

MPS 2020 Test Solutions

The Physics of Particles (University of Western Australia)

Dear Class!

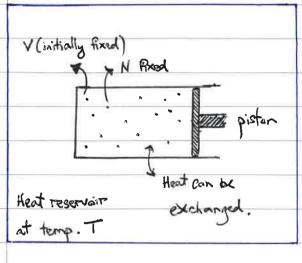
The solutions that follow are written in an a v. detailed form with the intention of improving your understanding.

They contain, in many cases, more material than is necessary. In a test exam setting you must pick and choose the important parts to demonstrate your understanding. I hope this helps your revision.

Cheers Daniel

2020 Many Particle Systems Test - SOLUTIONS

Q1 Consider the following system, fixed in volume (initially) or with a fixed number of particles in equilibrium with a heat reservoir:



A sample of an ideal, monatonic gas is initially confined to this volume with the piston locked in position.

(a) Upon release of the piston, the gas in general will enter a non-equilibrium state. However, once equilibrium is

re-established the direction of spontaneous

change is determined by the requirement that dFgs < 0, or AFgs < 0 (macroscopies

The given expression for $F(T,V,N) = \frac{3}{2}Nk_BT \ln\left(\frac{2\pi k^2N^{2/2}}{mk_BT}\right) - Nk_BT \ln V$.

So that:
$$\Delta F = F(T, V_f, N) - F(T, V_i, N)$$

$$= -Nk_BT(lnV_f - lnV_i) = Nk_BTln(V_i/V_f).$$
Now $\Delta F < O \Rightarrow ln(V_i) < O \Rightarrow V_f > V_i$

So upon release, spontaneous change will occur so that the piston 4 marks moves to increase the volume of the monatomic gas.

(b). For the above process determine
$$\Delta S_{system}$$
. We have $F_{sys} = \overline{E}_{sys} - T S_{sys}$. With $\overline{E}_{sys} = \frac{3}{2} N k_R T$

and $\Delta F_{sys} = \Delta E_{sys} - T\Delta S_{sys}$ as the process is isothermal.

Furthermore: $\Delta E_{sys} = \frac{3}{2} Nk_B \Delta T = 0$.

So $\Delta F_{sys} = -T\Delta S_{sys} < O \Rightarrow \Delta S_{sys} > O$.

4 marks

So upon release of the piston the entropy of the monatomic gas must increase. ie microscopically, the larger valume leads to more quantum stodes accessable to the gas atoms.

Q2. The system is N, indistinguishable (eg, as in a gas), identical, non-interacting, spin O bosons. It occupies a volume V and is in contact with a heat reservoir at temperature T. As they are Bosons, there is no restriction as to how many particles can occupy a single particle eigenstate. The single particle spectrum consists of only two eigenstates 10> 4 1€>.

(a) The list of the possible N-particle eigenstates for the system is as follows:

micro-slade	10>	1 <i>e</i> >	Estate = Nig 0+Nig) E	14>
1	<u>AAA</u> A	_	0	10/10/210/310/N
2	AAA A	A	€	107,1072 107, 167, +
3	AAA A N-2	AA	2€	107,1072 167, 167, 167, + appropriate terms
;		N-1	•	<u></u>
N	A	A AAA	(N-1) €	107, 167, 167, 107, 167, 107, 167, +
N+1	_	A AAA	Ne.	(e), 16/2 (e),

THIS COLUMN FOR INTEREST

Lt marks

ONTA

So there are N+1, N particle energy eigenstates as above with energies ne with n an integer 0 & n & N.

	So! lim S = 0.	On physi	cal ground	s this makes	sense as
2 marks	So! lim S = 0.	the system	= freeze	e out "into	the N
Ĭ.		particle at	round slak	. or microsol	x 1
	in the toble in (a). There is no the	ernal energy	available	(KBT - O) from
	the heat reservoir to excite the syste	m into N Do	oticle micr	states > 1	
	Ac we know with complete certain	the the state	of the 34	stom the	entropy "15
	As we know with complete certain O or S = kgln D = kgln 1 =	0 JK-1	1		B
	(f). For N distinguishable partic	loe			
-	- (- · · ·) - · (-	N	/	- E/KRT) N	
4 martis	Z(T,V,N) = Z(T	, 1, 1) =	(1+e	/ '-'s' /	
					7
	Q3. Our system conside of 3 p.	articles whose	oceupanc	y of a sing	h
	particle energy eigenslate is	restricted to	being O	, 1 or 2.	
	The single particle energy species simply illustrated to the r	drum	1		
	is simply illustrated to the r	ight: = 3		3€	37
	31	-	R	<u>2e</u>	127
	if the particles are indentical	part single		E	11>
	industinguishable find ZCT, V			41-	
19	for our 3 particle system.				
) '	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	12>	13>	Estate
	Possible system configurations are as	AA	A	-	40
	listed.	AA	-	A	5€
		Α	AA	-	5€
	ZCT,V,3) = Z e - Egholykist.	tum-	AA	A	76
3	three porticle cretgy eigenstals	A	_	AA	76
part service.	and a Sources	gar en.	A	AA	8€
***		Α	A	A	66.

7 (T v 2) -	0-46/KBT +	20-5e/kgT+	e-66/kgT+	2e-76/kgT+	e-86/kg
~ (, 1, 3)		~			_

8 marks = z4 + 2z5 + z6 + 2z7 + z8

Z = e - E/KRT

Q3. Here the system is 3 non-interacting E1

particles, each with a single particle

energy spectrum ne with

n = 0, 1, 2, 3, ... as shown to the right.

3 ∈ 13)

(like the simple Harmonic oscillator but 20 12)

without the zero point energy).

(a). The system is isolated from it's

environment wife total energy E = 3E. Under these circumstances, the micro-cannonical distribution applies and we need to list all the accessible micro-states such that the total energy E = E particle 1 + E particle 2 + E particle 3 + E

(i) The 3 particles are distinguishable (as in sites in a lattice)

3€. The entropy will then be S = kBln Ds

-	0	e	26	3€	microstate
	AB		,	C	1
	BC	7	Wins .	A	2 3! parmulations
	CA	-	gallerio .	13	3 of ABC lends
	A	B	C	-	it, 5, 6, 7, 8, 9) to 6 microstotes
	_	ABC		_	10.

3 marks

So there are 10 microstoles s.t. the total system energy is 3E

=> S= kgln 10 JK-1

	Since the single particle energy spectrum is unbounded	(countably infinite) it
	should be intuitively obvious that there will be an infi	ite number of 3-particle
	states contributing to I(T, V, 3).	,
	Some of the low energy 3 - particle states are tobula	ded below:
	10% 117 127 137	Estate
	AAA	0
	AA A —	Ę
	AA - A -	2€
	A A -	26
	AA A	3e
	A A A	36
	AAA	3€
		Š
	Although it is tempting to speculate that Z(T, V, 3) = 2	ine-ne this pattern
	is broken at higher n (try n=8).	140
	The best use can do is appeal to the Generating Func	tional from lectures.
	So let the occupation numbers of the single particles	tooks be (No, M, N2,)
		10> 11> 12>
	Then $Z(T,V,x) = \sum_{N \in Q} x^N Z(T,V,N)$	
	Neo 4	
	= \(\sum_{\mathbb{N}_0 + \mu_1 + \ldots} = \ldots \(\mu_0 \) \(\mu_0 \) + \(\mu_1 \) \(\mu_0 \)	+ N2.ZE +)/KBT.
	(No, N1,)	. 4
	= (\sum_{N=0}^{\infty} \times_{N=0}^{\infty} \times_{N=0}^{\infty} \times_{N=0}^{\infty} \times_{N=0}^{\infty}	NIE-NIERT) 8 = e-82
	No = 0 No 20	, ,
	$= \left(\frac{1}{1-\infty}\right)\left(\frac{1}{1-\infty^2}\right)\left(\frac{1}{1-\infty}\right)$	
در	1-02 1-07 1-0	C. C.
7	$= \frac{1}{1100} \frac{1}{(1-\omega_{5}^{-1})}$	So ! (AT.O)
	n== (1-025)	(1,1,0)
	, and the state of	

_	
	To determine Z(T, V, 3) we could expand this series and collect the
	coefficients of x3.
×.	A more economical way of saying this is to evaluate the following
	for the case N=3.
	and the second s
I'm ark	$Z(T,V,N) = \frac{1}{N!} \frac{\partial}{\partial x} Z(T,V,\alpha)$
	$Z(T,V,N) = \frac{1}{N!} \frac{\partial^{N}}{\partial x^{N}} Z(T,V,\alpha)$ $\alpha = 0$
	This is obviously more easier said than done . Any arguments hinting at this
	This is obviously more easier said than done . Any arguments hinting at this attracts some marks -
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