



# **18PYB101J MODULE-5**

## **LECTURE 2**

- **CHARACTERISTICS OF LASER**
- **AMPLIFICATION OF LIGHT BY  
POPULATION INVERSION**

# CHARACTERISTICS OF LASER

## Characteristics of laser

Laser is basically a light source. Laser light has the following important characteristics

- (i) High directionality
- (ii) High intensity
- (iii) Highly monochromatic
- (iv) Highly coherent

### (i) High directionality

An ordinary light source emits the light in all directions. But, a laser source emits light in only one direction. The divergence of laser beam is very small (fig. 7.12). So, laser light has high directionality.

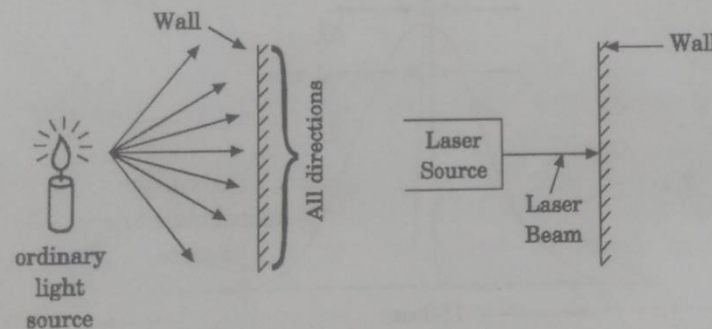


Fig. 7.12. High directionality

# CHARACTERISTICS OF LASER

## (ii) High Intensity

Laser source emits light as a narrow beam and its energy is concentrated in a small region (spot). This concentration of energy gives a high intensity to the laser light. (fig. 7.13).

## (iii) Highly monochromatic

Ordinary light spreads over a wavelength range of the order of 100 nm.

# CHARACTERISTICS OF LASER

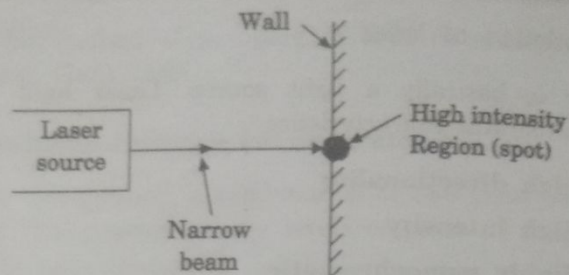


Fig. 7.13. High intensity

But a laser beam has very little spreading of the order of 1 nm. (fig. 7.14). Thus, laser beam is highly monochromatic. i.e., it emits only one colour of light.

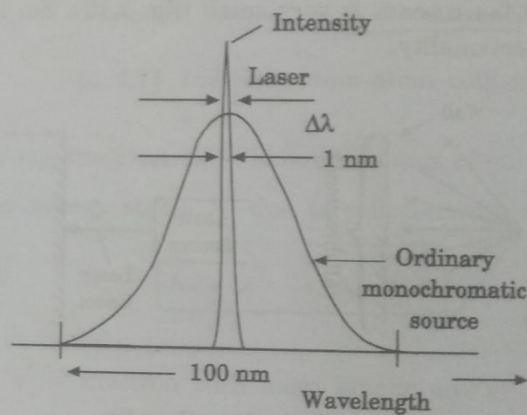


Fig. 7.14 Spectral width of laser

# CHARACTERISTICS OF LASER

## (iv) Highly Coherent

The light emitted from a laser source consists of wave trains. These wave trains have same frequency, phase and direction. So, they are coherent (fig. 7.15).

Laser light has a high degree of coherence. The coherence of laser emission results in extremely high intensity and hence more power.



## **Difference between spontaneous emission and stimulated emission**

<b>Property</b>	<b>Spontaneous emission (ordinary light)</b>	<b>Stimulated emission (laser light)</b>
Stimuli	Not required	Required
Monochromaticity	Less	High
Directionality	Less	High
Intensity	Less	High
Coherence	Less	High



## Population inversion-Negative temperature condition

- ❖ Boltzmann distribution law specifies what fraction of atoms are found in any particular energy state for any given equilibrium temperature
- ❖ If  $N_0$  is the number of atoms in the ground state,  $N_1$  is the number of atoms in the excited state of energy  $E_2$  measured relative to the ground state, then (ignoring degeneracy)





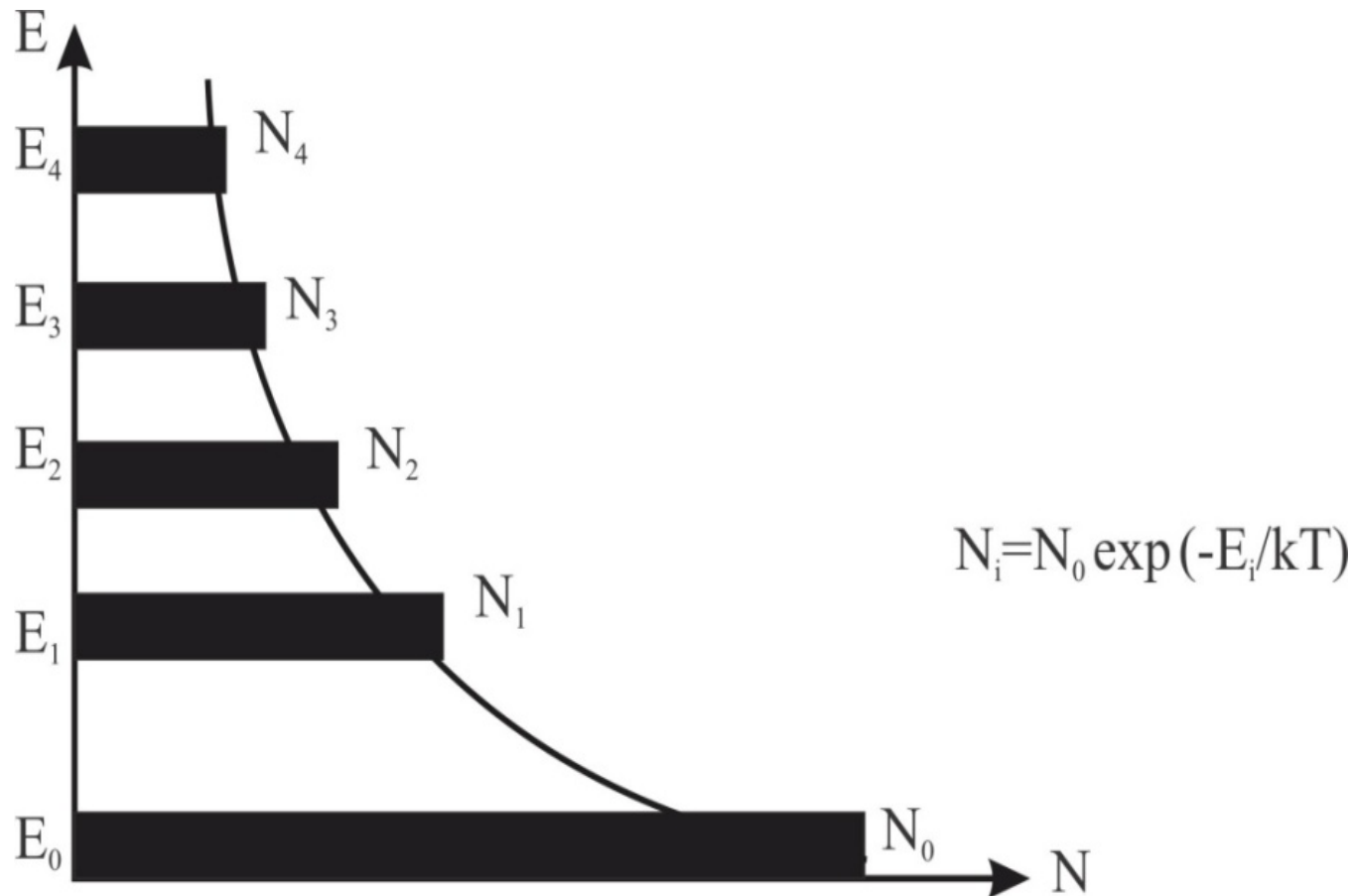
$$\frac{N_i}{N_o} = \exp\left(\frac{-E_i}{kT}\right)$$

where T is the absolute temperature in degree kelvin, and  
 $k = 1.38 \times 10^{-23}$  K (Boltzmann constant)

Boltzmann distribution is graphically represented in fig

- For laser action,  $N_1 > N_0$  (i.e., absorption < stimulated emission)
- The establishment of  $N_1 > N_0$  is known as population inversion.





**Boltzmann distribution for several energy levels**

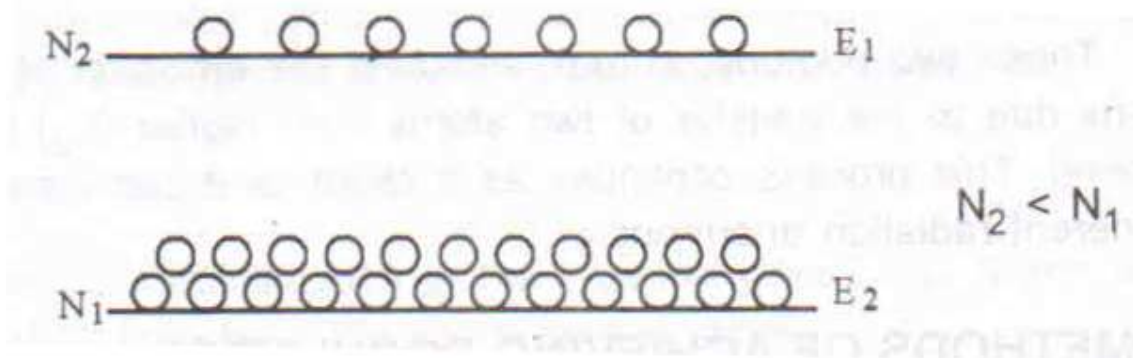


- The population inversion condition required for light amplification is a non-equilibrium distribution of atoms among the various energy levels of the atomic system.
- i.e., a *negative temperature condition* which establishes  $N_1 > N_0$  is known as population inversion.



## Normal population

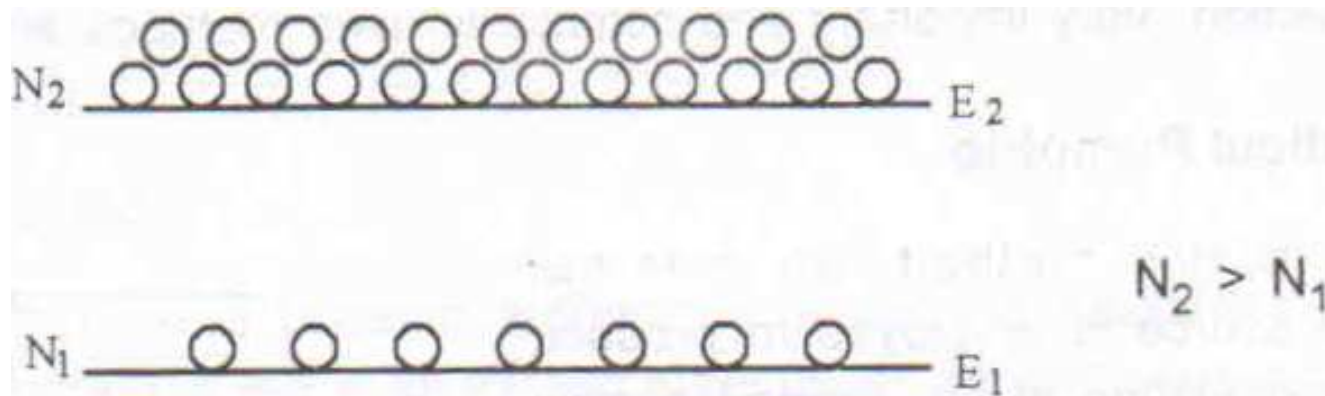
- Consider a two level energy system ( $E_1$  and  $E_2$ ). Suppose a photon of energy same as the energy difference between the two levels is incident on the system, absorption and emission processes are both equally probable.
- Usually the number of atoms or molecules or particles  $N_2$  of higher energy level is less than the population  $N_1$  of lower energy level and it is called 'Normal Population'.





# Population Inversion

- Making the number of particles  $N_2$  more in higher energy level than the number of particles  $N_1$  in lower energy level is called Population Inversion or Inverted Population.



# Laser Action



- During stimulated emission of radiation, the incident photon and the emitted photons all have same energy, phase, frequency and direction.
- A single photon is incident on the atoms in the state  $E_2$ . It releases a photon along with this incident photon. This results in the emission of two photons.



# Laser Action

- These two photons, in turn, stimulate the emission of atoms, two more photons due to the transfer of two atoms from higher ( $E_2$ ) level to the lower ( $E_1$ ) level.
- This process continues as a chain, and can increase the intensity of coherent radiation enormously.

