

Humus

In classical soil science, **humus** is the dark organic matter in soil that is formed by the decomposition of plant and animal matter. It is a kind of soil organic matter. It is rich in nutrients and retains moisture in the soil. Humus is the Latin word for "earth" or "ground". [2]

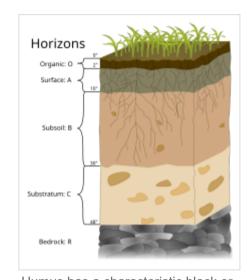
In <u>agriculture</u>, "humus" sometimes also is used to describe mature or natural <u>compost</u> extracted from a woodland or other spontaneous source for use as a <u>soil conditioner</u>. It is also used to describe a <u>topsoil horizon</u> that contains organic matter (humus type, [4] humus form, [5] or humus profile.

Humus has many <u>nutrients</u> that improve the health of soil, <u>nitrogen</u> being the most important. The ratio of <u>carbon</u> to <u>nitrogen</u> (C:N) of humus commonly ranges between 8:1 and 15:1 with the median being about 12:1. It also significantly improves (decreases) the <u>bulk density</u> of soil. Humus is <u>amorphous</u> and lacks the cellular structure characteristic of organisms.

The <u>solid</u> residue of <u>sewage</u> sludge treatment, which is a secondary phase in the <u>wastewater treatment</u> process, is also called humus. [10] When not judged <u>contaminated</u> by <u>pathogens</u>, toxic <u>heavy metals</u>, or <u>persistent organic pollutants</u> according to standard tolerance levels, it is sometimes composted and used as a soil amendment. [11]

Description

The primary materials needed for the process of humification are plant detritus and dead animals and microbes, excreta of all soil-dwelling organisms, and also <u>black carbon</u> resulting from past fires. The composition of humus varies with that of primary (plant) materials and secondary microbial and animal products. The decomposition rate of the different compounds will affect the composition of the humus. 13



Humus has a characteristic black or dark brown color and is an accumulation of organic <u>carbon</u>. Besides the three major soil horizons of (A) surface/topsoil, (B) <u>subsoil</u>, and (C) substratum, some soils have an organic horizon (O) on the very surface. Hard bedrock (R) is not in a strict sense soil.

It is difficult to define humus precisely because it is a very complex substance which is still not fully understood. Humus is different from decomposing soil organic matter. The latter looks rough and has visible remains of the original plant or animal matter. Fully humified humus, on the contrary, has a uniformly dark, spongy, and jelly-like appearance, and is amorphous; it may gradually decay over several years or persist for millennia. It has no determinate shape, structure, or quality. However, when examined under a microscope, humus may reveal tiny plant, animal, or microbial remains that have been mechanically, but not chemically, degraded. This suggests an ambiguous boundary between humus and soil organic matter, leading some

authors to contest the use of the term *humus* and derived terms such as $\underline{humic\ substances}$ or $\underline{humification}$, proposing the $Soil\ Continuum\ Model\ (SCM)$. However, humus can be considered as having distinct properties, mostly linked to its richness in $\underline{functional\ groups}$, justifying its maintenance as a specific term. $\underline{[17]}$

Fully formed humus is essentially a collection of very large and complex molecules formed in part from <u>lignin</u> and other <u>polyphenolic</u> molecules of the original plant material (foliage, wood, bark), in part from similar molecules that have been produced by <u>microbes</u>. During decomposition processes these polyphenols are modified chemically so that they are able to join up with one another to form very large molecules. Some parts of these molecules are modified in such a way that protein molecules, amino acids, and amino sugars are able to attach themselves to the polyphenol "base" molecule. As protein contains both nitrogen and sulfur, this attachment gives humus a moderate content of these two important plant nutrients. [19]

Radiocarbon and other dating techniques have shown that the polyphenolic base of humus (mostly <u>lignin</u> and <u>black carbon</u>) can be very old, but the <u>protein</u> and <u>carbohydrate</u> attachments much younger, while to the light of modern concepts and methods the situation appears much more complex and unpredictable than previously thought. It seems that microbes are able to pull protein off humus molecules rather more readily than they are able to break the polyphenolic base molecule itself. As protein is removed its place may be taken by younger protein, or this younger protein may attach itself to another part of the humus molecule. 121

The most useful functions of humus are in improving <u>soil structure</u>, all the more when associated with <u>cations</u> (e.g. <u>calcium</u>), $\underline{^{[22]}}$ and in providing a very large surface area that can hold nutrient elements until required by plants, an ion exchange function comparable to that of clay particles. $\underline{^{[23]}}$

Soil <u>carbon sequestration</u> is a major property of the soil, also considered as an <u>ecosystem service</u>. Only when it becomes stable and acquires its multicentury permanence, mostly via multiple interactions with the <u>soil matrix</u>, molecular soil humus should be considered to be of significance in removing the atmosphere's current carbon dioxide overload.

There is little data available on the composition of humus because it is a complex mixture that is challenging for researchers to analyze. Researchers in the 1940s and 1960s tried using chemical separation to analyze plant and humic compounds in forest and agricultural soils, but this proved impossible because extractants interacted with the analysed organic matter and created many artefacts. [26] Further research has been done in more recent years, though it remains an active field of study. [27]

Humification

<u>Microorganisms</u> decompose a large portion of the soil organic matter into inorganic minerals that the roots of plants can absorb as <u>nutrients</u>. This process is termed <u>mineralization</u>. In this process, <u>nitrogen</u> (<u>nitrogen cycle</u>) and the other nutrients (<u>nutrient cycle</u>) in the decomposed organic matter are recycled. Depending on the conditions in which the decomposition occurs, a fraction of the organic matter does not mineralize and instead is transformed by a

process called *humification*. Prior to modern analytical methods, early evidence led scientists to believe that humification resulted in concatenations of organic polymers resistant to the action of microorganisms, $\frac{[28]}{}$ however recent research has demonstrated that microorganisms are capable of digesting humus. $\frac{[29]}{}$

Humification can occur naturally in <u>soil</u> or artificially in the production of <u>compost</u>. Organic matter is humified by a combination of <u>saprotrophic</u> fungi, bacteria, microbes and animals such as earthworms, nematodes, protozoa, and arthropods (see <u>Soil biology</u>). Plant remains, including those that animals digested and excreted, contain organic compounds: sugars, starches, proteins, carbohydrates, lignins, waxes, resins, and organic acids. Decay in the soil begins with the decomposition of sugars and starches from carbohydrates, which decompose easily as <u>detritivores</u> initially invade the dead plant organs, while the remaining <u>cellulose</u> and <u>lignin</u> decompose more slowly. Simple proteins, organic acids, starches, and sugars decompose rapidly, while crude proteins, fats, waxes, and resins remain relatively unchanged for longer periods of time. [30]

Lignin, which is quickly transformed by white-rot fungi, [31] is one of the primary precursors of humus, [32] together with by-products of microbial animal animal activity. The humus produced by humification is thus a mixture of compounds and complex biological chemicals of plant, animal, and microbial origin that has many functions and benefits in soil. Some judge earthworm humus (vermicompost) to be the optimal organic manure.

Stability

Much of the humus in most soils has persisted for more than 100 years, rather than having been decomposed into CO_2 , and can be regarded as stable; this organic matter has been protected from decomposition by microbial or enzyme action because it is hidden (occluded) inside small aggregates of soil particles, or tightly sorbed or complexed to clays. Most humus that is not protected in this way is decomposed within 10 years and can be regarded as less stable or more labile. The mixing activity of soil-consuming invertebrates (e.g. earthworms, termites, some millipedes) contribute to the stability of humus by favouring the formation of organo-mineral complexes with clay at the inside of their guts, labeled where more carbon sequestration in humus forms such as mull and amphi, with well-developed mineral-organic horizons, when compared with moder where most organic matter accumulates at the soil surface. Surface.

Stable humus contributes few plant-available nutrients in soil, but it helps maintain its physical structure. A very stable form of humus is formed from the slow oxidation (redox) of soil carbon after the incorporation of finely powdered charcoal into the topsoil, suggested to result from the grinding and mixing activity of a tropical earthworm. This process is speculated to have been important in the formation of the unusually fertile Amazonian terra preta do Indio. However, some authors suggest that complex soil organic molecules may be much less stable than previously thought: "the available evidence does not support the formation of large-molecular-size and persistent 'humic substances' in soils. Instead, soil organic matter is a continuum of progressively decomposing organic compounds."

Horizons

Humus has a characteristic black or dark brown color and is organic due to an accumulation of organic carbon. Soil scientists use the capital letters O, A, B, C, and E to identify the master <u>soil horizons</u>, and lowercase letters for distinctions of these horizons. Most soils have three major horizons: the surface horizon (A), the subsoil (B), and the substratum (C). Some soils have an organic horizon (O) on the surface, but this horizon can also be buried. The master horizon (E) is used for subsurface horizons that have significantly lost minerals (<u>eluviation</u>). Bedrock, which is not soil, uses the letter R. The richness of soil horizons in humus determines their more or less dark color, generally decreasing from O to E, to the exception of deep horizons of podzolic soils enriched with colloidal humic substances which have been leached down the soil profile.

Benefits of soil organic matter and humus

The importance of chemically stable humus is thought by some to be the <u>fertility</u> it provides to soils in both a physical and chemical sense, though some agricultural experts put a greater focus on other features of it, such as its ability to suppress disease. It helps the soil retain moisture by increasing microporosity and encourages the formation of good <u>soil structure</u>. The incorporation of <u>oxygen</u> into large organic molecular assemblages generates many active, negatively charged sites that bind to positively charged <u>ions</u> (cations) of <u>plant nutrients</u>, making them more available to the plant by way of ion exchange. Humus allows soil organisms to feed and reproduce and is often described as the "life-force" of the soil. Is a life of the soil.

- The process that converts soil organic matter into humus feeds the population of microorganisms and other creatures in the soil, and thus maintains high and healthy levels of soil life. [53][54]
- The rate at which soil organic matter is converted into humus promotes (when fast, e.g. <u>mull</u>) or limits (when slow, e.g. <u>mor</u>) the coexistence of plants, animals, and microorganisms in the soil. [55]
- "Effective humus" and "stable humus" are additional sources of nutrients for microbes: the former provides a readily available supply, and the latter acts as a long-term storage reservoir. [56]
- Decomposition of dead plant material causes complex organic compounds to be slowly oxidized (lignin-like humus) or to decompose into simpler forms (sugars and amino sugars, and aliphatic and phenolic organic acids), which are further transformed into microbial biomass (microbial humus) or reorganized and further oxidized into humic assemblages (fulvic acids and humic acids), which bind to clay minerals and metal hydroxides. The ability of plants to absorb humic substances with their roots and metabolize them has been long debated. There is now a consensus that humus functions hormonally rather than simply nutritionally in plant physiology, and that organic sunstances exuded by roots and transformed in humus by soil organisms are an evolved strategy by which plants "talk" to the soil. [61]
- Humus is a negatively <u>charged colloidal</u> substance which increases the <u>cation-exchange capacity</u> of soil, hence its ability to store nutrients by <u>chelation</u>. [62] While these nutrient cations are available to plants, they are held in the soil and prevented from being leached by rain or irrigation. [52]
- Humus can hold the equivalent of 80–90% of its weight in moisture and therefore increases the soil's capacity to withstand drought. [63]
- The biochemical structure of humus enables it to moderate, i.e. buffer, excessive acidic or alkaline soil conditions. [64]

- During humification, microbes secrete sticky, gum-like <u>mucilages</u>; these contribute to the crumby structure (<u>tilth</u>) of the soil by adhering particles together and allowing greater <u>aeration</u> of the soil. [65] Toxic substances such as <u>heavy metals</u> and excess nutrients can be chelated, i.e., bound to the organic molecules of humus, and so prevented from leaching away. [66]
- The dark, usually brown or black, color of humus helps to warm cold soils in spring. [67]
- Humus can contribute to <u>climate change mitigation</u> through its <u>carbon sequestration</u> potential. Artificial humic acid and artificial fulvic acid synthesized from agricultural litter can increase the content of dissolved organic matter and total organic carbon in soil. [69]

See also

- Biochar
- Biomass
- Biotic material
- Detritus
- Glomalin
- Humic acid
- Immobilization (soil science)

- Mineralization (soil science)
- Mycorrhizal fungi and soil carbon storage
- Organic matter
- Plant litter
- Soil horizon
- Soil science
- Terra preta

References

- 1. Popkin, Gabriel (27 July 2021), <u>A soil-science revolution upends plans to fight climate change</u> (https://www.quantamagazine.org/a-soil-science-revolution-upends-plans-to-fight-climate-change-20210727/), Quanta Magazine, retrieved 9 June 2024, ""The latest edition of The Nature and Properties of Soils, published in 2016, cites Lehmann's 2015 paper and acknowledges that "our understanding of the nature and genesis of soil humus has advanced greatly since the turn of the century, requiring that some long-accepted concepts be revised or abandoned"."
- 2. "Humus" (https://www.dictionary.com/browse/humus). Retrieved 9 June 2024 via Dictionary.com *Random House Dictionary Unabridged*.
- 3. "Humus" (https://www.britannica.com/EBchecked/topic/276408/humus). *Encyclopaedia Britannica Online*. 2011. Retrieved 9 June 2024.
- 4. Chertov, Oleg G.; Komarov, Alexander S.; Crocker, Graham; Grace, Peter; Klir, Jan; Körschens, Martin; Poulton, Paul R.; Richter, Daniel (1997). "Simulating trends of soil organic carbon in seven long-term experiments using the SOMM model of the humus types" (https://fr.articles.sk/book/17536355/c35523). Geoderma. 81 (1–2): 121–135. Bibcode:1997Geode..81..121C (https://ui.adsabs.harvard.edu/abs/1997Geode..81..121C). doi:10.1016/S0016-7061(97)00085-2 (https://doi.org/10.1016%2FS0016-7061%2897%2900085-2). Retrieved 9 June 2024.
- Brêthes, Alain; Brun, Jean-Jacques; Jabiol, Bernard; Ponge, Jean-François; Toutain, François (1995). "Classification of forest humus forms: a French proposal" (https://www.researchgate.net/publication/45341270). Annales des Sciences Forestières. 52 (6): 535–46. doi:10.1051/forest:19950602 (https://doi.org/10.1051%2Fforest%3A19950602). Retrieved 16 June 2024.

- Bernier, Nicolas (1998). "Earthworm feeding activity and development of the humus profile" (https://www.academia.edu/34816078). Biology and Fertility of Soils. 26 (3): 215–23. Bibcode:1998BioFS..26..215B (https://ui.adsabs.harvard.edu/abs/1998BioFS..26..215B). doi:10.1007/s003740050370 (https://doi.org/10.1007%2Fs003740050370). Retrieved 16 June 2024.
- 7. Brady, Nyle C. (1984). *The nature and properties of soils* (https://www.ac ademia.edu/23641831) (9th ed.). New York, New York: Macmillan Publishing Company. p. 269. ISBN 978-0029460306. Retrieved 1 September 2024.
- 8. Bauer, Armand (1974). "Influence of soil organic matter on bulk density and available water capacity of soils" (https://library.ndsu.edu/ir/bitstrea m/handle/10365/24299/ndfr_19740501_v31_iss05_044.pdf) (PDF). Farm Research. 31 (5): 44–52. Retrieved 23 June 2024.
- 9. Whitehead, D. C.; Tinsley, J. (1963). "The biochemistry of humus formation" (https://fr.articles.sk/book/1689524/861585). Journal of the Science of Food and Agriculture. 14 (12): 849–57.

 Bibcode:1963JSFA...14..849W (https://ui.adsabs.harvard.edu/abs/1963JSFA...14..849W). doi:10.1002/jsfa.2740141201 (https://doi.org/10.1002%2Fjsfa.2740141201). Retrieved 23 June 2024.
- 10. "Sewage treatment" (https://library.e.abb.com/public/19d4b5f59e87bdd9c 12569580054d17e/3 sewage.pdf) (PDF). Retrieved 30 June 2024.
- 11. Brinton, William F. (2020). "Compost quality standards and guidelines, final report" (https://compost.css.cornell.edu/Brinton.pdf) (PDF). Ithaca, New York: Cornell University. Retrieved 7 July 2024.
- 12. Guggenberger, Georg (2005). "Humification and mineralization in soils". In Buscot, François; Varma, Ajit (eds.). *Microorganisms in soils: roles in genesis and Functions* (http://ndl.ethernet.edu.et/bitstream/123456789/7 5774/1/Franc%C2%B8ois%20Buscot.pdf#page=102) (PDF). Soil biology. Vol. 3. Dordrecht, The Netherlands: Springer. pp. 85–106. doi:10.1007/3-540-26609-7_4 (https://doi.org/10.1007%2F3-540-26609-7_4). ISBN 978-3-540-26609-9. Archived (https://web.archive.org/web/202407 07083204/http://ndl.ethernet.edu.et/bitstream/123456789/75774/1/Franc%C2%B8ois%20Buscot.pdf#page=102) (PDF) from the original on 7 July 2024. Retrieved 8 September 2024.

- 13. Kögel-Knabner, Ingrid; Zech, Wolfgang; Hatcher, Patrick G. (1988).

 "Chemical composition of the organic matter in forest soils: the humus layer" (https://fr.articles.sk/book/34689934/0bcef3). Journal of Plant Nutrition and Soil Science. 151 (5): 331–40.

 doi:10.1002/jpln.19881510512 (https://doi.org/10.1002%2Fjpln.19881510512). Retrieved 14 July 2024.
- 14. Waksman, Selman A. (1936). *Humus: origin, chemical composition and importance in nature* (https://citeseerx.ist.psu.edu/document?repid=rep1 &type=pdf&doi=45020e04c07d0fa28dca0093772951b65197eb2e).
 Baltimore, Maryland: Williams & Wilkins. ISBN 9780598966629.
 Retrieved 14 July 2024.
- 15. Bernier, Nicolas; Ponge, Jean-François (1994). "Humus form dynamics during the sylvogenetic cycle in a mountain spruce forest" (https://www.researchgate.net/publication/46312511). Soil Biology and Biochemistry. 26 (2): 183–220. Bibcode:1994SBiBi..26..183B (https://ui.adsabs.harvard.edu/abs/1994SBiBi..26..183B). doi:10.1016/0038-0717(94)90161-9 (https://doi.org/10.1016%2F0038-0717%2894%2990161-9). Retrieved 14 July 2024.
- Lehmann, Johannes; Kleber, Markus (2015). "The contentious nature of soil organic matter" (https://themarea.org/wp-content/uploads/2018/08/Lehmann-and-Kebbler-2015.pdf) (PDF). Nature. 528 (7580): 60–68. Bibcode:2015Natur.528...60L (https://ui.adsabs.harvard.edu/abs/2015Natur.528...60L). doi:10.1038/nature16069 (https://doi.org/10.1038%2Fnature16069). PMID 26595271 (https://pubmed.ncbi.nlm.nih.gov/26595271). Retrieved 14 July 2024.
- 17. Ponge, Jean-François (2022). "Humus: dark side of life or intractable "aether"?" (https://www.researchgate.net/publication/360175852). Pedosphere. **32** (4): 660–64. Bibcode:2022Pedos..32..660P (https://ui.adsabs.harvard.edu/abs/2022Pedos..32..660P). doi:10.1016/S1002-0160(21)60013-9 (https://doi.org/10.1016%2FS1002-0160%2821%2960013-9). Retrieved 14 July 2024.
- Dou, Sen; Shan, Jun; Song, Xiangyun; Cao, Rui; Wu, Meng; Li, Chenglin; Guan, Song (April 2020). "Are humic substances soil microbial residues or unique synthesized compounds? A perspective on their distinctiveness" (https://www.researchgate.net/publication/338991840). Pedosphere. 30 (2): 159–67. Bibcode:2020Pedos..30..159D (https://ui.adsabs.harvard.edu/abs/2020Pedos..30..159D). doi:10.1016/S1002-0160(20)60001-7 (https://doi.org/10.1016%2FS1002-0160%2820%2960001-7). Retrieved 21 July 2024.

- 19. Das, Subhasich; Bhattacharya, Satya Sundar (2017). "Significance of soil organic matter in relation to plants and their products". In Siddiqui, Mohammed Wasim; Bansal, Vasudha (eds.). *Plant secondary metabolites. Volume 3. Their roles in stress ecophysiology* (https://www.academia.edu/82083954). Palm Bay, Florida: Apple Academic Press. pp. 39–61. ISBN 978-1-77188-356-6. Retrieved 26 August 2024.
- 20. Piccolo, Alessandro (December 2002). "The supramolecular structure of humic substances: a novel understanding of humus chemistry and implications in soil science" (https://www.researchgate.net/publication/22 2526145). Advances in Agronomy. **75**: 57–134. doi:10.1016/S0065-2113(02)75003-7 (https://doi.org/10.1016%2FS0065-2113%2802%2975 003-7). ISBN 978-0-12-000793-6. Retrieved 4 August 2024.
- 21. Paul, Eldor A. (2016). "The nature and dynamics of soil organic matter: plant inputs, microbial transformations, and organic matter stabilization" (https://www.nrel.colostate.edu/assets/nrel_files/labs/paul-lab/docs/Paul_SBBreview2016.pdf) (PDF). *Soil Biology and Biochemistry*. **98**: 109–26. Bibcode:2016SBiBi..98..109P (https://ui.adsabs.harvard.edu/abs/2016SBiBi..98..109P). doi:10.1016/j.soilbio.2016.04.001 (https://doi.org/10.1016%2Fj.soilbio.2016.04.001). Retrieved 11 August 2024.
- 22. Huang, Xue Ru; Li, H.; Li, Song; Xiong, Hailing; Jiang, Xianjun (May 2016). "Role of cationic polarization in humus-increased soil aggregate stability" (https://www.researchgate.net/publication/303509978). European Journal of Soil Science. 67 (3): 341–50.
 Bibcode:2016EuJSS..67..341H (https://ui.adsabs.harvard.edu/abs/2016EuJSS..67..341H). doi:10.1111/ejss.12342 (https://doi.org/10.1111%2Fejs s.12342). Retrieved 11 August 2024.
- 23. Shoba, V. N.; Chudnenko, K. V. (August 2014). "Ion exchange properties of humus acids" (https://www.researchgate.net/publication/269385340). Eurasian Soil Science. 47 (8): 761–71. Bibcode: 2014EurSS..47..761S (https://ui.adsabs.harvard.edu/abs/2014EurSS..47..761S). doi:10.1134/S1064229314080110 (https://doi.org/10.1134%2FS1064229314080110). Retrieved 11 August 2024.
- 24. Lal, Rattan; Negassa, Wakene; Lorenz, Klaus (August 2015). "Carbon sequestration in soil" (https://www.researchgate.net/publication/283457192). Current Opinion in Environmental Sustainability. 15: 79–86. Bibcode:2015COES...15...79L (https://ui.adsabs.harvard.edu/abs/2015COES...15...79L). doi:10.1016/j.cosust.2015.09.002 (https://doi.org/10.1016/92Fj.cosust.2015.09.002). Retrieved 18 August 2024.

- 25. Dynarski, Katherine A.; Bossio, Deborah A.; Scow, Kate M. (13 November 2020). "Dynamic stability of soil carbon: reassessing the "permanence" of soil carbon sequestration" (https://doi.org/10.3389%2Ff envs.2020.514701). Frontiers in Environmental Science. 8 (714701). doi:10.3389/fenvs.2020.514701 (https://doi.org/10.3389%2Ffenvs.2020.514701).
- 26. Kleber, Markus; Lehmann, Johannes (8 March 2019). "Humic substances extracted by alkali are invalid proxies for the dynamics and functions of organic matter in terrestrial and aquatic ecosystems" (https://acsess.onlinelibrary.wiley.com/doi/pdfdirect/10.2134/jeq2019.01.0036). Journal of Environmental Quality. 48 (2): 207–16.

 Bibcode:2019JEnvQ..48..207K (https://ui.adsabs.harvard.edu/abs/2019JEnvQ..48..207K). doi:10.2134/jeq2019.01.0036 (https://doi.org/10.2134%2Fjeq2019.01.0036). PMID 30951127 (https://pubmed.ncbi.nlm.nih.gov/30951127). Retrieved 25 August 2024.
- 27. Baveye, Philippe C.; Wander, Michelle (6 March 2019). "The (bio)chemistry of soil humus and humic substances: why is the "new view" still considered novel after more than 80 years?" (https://doi.org/10. 3389%2Ffenvs.2019.00027). Frontiers in Environmental Science. 7 (27). doi:10.3389/fenvs.2019.00027 (https://doi.org/10.3389%2Ffenvs.2019.00027).
- 28. Brady, Nyle C. (1984). *The nature and properties of soils* (https://www.academia.edu/23641831) (9th ed.). New York, New York: Macmillan Publishing Company. p. 265. ISBN 978-0029460306. Retrieved 1 September 2024.
- 29. Popkin, Gabriel (2021). "A soil-science revolution upends plans to fight climate change" (https://www.quantamagazine.org/a-soil-science-revolution-upends-plans-to-fight-climate-change-20210727/). Quanta Magazine. Retrieved 1 September 2024. "Soil researchers have concluded that even the largest, most complex molecules can be quickly devoured by soil's abundant and voracious microbes"
- 30. Krishna, M. P.; Mohan, Mahesh (July 2017). "Litter decomposition in forest ecosystems: a review" (https://www.academia.edu/119720860). Energy, Ecology and Environment. 2 (3): 236–49. doi:10.1007/s40974-017-0064-9 (https://doi.org/10.1007%2Fs40974-017-0064-9). Retrieved 8 September 2024.

- 31. Levin, Laura; Forchiassin, Flavia (9 May 2001). "Ligninolytic enzymes of the white rot basidiomycete *Trametes trogii*" (https://www.academia.edu/120239930). *Acta Biotechnologica*. **21** (2): 179–86. doi:10.1002/1521-3846(200105)21:2<179::AID-ABIO179>3.0.CO;2-2 (https://doi.org/10.1002/2F1521-3846%28200105%2921%3A2%3C179%3A%3AAID-ABIO179%3E3.0.CO%3B2-2). Retrieved 15 September 2024.
- 32. González-Pérez, Martha; Vidal Torrado, Pablo; Colnago, Luiz A.; Martin-Neto, Ladislau; Otero, Xosé L.; Milori, Débora M. B. P.; Haenel Gomes, Felipe (31 August 2008). "13C NMR and FTIR spectroscopy characterization of humic acids in spodosols under tropical rain forest in southeastern Brazil" (https://www.academia.edu/14026276). Geoderma. 146 (3–4): 425–33. Bibcode:2008Geode.146..425G (https://ui.adsabs.harvard.edu/abs/2008Geode.146..425G). doi:10.1016/j.geoderma.2008.06.018 (https://doi.org/10.1016%2Fj.geoderma.2008.06.018). Retrieved 15 September 2024.
- 33. Knicker, Heike; Almendros, Gonzalo; González-Vila, Francisco Javier; Lüdemann, Hans-Dietrich; Martín, Fracisco (November–December 1995). "13C and 15N NMR analysis of some fungal melanins in comparison with soil organic matter" (https://www.academia.edu/780095 67). Organic Geochemistry. 23 (11–12): 1023–28.

 Bibcode:1995OrGeo..23.1023K (https://ui.adsabs.harvard.edu/abs/1995 OrGeo..23.1023K). doi:10.1016/0146-6380(95)00094-1 (https://doi.org/10.1016%2F0146-6380%2895%2900094-1). Retrieved 15 September 2024.
- 34. Muscolo, Adele; Bovalo, Francesco; Gionfriddo, Francesco; Nardi, Serenella (August 1999). "Earthworm humic matter produces auxin-like effects on *Daucus carota* cell growth and nitrate metabolism" (https://www.academia.edu/78825632). *Soil Biology and Biochemistry*. **31** (9): 1303–11. Bibcode:1999SBiBi..31.1303M (https://ui.adsabs.harvard.edu/abs/1999SBiBi..31.1303M). doi:10.1016/S0038-0717(99)00049-8 (https://doi.org/10.1016%2FS0038-0717%2899%2900049-8). Retrieved 15 September 2024.
- 35. Oyege, Ivan; Sridhar, B. B. Maruthi (10 November 2023). "Effects of vermicompost on soil and plant health and promoting sustainable agriculture" (https://doi.org/10.3390%2Fsoilsystems7040101). Soil Systems. 7 (4): 101. doi:10.3390/soilsystems7040101 (https://doi.org/10.3390%2Fsoilsystems7040101).

- Dungait, J. A.; Hopkins, D. W.; Gregory, A. S.; Whitmore, A. P. (14 February 2012). "Soil organic matter turnover is governed by accessibility not recalcitrance" (https://www.desmog.com/wp-content/uploads/files/Dungait%20SOM%20article.pdf) (PDF). Global Change Biology. 18 (6): 1781–96. doi:10.1111/j.1365-2486.2012.02665.x (https://doi.org/10.1111%2Fj.1365-2486.2012.02665.x). Retrieved 22 September 2024.
- 37. Baldock, Jeffrey A.; Skjemstad, Jan Otto (July 2000). "Role of the soil matrix and minerals in protecting natural organic materials against biological attack" (https://www.academia.edu/78009563). Organic Geochemistry. 31 (7): 697–710. doi:10.1016/S0146-6380(00)00049-8 (https://doi.org/10.1016%2FS0146-6380%2800%2900049-8). Retrieved 22 September 2024.
- 38. Angst, Šárka; Mueller, Carsten W.; Cajthaml, Tomáš; Angst, Gerrit; Lhotáková, Zuzana; Bartuška, Martin; Špaldoňová, Alexandra; Frouz, Jan (1 March 2017). "Stabilization of soil organic matter by earthworms is connected with physical protection rather than with chemical changes of organic matter" (https://www.academia.edu/80259832). Geoderma. 289: 29–35. doi:10.1016/j.geoderma.2016.11.017 (https://doi.org/10.1016%2Fj.geoderma.2016.11.017). Retrieved 6 October 2024.
- 39. Brauman, Alain (July 2000). "Effect of gut transit and mound deposit on soil organic matter transformations in the soil feeding termite: a review" (https://fr.articles.sk/book/18221151/100de8). European Journal of Soil Biology. 36 (3–4): 117–25. doi:10.1016/S1164-5563(00)01058-X (https://doi.org/10.1016%2FS1164-5563%2800%2901058-X). Retrieved 6 October 2024.
- Andreetta, Anna; Ciampalini, Rossano; Moretti, Pierpaolo; Vingiani, Simona; Poggio, Giorgio; Matteucci, Giorgio; Tescari, Francesca; Carnicelli, Stefano (2011). "Forest humus forms as potential indicators of soil carbon storage in Mediterranean environments" (https://www.researchgate.net/publication/226417489). Biology and Fertility of Soils. 47: 31–40. doi:10.1007/s00374-010-0499-z (https://doi.org/10.1007%2Fs00374-010-0499-z). Retrieved 6 October 2024.

- 41. Oades, J. Malcolm (February 1984). "Soil organic matter and structural stability: mechanisms and implications for management" (https://edisciplinas.usp.br/pluginfile.php/5168021/mod_resource/content/1/Grupo%206_Oades%2C%201984.%20Soil%20organic%20matter%20and%20structural%20stability%20mechanisms%20and%20implications%20for%20man agement.pdf) (PDF). *Plant and Soil.* **76** (1–3): 319–337. Bibcode: 1984PlSoi..76..319O (https://ui.adsabs.harvard.edu/abs/1984PlSoi..76..319O). doi:10.1007/BF02205590 (https://doi.org/10.1007%2FBF02205590). S2CID 7195036 (https://api.semanticscholar.org/CorpusID:7195036). Retrieved 13 October 2024.
- 42. Ponge, Jean-François; Topoliantz, Stéphanie; Ballof, Sylvain; Rossi, Jean-Pierre; Lavelle, Patrick; Betsch, Jean-Marie; Gaucher, Philippe (July 2006). "Ingestion of charcoal by the Amazonian earthworm Pontoscolex corethrurus: a potential for tropical soil fertility" (https://www.researchgate.net/publication/44735820). Soil Biology and Biochemistry. 38 (7): 2008–9. doi:10.1016/j.soilbio.2005.12.024 (https://doi.org/10.1016%2Fj.soilbio.2005.12.024). Retrieved 13 October 2024.
- 43. Arroyo-Kalin, Manuel (July 2017). "Amazonian Dark Earths". In Nicosia, Cristiano; Stoops, Georges (eds.). *Archaeological soil and sediment micromorphology* (https://www.researchgate.net/publication/319444794). Hoboken, New Jersey: Wiley. pp. 345–57. doi:10.1002/9781118941065.ch33 (https://doi.org/10.1002%2F9781118941065.ch33). ISBN 9781118941065. Retrieved 13 October 2024.
- 44. Gerlach, Renate; Fischer, Peter; Eckmeier, Eileen; Hilgers, Alexandra.

 "Buried dark soil horizons and archaeological features in the Neolithic settlement region of the Lower Rhine area, NW Germany: formation, geochemistry and chronostratigraphy" (https://www.academia.edu/77065464). Quaternary International. 265: 191–204. doi:10.1016/j.quaint.2011.10.007 (https://doi.org/10.1016%2Fj.quaint.2011.10.007). Retrieved 20 October 2024.
- 45. Sanborn, Paul; Lamontagne, Luc; Hendershot, William. "Podzolic soils of Canada: genesis, distribution, and classification" (https://doi.org/10.414 1%2Fcjss10024). Canadian Journal of Soil Science. 91 (5): 843–80. doi:10.4141/cjss10024 (https://doi.org/10.4141%2Fcjss10024).

- 46. Hargitai, László (December 1993). "The role of organic matter content and humus quality in the maintenance of soil fertility and in environmental protection" (https://fr.articles.sk/book/19958903/07e3a6). Landscape and Urban Planning. 27 (2–4): 161–67.

 Bibcode:1993LUrbP..27..161H (https://ui.adsabs.harvard.edu/abs/1993LUrbP..27..161H). doi:10.1016/0169-2046(93)90044-E (https://doi.org/10.1016%2F0169-2046%2893%2990044-E). Retrieved 27 October 2024.
- 47. Hoitink, Harry A. J.; Fahy, Peter C. (September 1986). "Basis for the control of soilborne plant pathogens with composts" (https://fr.articles.sk/book/18744629/4076fa). Annual Review of Phytopathology. 24: 93–114. doi:10.1146/annurev.py.24.090186.000521 (https://doi.org/10.1146%2Fannurev.py.24.090186.000521). Retrieved 27 October 2024.
- 48. Lal, Rattan (September 2020). "Soil organic matter and water retention" (https://www.researchgate.net/publication/341213360). *Agronomy Journal.* **116** (5): 3265–77. doi:10.1002/agj2.20282 (https://doi.org/10.1002/2Fagj2.20282). Retrieved 27 October 2024.
- de Macedo, José Ronaldo; do Amaral Meneguelli, Neli; Ottoni Filho, Theophilo Benedicto; Lima, Jorge Araújo de Sousa (February 2007).
 "Estimation of field capacity and moisture retention based on regression analysis involving chemical and physical properties in Alfisols and Ultisols of the state of Rio de Janeiro" (https://fr.articles.sk/book/3290217 7/18be8b). Communications in Soil Science and Plant Analysis. 33 (13–14): 2037–55. Bibcode:2002CSSPA..33.2037D (https://ui.adsabs.harvard.edu/abs/2002CSSPA..33.2037D). doi:10.1081/CSS-120005747 (https://api.semanticscholar.org/CorpusID:98466747). Retrieved 27 October 2024.
- 50. Hempfling, Reinhold; Schulten, Hans-Rolf; Horn, Rainer (June 1990).

 "Relevance of humus composition to the physical/mechanical stability of agricultural soils: a study by direct pyrolysis-mass spectrometry" (https://fr.articles.sk/book/8543777/01704b). Journal of Analytical and Applied Pyrolysis. 17 (3): 275–81. doi:10.1016/0165-2370(90)85016-G (https://doi.org/10.1016%2F0165-2370%2890%2985016-G). Retrieved 3 November 2024.
- 51. Piccolo, Alessandro (1996). "Humus and soil conservation". In Piccolo, Alessandro (ed.). *Humic substances in terrestrial ecosystems* (https://www.researchgate.net/publication/281451183). Amsterdam, The Netherlands: Elsevier. pp. 225–64. doi:10.1016/B978-044481516-3/50006-2 (https://doi.org/10.1016%2FB978-044481516-3%2F50006-2). ISBN 978-0-444-81516-3. Retrieved 3 November 2024.

- 52. Szalay, Alex (October–November 1964). "Cation exchange properties of humic acids and their importance in the geochemical enrichment of UO2++ and other cations" (https://fr.articles.sk/book/19639940/3bac09). Geochimica et Cosmochimica Acta. 28 (10–11): 1605–14.

 Bibcode:1964GeCoA..28.1605S (https://ui.adsabs.harvard.edu/abs/1964GeCoA..28.1605S). doi:10.1016/0016-7037(64)90009-2 (https://doi.org/10.1016%2F0016-7037%2864%2990009-2). Retrieved 3 November 2024.
- 53. Elo, Seija; Maunuksela, Liisa; Salkinoja-Salonen, Mirja; Smolander, Aino; Haahtela, Kielo (February 2000). "Humus bacteria of Norway spruce stands: plant growth promoting properties and birch, red fescue and alder colonizing capacity" (https://doi.org/10.1111%2Fj.1574-6941.2000.t b00679.x). FEMS Microbiology Ecology. 31 (2): 143–52. doi:10.1111/j.1574-6941.2000.tb00679.x (https://doi.org/10.1111%2Fj.1574-6941.2000.tb00679.x). PMID 10640667 (https://pubmed.ncbi.nlm.nih.g ov/10640667).
- 54. Vreeken-Buijs, Madelein J.; Hassink, Jan; Brussaard, Lijbert (1998). "Relationships of soil microarthropod biomass with organic matter and pore size distribution in soils under different land use" (https://www.academia.edu/65368490). Soil Biology and Biochemistry. 30 (1): 97–106. Bibcode:1998SBiBi..30...97V (https://ui.adsabs.harvard.edu/abs/1998SBiBi..30...97V). doi:10.1016/S0038-0717(97)00064-3 (https://doi.org/10.1016/S2FS0038-0717%2897%2900064-3). Retrieved 3 November 2024.
- 55. Ponge, Jean-François (July 2003). "Humus forms in terrestrial ecosystems: a framework to biodiversity" (https://www.academia.edu/205 08983). Soil Biology and Biochemistry. **35** (7): 935–45. doi:10.1016/S0038-0717(03)00149-4 (https://doi.org/10.1016%2FS0038-0717%2803%2900149-4). Retrieved 10 November 2024.
- 56. Hodges, R. D. (1991). "Soil organic matter: its central position in organic farming". In Wilson, W. S. (ed.). *Advances in soil organic matter research: the impact on agriculture and the environment* (https://fr.articles.sk/book/81383430/4b29a3). Sawston, United Kingdom: Woodhead Publishing. pp. 355–64. doi:10.1016/b978-1-85573-813-3.50040-8 (https://doi.org/10.1016%2Fb978-1-85573-813-3.50040-8). ISBN 978-1-85573-813-3. Retrieved 10 November 2024.
- 57. Gunina, Anna; Kuzyakov, Yakov (April 2022). "From energy to (soil organic) matter" (https://doi.org/10.1111%2Fgcb.16071). *Global Change Biology*. **28** (7): 2169–82. doi:10.1111/gcb.16071 (https://doi.org/10.1111/12Fgcb.16071).

- 58. Senn, T. L.; Kingman, Alta R.; Godley, W. C. (1973). "A review of humus and humic acids" (https://www.humintech.com/fileadmin/content_images/agriculture/information/articles_pdf/A-Review-of-Humus-and-Humic-Acids_T.L.Senn_A.R.Kingsmann.pdf) (PDF). Research Series, South Carolina Agricultural Experiment Station. 145. Retrieved 24 November 2024.
- 59. Eyheraguibel, Boris; Silvestre, Jérôme; Morard, Philippe (July 2008). "Effects of humic substances derived from organic waste enhancement on the growth and mineral nutrition of maize" (https://hal.science/hal-009 40093/file/Eyheraguibel_10804.pdf) (PDF). Bioresource Technology. 99 (10): 4206–12. Bibcode:2008BiTec..99.4206E (https://ui.adsabs.harvard.edu/abs/2008BiTec..99.4206E). doi:10.1016/j.biortech.2007.08.082 (https://biorog/10.1016%2Fj.biortech.2007.08.082). PMID 17962015 (https://pubmed.ncbi.nlm.nih.gov/17962015). Retrieved 17 November 2024.
- 60. Zandonadi, Daniel Basilio; Santos, Mirella Pupo; Busato, Jader Galba; Peres, Lázaro Eustáquio Pereira; Façanha, Arnoldo Rocha (2013). "Plant physiology as affected by humified organic matter" (https://doi.org/10.1590%2FS2197-00252013000100003). Theoretical and Experimental Plant Physiology. 25 (1): 12–25. doi:10.1590/S2197-00252013000100003 (https://doi.org/10.1590%2FS2197-00252013000100003).
- 61. Nardi, Serenella; Ertani, Andrea; Francioso, Ornella (February 2017).

 "Soil–root cross-talking: the role of humic substances" (https://www.academia.edu/102119488). Journal of Plant Nutrition and Soil Science. 180 (1): 5–13. doi:10.1002/jpln.201600348 (https://doi.org/10.1002%2Fjpln.201600348). hdl:2318/1731194 (https://hdl.handle.net/2318%2F1731194). Retrieved 17 November 2024.
- 62. Shoba, V. N.; Chudnenko, Konstantin V. (August 2014). "Ion exchange properties of humus acids" (https://www.researchgate.net/publication/269 385340). Eurasian Soil Science. 47 (8): 761–71. doi:10.1134/S1064229314080110 (https://doi.org/10.1134%2FS1064229 314080110). Retrieved 24 November 2024.
- 63. Olness, Alan; Archer, David (February 2005). "Effect of organic carbon on available water in soil" (https://fr.articles.sk/book/54264627/14eda9). Soil Science. 170 (2): 90–101. Bibcode:2005SoilS.170...90O (https://ui.a dsabs.harvard.edu/abs/2005SoilS.170...90O). doi:10.1097/00010694-200502000-000 (https://doi.org/10.1097%2F00010694-200502000-00002). S2CID 95336837 (https://api.semanticscholar.org/CorpusID:95336837). Retrieved 24 November 2024.

- 64. Kikuchi, Ryunosuke (February 2004). "Deacidification effect of the litter layer on forest soil during snowmelt runoff: laboratory experiment and its basic formularization for simulation modeling" (https://fr.articles.sk/book/1 6655436/acdc92). Chemosphere. 54 (8): 1163–69.

 Bibcode:2004Chmsp..54.1163K (https://ui.adsabs.harvard.edu/abs/2004 Chmsp..54.1163K). doi:10.1016/j.chemosphere.2003.10.025 (https://doi.org/10.1016%2Fj.chemosphere.2003.10.025). PMID 14664845 (https://pubmed.ncbi.nlm.nih.gov/14664845). Retrieved 24 November 2024.
- 65. Caesar-Tonthat, Thecan C. (August 2002). "Soil binding properties of mucilage produced by a basidiomycete fungus in a model system" (https://fr.articles.sk/book/20537646/09bc94). *Mycological Research*. **106** (8): 930–37. doi:10.1017/S0953756202006330 (https://doi.org/10.1017%2FS 0953756202006330). Retrieved 24 November 2024.
- 66. Zhu, Rui; Wu, Min; Yang, Jian (February 2011). "Mobilities and leachabilities of heavy metals in sludge with humus soil" (https://fr.articles.sk/book/14270592/0155c3). Journal of Environmental Sciences. 23 (2): 247–54. doi:10.1016/S1001-0742(10)60399-3 (https://doi.org/10.1016%2FS1001-0742%2810%2960399-3). Retrieved 24 November 2024.
- 67. Ludwig, J. W.; Harper, John L. (July 1958). "The influence of the environment on seed and seedling mortality. VIII. The influence of soil colour" (https://fr.articles.sk/book/57099612/546f22). *Journal of Ecology*. **46** (2): 381–89. doi:10.2307/2257402 (https://doi.org/10.2307%2F2257402). Retrieved 1 December 2024.

- Amelung, Wulf; Bossio, Deborah; De Vries, Wim; Kögel-Knabner, Ingrid; Lehmann, Johannes; Amundson, Ronald; Bol, Roland; Collins, Chris; Lal, Rattan; Leifeld, Jens; Minasny, Budiman; Pan, Gen-Xing; Paustian, Keith; Rumpel, Cornelia; Sanderman, Jonathan; Van Groenigen, Jan Willem; Mooney, Sacha; Van Wesemael, Bas; Wander, Michelle; Chabbi, Abbad (27 October 2020). "Towards a global-scale soil climate mitigation strategy" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC7591914).
 Nature Communications. 11 (1): 5427. Bibcode:2020NatCo..11.5427A (https://ui.adsabs.harvard.edu/abs/2020NatCo..11.5427A). doi:10.1038/s41467-020-18887-7 (https://doi.org/10.1038%2Fs41467-020-18887-7). ISSN 2041-1723 (https://search.worldcat.org/issn/2041-1723). PMC 7591914 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC7591914). PMID 33110065 (https://pubmed.ncbi.nlm.nih.gov/33110065).
- 69. Tang, Chunyu; Li, Yuelei; Song, Jingpeng; Antonietti, Markus; Yang, Fan (25 June 2021). "Artificial humic substances improve microbial activity for binding CO2" (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8387571). iScience. 24 (6): 102647. Bibcode:2021iSci...24j2647T (https://ui.adsabs.harvard.edu/abs/2021iSci...24j2647T). doi:10.1016/j.isci.2021.102647 (https://doi.org/10.1016%2Fj.isci.2021.102647). ISSN 2589-0042 (https://search.worldcat.org/issn/2589-0042). PMC 8387571 (https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8387571). PMID 34466779 (https://pubmed.ncbi.nlm.nih.gov/34466779).

External links

- Weber, Jerzy. <u>"Types of humus in soils" (http://karnet.up.wroc.pl/~weber/typy2.htm)</u>. Agricultural University of Wroclaw, Poland. Retrieved 8 December 2024.
- Wershaw, Robert L. "Evaluation of conceptual models of natural organic matter (humus) from a consideration of the chemical and biochemical processes of humification" (https://pubs.usgs.gov/sir/2004/5121/pdf/sir2004-5121.pdf) (PDF). *Pubs.USGU.gov*. United States Geological Survey. Retrieved 8 December 2024.
- "What are humic substances?" (https://humic-substances.org/). International Humic Substances Society. Retrieved 8 December 2024.