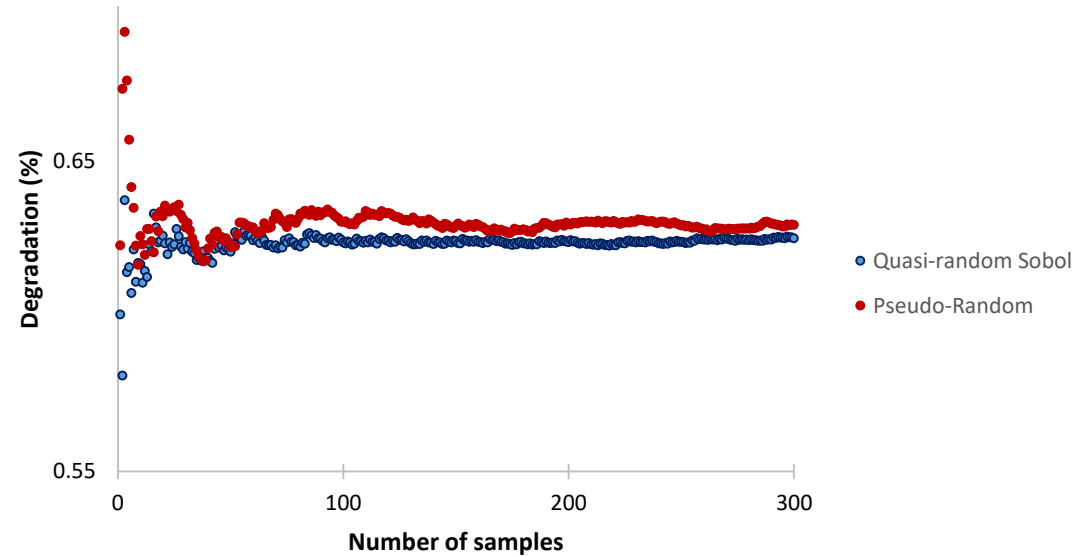
- Mean=60%
- STD=20%



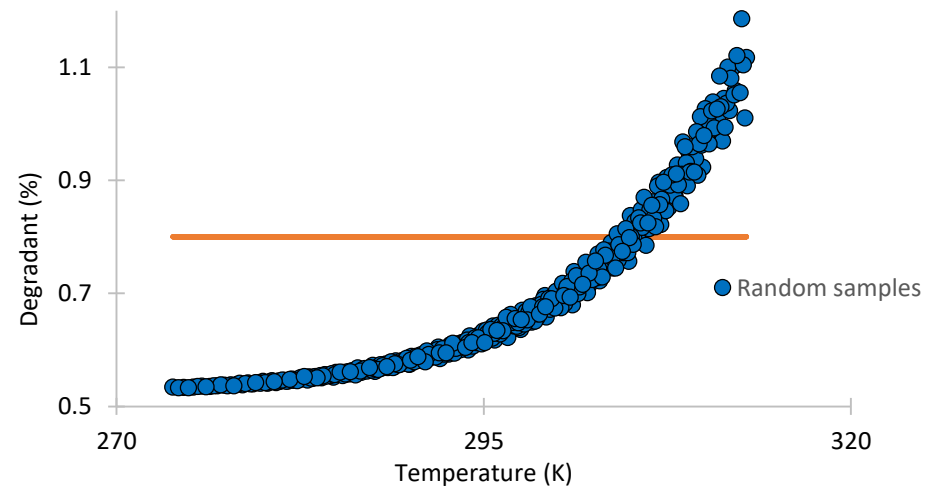
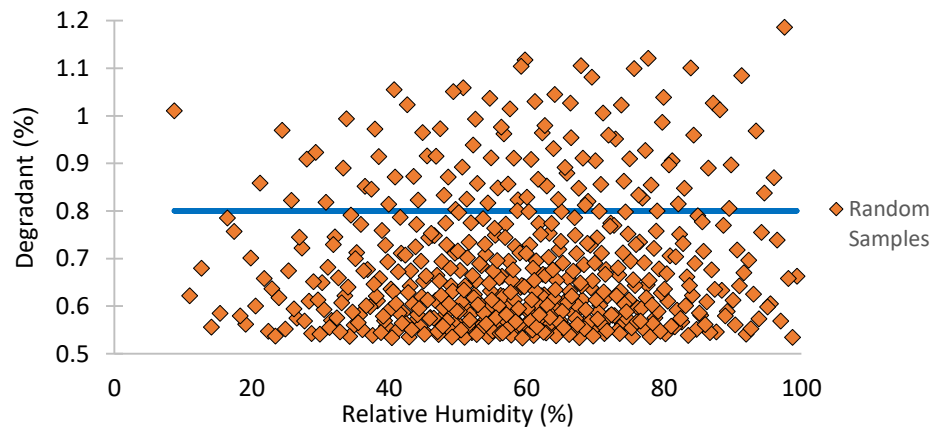
# Shelf life analysis - Environmental variability

Applying the framework

Why 500 samples  
with quasi-random  
Sobol sampling?

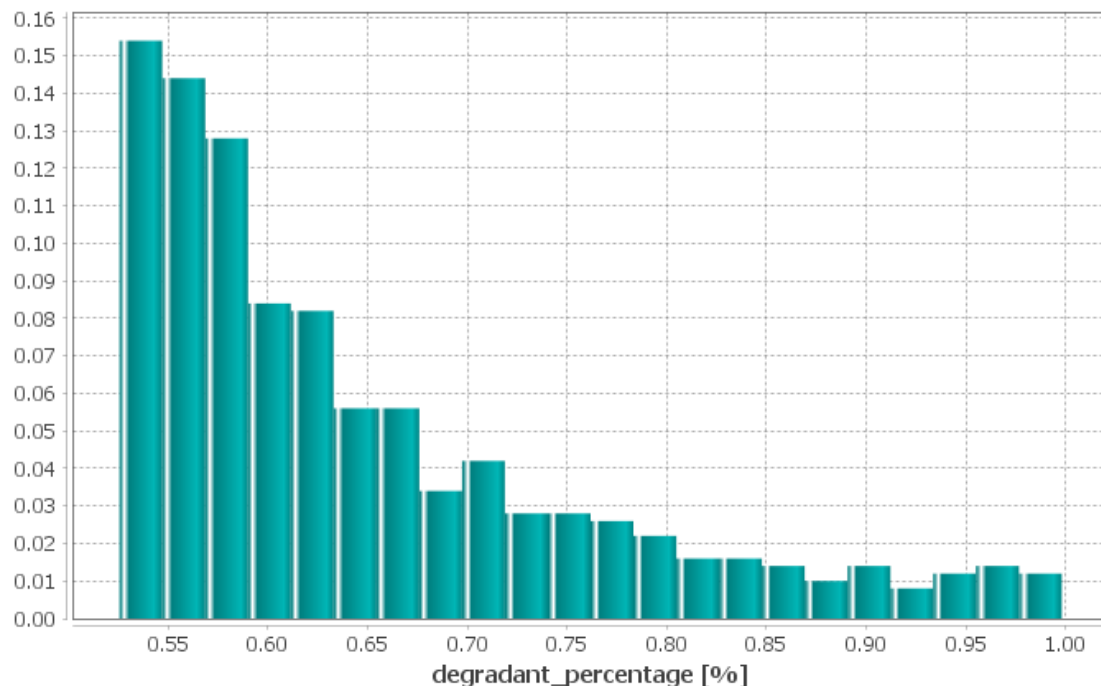


Analysis performed on the 2<sup>th</sup> year of shelf-life



# Shelf life analysis - Environmental variability

## Applying the framework



Degradation %	
Expected value	0.647
STD	0.114
Minimum	0.525
Maximum	0.998

**Statistical analysis target after 2 years of the suggested shelf-life:**

- **12.4%** of all samples **are already over the spec. limit** of 0.8% degradant
- A significant number of samples will soon be over the specification limit

**Incorporating uncertainty is vital in order to draw valid conclusions, and should become an integral part of modelling workflows in the future**

# Shelf life analysis – Parameter uncertainty

Applying the framework



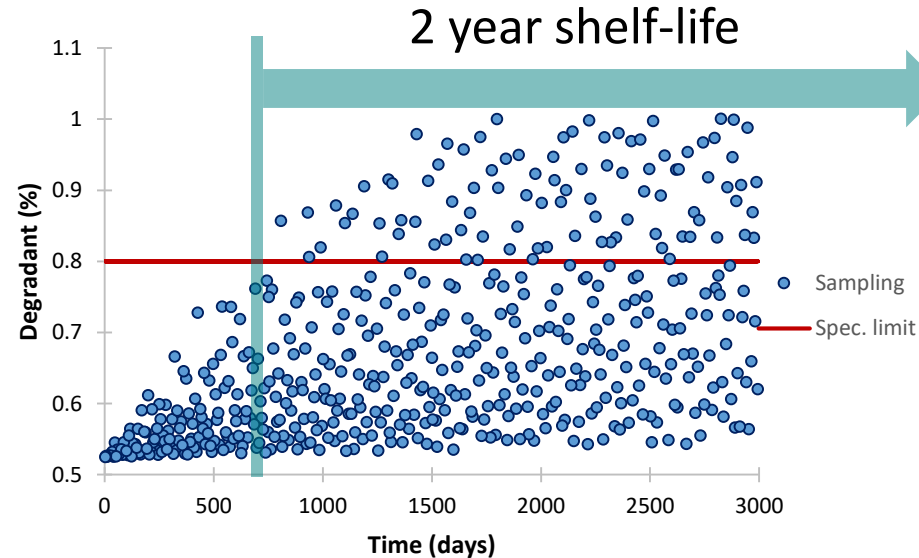
## Normal distribution

### Temperature:

- Mean= 298K
- STD= 12K

### Relative humidity:

- Mean=60%
- STD=20%



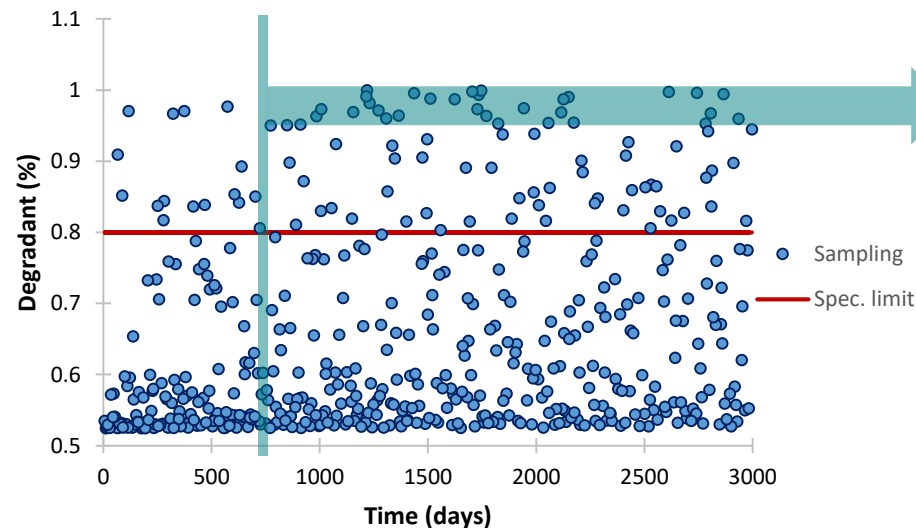
**100 000 samples**

$P(\text{failing spec})=12.4\%$

## Uniform distribution on kinetic parameters

$$\begin{aligned} 16.73 < E_a < 20.59 \\ 19.18 < \ln(A) < 24.64 \\ 0.0018 < B < 0.0055 \end{aligned}$$

95% confidence intervals



**100 000 samples**

$P(\text{failing spec})=35.8\%$

# Shelf life analysis – Factor sensitivity

Applying the framework



Temperature is the key driver in the response space amongst the external parameters

	Degradation (%)	
	1 <sup>st</sup> Order	Total effect
<b>RH</b>	0.00397	0.00662
<b>T</b>	0.9935	0.99571

Activation energy is the key driver in the response space amongst the kinetic parameters

	Degradation (%)	
	1 <sup>st</sup> Order	Total effect
<b>Ea</b>	0.6117	0.7630
<b>Ln(A)</b>	0.4731	0.6262
<b>B</b>	0.00166	0.00300



# Future directions



- Recreate ASAP functionality within gFP environment
  - very basic model
  - make use of advanced gPROMS platform capabilities
    - parameter estimation – model calibration using data from accelerated tests
    - GSA – accounting for parameter uncertainty and variability in env. conditions
    - optimisation
- Development of more mechanistic (not first principles) models
  - account more explicitly for type of dosage form
  - implement models based on hypotheses from experts within alliance
  - use experimental data to discriminate between these model
- Connectivity with manufacturing models – sensitivity of stability with respect to manufacturing decisions and disturbances

Thank you

