

Define reactants and species to be considered

Reactants

Natural Gas + Air

▼

Products

Soot formation

▼

List of Species

CO2

CO

H2O

H2

OH

▼

% Fuel

8.367

O/F

2.3

Phi

1

Species	N° moles	Mole fraction	Type	Temperature [K]	
N2	8.6524	0.7239	Inert	300	
O2	2.3000	0.1924	Oxidizer	300	
CH4	0.8500	0.0711	Fuel	300	
C2H6	0.1000	0.0084	Fuel	300	
C3H8	0.0500	0.0042	Fuel	300	

Select Problem Type

HP: Adiabatic T and composition at constant P

▼

☐ Frozen chemistry

☐ Ionized species

Define state of reactants and products

Reactants

300

Temperature [K]

1

Pressure [bar]

Products

1

Additional constraints

Products

Constant Enthalpy: $h_P = h_R$

Calculate

Clear

Welcome to Combustion Toolbox v0.5 --- A MATLAB-GUI based open-source tool for solving combustion problems.