

MPS PRACTICE PROBLEM SET 7

4

SOLUTIONS

1(a) Number of helium atoms per unit volume

$$= 10^9 \text{ kg m}^{-3} \times \frac{1 \text{ helium atom}}{7 \times 10^{-27} \text{ kg}}$$

$$= 1.4 \times 10^{35} \text{ helium atoms per m}^3.$$

Each helium atom is completely ionized, and so contributes two electrons to the Fermi gas

$$\Rightarrow \text{density of electron gas} = 2.8 \times 10^{35} \text{ electrons/m}^3$$

(b) From lecture

$$\epsilon_F = \frac{\hbar^2}{2m} \left(3\pi^2 \frac{\bar{N}}{V} \right)^{2/3}$$

$$\text{From (a), } \frac{\bar{N}}{V} = 2.8 \times 10^{35} \text{ m}^{-3}$$

$$m = \text{electron mass} = 9.1 \times 10^{-31} \text{ kg}$$

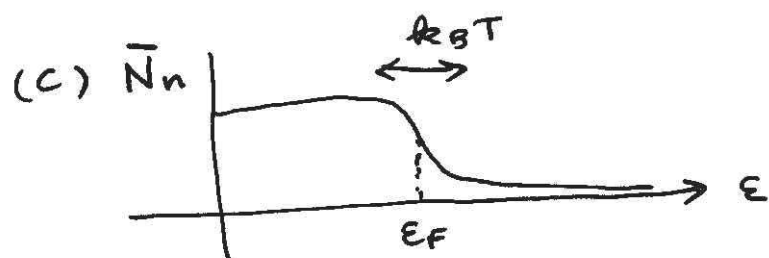
$$\hbar = \frac{h}{2\pi} = \frac{1}{2\pi} \times 6.6 \times 10^{-34} \text{ Js.}$$

$$\begin{aligned} \Rightarrow \epsilon_F &= \frac{1}{4\pi^2} \times \frac{(6.6 \times 10^{-34} \text{ kg m}^2 \text{ s}^{-1})^2}{2 \times 9.1 \times 10^{-31} \text{ kg}} \times (3\pi^2)^{2/3} \\ &\quad \times (2.8 \times 10^{35} \text{ m}^{-3})^{2/3} \end{aligned}$$

$$= 2.5 \times 10^{-14} \text{ kg m}^2 \text{ s}^{-2}$$

$$= 2.5 \times 10^{-14} \text{ J} \times \frac{1 \text{ eV}}{1.6 \times 10^{-19} \text{ J}}$$

$$= 1.6 \times 10^5 \text{ eV}$$



The electron gas is degenerate if $k_B T \ll \epsilon_F$.

From (b), $\epsilon_F \approx 1.6 \times 10^5 \text{ eV}$.

$$k_B T = 1.4 \times 10^{-23} \text{ J K}^{-1} \times 10^7 \text{ K}$$

$$= 1.4 \times 10^{-16} \text{ J} \times \frac{1 \text{ eV}}{1.6 \times 10^{-19} \text{ J}}$$

$$= 8.8 \times 10^2 \text{ eV}$$

$$\ll \epsilon_F.$$

So the electron gas can be treated as being degenerate i.e. its properties will be similar to those for an electron gas at $T=0$.

$$\begin{aligned}
 (d) \quad \frac{GM^2}{R^4} &\sim \text{electron gas pressure} \\
 &\sim \epsilon_F^{5/2} \quad \text{from lecture notes} \\
 &\sim \left(\frac{\bar{N}}{V}\right)^{5/3}
 \end{aligned}$$

Since $V = R^3$, and the number of electrons \bar{N} is twice the mass of the star divided by the mass of a helium atom

$$\frac{GM^2}{R^4} \sim \left(\frac{M}{R^3}\right)^{5/3} = \frac{M^{5/3}}{R^5}$$

$$\Rightarrow M^{5/3} - 2 \sim R^{5-4}$$

$$\Rightarrow M^{-1/3} \sim R$$

$$\begin{aligned}
 (e) \quad \text{Density } \rho &= \frac{\text{mass}}{\text{volume}} \\
 &= \frac{M}{\frac{4}{3}\pi R^3} \\
 &\sim \frac{M}{M^{-1}} \\
 &= M^2
 \end{aligned}$$