## ACID RAIN

#### DEFINITION

- The term acid rain means any form of precipitation like rain, fog, snow, or hail that contains harmful substances such as nitrogen and sulphur oxides.
- It is a precipitation that is very acidic pH<5 which has harmful effects on plants, aquatic animals and buildings.

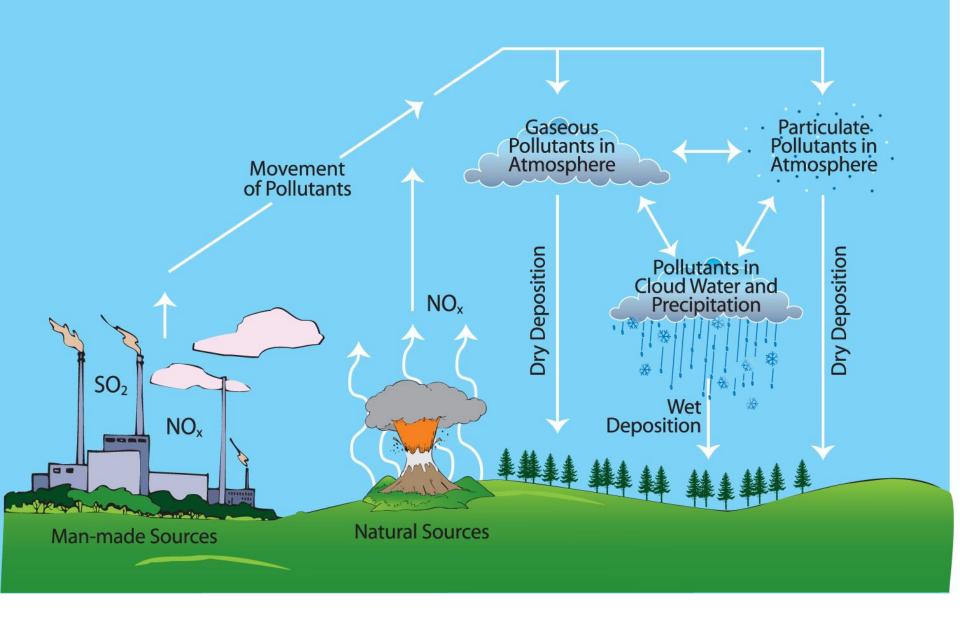
### CAUSES OF ACID RAIN

- •Oxides of sulphur (SO<sub>x</sub>)
- •Oxides of nitrogen (NO<sub>x</sub>)
- These are emitted primarily from automobiles and industries.

It is due to fossil fuel burning which emits these dangerous air pollutants.

### Effect of Pollutants

- These pollutants coming down with rain is called wet deposition.
- The same pollutants reaching the ground by gravity during dry intervals is called dry deposition.
- They are carried over thousands of km by wind and get deposited on soil, vegetation, water and property causing great damage.



### Reactions taking place...

■ 
$$SO_2(g) + H_2O(1) \rightarrow H_2SO_3(1)$$

■ 
$$SO_3(g) + O_2 + H_2O(1) \rightarrow H_2SO_4(1)$$

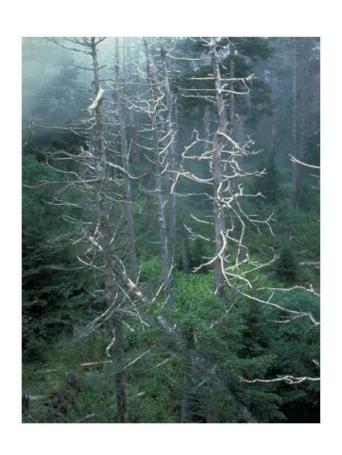
■ 
$$NO_3(g) + O_2 + H_2O(1) \rightarrow HNO_3(1)$$

■ 
$$HCl(g) + H_2O(l) \rightarrow HCl(aq)$$

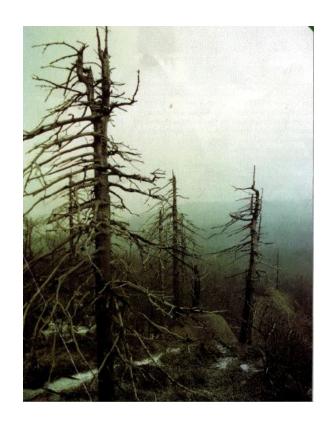
### EFFECTS OF ACID RAIN

- Affect the growth of aquatic life
- Retardation of terrestrial growth
- Contamination of air, water and food
- Corrosive to metals, buildings, limestone, marble, etc
- Increases acidity of the soil
- Accumulation of toxic elements

## Effect on plants...







# Effect on buildings...





# Effect on aquatic life...



