MANY PARTICLE SYSEMS PRACTICE PROBLEMS 5- SOLUTIONS

(a) For distinguishable particles,

Maxwell - Boltzmann statistic apply.

Liabel the patriles A and B, and

the energy levels by E and ZE.

The possible states are:

ع:	22	TOTAL ENERGY
AB	_	2 &
	AB	48
A	В	3 €
8	A	3 8

$$Z = e^{-25/k_BT} + e^{-45(k_BT)} + 2e^{-35/k_BT}$$

$$= \left(e^{-5/k_BT} + e^{-25/k_BT}\right)^2$$

$$= Z_1^2$$

where Z, is the single particle canonical partition Function. The

and Z1 = e = Ellest + e -25/lest as a single patile can occupy energy levelo E, and Ez.

(b) Modified Maxwell-Beltzmann first the patieles as distinguishable, and then try to correct for overcounting of states by dwiding by N! . In this case,

So Z ~ 1 (Z1)2

(result from Maxwell Boltzmann state

(c) Bose - Einstein statistics

$$\frac{\mathcal{E}}{AA} - \frac{2\mathcal{E}}{AA} - \frac{2\mathcal{E}}{AA} + \frac{3\mathcal{E}}{AA} + \frac{2\mathcal{E}}{AA} + \frac{2\mathcal{$$

Note that only the third term in this expression is comethly reproduced by modified Maxwell-Boltzmann statistics, because the particles are in different energy levels, and the order $\frac{22}{AB}$ or $\frac{24}{BA}$ correspond to two different states if the particles are distinguishable.

(d) Fermi - Derac Statistics: two identical perticles cannot occupy a guier energy level

Zp = e -3E/RBT

$$\overline{E} = \frac{(28e^{-28/RBT} + 48e^{-48/RBT} - 38/RBT)}{(e^{-28/RBT} + e^{-48/RBT} + e^{-38/RBT})}$$

$$= -\frac{3\varepsilon}{T} + \frac{3\varepsilon}{T}$$

(9) $5 = k_B \ln \Omega(E)$ where $\Omega(E)$ is the number of energy eigenstates of energy E.

Even though the system is in contact with a heat reservoir, the Pauli exclusion principle means there is only one evergy eigenstate, with E = 3E.

So $S = k_B \ln I = 0$

This is a general result - if a system of particles has only one energy eigenstate; the entropy is zero.

2 (a) The probability of occupancy of the energy eigenstate with energy ε_{λ} $(\lambda=1,2)$ Pi = e = Z with Z = e = \(\in \(\) + e = \(\) . The mean energy is E = Zi En e ErilkBT = E, e-E,/BBT + Ez e -Ez/kBT

 $= (E_{1}e^{-E_{1}/k_{B}T} + E_{2}e^{-E_{2}/k_{B}T})$ $= (e^{-E_{1}/k_{B}T} + e^{-E_{2}/k_{B}T})$

(b) E_2-E_1 is the energy required to produce a transition from the groundstate to the excited state $(E_1 \rightarrow E_2)$.

RBT is a measure of the average thermal

energy available.

. High temperature regime:

average thermal energy >> the energy vegured to produce a transition from the groundstate to exceled state

→ RBT>> E2-E.

 $=) \frac{(E_2 - E_1)}{ABT} << 1$

. Low temperature regime:

overage thermal energy << energy vegured to produce a transition from the groundstate to the excited date

→ RBT << E2-€1

=) (E2-E1) RBT >>1.

$$E = \frac{(\Xi_{1} e^{-\Xi_{1}/RBT} + \Xi_{2} e^{-\Xi_{2}/RBT})}{e^{-\Xi_{1}/RBT} + e^{-\Xi_{2}/RBT}}$$

$$= \frac{e^{-\Xi_{1}/RBT} (\Xi_{1} + \Xi_{2} e^{-\Xi_{2}/RBT})}{(\Xi_{1} + \Xi_{2} e^{-\Xi_{1}/RBT})}$$

$$= \frac{e^{-\Xi_{1}/RBT} (\Xi_{1} + \Xi_{2} e^{-\Xi_{1}/RBT})}{(\Xi_{2} - \Xi_{1}/RBT)}$$

$$= \frac{(\Xi_{1} + \Xi_{2} e^{-\Xi_{1}/RBT})}{(\Xi_{1} - \Xi_{1}/RBT)}$$

$$=\frac{(\varepsilon_1+\varepsilon_2e^{-(\varepsilon_2-\varepsilon_i)/k_BT})}{(1+e^{-(\varepsilon_2-\varepsilon_i)/k_BT})}$$

High temperature limit
$$\frac{(\Xi_{2}-\Xi_{1})}{\Re BT} <<1$$

$$\Rightarrow e^{-(\Xi_{2}-\Xi_{1})/RBT} \approx 1-\frac{(\Xi_{2}-\Xi_{1})}{\Re BT}$$

$$= \sum_{i=1}^{\infty} \frac{\mathcal{E}_{1} + \mathcal{E}_{2} \left(1 - \frac{(\mathcal{E}_{2} - \mathcal{E}_{1})}{\mathcal{R}_{B}T}\right)}{1 + 1 - \frac{(\mathcal{E}_{2} - \mathcal{E}_{1})}{\mathcal{R}_{B}T}}$$

$$\approx \frac{\varepsilon_1 + \varepsilon_2}{2}$$
 since $\frac{(\varepsilon_2 - \varepsilon_1)}{RB7} < < 1$

Low temperature limit:

$$\frac{(E_2 - E_1)}{RBT} >> 1$$

$$\frac{RBT}{RBT} >> 1$$

$$e^{-(E_2 - E_1)/RBT} << 1$$

$$= \frac{E_1}{RBT} = E_1$$

(d) High temperature limit:

There is a lot of thermal energy

available

The system is readily broked from

the groundstate to the excited state

both state have approximately

equal probability to be occupied

From

Ex = 2 = 1 + 2 = 2

There is little thermal energy avaluable

There is little thermal energy avaluable

The probability of excitation of the

system from the groundstate energy E;

to the excited state energy Ez is low

the system is very likely to be in

the groundstate with energy E;

⇒ E ≈ E