



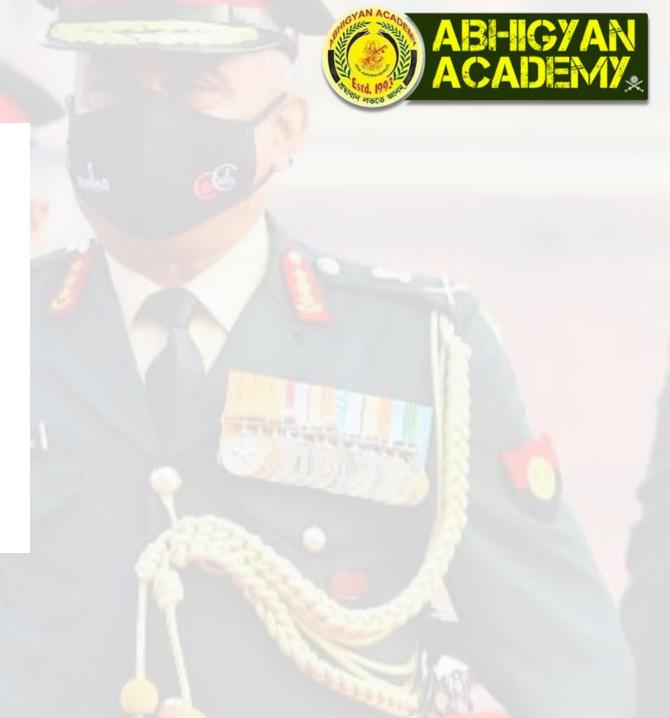
Which one of the following is a reduction reaction?

(a)
$$2 \text{ Mg (s)} + O_2(g) \rightarrow 2 \text{ MgO (s)}$$

(b)
$$S(s) + O_2(g) \rightarrow SO_2(g)$$

(c)
$$2 \text{ HgO (s)} \xrightarrow{\text{heat}} 2 \text{ Hg (l)} + O_2 (g)$$

(d)
$$Mg(s) + S(s) \rightarrow MgS(s)$$

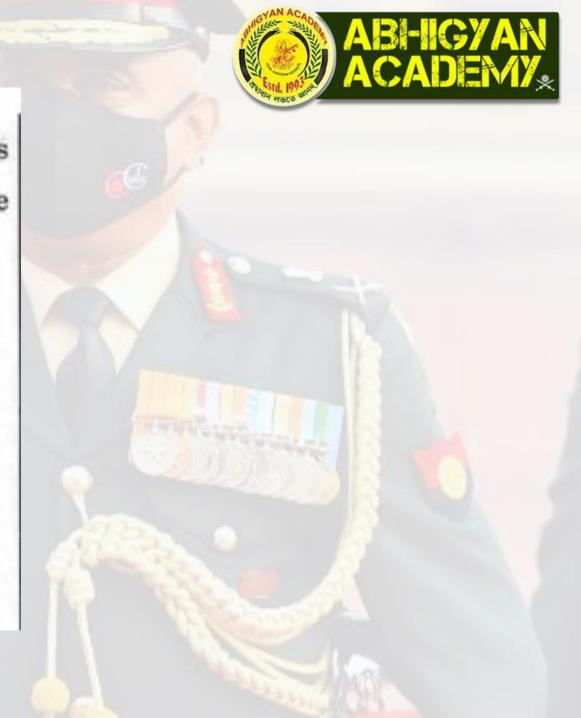


In the reaction $ZnO + C \rightarrow Zn + CO$, 'C' acts as

- (a) an acid
- (b) a base
- (c) an oxidising agent
- (d) a reducing agent

Which one of the following compounds does not exhibit a different oxidation number of the same element?

- (a) Pb_3O_4
- (b) Fe_3O_4
- (c) Fe_2O_3
- (d) Mn₃O₄





What is the oxidation state of Vanadium in

V2O5?

(a) +2

(b) +4

(c) +3

(d) +5