



Practice Problem Set 5 Solutions

The Physics of Particles (University of Western Australia)

11

MANY PARTICLE SYSTEMS

PRACTICE PROBLEMS 5 - SOLUTIONS

- 1(a) For distinguishable particles, Maxwell - Boltzmann statistics apply. Label the particles A and B, and the energy levels by ϵ and 2ϵ . The possible states are:

| ϵ | 2ϵ | <u>TOTAL ENERGY</u> |
|------------|-------------|---------------------|
| AB | — | 2ϵ |
| — | AB | 4ϵ |
| A | B | 3ϵ |
| B | A | 3ϵ |

$$\begin{aligned}
 Z &= e^{-2\epsilon/k_B T} + e^{-4\epsilon/k_B T} + 2e^{-3\epsilon/k_B T} \\
 &= \left(e^{-\epsilon/k_B T} + e^{-2\epsilon/k_B T} \right)^2 \\
 &= Z_1^2
 \end{aligned}$$

where Z_1 is the single particle canonical partition function. The

alternative way to solve this is
to note the for Maxwell-Boltzmann
statistics with two identical particles,

$$Z = (Z_1)^2$$

and $Z_1 = e^{-\epsilon/k_B T} + e^{-2\epsilon/k_B T}$ as

a single particle can occupy energy
levels ϵ_1 and ϵ_2 .

(b) Modified Maxwell-Boltzmann
statistics: we ^{first} treat the particles
as distinguishable, and then try
to correct for overcounting of states
by dividing by $N!$. In this case,
 $N = 2$.

$$\text{So } Z \approx \frac{1}{2!} (Z_1)^2$$

↑ result from Maxwell
Boltzmann stats

$$= \frac{1}{2!} (e^{-\epsilon/k_B T} + e^{-2\epsilon/k_B T})^2$$

$$= \frac{1}{2} e^{-2\epsilon/k_B T} + \frac{1}{2} e^{-4\epsilon/k_B T}$$

$$+ e^{-3\epsilon/k_B T}$$

(c) Bose - Einstein statistics

| ϵ | 2ϵ | <u>TOTAL ENERGY</u> |
|------------|-------------|---------------------|
| AA | - | 2ϵ |
| - | AA | 4ϵ |
| A | A | 3ϵ |

$$Z_{BE} = e^{-2\epsilon/k_B T} + e^{-4\epsilon/k_B T} + e^{-3\epsilon/k_B T}$$

Note that only the third term in this expression is correctly reproduced by modified Maxwell-Boltzmann statistics, because the particles are in different energy levels, and the order $\frac{\epsilon}{A} \frac{2\epsilon}{B}$ or $\frac{\epsilon}{B} \frac{2\epsilon}{A}$ correspond to two different states if the particles are distinguishable.

(d) Fermi - Dirac statistics: two identical particles cannot occupy a given energy level

| ϵ | 2ϵ | <u>TOTAL ENERGY</u> |
|------------|-------------|---------------------|
| A | A | 3ϵ |

$$Z_{FD} = e^{-3\epsilon/k_B T}$$

$$(c) \bar{E} = \sum_n E_n P_n = \sum_n E_n \frac{e^{-E_n/k_B T}}{Z}$$

• For case (c)

$$\bar{E} = \frac{(2\varepsilon e^{-2\varepsilon/k_B T} + 4\varepsilon e^{-4\varepsilon/k_B T} + 3\varepsilon e^{-3\varepsilon/k_B T})}{(e^{-2\varepsilon/k_B T} + e^{-4\varepsilon/k_B T} + e^{-3\varepsilon/k_B T})}$$

• For case (d)

$$\bar{E} = \frac{3\varepsilon e^{-3\varepsilon/k_B T}}{e^{-3\varepsilon/k_B T}} = 3\varepsilon$$

$$\bullet S = - \frac{\partial F}{\partial T}$$

$$= - \frac{\partial}{\partial T} (-k_B T \ln Z)$$

$$= k_B \ln Z + k_B T \frac{1}{Z} \frac{\partial Z}{\partial T}$$

$$\text{For (d), } Z_{FD} = e^{-3\varepsilon/k_B T}$$

$$\Rightarrow S = k_B \left(\frac{-3\varepsilon}{k_B T} \right) + k_B T \frac{1}{e^{-3\varepsilon/k_B T}}$$

$$= -\frac{3\varepsilon}{T} + \frac{3\varepsilon}{T}$$

$$= 0$$

(9) $S = k_B \ln \Omega(E)$ where $\Omega(E)$ is the number of energy eigenstates of energy E .

Even though the system is in contact with a heat reservoir, the Pauli exclusion principle means there is only one energy eigenstate, with $E = 3\varepsilon$.

$$\text{So } S = k_B \ln 1 = 0$$

This is a general result - if a system of particles has only one energy eigenstate, the entropy is zero.

2 (a) The probability of occupancy of the energy eigenstate with energy ϵ_i ($i=1, 2$) is

$$P_i = \frac{e^{-\epsilon_i/k_B T}}{Z}$$

with $Z = e^{-\epsilon_1/k_B T} + e^{-\epsilon_2/k_B T}$.

The mean energy is

$$\begin{aligned} \bar{E} &= \sum_i \epsilon_i e^{-\epsilon_i/k_B T} \\ &= \epsilon_1 \frac{e^{-\epsilon_1/k_B T}}{Z} + \epsilon_2 \frac{e^{-\epsilon_2/k_B T}}{Z} \\ &= \frac{(\epsilon_1 e^{-\epsilon_1/k_B T} + \epsilon_2 e^{-\epsilon_2/k_B T})}{(e^{-\epsilon_1/k_B T} + e^{-\epsilon_2/k_B T})} \end{aligned}$$

(b) $\epsilon_2 - \epsilon_1$ is the energy required to produce a transition from the groundstate to the excited state ($\epsilon_1 \rightarrow \epsilon_2$).

$k_B T$ is a measure of the average thermal energy available.

- High temperature regime:

average thermal energy \gg the energy required to produce a transition from the groundstate to excited state

$$\Rightarrow k_B T \gg E_2 - E_1$$

$$\Rightarrow \frac{(E_2 - E_1)}{k_B T} \ll 1$$

- Low temperature regime:

average thermal energy \ll energy required to produce a transition from the groundstate to the excited state

$$\Rightarrow k_B T \ll E_2 - E_1$$

$$\Rightarrow \frac{(E_2 - E_1)}{k_B T} \gg 1.$$

$$c) \bar{E} = \frac{(\epsilon_1 e^{-\epsilon_1/k_B T} + \epsilon_2 e^{-\epsilon_2/k_B T})}{e^{-\epsilon_1/k_B T} + e^{-\epsilon_2/k_B T}}$$

$$= \frac{e^{-\epsilon_1/k_B T} (\epsilon_1 + \epsilon_2 e^{-(\epsilon_2 - \epsilon_1)/k_B T})}{e^{-\epsilon_1/k_B T} (1 + e^{-(\epsilon_2 - \epsilon_1)/k_B T})}$$

$$= \frac{(\epsilon_1 + \epsilon_2 e^{-(\epsilon_2 - \epsilon_1)/k_B T})}{(1 + e^{-(\epsilon_2 - \epsilon_1)/k_B T})}$$

High temperature limit

$$\frac{(\epsilon_2 - \epsilon_1)}{k_B T} \ll 1$$

$$\Rightarrow e^{-(\epsilon_2 - \epsilon_1)/k_B T} \approx 1 - \frac{(\epsilon_2 - \epsilon_1)}{k_B T}$$

$$\Rightarrow \bar{E} \approx \frac{\epsilon_1 + \epsilon_2 (1 - \frac{(\epsilon_2 - \epsilon_1)}{k_B T})}{1 + 1 - \frac{(\epsilon_2 - \epsilon_1)}{k_B T}}$$

$$\approx \frac{\epsilon_1 + \epsilon_2}{2} \quad \text{since} \quad \frac{(\epsilon_2 - \epsilon_1)}{k_B T} \ll 1$$

Low temperature limit :

$$\frac{(E_2 - E_1)}{k_B T} \gg 1$$

$$\Rightarrow e^{-(E_2 - E_1)/k_B T} \ll 1$$

$$\Rightarrow \bar{E} \approx \frac{E_1}{1} = E_1$$

(d) High temperature limit :

There is a lot of thermal energy available

\Rightarrow the system is readily knocked from the groundstate to the excited state

\Rightarrow both states have approximately equal probability to be occupied

$$\Rightarrow \bar{E} \approx \frac{1}{2} E_1 + \frac{1}{2} E_2$$

Low temperature limit

There is little thermal energy available

\Rightarrow the probability of excitation of the system from the groundstate energy E_1 to the excited state energy E_2 is low

\Rightarrow the system is very likely to be in the groundstate with energy E_1

$$\Rightarrow \bar{E} \approx E_1$$