DEPARTMENT OF CHEMISTRY

SRM Institute of Science and Technology B.Tech(2020-2021)

SEMESTER-I

18CYB101J-CHEMISTRY LAB

LIST OF EXPERIMENTS

- 1. Determination of the amount of sodium carbonate and sodium hydroxide in a mixture by titration.
- 2. Determination of strength of an acid by conductometry.
- 3. Determination of hardness (Ca²⁺) of water using EDTA complexometric method.
- 4. Determination of ferrous ion using Potassium dichromate by potentiometric titration.
- 5. Estimation of amount of chloride content of a water sample.
- 6. Determination of molecular weight of polymer by viscosity average method.
- 7. Determination of the strength of a mixture of acetic acid and hydrochloric acid by conductometry.
- 8. Determination of strength of an acid using pH meter.

DEPARTMENT OF CHEMISTRY COLLEGE OF ENGINEERING AND TECHNOLOGY

SRM Institute of Science and Technology Chemistry Lab[Virtual] - Lesson Plan

Academic Year: 2020-21 Semester : I

Total Weeks Available: 10 Total Hours [As per virtual Class Time table]: 15 Hours

S.No Cours	Experiment Detailshemistry	Week	Work schedule Course BS	e for two pours
Code	Name	C	ategory Scien	ices 3 1 2 5
1.	Introduction	I		
2.	Determination of the amount of sodium carbonate and sodium hydroxide in a mixture by titration.	II	Virtual	Practice
3.	Determination of strength of an acid by conductometry.	III	session [60 min]	session* [30min]
4.	Determination of hardness (Ca ²⁺) of water using EDTA – complexometric method.	IV		
5.	Determination of ferrous ion using Potassium dichromate by potentiometric titration.	V		
6.	Estimation of amount of chloride content of a water sample.	VI		
7.	Determination of molecular weight of polymer by viscosity average method.	VII		
8.	Determination of the strength of a mixture of acetic acid and hydrochloric acid by conductometry.	VIII		
9.	Determination of strength of an acid using pH meter.	IX		
10.	Viva - Voce	X		

Model Practical exam: 9/12/2020 [V day order /Wed]

University Practical exam: 23/12/2020 [V day order /Wed]

^{*}Practice session –The students shall be asked to do calculations by giving model observation values for each experiment.